Identification and Differentiation of Species in Cooked Meat by Vertical Plate Gel Electrophoresis

Hamad Al-Awfy
University of Rhode Island

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IDENTIFICATION AND DIFFERENTIATION OF SPECIES IN COOKED MEAT BY VERTICAL PLATE GEL ELECTROPHORESIS

By

HAMAD AL-AWFY

A THESIS SUBMITTED IN PARTIAL FULFILLMENT OF THE REQUIREMENTS FOR THE DEGREE OF MASTER OF SCIENCE IN FOOD SCIENCE AND NUTRITION

UNIVERSITY OF RHODE ISLAND

1981
ABSTRACT

Since quality control for adulteration in meat is very important, the vertical plate polyacrylamide gel electrophoretic technique was studied for the detection and identification of cooked and canned species extracts.

Meat samples from beef, pork, lamb and horse were heated for one hour from 65-75°C at intervals of about 5°C. Because most of the research on the effect of cooking meats on the denatured serum proteins and the resultant electrophoretic patterns were done on beef, beef cooked at 70°C was selected as an initial trial for protein solubilization in order to choose an extraction and applying procedure for cooked meats. The beef was extracted with a variety of potential protein solubilizing agents such as sodium dodecyl sulfate (SDS), urea, Triton X-100 and BDI reagent in an attempt to solubilize the proteins. A single gel containing urea or SDS in conjunction with a discontinuous buffer system was studied as the electrophoresis system. Amido black 10B and Coomassie brilliant blue R-250 were compared as staining dyes. Constant power and constant voltage were compared as power supplies. Polyacrylamide gel electrofocusing was tried. Raw meat extracts were utilized as reference patterns.
A system of tris-chloride buffer at pH 6.7 containing 2M urea and 10% glucose gave the best results as an extraction procedure. A single 7% cyanogum gel containing 4M urea in conjunction with a discontinuous buffer system was utilized as the electrophoresis system. Amido black was found to be more sensitive so it was selected as staining procedure. Also, additions of mercaptoethanol to the gel and sample improved the electrophoretic patterns. It was found that with constant power, the front solvent was a very sharp straight line and the bands were sharper than with constant voltage.

Characteristic differences were discernible between beef and lamb versus pork and horse meat heated at 65°C, but the distinction was decreased with increasing temperature. Canned beef (corned beef) and mixture of canned pork and ham (SPAM) gave some bands but did not give proper separation of the bands. So with more study on solubility and gel electrophoretic patterns heat denatured proteins could be improved and achieved. Use of a densitometer might also improve the utility of the procedure.

Polyacrylamide gel electrofocusing did not work and that may be because the thickness of the gel which generates heat, or other causes, as discussed in Appendix C.
ACKNOWLEDGEMENT

I wish to express my sincere gratitude to Dr. Arthur G. Rand, of the Department of Food Science and Technology, Nutrition and Dietetics, for his guidance, valuable counsel, and his suggestions on the preparation and review of this thesis. I also would like to thank Dr. Charles E. Olney and Dr. Murn M. Nippo for serving on my thesis committee.

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Appreciation is expressed to my Government for granting me a scholarship. I deeply thank the officials of our Ministry, CODOT, and the Saudi Arabian Educational Mission for their constant support and encouragement throughout this research.
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Preface

This thesis has been prepared according to the manuscript thesis plan, following Journal of Food Science.
Identification and Differentiation of Species in Cooked Meats by Vertical Plate Gel Electrophoresis

Abstract

Since quality control for adulteration in meat is very important, the vertical plate polyacrylamide gel electrophoretic technique was studied for the detection and identification of cooked and canned species extracts.

Meat samples from beef, pork, lamb and Iberian meat were heated for one hour from 65-75°C at intervals of about 5°C. Extractions were performed with a variety of potential protein solubilizing agents in an attempt to solubilize the proteins. Amide black 10B and Comassie brilliant blue R-250 were compared as staining dyes. Constant power and constant voltage were compared as power supplies. Raw meat extracts were utilized as reference patterns.

A system of tri-citrate buffer at pH 6.7 containing 2M urea and 10C glucose gave the best results as an extraction procedure. Amide black was found to be more sensitive. Also, addition of mercaptoethanol to the gel and sample improved the electrophoretic pattern. Capacitive power gave better resolution.
Identification and differentiation of species in cooked meats by vertical plate gel electrophoresis.

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University of Rhode Island
Kingston, Rhode Island 02881

ABSTRACT

Since quality control for adulteration in meat is very important, the vertical plate polyacrylamide gel electrophoretic technique was studied for the detection and identification of cooked and canned species extracts.

Meat samples from beef, pork, lamb and horse meat were heated for one hour from 65-75°C at intervals of about 5°C. Extractions were performed with a variety of potential protein solubilizing agents in an attempt to solubilize the proteins. Amido black 10B and Coomassie brilliant blue R-150 were compared as staining dyes. Constant power and constant voltage were compared as power supplies. Raw meat extracts were utilized as reference patterns.

A system of tris-chloride buffer at pH 6.7 containing 2M urea and 10% glucose gave the best results as an extraction procedure. Amido black was found to be more sensitive. Also, addition of mercaptoethanol to the gel and sample improved the electrophoretic patterns. Constant power gave better resolution.
Characteristic differences were discernible between beef and lamb versus pork and horse meat heated at 65°C but the distinction was decreased with increasing temperature. Canned meats did not give good resolution.
INTRODUCTION

With increases in world population and the limited ability of growing animals to produce meat, meat prices continue to increase, especially some of the kinds which many people prefer (i.e., beef and lamb).

Since quality control for adulteration in meat is very important, electrophoresis with polyacrylamide gel has been studied and successfully applied by Payne (1963) to raw products. This method, the disc polyacrylamide gel electrophoretic technique, developed by Ornstein (1964) and Davis (1964), has proven to be successful for separation and identification of animal serum proteins. Distinct electrophoretic patterns were obtained between meat and fish species using this technique (Payne, 1963; Mancusa, 1964).

The vertical plate polyacrylamide gel electrophoretic (VPE) technique, which was developed by Raymond (1964) has also been successfully used for the detection and identification of raw meat and fish species (Coduri and Rand, 1972a, 1972b).

