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## The Influence of Formulation and Device Variables on the Intransasal Absorption of Salmon Calcitonin

Ramneik Dua  
*University of Rhode Island*

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THE INFLUENCE OF FORMULATION AND DEVICE VARIABLES  
ON THE INTRANASAL ABSORPTION OF  
SALMON CALCITONIN

BY  
RAMNEIK DUA

A DISSERTATION SUBMITTED IN PARTIAL FULFILLMENT OF THE  
REQUIREMENTS FOR THE DEGREE OF  
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IN  
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1995

DOCTOR OF PHILOSOPHY DISSERTATION  
OF  
RAMNEIK DUA

APPROVED:

Dissertation Committee

Major Professor *John S. Woodman*

*Amos Fia*

*Edward Carey*

*Ronald D. Hedlund*

DEAN OF THE GRADUATE SCHOOL

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## ABSTRACT

The nasal administration of drugs, particularly peptides, is very beneficial but like other mucosal routes, suffers from low bioavailability for the higher molecular weight compounds. The nasal bioavailability of peptides and proteins is influenced by the dosage form as well as the devices used for administration. In the present study an effort was made to investigate the effect of formulation and device variables on the intranasal delivery and absorption of salmon calcitonin (sCT). The formulations were designed as nasal sprays with viscosity of 1 and 76 cps using methylcellulose as a viscosity enhancing agent at 0 and 1% w/w; tonicity of 100 mOsm, 300 mOsm and 600 mOsm using sodium chloride as a tonicity adjusting agent; an acidic phospholipid, Dimyristoylphosphatidylglycerol (DMPG), as an absorption enhancer at 1% w/w and chlorobutanol as the preservative. The formulations were investigated for viscosity by using a cone and plate viscometer and for droplet size distribution with a Malvern laser sizer. The selected formulations were delivered to the nasal cavity of healthy male New Zealand rabbits using a commercially available metered spray pump and a prototype device, nasal micron spray pump, to facilitate a uniform distribution of the spray in the nasal cavity. The components of the devices, mechanism of pump and actuator design were evaluated for dose accuracy as well as their ability to generate a uniform distribution of spray in the nasal cavity. A full factorial design with tonicity at three levels, viscosity at two levels and concentration of enhancer at two levels was used as the experimental design. Serum levels of sCT were determined using a double antibody radioimmunoassay. The area under the curve (AUC) and bioavailability were determined using pharmacokinetic software MKDATA. The pharmacodynamic effect of salmon calcitonin of lowering of calcium was

measured, by complexation with arsenazo III, using a visible spectrophotometric technique at 650 nm. The data was analyzed using ANOVA and the comparisons among the treatments were performed using Scheffe's multiple comparison analysis and Fisher's least significance test. Deviation from isotonicity (300 mOsm) enhanced the intranasal bioavailability of sCT by 4-5 fold while the addition of DMPG at a concentration of 1% w/w doubled the bioavailability. Variation in the viscosity of formulations did not influence the bioavailability of salmon calcitonin. The data was interpreted using response surface methodology (RSM) with canonical analysis parameters demonstrating that optimum formulations could be designed.

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Graduate studies would not have been imaginable without the reassurance and unfailing support of my father, and my sisters, Puneit and Sumeit.

## PREFACE

I elected to write this dissertation following the format of the manuscript plan described in section 11-3 of the graduate manual at the University of Rhode Island. This option was most appropriate to present my results in various sections.

Section I consists of a general introduction of the topic and the objectives of my investigation. The three manuscripts, in Section II are the core of this study. Section III contains a set of appendices A to C giving experimental details and supporting data for the three manuscripts presented in Section II.

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**Dedicated to memory of my mother**

## SECTION I

## INTRODUCTION

Recent years have seen enormous advances in the field of protein and peptide engineering and a greater understanding of the way in which biological response modifiers function in the body. It is now possible through the use of recombinant DNA techniques, or by solid phase protein synthesis, to produce significant quantities of a wide variety of proteins and polypeptide hormones that are therapeutically active. The list of these response modifiers includes interferons, macrophage activation factors, neuropeptides and proteins that may have potential in cardiovascular diseases, inflammation, contraception etc. Prospects of using some of these materials in medicine have reached the stage where products have either been approved by regulatory authorities or are the subject of applications as investigational drugs or as new therapeutic agents. In most cases these agents are administered by injection, however those agents which are indicated for chronic therapy need to be administered by appropriate delivery systems.

Unfortunately the research technology for development of delivery systems for peptides and proteins has not kept pace with the rapid progress in biotechnology and consequently there are presently few systems that are entirely appropriate for the administration of macromolecules. Furthermore essential pharmacokinetic and correlative pharmacodynamic data is often missing, for example questions like where and how does the peptide function?, what dose is needed and how often should it be given?.

Many of these obstacles are related to the properties that set peptides and proteins apart from the vast majority of drug substances in use today. These properties include molecular size, susceptibility to proteolytic breakdown, rapid plasma clearance, peculiar dose-response curves, immunogenicity,

biocompatibility and the tendency of peptide or protein to undergo aggregation, adsorption and denaturation. Therefore existing drug delivery systems must be modified and novel ones must be developed to overcome these obstacles<sup>1</sup>.

Although the biotechnology industry has prompted a resurgence of interest in noninvasive delivery systems, there have been studies performed as early as 1922 investigating the efficacy of insulin by intranasal, oral, rectal and vaginal routes<sup>2</sup>. This early work and that of others up to the 1960's met with little success. However these studies served to illustrate the major problems of noninvasive delivery which are low bioavailability, unreliable dosing, and acute local toxicity of formulations using permeation enhancers; these problems still remain the major challenges<sup>3-7</sup>.

Some of the noninvasive routes investigated in the recent years are oral, ocular, buccal, pulmonary, transdermal, rectal, vaginal and nasal. For each route of delivery, there are two potential barriers to absorption, poor permeability and enzymatic barriers. The problems of poor permeability, high inter subject variability and acute toxicity associated with delivering protein and peptide drugs by noninvasive routes may be minimized by using enzymatic inhibitors and/or novel absorption enhancers. The delivery by these routes may not be bioequivalent to invasive methods, however the convenience to patient will, in some cases, outweigh the demand for complete bioequivalence<sup>8</sup>.

The main emphasis is to rationally evaluate the problems for peptide and protein drug delivery systems and the options available for rational and

optimal drug delivery.

### **Oral Delivery**

A wide variety of routes of administration and delivery systems exist, but by far the most popular approach is oral delivery where the drug is intended to be absorbed from the gastrointestinal tract (GIT). However the delivery of peptides and proteins via the GIT is difficult because of their inherent instability due to the presence of proteolytic enzymes and the poor permeability of the intestinal mucosa to high molecular weight polypeptides and proteins<sup>9</sup>.

Although generally unsuccessful at present, the potential therapeutic and commercial benefits of successful oral delivery of peptides and proteins are sufficiently compelling to sustain research. The principal strategy has been the protection of a given peptide or protein from the proteolytic enzymes in the lumen of the gastrointestinal tract. Insulin was encapsulated with an impervious azopolymer film that remains intact in the GIT until it reaches the large intestine. Once in the colon, the azo bonds of the polymer are reduced by the indigenous microflora releasing the peptide. This strategy was used for vasopressin and insulin and biologically active amounts were absorbed into the blood stream of the rat. In spite of these encouraging results, the general utility of this approach in oral delivery is far from certain and requires additional information on the effects of both bacterial proteolytic enzymes on peptide stability and fecal matter on peptide absorption within the colon<sup>10</sup>.

The less than successful delivery of oral dosage forms for proteins and

polypeptides has led to the investigation of alternative noninvasive routes such as transdermal or mucosal routes. Peptides and proteins when delivered via these alternative routes may be exposed to a smaller surface area, hence a smaller fraction of enzymes are present, as compared to the oral route. Peptides and proteins are subjected to less dilution via these routes since the resident fluid volume is less, thereby maximizing the concentration gradient for absorption.

### **Transdermal Delivery**

The skin is composed of two major tissue layers. The outermost layer of the epidermis, the stratum corneum, is composed of thin flat, layered keratinized cells, 15 to 20 layers thick and is major barrier to skin permeation. The epidermis is much thinner than the dermis and is made up of stratified keratinized cells. The underlying layer, or dermis is extensively vascularized and contains a nerve network. While the stratum corneum lacks in enzymes, the next viable layers contain numerous enzymes that can degrade peptide and protein drugs<sup>11-12</sup>.

The transdermal route has been successful in systemically delivering small drug molecules, such as scopolamine, nitroglycerine, clonidine, and estradiol. The delivery of peptide and protein has shown a greater resistance to absorption, primarily because of impermeability to the stratum corneum. At the present time it seems highly unlikely that the simple application of a peptide in a vehicle to the skin will produce suitable clinical effects<sup>13</sup>. The enhancement of peptide/protein skin permeation via electric current, iontophoresis, has shown some promise for the transdermal delivery of these

drugs. This mode of delivery may have the potential to deliver the drug in a pulsatile manner, which is critical for the desired physiological response of some of the hormonal drugs<sup>14-15</sup>.

However the iontophoretic transport of gonadotrophin-releasing hormone (GnRH), produced erratic results<sup>16</sup>. The LHRH analog, leuprolide, was explored for iontophoretic facilitated transport and there was no measurable flux at 0.5V<sup>17</sup>. Results obtained with the thyrotropin releasing hormone in nude mouse skin<sup>18</sup> and insulin in albino rabbit skin<sup>19</sup> indicated that iontophoresis offers benefits only when the integrity of the stratum corneum is compromised. In contrast, Siddiqui et al, reported iontophoretic facilitated transport of insulin across the skin of diabetic rats without first stripping the stratum corneum, although there were significant inter-animal variations in the extent of blood glucose lowering<sup>20</sup>.

No explanation has been suggested for these discrepancies in the role of the stratum corneum in limiting transdermal insulin absorption during iontophoresis. These equivocal results, coupled with the uncertain long-term effects of applied current on the skin cast doubt on the viability of iontophoresis as a routine, noninvasive technique to improve transdermal peptide and protein delivery<sup>21</sup>.

### **Ocular Delivery**

The visible part of the eye consists of the iris and the sclera. The iris is covered by cornea and the sclera is covered by a conjunctival membrane. Absorption primarily occurs through the cornea or the conjunctival membrane.

The systemic absorption of peptides and proteins may occur through contact with the conjunctiva and the nasal mucosa, the latter occurring as the result of drainage through the nasolacrimal duct. For optimal systemic delivery the absorption through the conjunctival and nasal mucosa must be maximized.

The main difficulty in the delivery of drugs to the eye is the rapid removal of the substances by blinking and drainage through the lacrimal duct and the low permeability of the cornea<sup>22</sup>. There is a significant level of aminopeptidases in the eye fluids and tissues which may degrade the peptides<sup>23</sup>.

Several investigators have studied the delivery of peptides and proteins into the systemic circulation by the ocular route<sup>24-29</sup>. It has been demonstrated that it is not possible to achieve therapeutic levels of insulin without the use of a permeation enhancer. The bioavailability of insulin was improved by coadministration of permeation enhancers such as polyoxyethylene-9-lauryl ether, sodium deoxycholate, sodium glycocholate. Over 80% of the insulin reaching the systemic circulation was found to be absorbed through the nasal mucosa and the absorption pattern followed that reported for nasal delivery quite closely<sup>30</sup>. The coadministration of permeation enhancers offers promising results but their long term effects on the ocular physiology and histology need further characterization.

### **Pulmonary Delivery**

Pulmonary delivery systems have also been considered as a method of improving the bioavailability of peptide and proteins because of certain

advantageous features offered by the lung. The walls of the alveoli in the deep lung are extremely thin (0.1 to 1  $\mu\text{m}$ ) as compared to the capillary walls (7  $\mu\text{m}$ ) or red blood cells (8  $\mu\text{m}$ ). They are an order of magnitude thinner than typical mucosal or epithelial membranes, which are several millimeters thick. In addition the surface area of the lung is extremely large, approximately 75  $\text{m}^2$  for a 70 kg male. Furthermore one puff of an aerosol administered correctly can cover this surface area within seconds<sup>31</sup>.

However the lungs are rich in enzymes. Virtually all known peptide hydrolases (exopeptidases and proteinases) are present in the lung tissue. Several membrane-bound peptidases exist on pulmonary vascular surfaces and a wide variety of proteinases are present in other lung cells. Several peptidases are distributed throughout the lung and they may be impossible to saturate and can cause rapid degradation of peptides<sup>32</sup>.

The particle size distribution of aerosolized drug is very critical to the bioavailability of the aerosol formulations. The particle size often defined as mass median aerodynamic diameter (MMAD), must be approximately 5  $\mu\text{m}$  or less in order for drug particles to reach lung. A major concern with aerosol formulations is the denaturation of peptides and proteins during shaking and aerosolization, the dispersibility, the particle size as well as the more common concerns of chemical stability, safety and efficacy. A typical distribution of dose delivered from a metered dose inhaler indicated that only 4.7% of the delivered dose actually reaches the peripheral lung and 6.3% reaches the central lung<sup>33</sup>.

The absorption of the fraction of the drug that is able to reach the lung

depends upon factors singular to respiratory system, such as tidal volume, respiratory rate, respiratory pause, respiratory distress syndrome, lung volume, emboli, total lung burden, solubilization in alveolar macrophages, recirculation, pancreatitis, exercise, interstitial lung diseases etc<sup>34</sup>. Thus a lot of variables besides the formulation variables have to be taken into account for optimal delivery by pulmonary route

### **Buccal Delivery**

The buccal membrane is similar to the skin in that it is a stratified squamous epithelium, with cellular extrusion products filling the intercellular spaces. The buccal enzymatic barrier is greater than other routes and enzyme activity has been reported to be greater than at the vaginal or nasal mucosa.

However, the buccal route offers convenience and is generally more acceptable to the patient than other routes. This route of delivery offers a much larger and easily accessible area for placement of delivery systems such as adhesive tablets, but there may be a risk of drug degradation by accidental swallowing or by the salivary wash-out of the tablet. Therefore the preferred dosage form for buccal delivery is bioadhesive patches. The formulation aspects of bioadhesive patches include variables such as the type of polymers used, the viscosity grade of the polymers, and the amount of polymer per patch which can further aid in optimizing the formulation<sup>35</sup>.

Permeability offered by buccal route is often poor as compared to other mucosal routes. Investigation of the buccal delivery of protirelin indicated the buccal dose required to achieve an equivalent pharmacodynamic response

was about 200 times higher than that for iv injection of the peptide. Thus bioadhesive patches alone may not improve bioavailability sufficiently to reach therapeutic levels and may require the coadministration of permeation enhancers<sup>36</sup>.

### **Rectal Delivery**

The human rectum offers a surface area of 200 to 400 cm<sup>2</sup>. It is abundant in blood, lymphatic vessels, and microflora and has a fairly constant environment with respect to temperature, pH, and viscosity of rectal fluid<sup>37</sup>.

The enzymatic degradation of model enkephalin peptides has been shown to be greatest in rectum and an absorption enhancer is required to achieve therapeutic levels.

The dosage forms for rectal delivery of peptides and proteins are usually more conventional such as solutions, gels, and suppositories. Of these, gels are the most efficient because they offer a proper balance between retention at the site of administration and the rate of peptide release<sup>38</sup>. It was demonstrated that the rectal bioavailability of [Asu<sup>1, 7</sup>]-eel calcitonin in the rat was best from a 0.1% polyacrylic acid gel when compared with other dosage forms such as saline, a triglyceride fatty acid mixture base, and a polyethylene glycol 1000 base. Following administration, solutions tend to spread away from the rectum into the ascending colon and from there into portal rather than general circulation<sup>39</sup> thereby resulting in reduced peptide and protein bioavailability. Recently, more sophisticated systems such as the osmotic pumps and hydrogel cylinders have been investigated as rate-controlled rectal

drug delivery systems, and while the results have been positive, these dosage forms have not yet been tested with peptide and proteins<sup>40-41</sup>.

There does not seem to be significant absorption via rectal route without the use of a permeation enhancer as demonstrated by the delivery of des-glycinamide arginine-vasopressin which when enhanced with STDHF reached an optimum concentration of 4%<sup>42</sup>.

However the lack of acceptability of this route by many patients and the interruption of delivery by defecation makes this route less desirable than other routes<sup>43</sup>.

### **Nasal Delivery**

The epithelial cells of the nasal mucosa are covered with numerous microvilli and increase the surface available for drug absorption. The nasal cavity offers a high surface area of approximately 150 cm<sup>2</sup> for absorption. Furthermore the subepithelial layer is highly vascularized with large and fenestrated capillaries specially designed for the rapid passage of fluid through the vascular wall. The venous blood draining from the nose passes directly into the systemic circulation and is well suited for the absorption of drugs that are extensively metabolized<sup>44</sup>.

Nasal administration of nonpeptide drugs or peptides of ten residues or less has been very successful. Some examples are oxytocin, vasopressin and its analog desmopressin acetate (DDAVP); and luteinizing hormone-releasing hormone (LHRH) and its superanalogs buserelin, leuprolide, and nafarelin.

However when the number of amino acids is increased to 20 or greater, as in insulin, glucagon or growth hormone releasing hormone, low bioavailability results except when delivered with an enhancer<sup>45-46</sup>.

The nasal bioavailability of peptides and proteins is influenced by the dosage form as well as the technique of administration. The most popular dosage form has been the aqueous solution, although gels, powders and microspheres have been investigated

The nasal administration of drugs, particularly peptides, is very advantageous but like other routes suffers from low bioavailability. To achieve reproducibility with dosing and provide acceptable economy in the treatment, a bioavailability of 1-2% for an expensive recombinant peptide must be improved. It is necessary to optimize both the techniques of administration and the nasal formulations to provide a sufficient level of drug absorption.

The optimal nasal absorption delivery system should be well tolerated, inexpensive, should have low toxicity and reasonable bioavailability. Starch microspheres, water insoluble and swellable powders and phospholipids are novel absorption systems which have promising properties but need further characterization<sup>47</sup>.

There are a number of noninvasive routes available for the delivery of polypeptides and proteins but all suffer from the obvious disadvantage of proteolytic activity and poor permeability to higher molecular weight molecules. However, with the administration of either enzymes inhibitors or protective polymers, or perhaps permeation enhancers, bioavailability can be

substantially improved. Even though bioavailability by these routes may still be fairly low, the advantages of noninvasive therapy certainly outweigh the disadvantages.