The effect of cooking beef on the denatured serum proteins and on the resultant electrophoretic patterns has been studied. Laakkonen et al. (1970) found that changes in the electrophoretic patterns of bovine muscle took place during low temperature and long-time heating. Fogg and Harrison (1975) have studied the effect of two end points,
25°C and 45°C in semitendinous beef muscle on the electrophoretic patterns of the sarcoplasmic proteins. They concluded that heating to 25°C decreased the number and intensity of the slowest migration protein components of the sarcoplasmic fraction and the effects were more pronounced at 45°C. Lee et al. (1974) have employed polyacrylamide gel electrophoresis in the study of qualitative changes in electrophoretic patterns of water soluble protein extracts from bovine muscles cooked to final temperatures of 65, 70, 75, 80, 85 and 90°C. They concluded that increments of cooking temperature caused a progressive weakening and disappearance of the electrophoretic bonds. Caldironi and Bazan (1980) reported that when beef muscles were cooked at temperatures between 60-80°C, protein bonds gradually disappeared and could not be detected above 80°C.

Deschreider and Meaux (1974) stated that heating profoundly changed the electrophoregrams, but nevertheless allowed differentiation of various kinds of meat. Those authors reported that the disappearance of bands was precise in relation to temperature increase. Mafinen et al. (1979) reported that native and denatured meat proteins (pork, beef, chicken and turkey) were solubilized for polyacrylamide gel electrophoresis (PAGE) by SDS+Mercaptoethanol. SDS-PAGE patterns of each denatured sample resembled those of native samples, and all patterns of meat protein bands were so similar that species identification was impossible.
This paper describes the study of vertical plate gel electrophoresis with urea as solubilizing agent for the potential separation and identification of protein from cooked/canned meat, and the effect of meat cooking temperatures on the analysis of cooked/canned meat.

Fresh and canned meats (beef, pork and lamb) were collected from a local market. Fresh horse meat was provided by M and M Company, Hartford, Connecticut.

Results:

All the reagents used in this experiment were obtained from Fisher Scientific Company, except Dodecylalkanol and Thymol which were obtained from Eastman Organic Chemicals. Tween 80 from Sigma Chemical Laboratories, Anhydrous NaOH, Chromatography blue and brilliant blue from Bio-Rad Laboratories. Cyanoamine, 1,4-DNED and ammonium persulfate from E-C Apparatus Corporation and Triton X-100 from Hartman-Leduc Co. The cyanoamine, 5-n, N'-tetramethylylenediamine (1,4-DNED) and ammonium persulfate were stored under refrigeration. Deionized distilled water was used throughout.

Equipment:

Vertical gel electrophoresis EC-470 cell, EC-4001 comb- plate power supply, EC-650 constant voltage power supply, 1 EC-929 diffusion destaining and an electrophoresis destainer, were purchased from EC Apparatus Co.

Methods:

Canned Meat Protein Solubilization

Because most of the research on the effect of cook-
MATERIALS AND METHODS

Materials

Meat source.

Fresh and canned meats (beef, pork and lamb) were collected from a local market. Fresh horse meat was provided by M and R Company, Hartford, Connecticut.

Reagents:

All the reagents used in this experiment were obtained from Fisher Scientific Company, except mercaptoethanol and thymol which were obtained from Eastman Organic Chemicals, Tween 80 from Sigma Chemical Laboratories, Amido black 10B, Coomassie blue and bromophenol from Bio-Rad Laboratories, Cyanogum, TEMED and ammonium persulfate from E-C Apparatus Corporation and Triton X-100 from Hartman-Leddon Co. The cyanogum, N,N,N',N'-tetramethylethylene diamine (TEMED) and ammonium persulfate were stored under refrigeration. Deionized distilled water was used throughout.

Equipment:

Vertical gel electrophoresis EC-470 Cell, EC-400 constant power supply, EC-454 constant voltage power supply, EC-225 diffusion destainer and an electrophoretic destainer, were purchased from EC Apparatus Co.

Methods:

Cooked Meat protein solubilization

Because most of the research on the effect of cook-
ing meats on the denatured serum proteins and on the result-
ant electrophoretic patterns were done on beef, beef cooked at 70°C was selected as an initial trial for protein solubil-
ization in order to choose an extraction and applying proce-
dure for cooked meats, as follows:

Sample (beef) was put in a plastic bag and heated at 70°C in a water bath for one hour. Then extractions were per-
formed with a variety of potential protein solubilizing agents.

1. Blend 20g of the sample with 8 ml 0.1% sodium dodecyl sulfate (SDS), 8 ml 0.5M urea and 2 ml 0.1% mercaptoethanol (ME). (SUM)

2. Blend 20g of the sample with 15 ml 2% SDS and 5 ml of 1% ME. (SM)

3. Blend 20g of the sample with 20 ml 2% SDS. (SDS-1)

4. Blend 20g of the sample with 20 ml 2% SDS and heat it in a boiling water bath for 3 minutes. (Heat after blending.) (SDS-2)

5. Heat 20g of the sample with 20ml 2% SDS in a boiling water bath for 3 minutes and blend (heat before blending). (SDS-3)

6. Blend 20g of the sample with 5 ml 3% Triton X-100 and 15 ml 1% ME. (TM)

7. Blend 20g of the sample with 10 ml 3% Triton X-100 and 10 ml 1% SDS. (TS)
8. Blend 20g of the sample with 10 ml 3% Triton X-100 and 10 ml 0.5M urea. (TU)

9. Heat 15g of the sample with 10 ml of BDI reagent - 30g of Triton X-100 and 70g of sodium tetraphosphate made up to one liter with distilled water (Thomas et al., 1955) in boiling water bath for 15 minutes and blend. (BDI)

10. Blend 50g of the sample with 25 ml tris(2-Amino-2-hydroxymethyl-1,3-proponediol)-chloride buffer at pH 6.7 containing 2M urea - 10% glucose - 0.2% mercaptoethanol. (BGU)

11. 10g sample + 10M urea (ultra pure urea) was prepared according to Krzynowek and Wiggin (1979). (U)

Each mixture was transferred to a 50 ml plastic centrifuge tube and centrifuged in a Sorvall RC-2B refrigerated centrifuge at 6000 RPM for 20-30 minutes. The supernatants were filtered through Whatman #1 filter paper. The density of each solution, for introduction into the slots of the electrophoretic gel, was improved with the addition of 0.1-0.25g of glucose per ml of supernatant. A crystal of thymol was added for preservation, and the samples were stored at 2°C until utilized.

Raw Meat:

Raw meat extracts were utilized as reference patterns, and the solutions were prepared as described by Coduri and Rand (1972a).
**Heat Processed Meat:**

Samples of individual meats were put in plastic bags and heated to 65°, 70° and 75°C in a water bath for one hour; the samples were left to cool down at room temperature for more than one hour. This meat was extracted for analysis or frozen for later use. Canned meats which contained various additives were washed several times with distilled water to minimize possible interference.

Heat processed meats were prepared for electrophoresis by mixing 60g with 25 ml of extraction buffer at low speed for 30 sec in a jacketed Waring Blendor jar surrounded with ice to prevent heat generation. The samples were blended at high speed until a uniform mixture was obtained. The blended samples were transferred to 50 ml centrifuge tubes and clarified at 6000 rpm for 20-30 min in a Sorval RC-2B refrigerated centrifuge. The supernatants were filtered through Whatman #1 filter paper. The density of the samples, for introduction into the slots of the electrophoretic gel, were improved with the addition of 0.1 g/ml of glucose. A crystal of thymol was added for preservation, and the samples were stored at 1°C until utilized.

**Urea-polyacrylamide Gel Electrophoresis:**

The vertical plate polyacrylamide gel electrophoretic technique described by Coduri and Rand (1972a) for the differentiation of meat species was used with some modification. Formulation for all reagents used in this method are shown in Table 1.
Table 1
Reagent Formulation for VPE Method

1 - Extraction buffer

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tris</td>
<td>0.6 g</td>
</tr>
<tr>
<td>Glucose</td>
<td>5.0 g</td>
</tr>
<tr>
<td>Urea</td>
<td>6.0 g</td>
</tr>
<tr>
<td>ME</td>
<td>0.1 ml</td>
</tr>
<tr>
<td>3N HCl to pH 6.7</td>
<td></td>
</tr>
<tr>
<td>distilled water to make 50 ml</td>
<td></td>
</tr>
</tbody>
</table>

2 - Gel solution

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tris</td>
<td>3.5 g</td>
</tr>
<tr>
<td>Cyanogum</td>
<td>10.5 g</td>
</tr>
<tr>
<td>Urea</td>
<td>36.0 g</td>
</tr>
<tr>
<td>TEMED</td>
<td>0.15 ml</td>
</tr>
<tr>
<td>Ammonium persulphate</td>
<td>0.13 g</td>
</tr>
<tr>
<td>Tween 80 (10%)</td>
<td>0.2 ml</td>
</tr>
<tr>
<td>3N HCl to pH 8.9</td>
<td></td>
</tr>
<tr>
<td>distilled water to make 150 ml</td>
<td></td>
</tr>
</tbody>
</table>

3 - Electrode buffer

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tris</td>
<td>3.0 g</td>
</tr>
<tr>
<td>Glycine</td>
<td>14.4 g</td>
</tr>
<tr>
<td>distilled water to make 2L, pH 8.3</td>
<td></td>
</tr>
</tbody>
</table>

4 - Stain

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Amido black</td>
<td>2 g</td>
</tr>
<tr>
<td>Methanol</td>
<td>100 ml</td>
</tr>
<tr>
<td>Glacial acetic acid</td>
<td>30 ml</td>
</tr>
<tr>
<td>Water</td>
<td>100 ml</td>
</tr>
</tbody>
</table>
**Gel Preparation**

The gel solution was prepared prior to each run, according to the formulation shown in Table 1. The tris, Cyanogum, urea and TEMED were dissolved in 90 ml of water and adjusted to pH 8.9 with 3N HCl, followed by filtration through Whatman #1 filter paper. The volume of the gel solution was adjusted to 150 ml, and 0.2 ml of 10% Tween 80, 0.1ml mercaptoethanol and 0.13g of ammonium persulfate were added with gentle agitation on a magnetic stirrer. The gel solution was poured at room temperature into an EC Apparatus Co. EC-470 vertical electrophoresis cell which was assembled horizontally. A slot form for 10 samples was inserted, and the gel solution allowed to solidify for 40-50 min. The cell was then connected to a circulating refrigerated water bath, and coolant at 10°C was introduced into the cooling plates. The temperature in the refrigerated water bath was reduced to 0°C over a 1 hour period and maintained at that temperature for the balance of the run. The excess gel above the slot form was removed and the cell was placed in a vertical position. The tris-glycine electrode buffer, precooled overnight to 2°C was added to the cell and allowed to overflow into the lowest chamber. The slot form was then carefully removed, and bromophenol blue was added to the buffer in the upper chamber to mark the position of the glycine-chloride front.

**Conditions for Electrophoresis**

1 - Constant Voltage - Was carried out as described by Coduri and Rand (1972a).
2 - Constant Power. - The electrophoretic cell was connected to an EC Apparatus Co. EC 400 constant power supply with the lower electrode as positive, and a prerun at 20 watts for 15 min. was carried out until the front had entered the top of the slots. A 50 ul (0.05 ml) aliquot of each sample was carefully added to the slots, underneath the electrode buffer, with a glass syringe. An initial potential of 10 watts (50 ma) was applied until the front had cleared the slots (30 min.). The potential was then increased to 20 watts for about 1.5 hour until the front migrated 10 cm from the bottom of the slots.

Staining and Destaining

After completion of the run, the power was turned off and the cell was disconnected. The electrode buffer was drained off and the cell was disassembled. The gel was removed and stained for 15 minutes in the Amido black 10B stain shown in Table 1, with gentle shaking. Destaining was accomplished first by gently shaking the gel in a 50-50 mixture (v/v) of methanol and 5% acetic acid for 15 min to remove readily soluble dye, and then by transferring to destainer to clean the gel. An electrophoretic destainer which cleared the patterns in about 45 min. at 3.0 amps with 7% acetic acid recirculated over charcoal was used initially, but electrophoretic destaining was not recommended because the process caused substantial loss of the protein dye complex (Ahmadi, 1979). The preferred method employed an E-C
Apparatus (EC-225) diffusion destainer which cleared the patterns in about 8-10 hours with a mixture of 5:5:1 (v/v) water, methanol, and glacial acetic acid recirculated over charcoal.