A comparison of different routes for the delivery of protirelin via nasal, buccal, peroral and iv demonstrated that nasal administration was 10-times more efficient than the buccal route. Based on the pharmacodynamic effects observed, the nasal dose had to be five times higher than iv dose, whereas the buccal dose had to be forty times the nasal dose<sup>48</sup>.

Of all the alternative routes, the nasal route has received the most attention to date. Nasal delivery has the potential of being an extremely effective route for systemic administration of peptides and proteins. The selection of proper formulation parameters such as the pH of the vehicle, buffer species and concentration, ionic strength, viscosity, charge, osmolarity, type and concentration of preservatives as well as delivery with an appropriate device such as nasal spray pump, compressed air nebulizer, instillation catheter and dry powder insufflator could further influence and optimize the absorption of peptide and proteins.

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## OBJECTIVES

The objectives of the present research were:

1. To identify the formulation and device variables that are critical in enhancing the low intranasal absorption of the polypeptide, salmon calcitonin.
2. To systematically characterize the effect of formulation and device variables on the droplet size distribution and dynamics of spray profile.
3. To study the primary and interactive effects of varying levels of viscosity and tonicity on the nasal absorption of salmon calcitonin.
4. To study the effect of the novel absorption enhancer, Dimyristoylphosphatidylglycerol (DMPG) on the absorption of salmon calcitonin.
5. To characterize the optimal calcium lowering response of salmon calcitonin as a function of salt concentration, viscosity and concentration of absorption enhancer, dimyristoylphosphatidylglycerol.
6. To demonstrate the use of response surface methodology (RSM) as an aid in determining optimal formulations.

## SECTION II

**MANUSCRIPT I**

**EVALUATION OF FORMULATION AND DELIVERY DEVICE  
VARIABLES ON NASAL SPRAY CHARACTERISTICS**

## ABSTRACT

The aim of this study was to evaluate the effect of formulation variables; viscosity, tonicity, concentration of absorption enhancer and nasal delivery devices; mechanical metered nasal spray pumps, pump components and a prototype device, the nasal micron spray pump on the spray characteristics and dose administration of a model nasal product. Formulations were prepared within specified ranges of viscosity (1 to 380 cps), tonicity (100, 300 and 600 mOsm) and absorption enhancer (0 and 0.5% w/w). Viscosity was determined using a Brookfield cone and plate viscometer. The spray characteristics of the formulations were analyzed with a Malvern laser sizer and evaluated by characterizing the size distribution of the sprayed droplets using M50 and M90 parameters. For the metered nasal spray pumps, results indicated that among the formulation variables, increasing viscosity or addition of an absorption enhancer increased droplet size and produced a deviation from a Gaussian size distribution of the sprayed droplets. It was not possible to spray any solution greater than 2.12 cps and maintain a Gaussian spray distribution with a metered nasal spray pump. Tonicity did not demonstrate any significant effect on the droplet size distribution. Changes in pump based on two different mechanisms of spray delivery or actuator assembly design based on the orifice diameter or spray angle of spray insert did not produce any significant differences in spray characteristics. However with the use of a prototype device, the nasal micron spray pump, it was possible to spray solutions up to a viscosity of 76 cps as a fine spray.

**Key words:** Spray characteristics, Malvern laser sizer, Metered nasal spray pump, Formulation variables, Device variables, Nasal micron spray pump

## INTRODUCTION

x Several types of delivery devices such as nasal pump sprayers, compressed air nebulizers, instillation catheters and dry powder insufflators are used to administer drugs to the nose. There are mechanistic variations in these devices that produce differences in the delivery of a drug in terms of volume and reproducibility. The instillation catheter has high accuracy but only a small volume, less than 50  $\mu\text{l}$ , therefore limited doses can be delivered. An aerosol spray can deliver higher volumes, up to 140  $\mu\text{l}$  but the amount of active ingredients delivered is limited by drug solubility and viscosity. Dry powders offer the option of delivering the highest mass of active ingredients because they may not be formulated to contain a carrier vehicle. Thus 100% of mass delivered can be active ingredient. Reproducibility is also an important factor. Administration by instillation catheter and compressed air nebulizer have been shown to provide good reproducibility with respect to the amount of material delivered. On the other hand administration by a nasal spray pump or dry powder insufflator have been found to be considerably more variable particularly with respect to amount of material deposited in the nose<sup>1</sup>.

4 These devices influence the mode in which the drug is administered, i.e., whether the drug is delivered as drops, aerosol or spray. The mode of drug administration can affect drug distribution in the nasal cavity, which in turn influences bioavailability of a drug. It has been demonstrated that significant differences in drug distribution were observed when drug was administered by catheter, mechanical metered nasal spray pump and metered dose

pressurized aerosol. The results indicated that the metered nasal spray pump gave a constant dose and a very good nasal mucosal distribution<sup>2, 3.</sup> ] \*

\* [ Delivery devices as well as the density, shape, hygroscopicity of the particles and the particle size distribution affect the deposition site within the nasal cavity. The deposition site in turn influences the extent and rate of absorption which in turn is dependent on ciliary movement and/or diffusion in the mucous layer and subsequently the biological response in experimental animals and humans. ] [ A uniform distribution of particles throughout the nasal cavity may be achieved by delivering the particles from a nasal spray using a pressurized propellant. A metered dose delivery system developed for the nasal delivery of drugs could provide a consistent dose delivery and spray pattern which affects the deposition of droplets in the nasal cavity. Metered nasal spray devices have shown the ability to deposit well controlled doses in the nasal cavity which remain there for sufficient period to provide complete absorption<sup>4.</sup> ] \*

\* [ Experimental work has demonstrated that the relative bioavailability and biological response can be improved by administration of a drug via nasal spray pumps rather than by nasal drops. Nose drops disperse the dose throughout the length of the nasal cavity, from atrium to the nasopharynx. The nasal spray deposits droplets anteriorly in the nasal cavity with little or no dose reaching the turbinates. The drug deposited anteriorly in the nasal cavity remained in the nose longer while drops are cleared more rapidly than the dose administered as a spray<sup>5, 6.</sup> ] \*

\* [In addition to the proper choice of nasal device, it is imperative that studies be performed to select the variables such as concentration and volume of the dose to be sprayed for optimum intranasal delivery. Nasal bioavailability has been found to be a function of the volume and concentration of the nasal spray. It has been demonstrated that a formulation administered by two 50  $\mu$ l sprays gave a biological response superior to that of same drug administered by one 100  $\mu$ l dose<sup>7</sup>. Thus both administration devices and formulation design can affect drug distribution, deposition and the clearance process from the nasal cavity, leading to modifications in the biopharmaceutical aspects and bioavailability of the drug administered.] \*

Despite the fact that nasal delivery devices have been available for over a decade there is still very little fundamental information as to how variations in the design of a nasal pump assembly can interact with the formulation variables and affect the dynamics of spray plume, droplet size, spray deposition and distribution in the nasal cavity. This information along with an understanding of nasal formulation parameters could aid in optimized products designed to enhance bioavailability. This paper documents the nasal devices and their interaction with the formulation variables.

In this study two types of nasal spray devices; a metered nasal spray pump and its mechanical components i.e., mechanism of pump, design of actuator and a prototype device, the nasal micron spray pump were studied.

The metered nasal spray pump is a drug delivery device that administers a predetermined volume of formulation into the nasal cavity. It has been successfully utilized commercially in the administration of several locally

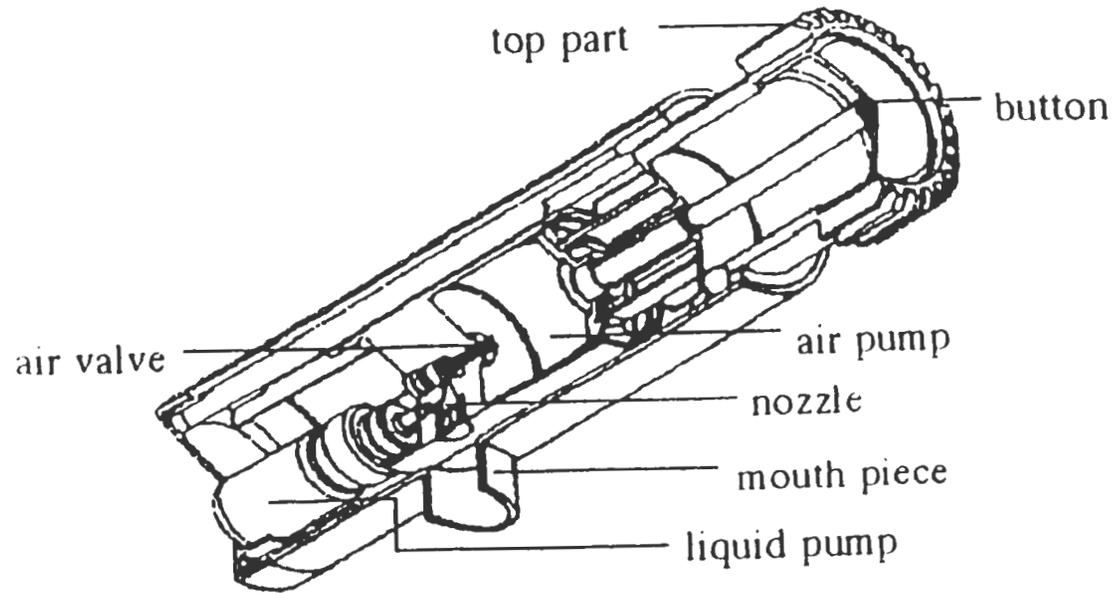
acting drugs, such as corticosteroids (Extracort, Syntaris), beclomethasone dipropionate (Aldecin, Beconase, Becotide, Vancenase), flunisolide (Nasalide), tramazoline (Tobispray), and nasal decongestants (Rhinospray, Otrivin, Afrin, Nostril, Nostrilla, Dristan); and is also used for the systemic delivery of calcitonin (Calcitonina, Calsynar), nafarelin acetate (Synarel), cromolyn acetate (Nasalcrom), insulin, desmopressin acetate, nitroglycerin. A typical nasal spray pump is composed of 10-13 spray components, to include metering chamber (5-7), actuator (2-3) and closure (2) components.

The metered nasal spray pump is available as (i) a ball and spring pump or (ii) a dual spring compression pump. As the names suggest, these pumps use two different mechanisms to aspirate the liquid from container and to release the spray. The ball and spring pump is composed of a cylinder, shut off valve (ball), spring, piston, stem, and a device for connection to the actuator. The dual spring compression or precompression pump is composed of piston, diptube and twin springs; these twin springs help to aspirate the liquid from container and also act as a shut off valve.

The actuator in a mechanical nasal spray pump aids in expelling the liquid mass aspirated by the pump assembly or metering chamber. Actuators are available as a one or two piece assembly and house a spray insert with a specified diameter to expel the liquid as spray. When a pump is actuated, a dose volume aliquot is released from the metering chamber and travels as a liquid mass until it encounters the restriction of the orifice of the spray insert. This restriction may have a series of spiral baffles, which begin the liquid breakup and impart a spinning motion to the liquid resulting in the spray mist. The spray pattern is determined by the number of spiral baffles and is

used to produce a wide (50-70°) or narrow (20-40°) spray cone. Droplet size is dependent upon force of the liquid mass striking the orifice, plus the design of the baffles, the diameter of the opening as well as the characteristics of the liquid itself.

It has been previously demonstrated that the addition of a viscosity enhancing agent increases contact time in the nose by reducing the rate of mucociliary clearance of the dosage form and may help to enhance the activity by increasing the "window" of absorption for the administered drug. Viscosity also seems to change the pattern of deposition and clearance<sup>8</sup>. However previous studies have investigated viscosity in the range of 1-3.5 cps, which can be considered as low viscosity solutions. The nasal micron spray pump (NMSP), a prototype device originally developed for administration of pulmonary inhalation solutions (figure 1) was used in this study to investigate the feasibility of using the device to deliver solutions of higher viscosity to the nose. The NMSP is a hand actuated propellant free spray device that is capable of producing a fine spray. The physical and technical requirements have been achieved by using an air blast nozzle (figure 2). The air blast nozzle is the heart of this spray device and its geometry is mainly responsible for the ability to generate small droplets especially for viscous solutions. The nozzle consists of a mixing chamber whose volume is partly filled by a polyurethane sponge with open pores. The NMSP also consists of two pumps, a liquid metering pump and an air pump. When the device is actuated by turning the top part in a rotational motion, a liquid volume is metered into the mixing chamber simultaneously. The device is now primed. By pressing the actuation button the air valve opens and air starts to flow through the nozzle and is compressed. This compressed



**View of the interior of  
Nasal Micron Spray Pump**

**Figure 1**

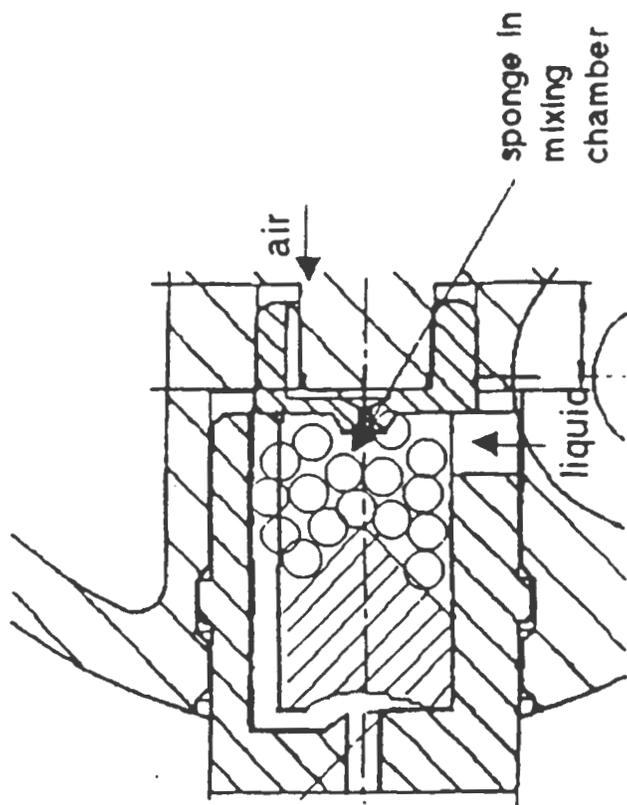


FIGURE2: Nozzle design for internal mixing of liquid and gas.

air expands through the nozzle system with increasing velocity. The solution is now accelerated by the expanding ambient air leaving the nozzle and the kinetic energy of the air stream pushes and shears the liquid which was previously dispersed in the sponge and thus disintegrates the liquid into small droplets<sup>9</sup>

The geometry of air blast nozzle of the NMSP provides a huge improvement as compared to the metered nasal spray pump which uses a mechanical actuator to spray the liquid mass. The use of this new device for nasal delivery could be invaluable for administering the drugs in high viscosity solutions because it is able to generate a fine mist of spray for distribution into the nasal cavity<sup>9</sup>.

Nasal devices are capable of generating different spray profiles because of the different types of components which makeup the functional unit. The literature shows no investigation of the influence of nasal administration device variables on spray characteristics. Likewise the spray characteristics may also be affected by the formulation variables.

Since it is strongly postulated that the functional components of a nasal pump assembly can interact with the formulation variables and affect the dynamics of spray profile, spray deposition and distribution in the nasal cavity; it was decided to evaluate the influence of the formulation variables: viscosity, tonicity, concentration of absorption enhancer with the nasal administration device variables: mechanism of pump (ball and spring versus dual spring compression pump), design of actuator (orifice diameter of spray insert, spray angle), on the spray characteristics and droplet size distribution.

## **EXPERIMENTAL**

### **Materials**

Methylcellulose, Methocel A15C Premium (Dow Chemical Co., Midland, MI) was used as a viscosity enhancing agent in the concentration of 0.25%-2% w/w. Sodium Chloride (Fisher Scientific, Fairlawn, NJ) was used to adjust tonicity. Dimyristoylphosphatidylglycerol (Avanti Polar Lipids, Alabaster, AL) was used in the concentration of 0.5% as absorption enhancer. Chlorbutanol 0.5% w/w (Sigma Chemical Co., St. Louis, MO) was used as a preservative.

### **Methodology**

#### **Selection of formulation variables**

##### **Viscosity**

Methylcellulose (Methocel), a natural carbohydrate that contains a repeating structure of anhydroglucose units was chosen as a viscosity enhancing agent. It is odorless and colorless. The premium grade was chosen as it is a GRAS substance and approved by FDA. Formulations containing methylcellulose were prepared in concentrations ranging from 0.25% to 2%.

##### **Tonicity**

Tonicity was investigated at three levels using sodium chloride as a tonicity adjusting agent. Formulations were prepared at 600 mOsms (hypertonic), 300 mOsm (isotonic) and 100 mOsms (hypotonic).

##### **Absorption Enhancer**

Dimyristoylphosphatidylglycerol, an acidic phospholipid, a novel permeation enhancer was used at a concentration of 0.5% w/w.

### **Preservative**

Chlorbutanol, a lipophilic preservative, 0.5% w/w was used as a preservative.

### **Preparation of Formulations**

The formulations were prepared with a concentration of methylcellulose ranging from 0.25% to 2% in a sodium acetate-acetic acid buffer at pH 4. Methylcellulose was dispersed in hot water (90-95° C) and stirred until all the particles were hydrated. Cold water was added slowly to the above mixture, the mixture was kept in an ice bath. The above dispersion was agitated for 3-4 hrs, the specified quantity of sodium chloride was added depending upon the osmolarity required for the formulation and then stored in a refrigerator overnight. The dispersion was brought to the pH 4 with buffer after bringing it to the room temperature. All the formulations were filled into 4 ml clear, round, Type I glass serum vials equipped with metered pumps.

### **Determination of Viscosity**

The viscosity of all the formulations was determined using a Brookfield cone-plate viscometer, Model DV II. This cone-plate viscometer is equipped with a detachable "cone" and can be used with cones ranging from 0.8 to 30° radians. Cone, CP 52D, diameter of 2.4 cms and a cone of 30° was selected to carry out the viscosity determinations at 25°C. The viscometer was calibrated using standards ranging from 1 to 500 cps. Viscosity was determined at rotational speeds ranging from 0.5 to 100 rpm. Shear torque readings were recorded at each rotational speed.