Photography

Amido black stained gels were photographed using background lighting on Polaroid 3000 speed black and white land pack film type 107C, employing a Polaroid CU-5 camera.
RESULTS AND DISCUSSION

In the initial trials, beef cooked at about 70°C was extracted with different solutions in an attempt to solubilize the proteins. Figures 1 and 2 show the effect of different extraction solutions on the electrophoretic patterns. As seen in Figure 1, cooked beef extraction with SDS only without heating (SDS-1), SDS with heating after blending (SDS-2), SDS with heating before blending (SDS-3) and SDS with mercaptoethanol (SM), all gave faint bands, whereas combination of SDS, urea and mercaptoethanol enhanced the appearance of the bands; but tris-HCl buffer containing 2M urea and 10% glucose (BGU) improved the occurrence and sharpness of the bands. As shown in Figure 2 cooked beef extracted with Triton-X-100 with mercaptoethanol (TM), Triton X-100 with SDS (TS), Triton X-100 with urea (TU) and BDI reagent all give very faint bands, whereas 10M urea gave bands but did not give proper separation of the bands. Those figures show tris-HCl buffer containing 2M urea and 10% glucose (BGU) produced electrophoretic patterns which were better and more sharply defined than other extraction media.

The concentration of urea in the extraction buffer was examined to select the optimum concentration. As shown in Figure 3, with a cooked beef meat heated to about 70°C, 6M and 8M urea did not give proper separation of the bands, whereas 1M, 2M and 4M urea gave excellent migration. The optimum extraction concentration appeared to be 2M urea with
Figure 1. Electrophoretic patterns obtained from cooked beef meat extracts comparing SDS based extraction media with buffer-glucose-urea (BGU)

Samples:
(1) SUM, (2 and 4) MS, (3 and 5) SDS-1, (6 and 8) SDS-2, (7 and 9) SDS-3, (10) BGU
Figure 2. Electrophoretic patterns obtained from cooked meat extracts comparing Triton X-100, urea, BDI and BGU extraction media.

(1) Triton X-100 with ME (TM), (2) Triton X-100 with SDS (TS), (3) Triton X-100 with urea (TU), (6) 10M urea (U), (7) BDI reagents, (9) tris-HCl buffer containing 2M urea and 10% glucose (BGU).
Figure 3. Effect of urea concentration on cooked beef meat extracts, in presence of mercaptoethanol.

(1) 1M urea, (2) 2M urea, (3) 4M urea, (4) 6M urea, (5) 8M urea, (6,7) 2M urea without ME.
mercaptoethanol which provided a slightly better pattern.

The addition of mercaptoethanol to the running gel was also studied. Figure 4a shows electrophoretic patterns obtained from cooked meats without addition of mercaptoethanol to the running gel. Figure 4b shows electrophoretic patterns obtained from the same extracts of cooked meats, but with addition of 0.1 ml concentrated mercaptoethanol to the running gel. Those two figures indicated that addition of mercaptoethanol to the running gel improved the number of bands separated. Also different amounts (0.2 ml, 0.3 ml) of mercaptoethanol were tried to see if increasing and addition of mercaptoethanol to the running gel provided further improvement, but there was no change in the resolutions.

The effect of Amido black 10B and Coomassie brilliant blue R-250 on the electrophoresis of meat protein as a staining procedure was compared. Cooked beef meat extracted with extraction buffer containing 1M, 2M and 4M urea containing mercaptoethanol and 2M urea without mercaptoethanol were applied to a gel in duplicate. After the electrophoretic run was accomplished, the gel was divided down the middle. One side of it was stained in Amido black 10B according to Coduri and Rand (1972a). The other half of the gel was stained for one hour at 25°C with 0.1% Coomassie brilliant blue solution freshly made up in 50% trichloroacetic acid and destained by repeated washing in 10% acetic acid for approximately 48 hours (Ogita and Markert, 1979). As seen in Figure 5, Amido black was sensitive and gave sharper
Figure 4a. Electrophoretic patterns obtained from cooked meats without addition of ME to the running gel.

Samples:

(1) Beef cooked at 70°C  (2) Pork cooked at 70°C  (3) Horsemeat cooked at 70°C  (4) Lamb cooked at 70°C  (5) Beef cooked at 75°C  (6) Pork cooked at 75°C  (7) Horse meat cooked at 75°C  (8) Lamb cooked at 75°C.
Figure 4b. Electrophoretic patterns obtained from cooked meats with addition of 0.1 ml concentrated mercaptoethanol to the running gel.

Samples:

(1) Beef cooked at 70°C (2) Pork cooked at 70°C (3 and 4) Horse meat cooked at 70°C (5) Lamb cooked at 70°C (6) Beef cooked at 75°C (7) Pork cooked at 75°C (8) Horse meat cooked at 75°C (9) Lamb cooked at 75°C.
resolution - better than Coomassie blue. Wilson (1979) also found the same results. Therefore, Amido black 10B was selected as the stain of preference, following the procedure of Coduri and Rand (1972).

The effect of constant power (Figures 6, 7 and 8) using a constant power supply was compared to the use of constant voltage in Figures 1, 2, 3, 4a, 4b and 5. It was found that the constant power the glycine-chloride front was a very sharp straight line, whereas in constant voltage the boundary was not always in a very straight line. The bands were sharper using constant power than constant voltage power. Also, the running time was cut from 3 hours to about 2 hours. Whereas the function of the constant power was to maintain mobility of the sample being separated and not overheat or underpower the separation nor require constant adjustments (E-C Apparatus Corporation), constant power was selected as the preferred method.

Employing the optimum extraction method developed (BGU), raw and cooked meats were analyzed by gel electrophoresis. Figures 6 and 7 show the effect of various cooking temperatures. At 65°C the fastest migrating protein band of beef became very weak and the intensity of the second band also decreased. The slowest disappeared but the thrid band did not change migrating; new secondary band appeared. The slowest migrating band of pork disappeared and the intensity of the fastest migrating band decreased. The intensity, visibility and sharpness of the two bands of lamb became more
Figure 5. Comparison between Amido black 10B and Coomassie brilliant blue R-250, as staining procedure.

Samples 1, 2, 3 and 4 stained with Amido black 10B; 7, 8, 9, and 10 stained with Coomassie Blue.
Figure 6. Polyacrylamide electrophoretograms of (1) raw beef, (2) beef cooked at 65°C, (3) beef cooked at 70°C, (4) beef cooked at 75°C, (5) raw pork, (6) pork cooked at 65°C, (7) pork cooked at 70°C, and (8) pork cooked at 75°C.
Figure 7. Polyacrylamide electrophoreograms of (1) raw lamb, (2) lamb cooked at 65°C, (3) lamb cooked at 70°C, (4) lamb cooked at 75°C, (5) raw horse, (6) horse cooked at 65°C, (7) horse cooked at 70°C, and (8) horse cooked at 75°C.
Figure 8. Polyacrylamide Electrophoretogram of raw and canned meats. (1) raw beef, (2) corned beef, (3) raw pork, and (4) canned pork and ham.
obvious. The slowest migrating band of horse meat disappeared whereas the two highest migrating bands became lighter and in fact the second band may have begun to change or disappear.