### **Selection of delivery systems for intranasal drug administration**

There has not been any investigation of the interaction of the components of a spray pump and the formulation variables on the spray profile and droplet size distribution. The evaluation of the metered nasal spray pumps was performed using the different available components of (i) pump mechanism ie., ball and spring (Pfeiffer); dual spring compression (Valois) (ii) orifice diameter of the actuator ie., 0.1 mm, 0.3 mm and 0.7 mm and (iii) angle of spray of the actuator ie., 20°, 35°, 60°.

### **Dose reproducibility studies**

After priming of the pumps the weight loss per spray actuation was recorded for 10 consecutive strokes and arithmetic mean and standard deviation were calculated. Since these devices would later be used to administer the drug in bioavailability studies and since the dose was to be administered within a interval of 60 minutes to the experimental animals, the dose accuracy was measured after 1 hr to determine if a second priming stroke was required before the second dose was administered. The procedure was also repeated, after 24 hrs.

### **Screening of Nasal Devices**

An initial qualitative screening was performed with formulations ranging in viscosity (1 cps to 380 cps), tonicity (100, 300 and 600 mOsm) and concentration of the absorption enhancer (0 and 0.5% w/w) in combination with the various nasal device components such as the type of pump and actuator design etc. The initial evaluation of the spray profile was obtained by spraying against a background of black and white. The spray pattern was categorized as a fine spray, stream or stream with drops. Only the nasal spray devices

producing a fine spray were later used to determine the droplet size distribution using a Malvern laser sizer.

#### \* Determination of Droplet size using the Malvern laser sizer

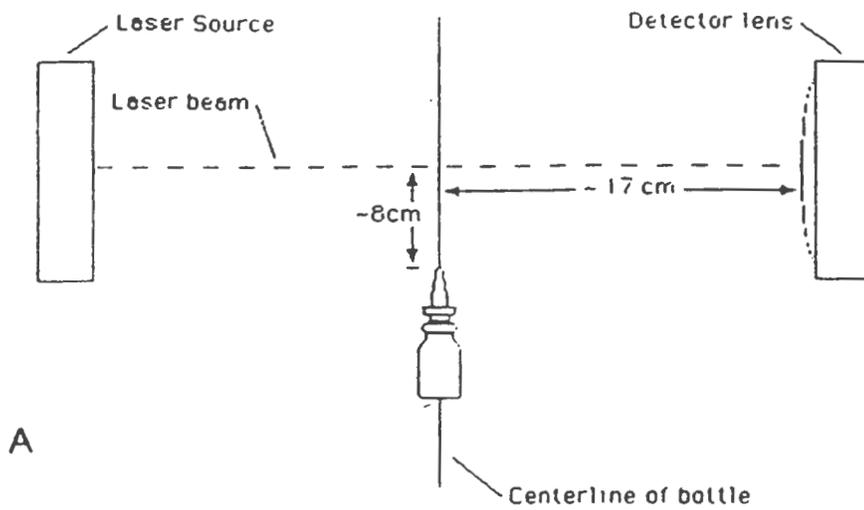
The samples were analyzed using a Malvern laser particle sizer (Malvern Instruments, Inc., Southborough, MA), equipped with a ~~2600c analyzing unit,~~ <sup>Digital PC 3000 computer</sup> and ~~NEC Powermate SX/16i computer,~~ <sup>DIGITAL PCXBV Color monitor</sup> ~~version B0 data-acquisition software,~~ <sup>Software: malvern masterizer software</sup> a

spray synchronizer (PS51), and infrared trigger sensor (PS57), NEC Multisync 2A monitor, and an Okidata Microline 320 dot matrix printer. An adjustable metal support bracket, to hold samples, was machined by Trans American Tool Co., Inc., Spring City, PA. The Malvern 2600c analyzing unit was equipped with a 300 mm range lens and configured as shown in figure 3. After performing a background measurement the sample bottle was positioned in the support at a distance of 10 cm from the actuator to the tip to the laser beam. The centerline of the bottle was 17 cm from the range lens. A "measure sample" command was executed from the key board prior to the actuation of the sample. Samples were actuated four times for each analysis. The analysis was performed in triplicate. The computer system was configured so that a total of 1000 sweeps (measurements) were executed during each analysis. The computer sweeps the detector at a rate of approximately 125 sweeps per second thereby giving the analyst sufficient time (about eight seconds) to perform four actuations per sample analysis of each sample.

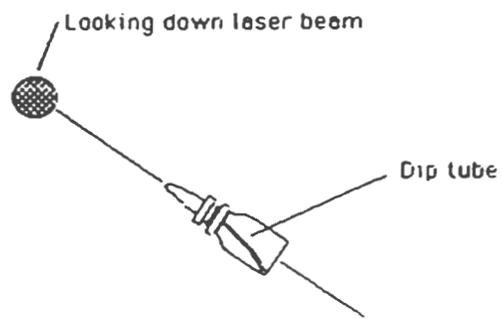
#### **Statistical Analysis**

In order to determine the influence of the experimental variables on the spray characteristics, the measured droplet size distributions were subjected to

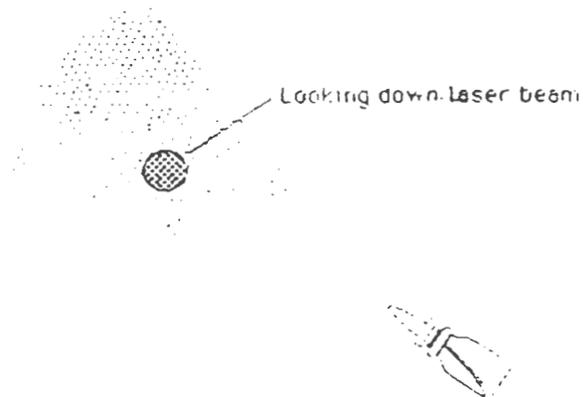
Figure 3 Configuration of Malvern laser sizer



A



B



C

ANOVA. A completely randomized design was used. The analysis was performed using Statistical Analysis Software (SAS) and subjected to an F Test and hypothesis testing at 0.01 and 0.05 significance level. According to the hypothesis testing it was assumed that there was no difference among the droplet size measured and in cases where  $F_0$  was greater than F at 0.01 level and 0.05 levels, the hypothesis testing was rejected. Scheffe's multicomparison test were performed to further indicate significant differences among groups

## **RESULTS AND DISCUSSION**

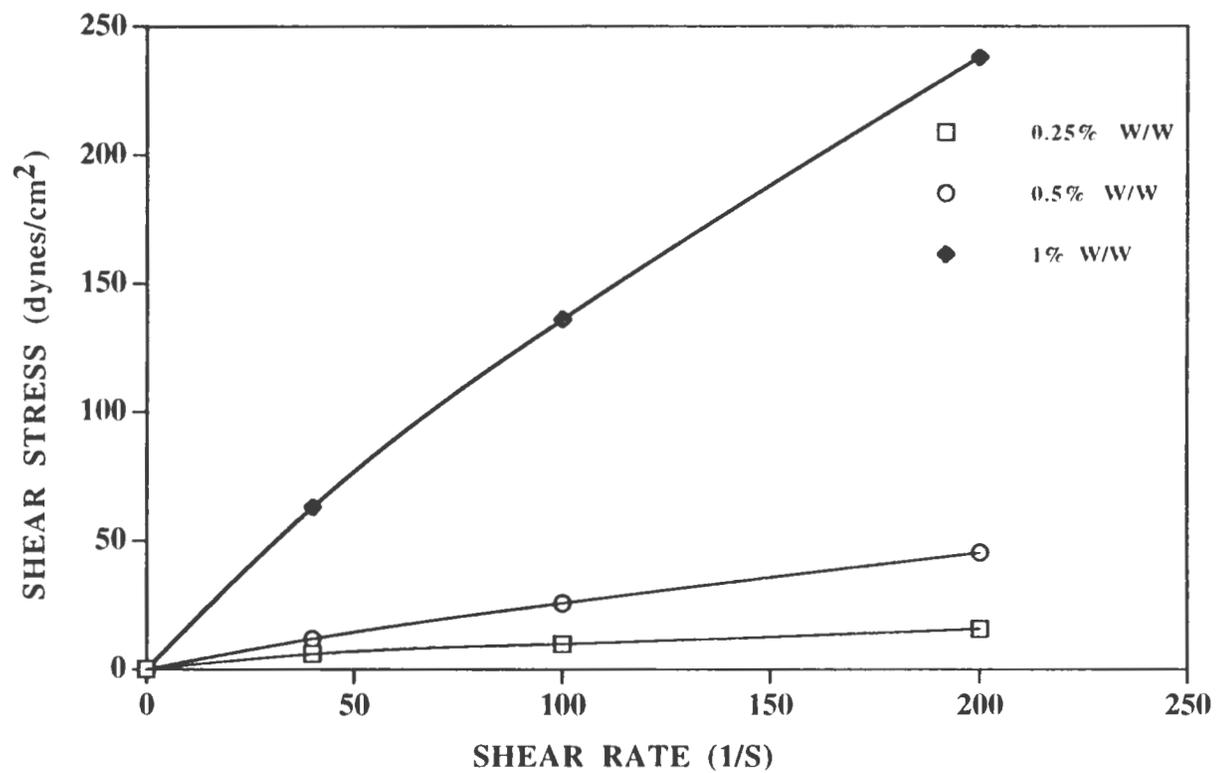
### **Viscosity studies**

The viscosity of the formulations varying in methylcellulose concentration was determined using a Brookfield cone and plate viscometer. The flow behavior was determined from a plot of shear rate versus shear stress (figure 4). Methylcellulose exhibited a non-Newtonian flow behavior. There was a nonlinear relationship between the shear rate and shear stress for the methylcellulose indicating a deviation from the Newtonian behavior. Thus the viscosity was determined at a fixed rotational speed of 50 rpm. The viscosity at 50 rpm was found to be 1 cps for formulations without methylcellulose. The viscosity was calculated to be 2.12 cps, 11.35 cps, 76 and 380 cps for formulations containing 0.25% w/w, 0.5% w/w, 1% w/w and 2% w/w of methylcellulose.

### **Qualitative screening of formulations**

All prepared formulations were qualitatively screened. When the nasal devices with various actuator designs were tested i.e., the metered spray pump

FIGURE 4 NON-NEWTONIAN FLOW EXHIBITED BY METHYLCELLULOSE



using either a ball and spring mechanism or a dual spring compression pump; the actuator designs with orifice diameters of spray insert of 0.1 mm, 0.3 mm and 0.7 mm and spray angle, 20°, 35° and 60°; all the test combinations generated a fine spray when the buffer solution with a viscosity of 1cps or a 0.25% w/w methylcellulose formulation with a viscosity of 2.12 cps was sprayed. All the formulations varying in tonicity, 300 mOsm, 600 mOsm and 100 mOsm or concentration of absorption enhancer, 0% w/w and 0.5% w/w produced a fine spray. However when formulations with 0.5% w/w methocel with a viscosity of 11.35 cps were tested with the various pump and actuator designs; all the test combinations generated a mixture of spray with drops.

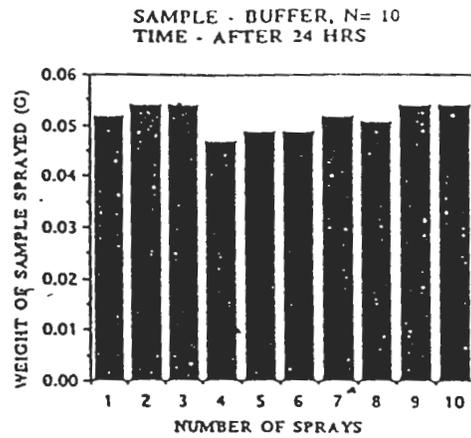
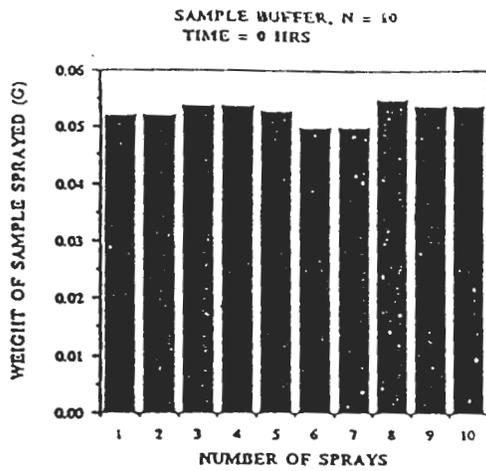
Formulations, with a viscosity greater than 11.35 cps were investigated using the (NMSP). It was determined that the NMSP was able to deliver all formulations up to viscosity of 76 cps generated by the 1% w/w methylcellulose as a fine spray. However the viscosity generated by 2% w/w methylcellulose (380 cps) and was released from the NMSP as a gel.

#### **Dose accuracy studies**

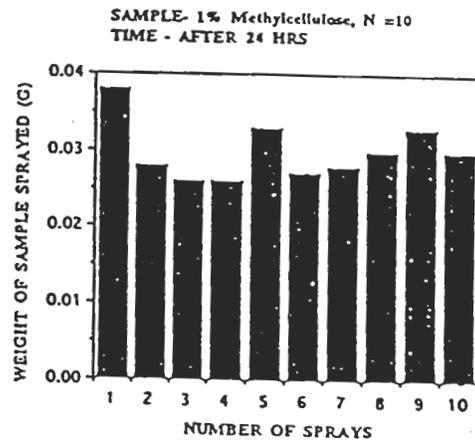
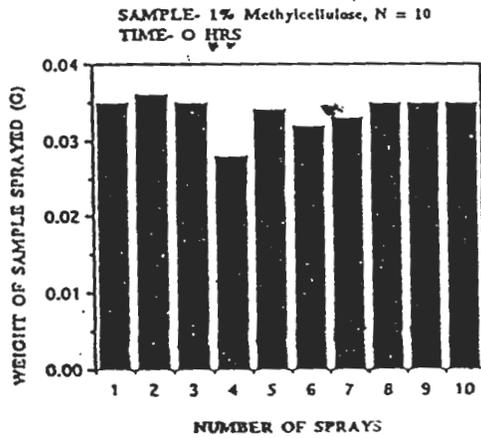
A total of 5-6 actuations were required to prime the metered nasal spray pump and ten actuations were required to prime the NMSP. This was determined by measuring the weight loss per spray actuation for a formulation possessing a mean weight of 50 mg for metered nasal spray pump and of 35 mg for the NMSP. The standard deviation was found to be  $\pm 2\%$  from the mean weight of the metered nasal spray pump and  $\pm 6\%$  from the mean weight of the NMSP immediately after priming the pump. However, while the deviation for the metered nasal spray pump was found

FIGURE 5 DOSE REPRODUCIBILITY STUDIES OF NASAL DEVICES

Metered Nasal Spray Pump



Nasal Micron Spray Pump (NMSP)



to be  $\pm 2\%$  from the mean weight; the NMSP showed a slight loss of primed value that increased the standard deviation to  $\pm 14\%$  from the mean weight after 24 hrs (figure 5).

### **Droplet size analysis**

#### **Metered nasal spray pump**

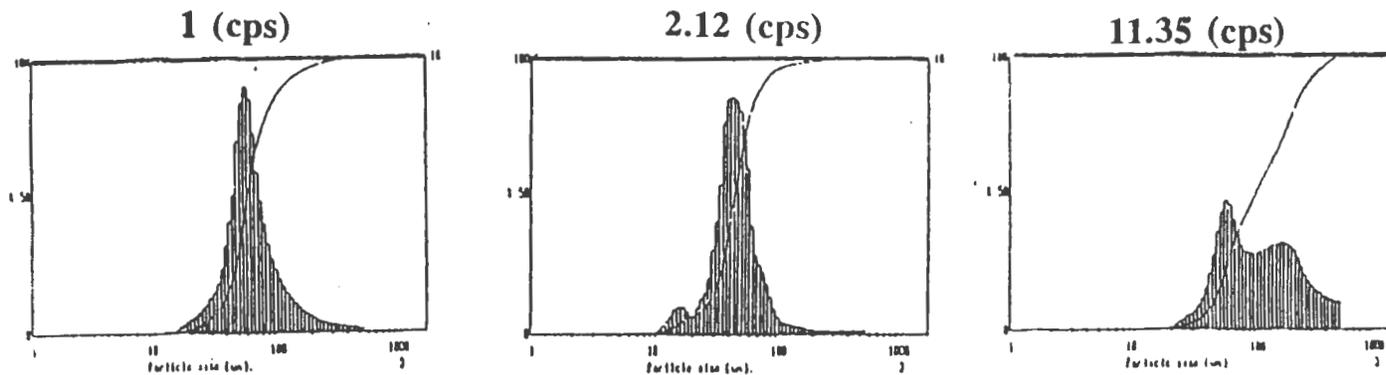
When using a Malvern laser sizer to measure the droplet produced by spray pump, the **M50** and **M90** seemed to best characterize the size distribution. The parameter **M50** represents average droplet size of 50% of the sprayed droplets and **M90** represents an average size of 90% of the sprayed droplet. The **M50** for buffer formulations with a viscosity of 1 cps was  $45.23 \pm 1.11 \mu\text{m}$ , and **M90** was  $68.41 \pm 7.94 \mu\text{m}$ . The **M50** for 0.25% methylcellulose formulation with an increased viscosity of 2.12 cps was  $56.97 \pm 3.45 \mu\text{m}$  and  $130.22 \pm 13.59 \mu\text{m}$  for the **M90** range. When increasing the methylcellulose to a 0.5% the viscosity was 11.35 cps and the **M50** increased to  $105.86 \pm 7.77 \mu\text{m}$  and the **M90** was  $305.04 \pm 18.54 \mu\text{m}$ .

The influence of the formulation variable, viscosity, on the spray profile and droplet size analysis (figure 6), demonstrated that the average droplet size increased with higher concentrations of methylcellulose. However with this increase in viscosity the profile of droplet size deviated from a Gaussian distribution. An ANOVA of these formulations varying in viscosity indicated a significant difference at the **M50** and at the **M90**. Scheffe's multiple comparison test and Fisher's Least Significance test indicated that the droplet size distribution obtained from 1 cps and 2.12 cps was significantly different from the distribution produced by solutions with a viscosity of 11.35 cps. The

FIGURE 6 THE INFLUENCE OF VISCOSITY ON THE DROPLET SIZE DISTRIBUTION  
USING METERED NASAL SPRAY PUMP

VISCOSITY						
	1 (cps)		2.12 (cps)		11.35 (cps)	
N = 12	M50	M90	M50	M90	M50	M90
AVERAGE	45.23	68.41	56.97	130.22	105.9	305
S. DEV	1.11	7.94	3.45	13.593	7.77	18.54

45



average at the **M90** level tends to produce a large standard deviation than that seen for the **M50**.

Isotonic formulations with a osmolarity of 300 mOsm generated a **M50** of  $45.23 \pm 1.15 \mu\text{m}$ , and **M90** was  $68.4 \pm 7.94 \mu\text{m}$ . Hypotonic formulations with a osmolarity of 100 mOsm generated a **M50** of  $45.07 \pm 0.34 \mu\text{m}$  and **M90** was  $76.32 \pm 2.49 \mu\text{m}$ . Hypertonic formulations with a osmolarity of 600 mOsm generated a **M50** of  $47.4 \pm 2.25 \mu\text{m}$ , and **M90** was  $130.37 \pm 2.25 \mu\text{m}$ . Thus the formulations varying in tonicity (figure 7) did not show any significant difference at **M50** but did show a significant difference at **M90**. Scheffe's multiple comparison test indicated that the droplet size distribution at **M90** obtained from the hypertonic formulation was significantly different from the isotonic and hypotonic formulations. The exact reason for this difference is not known.

The addition of the absorption enhancer, Dimyristoylphosphatidylglycerol (DMPG) to the formulation also showed an increase in the **M50** and **M90** (figure 8). The **M50** for formulations without DMPG was  $45.23 \pm 1.11 \mu\text{m}$ , and **M90** was  $66.79 \pm 6.87 \mu\text{m}$ . However the addition of 0.5% DMPG provided a **M50** of  $71.16 \pm 3.18 \mu\text{m}$ , and a **M90** of  $133.07 \pm 0.79 \mu\text{m}$ . This increase may be due to the fact that the phospholipid forms a dispersion as compared to the solution seen for the previous formulations. Formulations varying in the concentration of absorption enhancer indicated a significant difference at **M50** and at **M90** as shown by the Scheffe's multiple comparison test at  $p < 0.01$ .

Among the device variables, the droplet size distribution and spray profiles for the two different designs of the nasal spray pump are shown in figure 9. The **M50** for buffer formulations sprayed with a ball and spring pump was

FIGURE 7 THE INFLUENCE OF TONICITY ON THE DROPLET SIZE DISTRIBUTION  
USING METERED NASAL SPRAY PUMP

TONICITY						
	ISOTONIC		HYPOTONIC		HYPERTONIC	
N = 12	M50	M90	M50	M90	M50	M90
AVERAGE	45.23	68.41	45.07	76.32	47.4	130.4
S. DEV	1.15	7.94	0.342	2.49	2.25	27.96

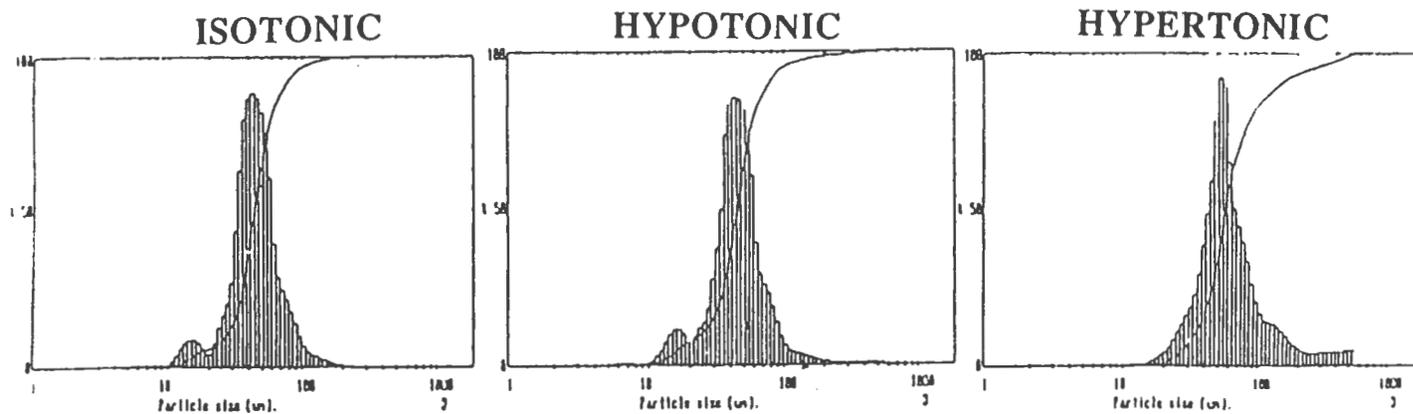


FIGURE 8 THE INFLUENCE OF CONCENTRATION OF ABSORPTION ENHANCER ON THE DROPLET SIZE DISTRIBUTION USING METERED NASAL SPRAY PUMP

CONCENTRATION OF ABSORPTION ENHANCER				
	0% w/w		0.5% w/w	
N = 12	M50	M90	M50	M90
AVERAGE	45.23	66.79	71.16	133.07
S. DEV	1.11	6.87	3.18	0.794

48

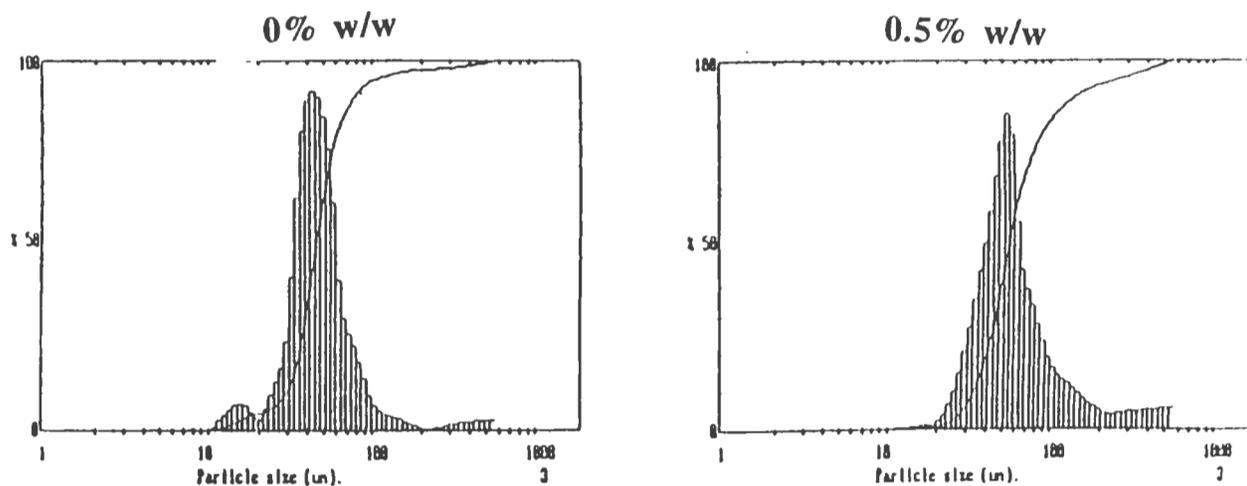
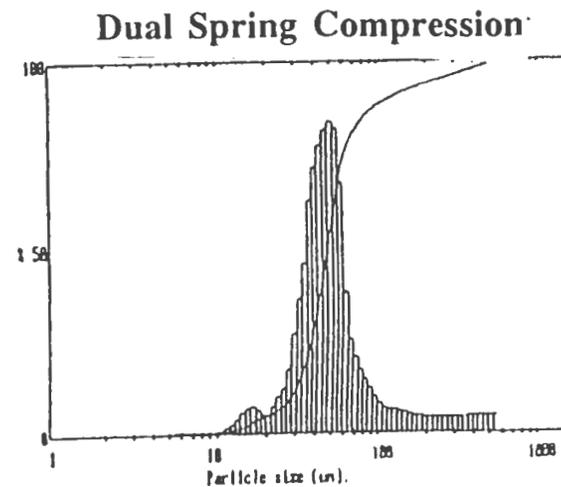
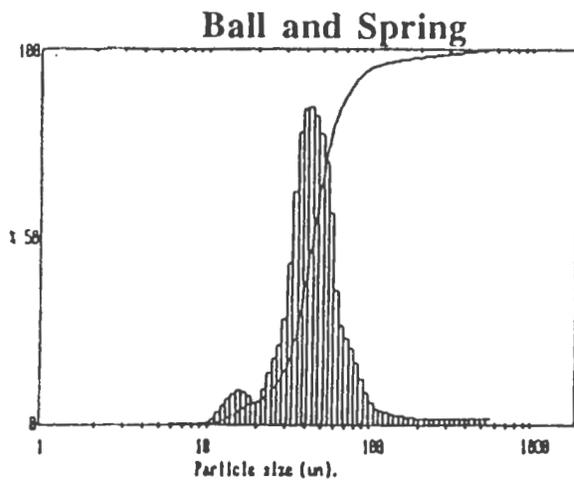


FIGURE 9 THE INFLUENCE OF MECHANISM OF PUMP ON THE DROPLET SIZE DISTRIBUTION FOR METERED NASAL SPRAY PUMP

MECHANISM OF PUMP				
	Pfeiffer		Valois	
	Ball and Spring		Dual Spring Compression	
N = 12	M50	M90	M50	M90
AVERAGE	44.37	76.64	47.42	106.54
S. DEV	0.459	1.01	2.243	35.221

49



44.37±0.459  $\mu\text{m}$ , and the **M90** was 76.64±1.01  $\mu\text{m}$ . The **M50** for buffer formulations sprayed using a dual spring compression pump was 47.42±2.24  $\mu\text{m}$ , and the **M90** was 106.54±35.22  $\mu\text{m}$ . The influence of the orifice diameter of the spray insert is shown in figure 10. The **M50** for buffer formulations sprayed using an actuator with 0.1 mm spray insert was 45.23±1.15  $\mu\text{m}$  and the **M90** was 68.41±7.94  $\mu\text{m}$ . The **M50** for buffer formulations sprayed using a spray insert with a diameter of 0.3 mm was 45.64±0.34  $\mu\text{m}$ , and the **M90** was 77.45±2.81  $\mu\text{m}$ . The **M50** for buffer formulations sprayed with a spray insert with a diameter of 0.7 mm was 47.40±2.25  $\mu\text{m}$  and the **M90** was 123.6±24.86  $\mu\text{m}$ . Similarly the effect of spray angle is shown in figure 11. The **M50** for buffer formulations sprayed from an orifice with a spray angle of 20° was 45.28±1.05  $\mu\text{m}$  and the **M90** was 67.51±7.94  $\mu\text{m}$ . Using spray angle of 35° the **M50** for buffer formulations sprayed was 45.07±0.34 $\mu\text{m}$ , and the **M90** was 76.56±2.4  $\mu\text{m}$ . The **M50** for buffer formulations sprayed with a spray angle of 60° was 47.57±3.4  $\mu\text{m}$ , and the **M90** was 124.93±40.34  $\mu\text{m}$ . Statistical analysis of the results indicated that there was not any significant difference at 0.01 level; between the **M50** and **M90** for two different design mechanisms for the pumps, on three different spray angles and the **M50** for the three sizes of spray inserts. However the **M90** for the 0.7 mm orifice insert was found to be significantly different from 0.1 mm and 0.3 mm.

The actuator and orifice diameter of the spray insert of metered nasal spray pump that were designed to be able to facilitate breakdown of the higher viscosity formulations from a stream to fine spray did not produce any significant differences in the droplet size distributions. This study emphasized the need to explore the effect of increased pressure on the higher viscosity

FIGURE 11 THE INFLUENCE OF SPRAY ANGLE ON THE DROPLET SIZE DISTRIBUTION  
USING METERED NASAL SPRAY PUMP

ACTUATOR DESIGN						
SPRAY ANGLE						
	20 degrees		35 degrees		60 degrees	
N = 12	M50	M90	M50	M90	M50	M90
AVERAGE	45.28	67.51	45.07	76.56	47.57	124.9
S. DEV.	1.05	7.94	0.342	2.4	3.4	40.34

52

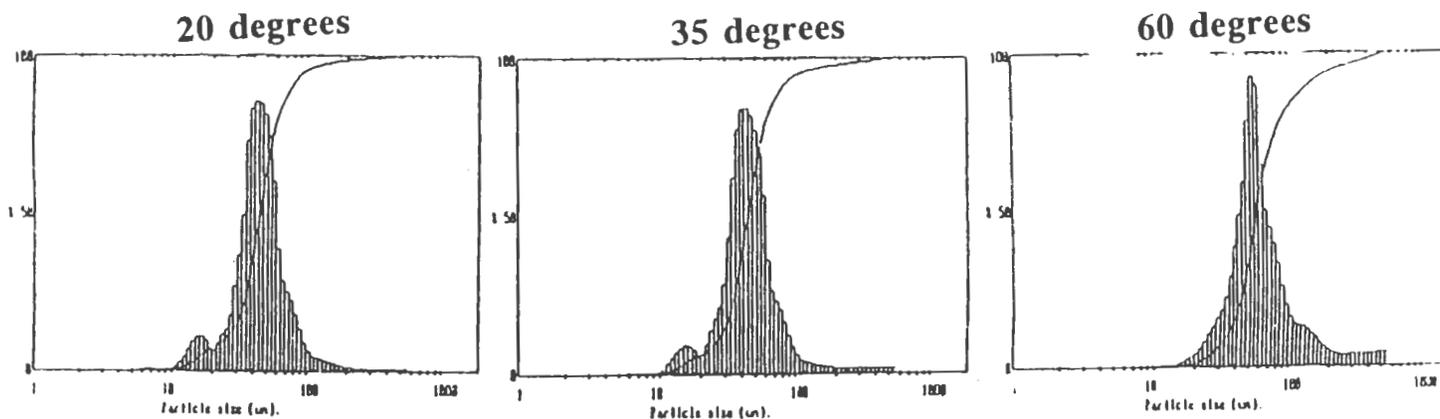


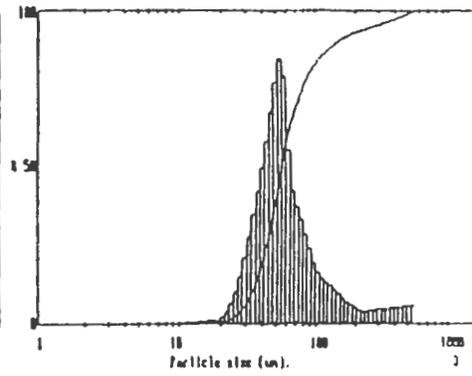
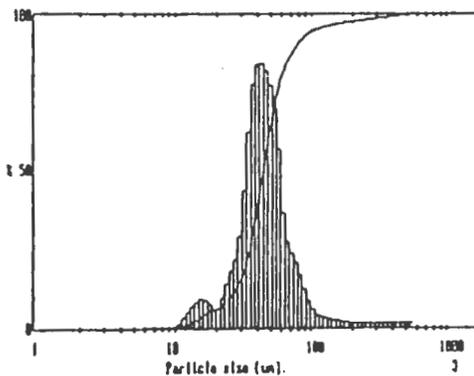
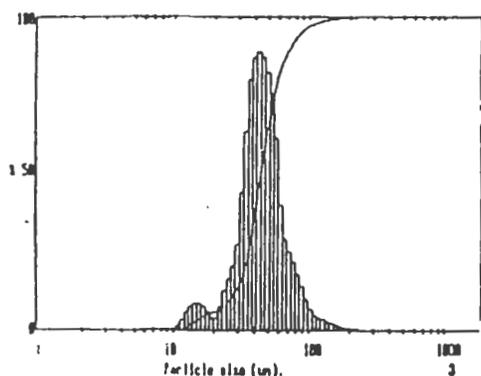
FIGURE 10 THE INFLUENCE OF ORIFICE DIAMETER ON THE DROPLET SIZE DISTRIBUTION FOR METERED NASAL SPRAY PUMP

ACTUATOR DESIGN						
EFFECT OF ORIFICE DIAMETER OF SPRAY INSERT						
	0.1 mm		0.3 mm		0.7 mm	
N = 12	M50	M90	M50	M90	M50	M90
AVERAGE	45.23	68.41	45.64	77.453	47.4	123.6
S. DEV.	1.15	7.94	0.342	2.81	2.25	24.86

0.1 mm

0.3 mm

0.7 mm



formulations. It was decided to evaluate the prototype nasal micron spray pump.

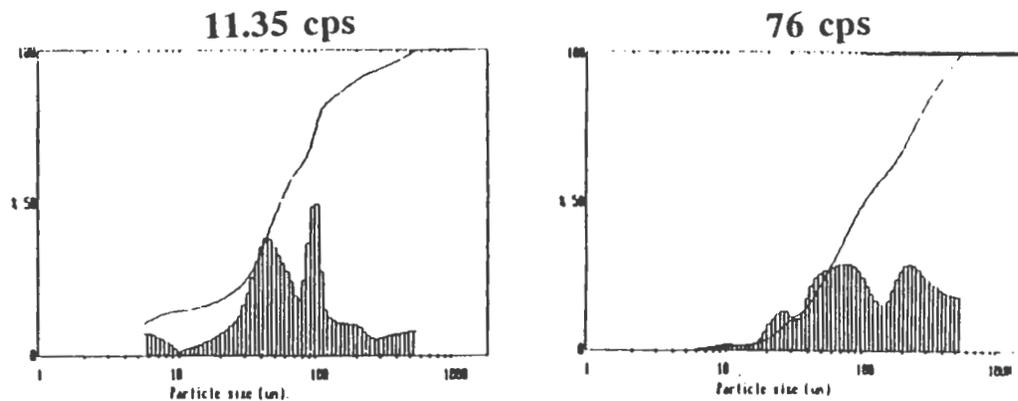
### **Nasal micron spray pump**

The nasal micron spray pump (NMSP) is a prototype device designed to reduce the particle size distribution by using the increased pressure generated by the of air blast nozzle. Since viscosity seemed to be the most influential factor in the determination of droplet size, the formulations varying in viscosity were delivered using this new prototype spray pump and droplet size determined. The NMSP was able generate a **M50** of  $53.27 \pm 3.38 \mu\text{m}$  and a **M90** of  $223.44 \pm 83.44 \mu\text{m}$  for a 1 cps buffer and also to deliver the 11.35 cps (0.5%w/w) and 76 cps (1% w/w) methylcellulose formulations with an average **M50** of  $54.25 \mu\text{m}$  and  $112.45 \mu\text{m}$  respectively (figure 12). These same formulations when delivered with the standard metered spray pump measured **M50** of  $105.86 \mu\text{m}$  for 0.5% w/w methylcellulose of 11.35 cps and a **M90** of  $382.38 \pm 45.82 \mu\text{m}$ . Thus nasal micron spray pump was able to reduce the average droplet size distribution for high viscosity solutions significantly. However the spray profile for the NMSP showed a bimodal curve which seems to be due to a portion of spray striking the mouthpiece upon release from the air blast nozzle. Interestingly the NMSP delivered the 76 cps solution formulation as a fine spray, but could not disperse solutions having higher viscosity than 76 cps as a fine spray. Nevertheless thos prototype spray device seems to have the potential of provideing more "narrowed " spray distribution for high viscosity formulations.