The major protein bands of native horsemeat stained lighter than other meat extracts, and the reason may be that the volume of extraction material obtained from horse meat was greater than the amount recovered from the other meats. The possibility that the increased volume resulted in dilution of the soluble material and there may be less soluble material present in horse meat were considered as probable causes for the weaker staining of extraction containing horse meat (Coduri and Rand, 1972a).

At 70°C, the fastest migrating bands of beef, pork and horse virtually disappeared, whereas lamb bands were less intensive. The intensity of the second migrating band of beef and lamb decreased to a great extent, while pork bands almost disappeared, but in the horse meat a new sharp band appeared.

At 75°C the fastest migrating band of all meats disappeared. The second band of pork and horse meat disappeared while beef and lamb were still present but very faint. Some new secondary bands appeared in both pork and horse.

Figure 8 shows that canned beef (corned beef) gave some bands but did not give proper separation of the bands. A mixture of canned pork and ham (SPAM) gave very faint bands.

The increased sharpness and appearance of more bands
after heating may be due to a low molecular weight fragment derived from one of the proteins initially present (Cohen, 1966).

These figures show the electrophoretic patterns obtained with VPE method using only one gel. Characteristic differences were discernible between beef and lamb versus pork and horse at 65°C, but the distinction decreased with an increase of the heat process temperature. Therefore, this study did not agree with Mafinen et al. (1979) who reported that electrophoretic patterns of each denatured sample resembled those of native samples, and all patterns of meat protein bands were so similar that species identification was impossible. However, these results agreed with Laakkonen et al. (1970), Fogg and Harrison (1975), Lee et al. (1974), Deschrreider and Meaux (1974), and Caldironi and Bizan (1980) who reported that the disappearance of bands was precise in relation to temperature increase.

More study on solubility using different solubilizing agents and different concentrations and studying their chemistry (i.e. interaction with proteins) should be done to measure the denatured proteins. Using different gel media and gel electrophoresis techniques could improve the resolution and the appearance of the electrophoretic patterns. Use of a densitometer also might improve the differentiation and increase the detection of the light bands.
References


APPENDIX A

LITERATURE REVIEW

With the vast increase in world population and the limited ability of grazing animals to produce meat, meat prices are rising higher day after day, especially some of the kinds which most of the people prefer (i.e., beef and lamb). Fennema (1967) predicts that the industrialists will seek more and more in areas of substituted, inexpensive or questionable foods as a possible solution to world food unbalance.

If these trends continue, the substitution of course can be predicted in the near future. The selection of any adulteration is ground and probably mild in one of the inevitable problems which may accompany the manipulation of their quality, particularly when canned. The study of this problem is very important especially from the point of consumer protection, and religious considerations in some countries where pig meat is not allowed in any forms. Protein analysis could be one of the logical solutions in different state most products.

Whereas, proteins are non-electrolytes, phe-phient, high molecular weight ions, and the side chains of certain amino acids are non-hydrolyzable sites on simple proteins, so if soluble proteins are charged at some pH values within their range of stability, and are therefore amenable to electro- phoretic analysis. Various electrophoretic techniques such as moving boundary electrophoresis, zone electrophore-
With the vast increase in world population and the limited ability of growing animals to produce meat, meat prices are getting higher day after day, especially some of the kinds which most of the people prefer (i.e., beef and lamb). Fennema (1976) predicts that the industrialists will look more and more to areas of substituted, fabricated or synthetic foods as a possible solution to world food problems.

If these trends continue, the substitution of meats can be predicted in these new kinds of foods. Thus, the detection of any adulteration in ground and combined meats is one of the difficult problems which may arise in the examination of their purity, particularly when canned. The study of this problem is very important especially from the point of consumer protection, and religious considerations in some countries where pig meat is not allowed in any ratio. Protein analysis could be one of the logical solutions to differentiate meat products.

Whereas proteins are polyelectrolytes, polyvalent, high molecular weight ions, and the side chains of certain amino acids are the ionizable sites on simple proteins, so all soluble proteins are charged at some pH values within their range of stability and are therefore amenable to electrophoretic analysis. Various electrophoretic techniques such as: moving boundary electrophoresis, zone electrophore-
sis, polyacrylamide gel electrophoresis and thin layer isoelectric focusing have been used for the separation and identification of proteins.

Since the quality control from any adulteration in meat is very important, electrophoresis with polyacrylamide gel has been successfully applied by Payne (1963) as a method for separation and identification of animal serum proteins using the disc polyacrylamide gel electrophoretic technique. As developed by Ornstein (1964) and Davis (1964), the method has proven to be a good method for the detection and identification of mixtures of animal proteins. Distinct electrophoretic patterns were obtained between meat and fish species using this technique (Payne, 1963; Mancuso, 1964). However, any possible application of the disc procedure for routine analysis of animal species would be hampered by the time-consuming process and the many manipulations required. A separate gel must be prepared for each sample, making it difficult to maintain uniform conditions for all samples. Thus, visual comparisons of several samples for the detection of differences in electrophoretic patterns may not always be valid, or possible, and the procedure would not appear to be satisfactory for daily analytical work (Coduti and Rand, 1972a).

The vertical plate polyacrylamide gel electrophoretic (VPE) technique, which was developed by Raymond (1964) has also been a useful method for protein analysis. This technique is preferable in contrast to a disc technique for several reasons. (1) The flat slab provides maximum surface
area for cooling the gel; (2) the resulting patterns are easier to quantitate in standard recording densitometers; (3) a large number of samples can be processed in a single gel, making the technique easier to carry out in the laboratory and facilitating the direct intercomparison of specimens processed under identical conditions all the way through to the final densitometer recording; (4) and most important, the flat slab permits the application of two-dimensional technique, which is impossible in the vertical tube apparatus (Raymond, 1964). Nakanichi and Raymond (1962) reported that VEP could be used in the routine process of blood proteins. Laakkonen et al. (1970) found this technique was useful for meat analysis.

Coduri and Rand (1972a) were the first to establish that distinct electrophoretic pattern could be obtained from the sarcoplasmic extracts of different meats -- beef, lamb, pork, and horse, using vertical plate technique. Also, they applied this technique for the differentiation on fish and shellfish species (1972b).

Coduri, Bonatti and Simpson (1979) applied a vertical plate gel electrophoresis to the separation of pigmented and nonpigmented trout and salmon species.