FIGURE 12 THE INFLUENCE OF VISCOSITY ON THE DROPLET SIZE DISTRIBUTION USING NASAL MICRON SPRAY PUMP

VISCOSITY				
	11.35 cps		76 cps	
N = 3	M50	M90	M50	M90
AVERAGE	54.25	257.73	112.48	382.38
S. DEV	4	110.02	27.6	45.82

54



The influence of formulation variables and devices on nasal drug delivery has been focus of many investigations. However there is no published information on how the different components of nasal spray pump can affect the spray characteristics and the droplet size. In this study an attempt was made to determine the influence of nasal administration device variables, such as design mechanism of the pumps, actuator, spray insert diameter as well as the various formulation parameters such as viscosity and presence of absorption enhancer on the droplet size parameters M<sub>50</sub> and M<sub>90</sub>. However the influence of these spray profile on bioavailability is not known and is yet to be elucidated.

#### **Acknowledgments**

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**MANUSCRIPT II**

**THE INFLUENCE OF TONICITY AND VISCOSITY ON THE INTRANASAL  
BIOAVAILABILITY OF SALMON CALCITONIN IN RABBITS.**

## ABSTRACT

As with any drug delivery system, a clear understanding of the physicochemical and formulation factors is necessary for rational design of the dosage form. Formulation factors require careful assessment to identify those which may influence pharmacological or physiological response and thus assure optimum therapeutic activity. On the basis of physicochemical and biopharmaceutical studies of peptide or protein drugs; a nasal formulation administered with an appropriate delivery device may be designed to provide optimal nasal activity. In the present study an attempt was made to investigate the effect of tonicity and viscosity on the intranasal absorption of salmon calcitonin (sCT). Formulations were designed as nasal sprays with a viscosity of 1 or 76 cps, using 0% w/w and 1% w/w methylcellulose as the viscosity enhancing agent; and with a tonicity of 100, 300 or 600 mOsm, using sodium chloride as the tonicity adjusting agent to provide hypotonic, isotonic and hypertonic formulations. Chlorbutanol was used as a preservative. The low viscosity formulations were delivered using a metered nasal spray pump and the high viscosity formulations were administered using a prototype device, the nasal micron spray pump, to facilitate a uniform distribution of the spray into the nasal cavity. Serum levels of sCT were determined in healthy male New Zealand rabbits after intranasal administration of 2000 IU of sCT in 200  $\mu$ l. The pharmacodynamic effect of salmon calcitonin of lowering blood calcium level was determined by a visible spectrophotometric technique. A full factorial design was employed with ANOVA and Scheffe's multiple comparison test used to analyze the data. Deviation from isotonicity was found to increase the bioavailability by 4-5 times. Variation in the viscosity did not influence the bioavailability of

salmon calcitonin. Response surface methodology and the canonical analysis parameters were applied to determine the optimum formulations.

**Key Words:** Calcitonin, Intranasal, Tonicity, Viscosity, Methylcellulose, Nasal spray, Osteoporosis, Spray pumps.

## INTRODUCTION

The recent phenomenal advancement in biotechnology has contrived stimulating opportunities for the use of peptides and proteins as novel therapeutic agents. The emergence of these peptides and proteins has raised questions as to the extent to which existing formulation and drug delivery technologies can be used and the type of new technologies that must be developed to ensure the maximum therapeutic activity. Peptides and proteins have been administered intravenously because of the low oral bioavailability. This is very inconvenient for patients suffering from chronic diseases. Therefore, the administration of peptides and proteins by way of the mucosal routes such as nasal, rectal, buccal, vaginal and ocular mucosae are being investigated<sup>1</sup>.

The use of the intranasal route for systemic delivery of peptide and protein drugs has several potential advantages including the large surface area of the nasal cavity and the highly vascularized nature of nasal mucosa. Nasal absorption of polypeptides such as oxytocin<sup>2</sup>, synthetic lysine vasopressin<sup>3</sup>, synthetic LH-RH and its analogues<sup>4</sup>, insulin<sup>5-7</sup>, enkephalins analogs<sup>8</sup>, growth hormone releasing factor<sup>9-10</sup> has been reported. The permeability of the nasal mucosa to peptides seems to decrease with increasing molecular weight. For peptides with a molecular weight of more than 1000D, bioavailabilities are in the range of 1-3%<sup>11-12</sup>. The nasal absorption efficiency of peptides and proteins is also influenced by the overall design of the dosage form as well as the devices used for administration. Some of the dosage forms which have been investigated are solutions<sup>13-14</sup> gels<sup>15</sup>, powders<sup>16</sup> and microspheres<sup>17</sup>. These dosage forms have been used for the administration of

insulin<sup>18-20</sup>, human growth hormone<sup>21-22</sup>, oxytocin<sup>23</sup>, desmopressin<sup>24</sup> and propranolol<sup>25</sup>. The earliest and the most classical form of nasal formulation is the solution. However solutions have often been shown to provide lower bioavailability because they are more readily cleared from the nasal cavity, particularly when administered as drops. Conversely sprays are confined to the nonciliated regions of the nasal cavity and therefore are retained longer than drops<sup>26</sup>. There is little information on how the formulation variables of a nasal spray such as buffer concentration, buffer species, ionization of buffer species, ionic strength, viscosity, charge, pH of vehicle, osmolarity, type and concentration of preservatives could further influence the absorption of peptide and proteins.

Salmon Calcitonin, a polypeptide hormone secreted by the parafollicular C cells of the thyroid gland, lowers serum calcium levels by decreasing bone and renal tubular reabsorption of calcium. Injectable formulations of salmon calcitonin have been used with some success in the management of metabolic bone disorders such as osteoporosis and Paget's disease but this form of administration is inconvenient and has been poorly tolerated by the patients. The development of an intranasal preparation of salmon calcitonin will provide a more convenient means of administering the drug<sup>27</sup>. Intranasal spray preparations containing salmon calcitonin have been shown to be effective in metabolic bone disease<sup>28</sup>. Unfortunately nasal bioavailability is significantly lower than that realized by injection. Salmon calcitonin in a spray form is now registered in some European countries, however a nasal spray of salmon calcitonin has not been approved in United States by FDA as yet. The available data indicate a reasonable absorption, but there is very little detail addressing the effects of formulation parameters, such as pH, ionization

of buffer salts, osmolarity, viscosity, etc., on the bioavailability<sup>29-30</sup>. To optimize drug activity, the factors influencing the drug, formulation components in combination with delivery devices must be identified and modified to realize significant drug activity.

Since the stability of calcitonin dictates such parameters as pH, buffer and preservatives; this study was designed to investigate the effect of the formulation variables, viscosity and tonicity, on the intranasal bioavailability of salmon calcitonin. Previous studies have been carried out with various drugs to evaluate the effects resulting from an increase in viscosity in a formulation. An increase in viscosity is postulated to increase the particle size of a spray formulation which leads to a more localized deposition and slower clearance from nose. However, the effects of increased viscosity on mucociliary clearance, are difficult to interpret. According to some authors, an increase in viscosity results in a decrease in the clearance rate<sup>31</sup>. In other cases a progressive increase in viscosity leads to an initial slowdown of clearance followed by an increase<sup>32</sup>. Increasing viscosity also seems to progressively lower the rate of diffusion of the drug from the deposited droplets resulting in delayed release and absorption without improvement in bioavailability. While the above data was generated using viscosity in the range of 1-11 cps, the present study will investigate the use of a significantly higher viscosity (~75 cps) formulation obtained by using 1% w/w methylcellulose. Previous studies have shown that metered nasal spray pumps are limited in their ability to spray higher viscosity solutions as a fine spray. It has been shown that it is not possible to achieve a gaussian droplet size distribution for solutions of viscosity greater than 2.12 cps<sup>33</sup>. However high viscosity formulations can be delivered as a fine spray using a prototype device, the

nasal micron spray pump, to facilitate the uniform distribution of spray in the nasal cavity. These results will be compared to those obtained using a low viscosity formulation, 1 cps, delivered from a commercially available metered dose spray pump. Additionally, the study will compare the droplet size distribution obtained with the two different devices, the effect on Tmax, the time required to obtain maximum concentration in the blood, and the bioavailability of salmon calcitonin.

Isotonic solutions are used to assist cells in maintaining a stable physiological environment. Deviation from isotonic conditions can cause tissue damage, irritation, hemolysis of blood cells and electrolyte imbalance. However very little information has been generated on the effect of varying the degree of tonicity for oral or mucosal drug delivery especially intranasal formulations. Absorption of quinine from the small intestine was found to decrease with increasing hypertonicity<sup>34</sup>. Malone et al, 1960, reported that the oral absorption of phenobarbitone in the rat was higher from hypotonic than from hypertonic solutions<sup>35</sup>. However Ohwaki et al, 1984<sup>36</sup>, demonstrated an increase in the bioavailability of secretin from the nasal cavity with hypertonicity. Based on the information obtained from these few experiments, it is difficult to interpret the effect of tonicity on the absorption in the nasal cavity. The literature does not cite any reference of the effect of tonicity on the bioavailability of salmon calcitonin. This study investigates the effect of tonicity at three levels, isotonicity (300 mOsm), hypotonicity (100 mOsm) and hypertonicity (600 mOsm). These formulations were obtained by using sodium chloride as tonicity adjusting agent. Since the formulation is meant for a chronic use, mild deviations from isotonic conditions was investigated as severe deviation can effect the integrity of the nasal mucosa.

## **EXPERIMENTAL**

### **Materials**

Salmon calcitonin (sp. activity 5384 I.U./mg) was a gift from Armour Pharmaceuticals, Kankanee, IL). RIA Kits were procured from (Diagnostics Systems laboratories, Webster, TX), Acetic acid and sodium acetate, analytical grade were purchased from (Fisher Scientific, Fair Lawn, NJ). The anesthetics, acepromazine maleate, ketamine hydrochloride (Aveco Co. Inc., Fort Dodge, IA), DMA Calcium Plus Reagent® (DMA, Houston, TX), Methylcellulose, Methocel A15C (Dow Chemical Company, Midland, MI), Chlorbutanol (Sigma Chemical Company, St. Louis, MO), Catheters and Serum separators (Baxter Health Care Corporation, Deerfield, IL), Heparin Sodium injection, USP (Elkins-Sinn Inc., Cherry Hill, NJ) were purchased and used as received. Distilled water was used in all nasal formulations.

A 50 µl metered nasal spray pump was chosen to deliver the low viscosity formulations (1 cps). A prototype device, the nasal micron spray pump (NMSP), capable of spraying 35 µl per actuation was employed to deliver formulations of higher viscosity (76 cps). Both the pumps were provided by Pfeiffer GMBH, Radolfzell, Germany.

### **Methods**

#### **Preparations of formulations**

The formulations were prepared with a concentration of methylcellulose ranging from 0% to 1% w/w in a 0.03 M sodium acetate-acetic acid buffer at pH 4 and kept in the refrigerator. A pH of 4 was selected for the formulation as salmon calcitonin has maximum stability between pH 3-5. The formulation

was refrigerated as salmon calcitonin has been reported to be most stable when stored at 2-4°C.

#### **Measurement of the osmotic pressure (tonicity)**

The osmotic pressure of the formulations was measured using a A-010 Microdiagnostics Osmometer. Standards were measured ranging from 100 mOsm-2000 mOsm and then formulations were measured in duplicate.

#### **Administration of nasal formulations to the rabbits**

It has been shown<sup>37</sup> that differences in volume, concentration, or dosing technique for an intranasal drug can provide a significant difference in bioavailability. Instilling a drug as a smaller volume in both nostrils as compared to a larger volume in one nostril demonstrated a lower coefficient of variation and slower clearance from the nose.

It has been demonstrated<sup>38</sup> that the addition of a viscosity enhancing agent increases the contact time of the dosage form to the mucosal membrane and may help to enhance the absorption by reducing the clearance of the administered drug. Viscosity also changes the pattern of deposition and clearance<sup>38</sup>. In these studies the viscosity used was in the range of 1-3.5 cps, which can be considered as low viscosity solutions. The NMSP, a prototype device originally developed for inhalation solution was investigated for its ability to deliver higher viscosity solutions as a fine spray. The lower droplet size distribution for the high viscosity solutions is achieved by the NMSP using an air blast nozzle as an actuator. By improving the geometry and therefore the dynamic properties within the spray nozzle, the droplet spectrum has been reduced to smaller diameters leaving the spray nozzle<sup>39</sup>.

A standard metered manual spray pump was chosen to deliver low viscosity formulations (1 cps). A metered 50 µl nasal spray pump based on a ball and spring mechanism was combined with a flanged actuator using a spray insert with a 35° spray angle was used. The total volume of 200 µl to contain the dose was instilled by spraying twice into each nostril. The NMSP was employed to deliver formulations of higher viscosity (76 cps). The volume delivered was 35 µl per actuation and a 210 µl volume to contain the dose was instilled by spraying 3 times into each nostril.

### **In vivo study**

#### **Animal Model**

The New Zealand white rabbit was selected as an animal model because it provides a well controlled animal model for screening the nasal absorption potential of sprayed drug formulations. The blood volume of the rabbit is sufficiently large to permit frequent blood samples and allow a full characterization of the absorption profile of the drug<sup>40</sup>. Rabbits were obtained from Milbrook farms (Amherst, MA) with a mean weight of about 3 kg. The study protocol was reviewed and approved by the Institutional Animal Care and Use Committee, University of Rhode Island, Kingston, RI

#### **Experimental procedure**

The rabbits were fasted for 36 hours prior to each experiment with free access to water. The rabbits were anesthetized with an i.m. dose of 0.25 mg/kg of ketamine hydrochloride and 2.5 mg/kg of acepromazine maleate. Rabbits were kept lying on their backs on thermal rugs during each experiment. A catheter was placed in the rabbit's median ear artery and a 0.8-1 ml blood sample was collected at -5, 10, 20, 30, 40, 50, 60, 75, 90, 120, 180, 240 and 300

minutes after administration of the formulation. The samples were allowed to clot for at least 30 minutes at room temperature and then centrifuged at 3000 rpm for 10 minutes and stored at -20°C until assayed.

The control group was used to measure any interference from the trauma of administration of anesthesia on the inherent calcitonin level. The control rabbits were subjected to the same conditions, however calcitonin was not administered. The values obtained were subtracted from the levels obtained after administration of calcitonin formulations.

### **Study Design**

Calcitonin was administered as spray to the rabbits as per a randomized cross-over design. A wash out period of at least one week was allowed between treatments.

### **Analytical methods**

#### **Serum calcitonin**

All serum samples were assayed with a commercial RIA kit developed at Diagnostics Systems Laboratories (Webster, TX). Guinea pig anti-sCT antibodies were used as primary antibodies and have demonstrated less than 2% w/w cross reactivity. The sigmoidal standard curve of B/Bo vs [sCT], typical of competitive binding assays was linearized using the log transformation. The best fit line was determined using a non weighted least square regression analysis. The radioimmunoassay quantitation range was 100-5000 pg/ml. The square of the coefficient of determination ( $r^2$ ) in this range for the B/Bo versus log[sCT] plots was greater than 0.993 for all assays performed in duplicate.

### Serum calcium levels

Serum calcium levels were determined using a DMA Calcium Plus Procedure® utilizing arsenazo III to bind the calcium at an acid pH and form a bluish purple complex. Serum calcium was quantified spectrophotometrically at 650 nm and the range of the linearity was determined to be between 5-25 mg/dl. The square of the correlation ( $r^2$ ) in this range for the absorbance versus concentration (mg/dl) was 0.999 for all the assays performed in duplicate during the course of study.

### Data Analysis

The concentration of calcitonin was measured in the serum over a period of 5 hours after administration of salmon calcitonin formulations. Salmon calcitonin activity was determined as a function of lowering of calcium levels. The nasal bioavailability of calcitonin was calculated relative to the serum calcitonin levels over a period of 5 hours after injecting calcitonin intravenously.  $C_{max}$  values are the peak serum calcitonin concentrations observed at time  $T_{max}$  after administration of calcitonin.

$$\% \text{Absolute Bioavailability} = \frac{AUC_{in} - AUC_{control}}{AUC_{iv} - AUC_{control}} \times \frac{Dose_{iv}}{Dose_{in}} \times 100$$

The hypocalcemic effect was measured in terms of maximal decrease ( $\%max_d$ ) which was calculated as the highest percentage of reduction in calcium levels as compared to the basal values. The total decrease in serum calcium level (D%) was calculated using a modified method, by Hirai et al, 1981<sup>41</sup>.

$$D\% = \frac{AUC_{control} - AUC_{in}}{AUC_{control}} \times 100$$

where  $AUC_{in}$ ,  $AUC_{control}$  and  $AUC_{iv}$  refer to the area under the curve calculated by the linear trapezoidal rule.  $AUC_{in}$  refers to area under the curve calculated after the intranasal administration of calcitonin formulations.  $AUC_{iv}$  refers to area under the curve calculated after the intravenous administration of calcitonin formulations.  $AUC_{control}$  refers to area under the curve calculated without administration of the calcitonin. The time at which this reduction takes place is  $T_{max}$ .

The data was analyzed by ANOVA using Statistical Analytical Software (SAS). The formulation effects were compared with the controls using Dunnett's test. Multiple comparison among the treatment effects were determined using Fisher's Least Significant Difference test (LSD) and/or Scheffe's multiple comparison test. Differences among the treatments were assumed to be significant for values of  $p < 0.05$ .

#### **Determination of optimum formulations**

One of the main objectives of this investigation was to study the effects of tonicity and viscosity on the intranasal bioavailability. Tonicity was investigated at 3 levels of 100, 300 and 600 mOsm and viscosity was studied at two levels of 1 and 76 cps. In order to optimize the effects of the independent variables of viscosity and tonicity on the dependent factor, area under the curve, a full factorial statistical design, was used. A factorial design was used to study the primary effects of viscosity and tonicity and the secondary effects or interactions of viscosity and tonicity on the intranasal absorption of sCT

In the simplest of cases, techniques used for optimization involve the determination of maximum or minima but when the relationship for the

response is given as a function of two or more than two independent variables, the system becomes more complicated. In the present case, Response surface methodology (RSM) was applied to estimate the maximum or minimum response so as to determine the optimum formulations for the nasal delivery of salmon calcitonin. The SAS/RSREG software was used to fit the parameters of the complete quadratic response surface obtained from the three levels of tonicity incorporated in the formulations. The fitted surface was analyzed to determine the level of the independent variable providing the optimum response. The predicted optimal value can be found from the estimated surface, which usually shows a minima, maxima or saddle point, if there is no clear maxima or minima. Since the independent variable should be at three or more levels, tonicity was designated an independent variable with area under the curve as a response surface. Viscosity was used as a covariate that was investigated at two levels.<sup>42</sup>

## **RESULTS AND DISCUSSION**

### **(A) Intravenous administration**

The average serum sCT concentration versus time profile for 5 I.U. of i.v. administered dose is shown in figure 1. The average  $C_{\max}$  was  $25 \pm 2.9$  ng/ml at a  $T_{\max}$  of  $14 \pm 2.4$  minutes and the area under the curve (AUC) was  $1102.1 \pm 19.8$  ng.min/ml as compared to control group in which average concentration was 2 ng/ml and the AUC was  $401 \pm 75.3$  ng.min/ml. The salmon calcitonin levels obtained in the serum were significantly different from the control level at  $p < 0.05$ . The pharmacokinetic parameters are summarized in Table I.

FIGURE 1 CONCENTRATION TIME PROFILE OF SALMON CALCITONIN AFTER IV ADMINISTRATION (N=5); DOSE 5 IU

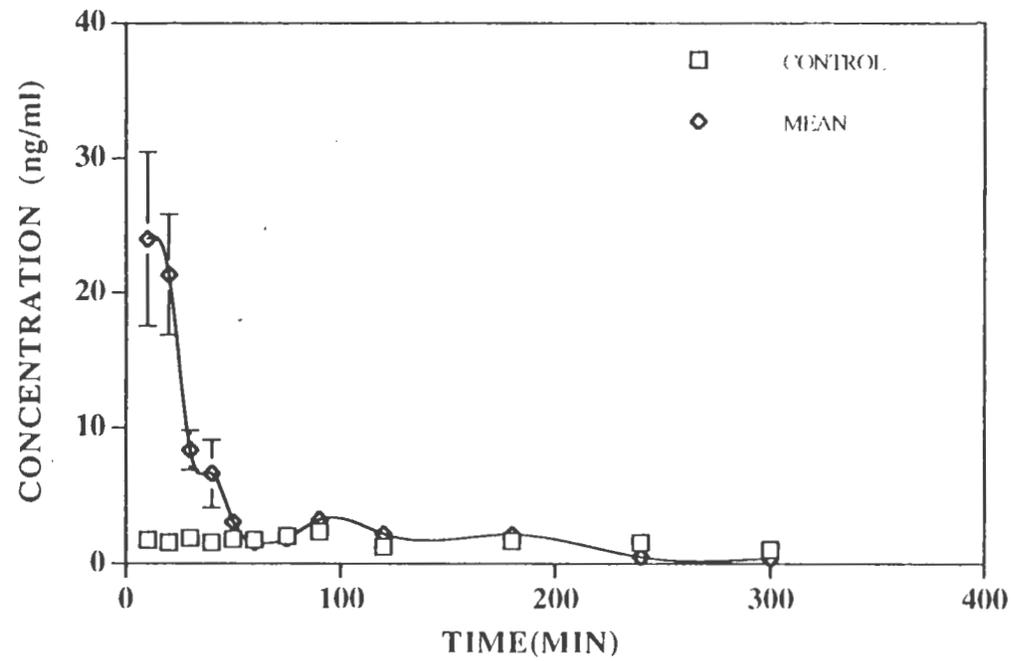


TABLE I PHARMACOKINETIC PARAMETERS AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN TO NEW ZEALAND RABBITS (N = 5)

VARIABLE	Tmax (min)	Cmax(ng/ml)	AUC(ng.min/ml)	%Bioavailability
CONTROL			401 ± 75.3	
I.V (5 I.U.)	14	25 ± 2.9	1102.1 ± 19.8 <sup>a</sup>	100
INTRANASAL LOW VISCOSITY DOSE 2000 I.U.				
ISOTONIC	40	10 ± 4.2	714.2 ± 15.7	0.16
HYPERTONIC	90	42 ± 19.6	3508 ± 1280 <sup>b</sup>	0.80
HYPOTONIC	38	26 ± 7.6	3171 ± 258 <sup>b</sup>	0.71
INTRANASAL HIGH VISCOSITY DOSE 2000 I.U.				
ISOTONIC	58	12 ± 9.7	604.6 ± 97.74	0.14
HYPERTONIC	90	32 ± 19.6	2183 ± 258 <sup>b</sup>	0.62
HYPOTONIC	120	23 ± 2.4	3579 ± 1030 <sup>b</sup>	0.81

a significantly different from control at  $p < 0.05$

b significantly different at low and high viscosity at  $p < 0.5$

The pharmacodynamic effect of sCT of lowering calcium is shown in figure 2 and summarized in Table II. The total reduction in calcium level is termed as D%. The hypocalcemic effect is also measured in terms of maximal percent decrease ( $\%max_d$ ), which is the highest percent of reduction in calcium levels as compared to the basal values. The time at which this reduction takes place is  $T_{max}$ . However between D% and  $\%max_d$ ; the D% is considered to be a more acceptable measure of hypocalcemic activity as it encompasses the overall reduction in activity rather than reduction at one time point.

It can be seen that the administration of the 5 IU iv dose resulted in a initial percent maximal decrease ( $\%max_d$ ) of 28% of calcium at 30 minutes ( $T_{max}$ ). After 30 minutes there was an increase in the calcium level, and then there was a gradual decrease after 75 minutes until 240 minutes. This pattern of concentration change may be explained on the basis that often administration of salmon calcitonin, there is a decrease in the calcium level, once the calcium level reaches a critical value, the parathyroid hormone already present in the body counteracts the effects of calcitonin and raises the calcium level. Thus the total lowering, D%, was calculated and was found to be 32% as compared to the control group. The AUC for i.v. administration was  $2732 \pm 271.39$  mg.min/dL as compared to the AUC for the control group of  $4045.40 \pm 255.17$  mg.min/dL and was significantly different at  $p < 0.05$ .

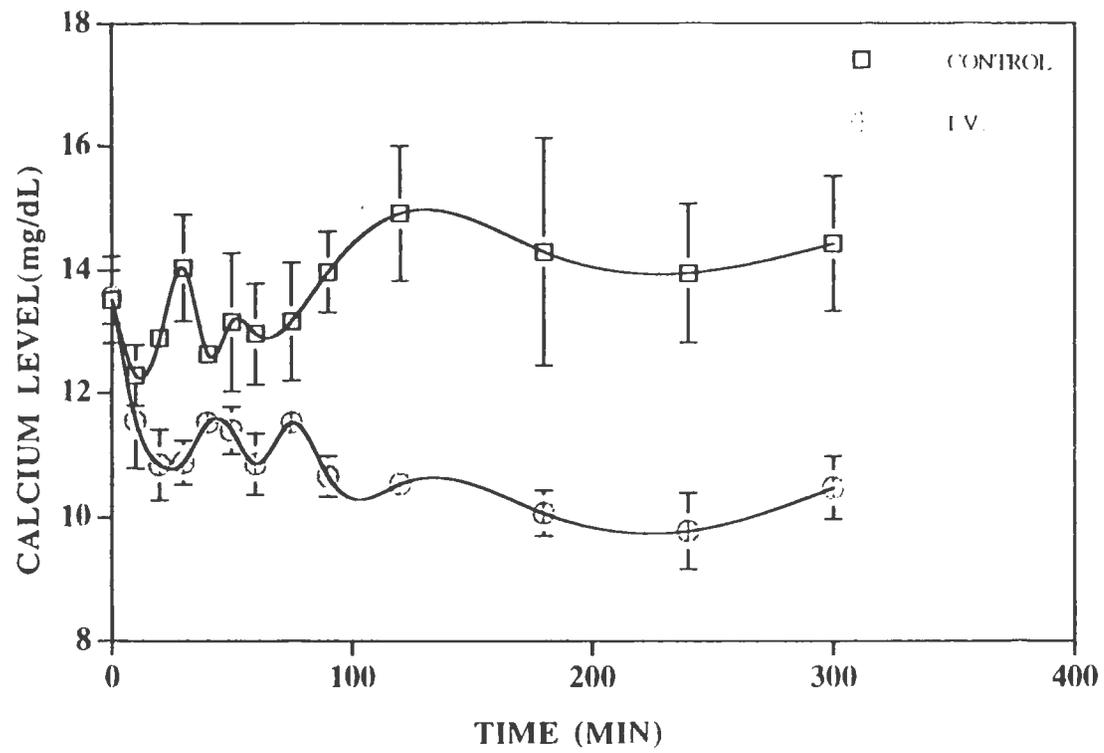
TABLE II PHARMACODYNAMIC PARAMETERS AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN TO NEW ZEALAND RABBITS (N =5); DOSE 2000 IU

<u>VARIABLE</u>	<u>Tmax(min)</u>	<u>Maxd (%)</u>	<u>AUC (mg.min/dL)</u>	<u>% D</u>
CONTROL			4045.40 ± 255.17	
I.V (5 IU)	30	28	2732 ± 271.39 <sup>a</sup>	32
<u>LOW VISCOSITY</u>				
ISOTONIC	240	12.96	3724.12 ± 120.99	7.9
HYPERTONIC	75	28.83	3267.74 ± 31.31 <sup>a, b</sup>	19
HYPOTONIC	180	28.12	3308.0 ± 42.45 <sup>a</sup>	18
<u>HIGH VISCOSITY</u>				
ISOTONIC	240	28.81	3242.83 ± 21.66	19
HYPERTONIC	90	23.52	3696.28 ± 53.24 <sup>a, b</sup>	9.0
HYPOTONIC	120	34.58	3154.47 ± 102.49 <sup>a</sup>	22

a = significantly different from control at p<0.05

b = significantly different at low and high viscosity p< 0.05

FIGURE 2 THE INFLUENCE OF IV ADMINISTRATION OF SALMON CALCITONIN ON THE CALCIUM LEVEL (N=5)



## **(B) Intranasal administration**

### **Effect of tonicity with varying viscosity**

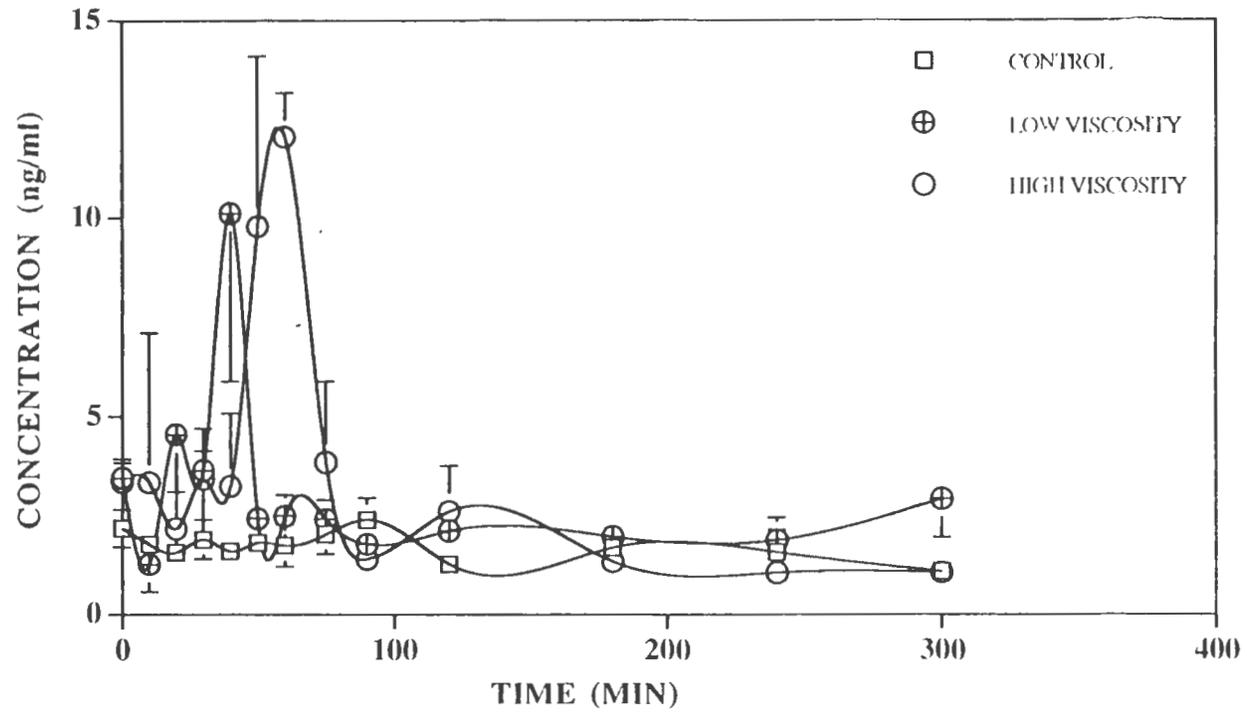
#### **Influence of isotonicity**

To determine the effect of tonicity on the absorption of salmon calcitonin, the base formulations containing a dose of 2000 IU salmon calcitonin in a isotonic 0.03 M acetate buffer at a pH 4 with a viscosity of 1 cps were administered to the rabbits. The AUC, C<sub>max</sub>, T<sub>max</sub> and bioavailability of all the formulations differing in tonicity are summarized in table I. The AUC of isotonic formulations at a viscosity of 1cps was 714.2±15.7 ng.min/ml with a bioavailability of 0.16. The C<sub>max</sub> obtained by the formulations was 10±4.2 ng/ml at a T<sub>max</sub> of 40 minutes.

In contrast higher viscosity (76 cps) isotonic (300mOsm) formulations were prepared with 1% methylcellulose at pH 4 containing 2000 IU salmon calcitonin in 200 µl sprayed into the nasal cavity of rabbits elicited a C<sub>max</sub> of 12±9.7 ng/ml at a T<sub>max</sub> of 58±8.3 minutes as shown in figure 3. The AUC for the higher viscosity isotonic formulations was 604.6±97.74 ng.min/ml with a bioavailability of 0.14. There was no significant difference in the low and high viscosity formulations at isotonic conditions.

The bioavailability obtained by the intranasal administration of salmon calcitonin for low and high viscosity isotonic formulations was extremely low and is typical of higher molecular weight polypeptides as shown by previous investigators<sup>11</sup>.

FIGURE 3 THE INFLUENCE OF VISCOSITY ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN IN ISOTONIC SOLUTIONS



### **Influence of hypertonicity**

The low viscosity hypertonic (600mOsm) formulations sprayed intranasally showed a  $C_{max}$  of  $42 \pm 19.6$  ng/ml at  $T_{max}$  of  $90 \pm 12$  minutes as shown in figure 4, as compared to a  $C_{max}$  of  $10 \pm 4.2$  ng/ml at a  $T_{max}$  of 40 minutes for low viscosity isotonic (300mOsm) formulations. Thus deviation from isotonicity elicited a higher concentration of sCT in the serum at a low viscosity of 1 cps. The AUC calculated for low viscosity hypertonic salmon calcitonin formulations was  $3508 \pm 1280$  ng.min/ml with a bioavailability of 0.80. The bioavailability obtained was about five times higher than that obtained for low viscosity isotonic formulations.