In order to import meat from many countries of the world the meat must be cooked to 69°C to insure destruction of the foot and mouth disease virus if present (Heidelbaugh and Graves, 1968), a number of methods therefore have been tried
to determine the cooking temperature such as extractability, coagulation test, determination of acid phosphatase activity and direct spectroscopy of extracted meat pigments. Cohen (1966) studied the protein changes related to ham processing temperatures and found that the rate of heating as well as the temperature reduced the amount of extractable proteins. In 1969 he studied the determination of acid phosphatase activity in canned hams as an indicator of temperatures attained during cooking. He found the method lacked accuracy when applied to hams processed to higher internal temperatures. Doesburg and Papendrof (1969) stated that the degree of heating of some fish muscle could be calculated from the coagulation test. Helmke and Froning (1971) studied the effect of endpoint cooking temperature on the color of turkey meat. They found that the extracted pigment from turkey meat, which was cooked to 82°C, had a spectra quite different than that observed at lower temperatures.

None of these methods, however, has proved to be acceptable because they were only applicable to narrow ranges of cooking temperature or other experimental conditions.

Since cooked meat will be denatured and protein solubility will decrease, solubilizing agents such as urea and sodium dodecyl sulphate (SDS) are necessary to solubilize the protein in cooked meat. Since SDS and urea are two of the most commonly used reagents in protein chemistry, their applications in protein denaturation, solubilization, dissociation and purification are much too numerous to summarize here.
An advantage of these agents is the ability to obtain constant electrophoretic mobilities of proteins independent of isoelectric point and amino acid composition, which can be slightly modified during cooking (Lee et al., 1974).

The effect of cooking beef on the denaturation of proteins and on the resultant electrophoretic patterns has been studied. Laakkonen et al. (1970) found that changes in the electrophoretic patterns of bovine muscle took place during low-temperature and long-time heating. Fogg and Harrison (1975) have studied the effect of two end points, 25°C and 45°C in semi-tendinous beef muscle on the electrophoretic patterns of the sarcoplasmic proteins. They conclude that heating to 25°C decreased the number and intensity of the slowest migration protein components of the sarcoplasmic fraction and that the effects were more pronounced at 45°C. Lee et al. (1974) have employed polyacrylamide gel electrophoresis in the study of qualitative changes in electrophoretic patterns of water soluble protein extracts from bovine muscles cooked to final temperatures of 65, 70, 75, 80, 85 and 90°C. They concluded that increments of cooking temperature caused a progressive weakening and disappearance of the electrophoretic bands. Also, Calderoni and Bazan (1980) reported that when beef muscles were cooked at temperatures between 60-80°C, protein bands gradually disappeared and could not be detected above 80°C.

Deschrreider and Meaux (1974) stated that heating profoundly changed the electrophorograms of various kinds of
meats. Those authors reported that the disappearance of bands was precise in relation to temperature increase. But Mafinen et al. (1979) reported that native and denatured meat proteins (pork, beef, chicken and turkey) were solubilized for polyacrylamide gel electrophoresis (PAGE) by SDS + Mercaptoethanol. SDS-PAGE patterns of each denatured sample resembled those of native samples, and all patterns of meat protein bands were so similar that species identification was impossible.

Gabucci and Flego (1976) studied the heat stability of beef, pork and horse albumin by electroimmunodiffusion and found that antigen activity of beef was lost after 15 min heating at 80°C, pork after 60 min at 100°C, and horse after 30 min at 70°C.

Krzynowek and Wiggin (1979) were able to achieve identification of species in cooked crabmeat by thin-layer isoelectric focusing.
APPENDIX B

SDS-Polyacrylamide gel electrophoresis for identification and differentiation of species in cooked meat by vertical plate gel electrophoresis.

Table 2

Stock Solutions Used in Electrophoresis

IA. Polyacrylamide-bis solution:

- Polyacrylamide: 19.6 g
- Bis: 1.0 g
- Glycerol: 20.0 ml
- 2H2O to make 100.0 ml

IB. Buffer solution:

- Tris-HCl-SDS buffer (1.5M Tris):
  - Tris: 12.3 g
  - MCl (conc.): 5.0 ml
  - 10% SDS solution: 4.2 ml
  - 2H2O to make 100.0 ml
SDS-Polyacrylamide gel electrophoresis for identification and differentiation of species in cooked meat by vertical plate gel electrophoresis.

Electrophoresis in polyacrylamide gel electrophoresis in the presence of the anionic detergent sodium dodecyl sulfate is a widely used technique in detection and identification of protein. Since research on the effect of cooking meats has successfully used SDS-polyacrylamide gel for separation and identification of the denatured serum proteins (Lee et al., 1974; Mafinen et al., 1979 and Calderoni and Bazan, 1980), and it appeared to be a good procedure to try for analysis of cooked meats. Therefore SDS-polyacrylamide gel electrophoresis was performed according to Ogita and Markert (1979) with some modifications as shown in Table 2.

Table 2

Stock Solutions Used in Electrophoresis

I: Stock solutions for running gel:

IA: Acrylamide-bis solution:

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Acrylamide</td>
<td>39.0 g</td>
</tr>
<tr>
<td>Bis</td>
<td>1.0 g</td>
</tr>
<tr>
<td>Glycerol</td>
<td>20.0 ml</td>
</tr>
<tr>
<td>H2O to make</td>
<td>100.0 ml</td>
</tr>
</tbody>
</table>

IB: Buffer solution:

Tris-HCl-SDS buffer (1.5M Tris)

<table>
<thead>
<tr>
<th>Component</th>
<th>Amount</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tris</td>
<td>18.3 g</td>
</tr>
<tr>
<td>HCl (conc.)</td>
<td>6.0 ml</td>
</tr>
<tr>
<td>10% SDS solution</td>
<td>4.0 ml</td>
</tr>
<tr>
<td>H2O to make</td>
<td>100.0 ml</td>
</tr>
</tbody>
</table>
IC: 0.4% (w/v) APS solution

Ammonium persulphate 0.4 g
H₂O to make 100.0 ml

ID: 0.8% (v/v) TEMED solution

TEMED 0.8 ml
H₂O to make 100.0 ml

II: Stock solution for stacking gel

IIA - Acrylamide-bis solution
Acrylamide 38.0 g
Bis 2.0 g
Glycerol 20.0 ml
H₂O to make 100.0 ml

IIB - Tris-HCl-SDS buffer (0.50M tris)
Tris 3.0 g
HCl (conc.) 2.0 ml
10% SDS 4.0 ml
H₂O to make 100.0 ml

IIC and IID as IC and ID.

III: Electrode Buffer (pH 8.3)

IIIA - Tris-glycine-SDS buffer (0.025 M tris)
Tris 3.0 g
Glycine 14.4 g
10% SDS 10.0 ml
H₂O to make 1.0 liter

IV: Stock solution used in sample preparation

2.5% SDS for treating samples
0.5M Tris-HCl buffer (pH 6.8) 3.0 ml
10% SDS solution 5.0 ml
Glycerine 10.0 ml
Mercaptoethanol 2.0 ml

Bromophenol blue solution (0.00.0%)

1% BrB Methanol 0.5 ml
H₂O 100.0 ml

Composition of running gels

Stock solutions of IA, IB, IC, and H₂O were mixed in the proportions as given in Table 2. The proportions of
of the IA solution to water were varied to give different gel concentrations up to 20% acrylamide gel (Table 3), then ID solution was added.

Preparation of the stacking gel. - The stacking gel mixture was prepared by mixing the solution as given in Table 3. A 4% acrylamide stacking gel was optional for nearly all experiments.

Preparation of the sample. - Cooked beef, lamb, pork and horse meats were prepared as described by Ogita and Markert (1979).

Gel protein staining and destaining. - The gel was stained for 1 hour at 25°C with a 0.1% Coomassie brilliant blue solution freshly made up in 50% trichloroacetic acid (TCA) and then destained by repeated washing in 10% acetic acid for several times until the gel became clear.

Procedure

A. Prior to assembling of the EC-470 vertical gel cell, coat each gasket with approximately 2.5 ml molten agarose. Assemble cell and place in 45° angle. Pour running gel into gel slab area, overlay carefully with distilled water. Allow 40 minutes for complete polymerization then pour off distilled water.

B. Stand cell in horizontal position. Pour stacking gel into gel slab area and insert the slot form.
Allow to get for 50 minutes, then connect the cooling plates to the circulating water bath at 4°C and turn it on.

C. Remove the excess gel above the slot form, place the cell in a vertical position, and fill the upper electrode chamber with electrode buffer. Carefully remove the slot form and add the remaining electrode buffer to the cell.

D. Add 5 ml of 0.001% bromophenol blue dye solution to the upper electrode chamber. Connect the electrophoresis cell to the power supply with lower electrode as positive.

E. Carefully add 50 ul (0.05 ml) of the sample to the cell. Turn the power on, and apply 400V until the front is migrated 10 cm from the bottom of the slots.

F. Turn off the power and disconnect the cell. Drain off the electrode buffer and disassemble the cell. Remove the gel and stain it and destain it.
Table 3
Comparison of Polyacrylamide Gel of Different Concentrations

<table>
<thead>
<tr>
<th>Composition of Running Gels*</th>
<th></th>
<th>Composition of Running Gels*</th>
</tr>
</thead>
<tbody>
<tr>
<td>Percentage gel</td>
<td>Stock solution IA(ml)</td>
<td>H₂O(ml)</td>
</tr>
<tr>
<td>3</td>
<td>1.5</td>
<td>8.5</td>
</tr>
<tr>
<td>4</td>
<td>2.0</td>
<td>8.0</td>
</tr>
<tr>
<td>5</td>
<td>2.5</td>
<td>7.5</td>
</tr>
<tr>
<td>6</td>
<td>3.0</td>
<td>7.0</td>
</tr>
<tr>
<td>7</td>
<td>3.5</td>
<td>6.5</td>
</tr>
<tr>
<td>8</td>
<td>4.0</td>
<td>6.0</td>
</tr>
<tr>
<td>9</td>
<td>4.5</td>
<td>5.5</td>
</tr>
<tr>
<td>10</td>
<td>5</td>
<td>5</td>
</tr>
<tr>
<td>11</td>
<td>5.5</td>
<td>4.5</td>
</tr>
<tr>
<td>12</td>
<td>6.0</td>
<td>4.00</td>
</tr>
<tr>
<td>13</td>
<td>6.5</td>
<td>3.5</td>
</tr>
</tbody>
</table>

*In preparation each concentration of the gel according to these tables. Aliquots of solutions IB, IC, and ID are also added in the ratio of 5 ml of IB, 2.5 ml of IC and 2.5 ml of ID.
RESULTS AND DISCUSSION

SDS was tried as the procedure for separation and identification of cooked meats. The meat extracted with this method and also the meat which was extracted with SDS (SDS-1, SDS-2, SDS-3, SUM and SM) in the initial trials for protein solubilization in order to choose an extraction and applying procedure for cooked meats were applied to this method using 7% running gel and 4% stacking gel. After staining and destaining, no bands appeared. This could be related to one of the following problems which have been studied and been associated with SDS. Stoklosa and Latz (1974) indicated that it was necessary to add SDS only to sample solutions. In 1975 they found that the amount of SDS added to sample solutions had a crucial effect on the electrophoretic behavior of protein bands. Nelles and Bamburg (1976) reported that the presence of a small amount of SDS in the upper buffer chamber was necessary to obtain decent results using SDS-polyacrylamide gel electrophoresis.

Kuba et al. (1979) found that the gel concentration had a significant effect on the electrophoresis of SDS modified proteins added in SDS-polyacrylamide gel electrophoresis or could be a staining problem, such as time, amount or kind or chemical resources. Because SDS dissociates the protein into subunits and completely unfolds each polypeptide chain
to form a long, rod-like SDS-polypeptide complex (Lehninger, 1976), 9 percent and 12% running gel have been tried, but also with no resolution.
APPENDIX C

Polyacrylamide gel electrofocusing for identification and differentiation of species in cooked meat by vertical plate gel electrophoresis.

Sample preparation

The samples were prepared as described previously employing the optimum extraction method developed, i.e.,

Electrode solutions

a/ Anode (lower compartment) 1M NaOH
b/ Cathode (upper compartment) 1M HAc

Solution for electrofocusing

plug gel solution - (10% polyacrylamide)

polyacrylamide 10.0 g

distilled water to 100.0 ml

running gel solution - (5X cyanogen with 0.5X ampholyte)

cyanogen 3.0 g

0.5X ampholyte solution 2.0 ml

APPENDIX C

Polyacrylamide gel electrofocusing for identification and differentiation of species in cooked meat by vertical plate gel electrophoresis.