The higher viscosity (76 cps) hypertonic (600mOsm) formulations prepared with 1% methylcellulose and sprayed into the nasal cavity of rabbits elicited a  $C_{max}$  of  $32 \pm 19.6$  ng/ml at  $T_{max}$  at  $90 \pm 12$  minutes as shown in figure 4. The area under the curve was  $2183 \pm 258$  ng.min/ml with a bioavailability of 0.62. Thus deviation from isotonicity at a higher viscosity also demonstrated an increased level of salmon calcitonin in the serum. The bioavailability obtained by high viscosity hypertonic salmon calcitonin formulations was significantly different from high viscosity isotonic formulations as determined at  $p < 0.05$

### **Influence of hypotonicity**

The low viscosity (1cps) hypotonic (100mOsm) formulations in an acetate buffer at pH 4 when sprayed into the nasal cavity of rabbits produced a  $C_{max}$  of  $26 \pm 7.6$  ng/ml at  $T_{max}$  of  $38 \pm 9.16$  minutes as shown in figure 5. The AUC obtained was  $3171.33 \pm 258$  ng.min/ml with a bioavailability of 0.71. The AUC obtained for low viscosity hypotonic formulations was almost of the same

FIGURE 4 THE INFLUENCE OF VISCOSITY ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN IN HYPERTONIC SOLUTIONS

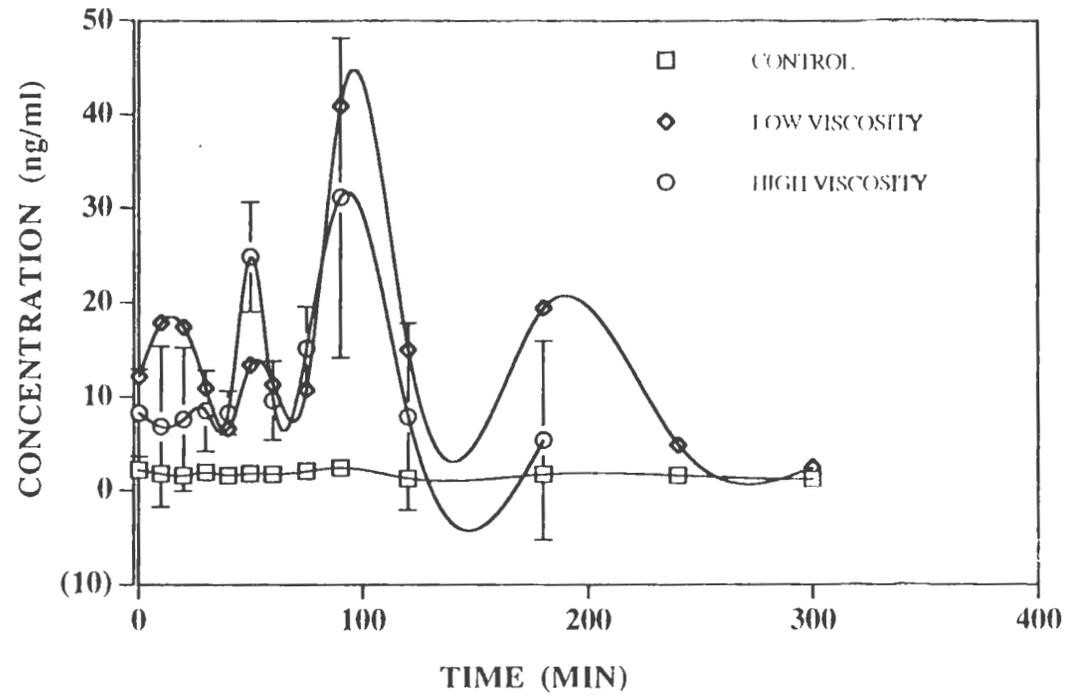
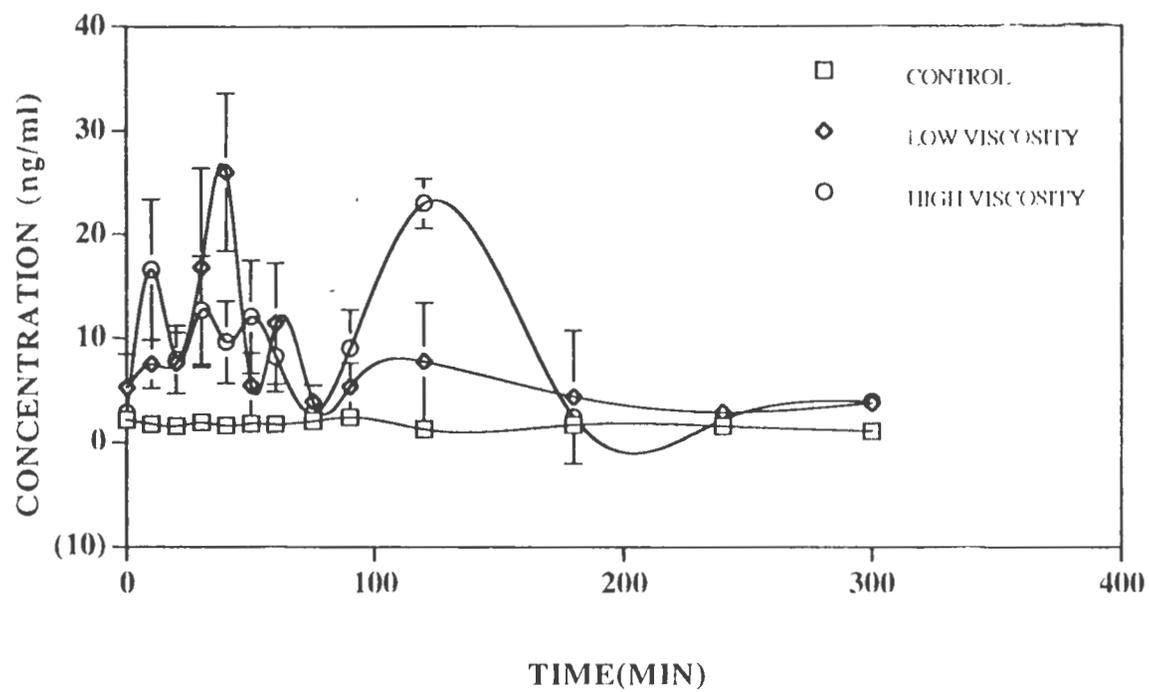


FIGURE 5 THE INFLUENCE OF VISCOSITY ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN FOR HYPOTONIC SOLUTIONS



magnitude as low viscosity hypertonic formulations and was significantly different from low viscosity isotonic formulations at  $p < 0.05$ .

The higher viscosity hypotonic (100mOsm) formulations prepared with 1% methylcellulose at a pH of 4 when sprayed into the nasal cavity of rabbits produced a  $C_{max}$  of  $23 \pm 2.4$  ng/ml at  $T_{max}$  of  $120 \pm 18$  minutes respectively. The AUC calculated was  $3579 \pm 1030$  ng/ml with a bioavailability of 0.81. Even at high viscosity, a deviation from isotonicity produced an increase in the bioavailability.

Thus, it can be seen that absorption of hypertonic and hypotonic formulations was significantly different from isotonic formulations at both low and high viscosity and bioavailability of the drug was enhanced in both the hypotonic and hypertonic conditions by approximately 4-5 fold.

There have been very few experiments performed to study the effect of tonicity, but the information generated from these studies indicates that deviation from the tonicity influences the absorption and this could be explained by the fact that the hypertonic solutions cause shrinkage of the cells and hypotonic solutions cause swelling of the epithelial cells as a result of increased water uptake which could alter the pore size within the cell junctions and the permeability of the cell walls thus leading to enhanced absorption<sup>43</sup>.

#### **Influence of viscosity**

It can be seen from the literature that low and high viscosity (1-11 cps) formulations when administered using a metered nasal spray pump that the

ability of this pump to produce a fine uniform spray is limited for high viscosity formulations. Therefore, in this study a metered dose spray pump based on a ball and spring mechanism was used to spray low viscosity (1cps) formulations and a nasal micron spray pump (NMSP) designed with a air blast spray nozzle was used to spray higher viscosity (76 cps) formulations. The droplet size distribution generated by both the devices had been previously measured using a Malvern sizer and categorized as **M<sub>50</sub>** which is the droplet size of 50% of the droplets sprayed<sup>33</sup>.

The average **M<sub>50</sub>** droplet size for formulations with a viscosity of 1 cps was  $45 \pm 1.11 \mu\text{m}$  and the droplet size **M<sub>50</sub>** of formulations of viscosity, 76 cps was  $112.48 \pm 27.6 \mu\text{m}$  when sprayed using a metered dose spray pump<sup>33</sup>. The **M<sub>50</sub>** generated by the NMSP was  $53.27 \pm 3.38 \mu\text{m}$  for the 1 cps and  $112.45 \pm 27.6 \mu\text{m}$  for the 76 cps formulations. The **M<sub>50</sub>** generated by two nasal delivery devices for the formulations differing in viscosity by 76 cps was approximately different by 2.5 fold ( $45 \mu\text{m}$  vs  $112.45 \mu\text{m}$ ). Thus nasal micron spray pump was very successful in lowering the average droplet size as well as reducing the range of the droplet size distribution. Interestingly, the distribution produced by metered nasal spray pump for 1 cps viscosity was Gaussian and unimodal while that produced by NMSP deviated slightly from Gaussian behavior and was bimodal<sup>33</sup>.

The **T<sub>max</sub>** produced by the low viscosity isotonic and hypertonic formulations was 40 and 38 minutes respectively as compared to that for the higher viscosity isotonic and hypertonic formulations of 58 and 120 minutes which agrees with literature that postulates that a higher droplet size leads to a more localized deposition and causes a slower clearance from the nose<sup>32</sup>.

However for the hypertonic formulations, the T<sub>max</sub> for both low viscosity and high viscosity formulations was 90 minutes. A possible explanation may be that the T<sub>max</sub> is not only a measure of higher viscosity but also is affected by the other excipients of the formulation, especially the tonicity adjusting agent, sodium chloride which could change the time required to achieve the maximum concentration by altering the physiological and pharmacological response generated by the formulation in the surrounding environment.

There was a decrease in the absorption of sCT from higher viscosity isotonic and hypertonic formulations. Similar results have been reported by Harris, 1989<sup>32</sup> who used methylcellulose as a viscosity enhancing agent and demonstrated no improvement in the bioavailability of the drug and actually reported a decrease in absorption possibly due to the delayed diffusion of drug from the formulation because of the higher viscosity. Variation in the viscosity did not show any significant difference in the absorption of sCT.

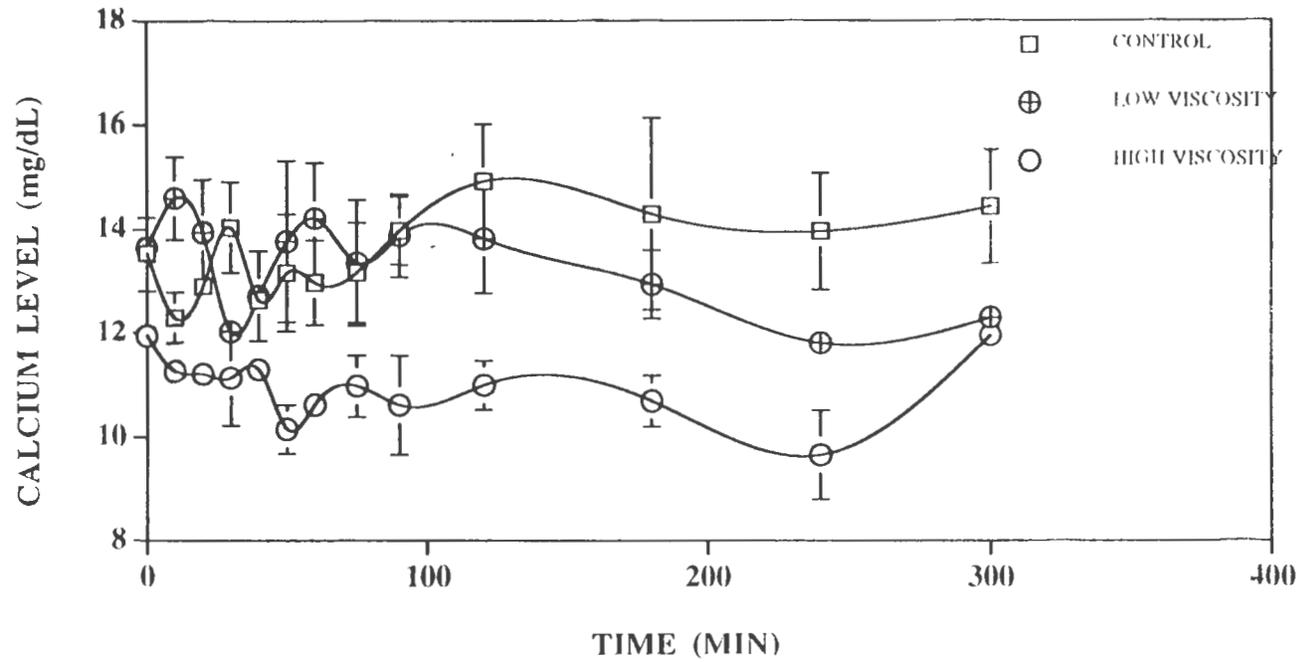
## **Pharmacodynamic effects**

### **Effect of tonicity with varying viscosity**

#### **Influence of Isotonicity**

The pharmacodynamic effect of sCT of the lowering of calcium for all the formulations varying in tonicity as a function of viscosity were calculated and are summarized in table II and shown in figures 6-8. The AUC for isotonic formulations at low viscosity was found to be 3724.12±120.99 mg.min/dL as compared to 4045±255.17 mg.min/dL for the control group and 2732±271.39 mg.min/dL for i.v administration. Isotonic low viscosity formulations

FIGURE 6 THE INFLUENCE OF VISCOSITY ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF ISOTONIC SALMON CALCITONIN SOLUTIONS



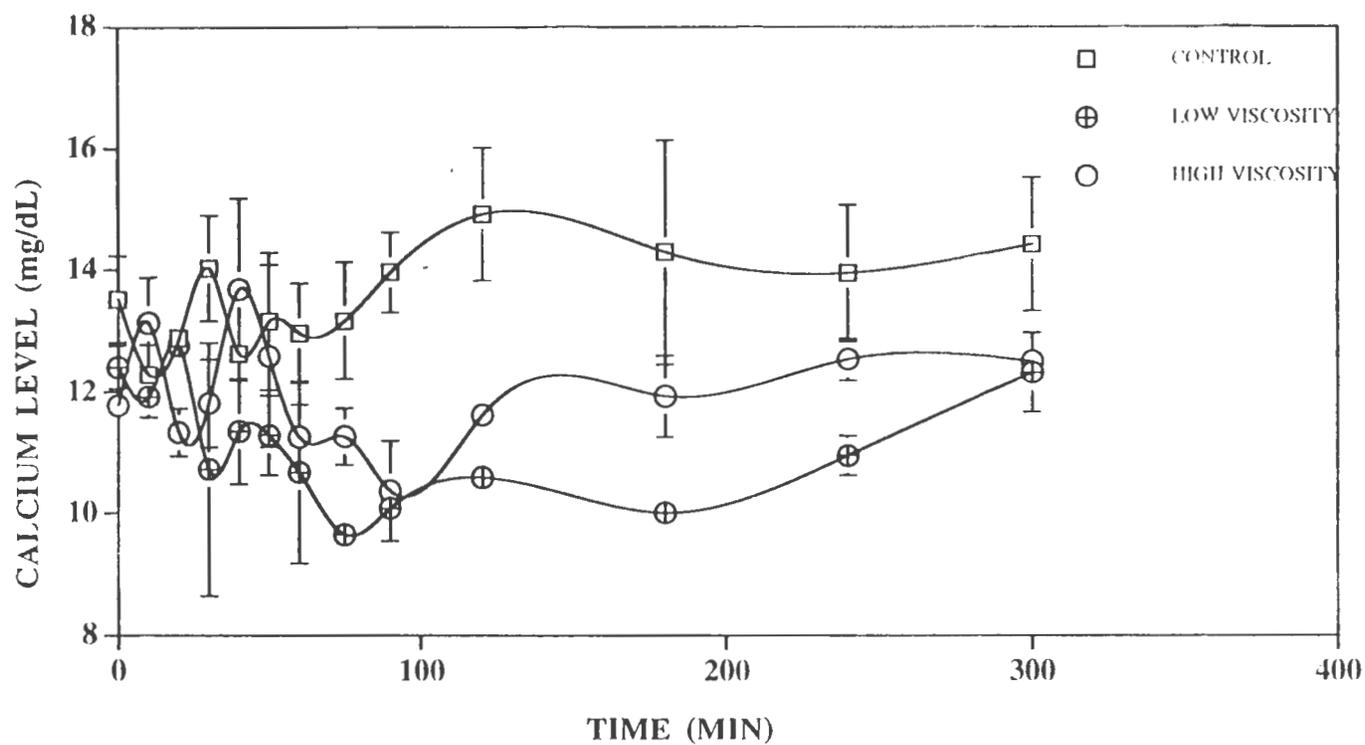
sprayed intranasally were able to produce a reduction in calcium activity (D%) of 7.9% and a maximum reduction, (%max<sub>d</sub>), of 12.96 at a T<sub>max</sub> of 240 minutes. However the low viscosity isotonic formulations were not able to produce a significant difference as compared to the control group at p<0.05.

The AUC for the high viscosity isotonic formulations was 3242.83±21.66 mg.min/dL and produced a total reduction in activity of 19% and a maximum reduction of 28.81 at a T<sub>max</sub> of 240 minutes. There was an initial maximal decrease at 60 minutes and the level of calcium started rising from 60 minutes until 90 minutes and again there was a gradual decrease until 240 minutes. The maximal percent decrease for the low and high viscosity isotonic formulations was obtained at 240 minutes as seen in figure 6. The high viscosity isotonic formulations generated a significant difference as compared to the control and low viscosity isotonic formulations. This may be due to the slower clearance that is postulated for higher viscosity formulations have longer contact with the mucosal membrane. The calcium lowering effect for high viscosity isotonic formulations was significantly different from low viscosity isotonic formulations and control group.

### **Influence of hypertonicity**

The AUC for the low viscosity hypertonic formulations was 3267.74±31.31 mg.min/dL as compared 4045±255.17 mg.min/dL for the control group. These hypertonic formulations at low viscosity when sprayed intranasally were able to produce a total reduction in calcium activity (D%) of 19% and a maximum reduction, %max<sub>d</sub>, of 28.83 at a T<sub>max</sub> of 75 minutes. The total calcium lowering effect of low viscosity hypertonic formulations was almost 2.5 times

FIGURE 7 THE INFLUENCE OF VISCOSITY ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN HYPERTONIC SOLUTIONS



more than low viscosity isotonic formulations and was also significantly different from control group.

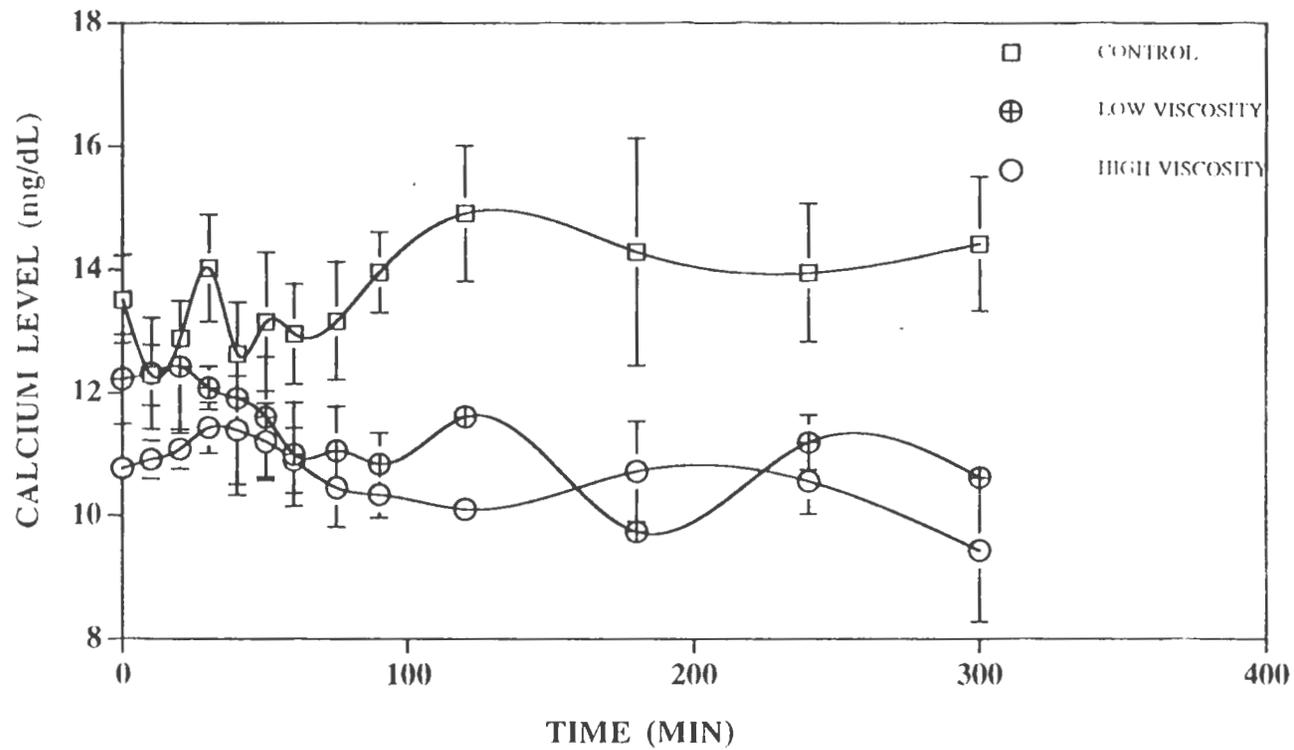
The AUC for high viscosity hypertonic formulations was  $3696.28 \pm 53.24$  mg.min/dL and when sprayed intranasally were able to produce a total reduction activity of 9% and a maximum reduction of 23.52 at a Tmax of 90 minutes. The calcium lowering effect for high viscosity formulations was significantly different from low viscosity formulations and was more comparable to the low viscosity isotonic formulations. The difference in the pattern of low and high viscosity hypertonic formulations on the calcium lowering effect is not known.