Sample preparation

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Electrode solutions

a/ Anode (lower compartment) 1M NaOH
b/ Cathode (upper compartment) 1M HAc

Solution for electrofocusing

plug gel solution - (10% polyacrylamide)

polyacrylamide 10.0 g

distilled water to 100.0 ml

running gel solution - (5X cyanogen with 0.5X ampholyte)

cyanogen 3.0 g

0.5X ampholyte solution 2.0 ml
Recently more interest has centered on isoelectric focusing in polyacrylamide gels either in disc or slab in detection and identification of protein. Therefore polyacrylamide gel electrofocusing as recommended by E-C Apparatus Corporation was tried with some modification as follows:

**Solution for electrofocusing**

<table>
<thead>
<tr>
<th>Component</th>
<th>Description</th>
<th>Amount(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>plug gel solution - (10% cyanogum 41)</strong></td>
<td>cyanogum-41</td>
<td>10.0 g</td>
</tr>
<tr>
<td></td>
<td>distilled water to TMED</td>
<td>100.0 ml</td>
</tr>
<tr>
<td></td>
<td>AP</td>
<td>0.1 ml</td>
</tr>
<tr>
<td></td>
<td></td>
<td>100.0 mg</td>
</tr>
<tr>
<td><strong>running gel solution - (5% cyanogum with 1.3% ampholyte)</strong></td>
<td>cyanogum-41</td>
<td>3.0 g</td>
</tr>
<tr>
<td></td>
<td>40% ampholyte solution</td>
<td>2.0 ml</td>
</tr>
<tr>
<td></td>
<td>water to TMED</td>
<td>60.0 ml</td>
</tr>
<tr>
<td></td>
<td>AP</td>
<td>0.4 ml</td>
</tr>
<tr>
<td></td>
<td></td>
<td>0.05 g</td>
</tr>
<tr>
<td><strong>gasket seal material - (0.5% agarose)</strong></td>
<td>agarose</td>
<td>0.05 g</td>
</tr>
<tr>
<td></td>
<td>distilled water to TMED</td>
<td>10.0 ml</td>
</tr>
</tbody>
</table>

**Sample preparation**

The samples were prepared as described previously employing the optimum extraction method developed, BGU.

**Electrode solutions**

a) Anode (lower compartment) 1M$_3$HPO$_4$

b) Cathode (upper compartment) 1M NaOH
Overlay solution
10% ampholyte

Fixation solution
methanol 900 ml
acetic acid 20 ml
water to 2 l

Staining solution - (0.05%)
Coomassie Brilliant Blue R-250 or G-250 0.5 g
Fixation solution 100 ml

PROCEDURE

A. Prior to assembly of the EC 470 vertical gel cell, coat each gasket with approximately 2.5 ml molten agarose. Assemble cell and place in 45° angle. Add catalysts to 100 ml of 10% gel plug solution and pour into gel slab area. Overlay carefully with the distilled water. Allow 30 minutes for complete polymerization.

B. Stand cell in vertical position. Test plug and gasket seal by filling gel area with distilled water. Observe if level is maintained. If not, place cell in 45° after pouring off distilled water and add 2 ml of gel plug solution as in step A with water overlay. Retest unit. If failure results again, begin at step A. If successful in original test or retest, pour off distilled water and shake excess droplets from gel slab area.
C. Stand cell in vertical position. Add catalysts to running gel solution and pour into gel slab area between cooling plates. Place slot former of desired number of teeth into position. Please note, a small weight on the slot former may be necessary to maintain correct position during polymerization. Allow 30-45 minutes for polymerization.

D. Place overflow tube into top buffer compartment into the hole provided. Plug buffer drain and recirculation holes in upper and lower buffer tanks with polyethylene plugs. Fill upper buffer compartment with distilled water. Carefully remove slot former from gel. Keeping cell upright, drain water completely from upper and lower compartments.

E. Add 25 ul samples to each slot.

F. Carefully add overlay solution to fill each slot.

G. Remove overflow tube and add bottom buffer to lower compartment with a long neck funnel. Add approximately 500 ml.

H. After filling lower buffer compartment use a polyethylene plug in lieu of the overflow tube to isolate the bottom buffer from the top buffer. Carefully add top compartment buffer, being sure not to disturb the samples.

I. Place safety cover on unit. Connect cooling system. Connect power supply with positive lead to the acid containing buffer compartment.
J. Energize power supply. Using the EC 400 constant power supply, set at 20 watts constant power.

K. After separation, drain and disassemble cell in the usual manner. Remove gel above the plug for fixation and staining.

L. Place gel in one liter of warm, 50°C, fixation solution for 4 hours minimum.

M. Stain gel for 1 hour or until bands are visible.

N. Destain either electrophoretically or by exhaustive washing in fixation solution.
RESULTS AND DISCUSSION

Polyacrylamide gel electrofocusing procedure as recommended by E-C Apparatus Corporation was tried with some modification. Initially the heat generation proved excessive after short time of the run at the recommended 20 watts (W). So, the run was repeated using 10W but heat generation proved excessive after a period of time and therefore, the power was cut to 2.5W. Even so, there was still a small heat effect on the gel. Figure 9 shows polyacrylamide gel electrofocusing of raw, cooked (beef, pork, lamb and horse meats) and canned meat (corned beef and mixture of pork and ham). It shows no migration or bands, and that may be because of the thickness of the gel which generates heat or because the sample contained salt, which contributed to generated heat, or both. Also it could be a problem caused by oxygen generated at the anode. This is partially dissolved in the medium, and may oxidize the SH-groups of protein. (Goal et al., 1980).

An overview of the times for running, fixing, staining, and destaining of electrophoresis and isoelectric focusing in polyacrylamide gel is shown in Table 4. These results indicated that electrophoresis is less consuming of time and more applicable for routine work.
Figure 9. Polyacrylamide gel electrofocusing of raw, cooked (at 75°) and canned meats.

Table 4

Time in Hours for Electrophoresis and Isoelectric Focusing in Polyacrylamide Gel

<table>
<thead>
<tr>
<th>Technique</th>
<th>Running Time</th>
<th>Fixing</th>
<th>Staining</th>
<th>Destaining</th>
<th>Total Time</th>
</tr>
</thead>
<tbody>
<tr>
<td>Electrophoresis</td>
<td>2</td>
<td>-</td>
<td>0.4</td>
<td>8-10</td>
<td>10.4-12.4</td>
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<tr>
<td>Isoelectrofocusing</td>
<td>6-24</td>
<td>6</td>
<td>1-10</td>
<td>4-6</td>
<td>17-46</td>
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APPENDIX D

Bibliography of the Complete Thesis
BIBLIOGRAPHY