#### **Influence of hypotonicity**

The AUC for the low viscosity hypotonic formulations was  $3308 \pm 42.45$  mg.min/dL and produced a total reduction in calcium activity of 18% and a maximum reduction of 28.12% at a Tmax of 180 minutes as seen in figure 8.. The calcium lowering effect produced by low viscosity hypotonic formulations was comparable to that produced by low viscosity hypertonic formulations and was significantly different from controls.

The AUC for high viscosity hypotonic formulations was  $3154.47 \pm 102.49$  mg.min/dL produced a total reduction in activity of 22% and a maximum reduction of 34.58 at a Tmax of 120 minutes. The calcium lowering effect of the high viscosity hypotonic formulations was also similar to low viscosity hypotonic formulations and significantly different from the low viscosity isotonic and control formulations.

FIGURE 8 THE INFLUENCE OF VISCOSITY ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN HYPOTONIC SOLUTIONS



The %max<sub>D</sub> of 28.83% produced by low viscosity hypertonic formulations (600 mOsm) and 28.12% produced by low viscosity hypotonic formulations (100 mOsm) was comparable to %max<sub>D</sub> produced by i.v. administration of 28%. However the total reduction in activity (D%) for the iv formulations were significantly greater than that seen after nasal administration.

The lowering of calcium for all formulations ranged from 7.9 to 22% of the basal value in intranasal formulations as compared to 32% for the iv solutions. The %max<sub>D</sub> intranasal delivery ranged from 12 to 34.5% as compared to 28% in iv administration. The similarity of the extent of the hypocalcemic effect observed after administration of different intranasal formulations of sCT may be attributed to the acute homeostatic mechanism between calcitonin and the parathyroid hormone. The parathyroid hormone counteracts the effects of calcitonin once a critical level of hypocalcemia is attained. Thus the extent of the hypocalcemic effect is controlled by the hormonal balance. The acute control of blood calcium is accomplished by the feedback action of calcitonin and the parathyroid hormone. This complex regulation of blood calcium concentration makes it difficult to evaluate the efficacy of sCT delivery systems by observation of the hypocalcemic effect<sup>44</sup>.

Although the measure of pharmacodynamic effects is useful clinically especially in the event of low bioavailability, the results obtained from the various biochemical markers for salmon calcitonin such as lowering in calcium, bone turnover and decrease in the urinary hydroxyproline are difficult to interpret as these parameters are also under hormonal regulation and may or may not differ for healthy subjects and diseased subjects.

It has been demonstrated that the hypocalcemic effect of intranasal salmon calcitonin is governed by the underlying rate of bone turnover and therefore the effect is minimal in healthy volunteers. That is healthy subjects do not show a significant change in bone turnover while a more pronounced action is observed in subjects with active osteoporosis or Paget's disease<sup>45</sup>.

On the other hand, quantitatively similar reductions in urinary hydroxyproline excretion were demonstrated, indicating similar levels of inhibition of osteoclastic bone resorption following intranasal administration of 200 to 400 IU salmon calcitonin in healthy volunteers aged less than 30 years and elderly patients with osteoporosis or Paget's disease<sup>46</sup>.

Since the hypocalcemic effects of calcitonin are governed by the underlying rate of bone turnover, the pharmacokinetic parameters which relate to the presence of salmon calcitonin in the serum should be a more reliable indicator of the nasal bioavailability of salmon calcitonin

### **Optimization of formulations**

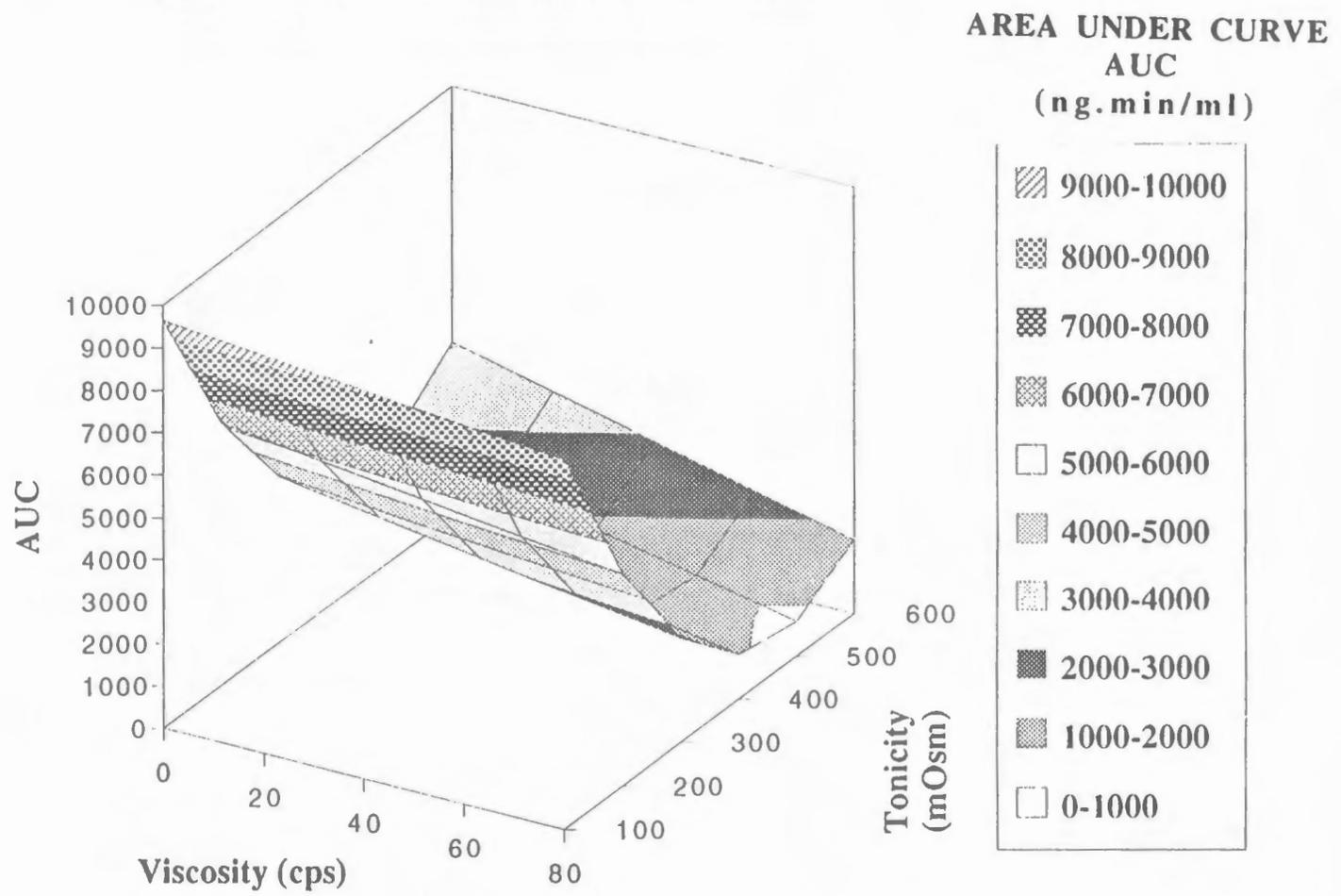
Six formulations resulting from using 3 levels of tonicity and 2 levels of viscosity were administered to rabbits. The area under the curve was calculated and analyzed by ANOVA. The results generated from the experimental data suggested that hypertonic and hypotonic formulations were significantly different from isotonic formulations at both low and high viscosity. The results were confirmed using Scheffe's multiple comparison test. Regression was carried out to define the relationship between area under the curve and the independent factors of tonicity and viscosity; and the optimum region was determined by the response surface analysis.

A saddle point curve (figure 9) was obtained which suggests that there is no maximum or minimum tonicity but that a deviation from the mid point (isotonicity) would favor an increase in the area under the curve and ultimately bioavailability. Viscosity, used as a covariate and plotted along with tonicity and area under the curve demonstrated that there was no influence of viscosity on the area under the curve and subsequently bioavailability. This lack of increasing viscosity effect is shown in figure 9 which indicates a AUC of 8000-10000 ng.min/ml for hypotonic formulations through out the viscosity range from 1 to 80 cps. There is a decrease in the AUC from 8000 to 2000 ng.min/ml as the tonicity increases to 300 mOsm. Again the surface shows a increase as the tonicity increases from 300mOsm to 600 mOsm there is a consistent increase in AUC from 2000-6000 ng.min/ml.

In conclusion it is feasible to enhance the absorption of sCT across the nasal mucosa using proper formulation variables. In spite of low plasma levels and bioavailability, there are clinical data which support the efficacy of intranasal formulations in conditions such as Paget's disease and osteoporosis. Clinical trials have consistently reported an excellent tolerability and acceptability of intranasal formulations<sup>11</sup>.

However further investigations are required for the evaluation of the toxicological effects of tonicity on the integrity of nasal mucosa, the chronic use of these formulations when administered intranasally may have the potential of damaging either the mucociliary system or the underlying epithelium.

**FIGURE 9 THREE DIMENSIONAL SURFACE PLOT OF TONICITY AND VISCOSITY VERSUS AREA UNDER THE CURVE**



In summary, the development and use of salmon calcitonin nasal spray holds a lot of promise for the future treatment and management of metabolic bone disorders.

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MANUSCRIPT III  
THE INFLUENCE OF DIMYRISTOYLPHOSPHATIDYLGLYCEROL ON  
THE NASAL ABSORPTION OF SALMON CALCITONIN

## **ABSTRACT**

The objective of this investigation was to study the effect of the phospholipid, dimyristoylphosphatidylglycerol (DMPG) on the intranasal absorption of polypeptide, salmon calcitonin (sCT). DMPG was included as an absorption enhancer for salmon calcitonin in a 0.03 M acetate buffer at pH 4 in a nasal spray formulation. The absorption of sCT was also studied as a function of NaCl salt concentration from 0.045-0.3 M. Serum calcitonin was determined using a double antibody RIA. The presence of the phospholipid demonstrated a significant difference in the intranasal bioavailability of salmon calcitonin. The results showed DMPG enhanced nasal absorption of sCT by approximately two fold at a salt concentration of 0.045 M. However higher salt concentration in the formulations demonstrated a decrease in the absorption enhancing effect of DMPG. The effect of DMPG on the calcium lowering effect of sCT was also studied. The serum calcium was quantified using arsenazo III complex spectrophotometrically at 650 nm. There was no significant difference in the hypocalcemic activity of sCT in the presence of DMPG. In addition it was found that increasing viscosity of nasal formulations containing DMPG could not further increase the bioavailability of sCT.

**Key Words:** Calcitonin, Dimyristoylphosphatidylglycerol, Absorption enhancer, Bioavailability, Viscosity, Salt concentration, Nasal micron spray pump.

## INTRODUCTION

The mammalian body possesses several mechanisms to restrict the entry of macromolecules including polypeptides and proteins. These include the presence of various epithelia that are poorly absorptive, the presence of significant levels of enzymatic activity at various locations between the point of entry into the systemic circulation and the target site of a peptide or protein, the availability of multiple enzymes to degrade peptides and proteins at a given location, and varying levels of immunoglobulins to neutralize peptides and proteins both before and after they are absorbed. The inevitable outcome is that the bioavailability of peptides and proteins is likely to be much less than that for a small molecule.

To improve and optimize absorption of peptide and proteins, certain approaches and strategies have been used and these include, the use of penetration enhancers to alter membrane permeability, coadministration of inhibitors to restrain the activity of proteolytic enzymes primarily at the absorption site, and the use of analogs that are metabolically stable and which in turn may be more bioavailable.

Among the strategies mentioned, the formulation of macromolecules with enhancer seems to be a realistic and viable alternative and has been a subject of many investigations. Penetration enhancers are compounds, generally of low molecular weight, that facilitate the absorption of solutes across biological membranes. With few exceptions and regardless of the nonparenteral route of administration, penetration enhancers are required for the absorption of peptides and proteins in pharmacologically active quantities.

Most of the penetration enhancers chosen for mucosal delivery have belonged to one of the four major classes. They are: *chelators* such as EDTA, citric acid, salicylates, N-acyl derivatives of collagen, and enamine(N-amino acyl derivatives of  $\beta$ -diketones); *surfactants* such as sodium lauryl sulfate, polyoxyethylene-9-lauryl ether, polyoxyethylene-20-cetyl ether; *bile salts* such as sodium deoxycholate, sodium glycocholate, and sodium taurocholate; and *fatty acids* such as oleic acid and monolein<sup>1</sup>.

Numerous absorption enhancers have been described for the use in the nasal route and their efficacy is very evident, although variable, depending on the peptide considered. Some of these absorption promoters include anionic and cationic surfactants<sup>2</sup>, bile salt-fatty acid mixed micelles<sup>3</sup>, fusidic acid derivatives<sup>4-6</sup>, medium chain fatty acid salts<sup>7</sup>, fatty acid derivatives of carnitines<sup>8</sup>, enamine derivatives of amino acids<sup>9-10</sup>, glycerine esters<sup>11</sup>, salicylates<sup>12-13</sup> and carboxylic acids<sup>14</sup>.

Different mechanisms of action have been proposed such as alteration in the mucus rheology, reduction in the nasal ciliary beat frequency, enhancement of transcellular transport by affecting membrane lipids and proteins, enhancement of paracellular transport, suppression of proteolytic activity or enhancement of the thermodynamic activity of peptides and proteins. Penetration enhancers improve peptide and protein absorption by one or more of the several mechanisms. Bile salts have been shown to reduce the viscosity of the mucus layer adhering to all mucosal surfaces, thereby facilitating peptide and protein diffusion towards the membrane surface<sup>15</sup>. Chelators<sup>16-19</sup> and to certain extent, polyoxyethylated nonionic surfactants<sup>20</sup> interfere with the ability of calcium ions to maintain the dimension of the

intercellular space, thereby permitting the paracellular transport of peptides and proteins which otherwise will be excluded from this pathway. Interestingly, most of the penetration enhancers are capable of increasing membrane fluidity either by creating disorder in the phospholipid domain in the membrane, as is the case of salicylates, oleic acid and monolein<sup>21-23</sup>, or by facilitating the leaching of proteins from the membrane, as in the case for surfactants and EDTA.

The efficiency of penetration enhancers depends on several factors, including the nature of peptides, the physicochemical properties of the delivery system as related to peptide and protein release, lipophilicity of the enhancer, the intrinsic ability of the enhancer to perturb membrane permeability, the site of application of enhancer and the animal species used.

**Nature of peptide:** The intrinsic ability of peptides to cross biological membranes is anticipated to be a complex function of their physicochemical properties. The bioavailability of peptide following nasal administration varies with their primary structure, although the precise relationship between absorption and physicochemical properties is as yet unknown. The coadministration of peptides with 1% sodium glycocholate elicited an increase in the absorption of a poorly absorbable peptide, namely leuprolide, However it had no effect on the absorption of metkephamid, which is relatively well absorbed peptide. Interestingly even in the presence of adjuvants, the bioavailability of leuprolide and insulin was far from complete<sup>24-26</sup>.

**Delivery systems:** The selection of a delivery system for a peptide and its

penetration enhancer must take into account the vast difference in physicochemical properties that may exist between these two substances due to the difference in their release rates from the system. For a penetration enhancer to be effective, it must be released either simultaneously with the peptide or shortly before the peptide itself is released. This requirement is supported by the findings of Nishhata et al<sup>27</sup> in that the bioavailability of insulin following rectal administration in the rat was reduced from 30.2% to 12.8% when the release of DL-phenylalanyl ethylacetoacetate, the adjuvant, from the suppository was delayed from 17.5 to 30 minutes prior to the administration of insulin.

**Nature of enhancer:** A penetration enhancer must be able to penetrate the membrane and at the same time, achieve a high enough concentration to perturb membrane structure<sup>28</sup>. Hirai et al<sup>29</sup> determined that somewhat lipophilic ester and ether type nonionic surfactants with HLB values of 10-14 appeared to optimize the nasal absorption of insulin in the rat when compared with extremely hydrophilic or lipophilic derivatives. The coadministration of monolein and bile salts has been found to bring about a synergistic effect in enhancing the oral absorption of heparin in the rat presumably due to the facilitation of the membrane penetration of the bile salt by monolein<sup>30</sup>. In spite of its attractiveness in optimizing the efficiency of adjuvants, the approach of using co-adjuvants that operate by different mechanisms has not been fully explored.

**Potency of enhancer:** In a series of phenylalanylenamine type enhancers, which act by chelating  $Ca^{2+}$ , the extent of hypoglycemia in rabbits following rectal administration of insulin was most pronounced for the most potent

chelators in that series<sup>31</sup>.

**Site of administration:** Because the permeability characteristics of each mucosal site vary, the extent of absorption of a peptide differs depending on the site of administration<sup>32</sup>. Polyoxyethylene-9-lauryl ether and sodium glycocholate were far more effective in promoting the nasal absorption of leuprolide rather than vaginal absorption<sup>18</sup>. This was not a surprising finding in light of the fact that the nasal epithelium is comprised of columnar cells whereas the vaginal epithelium is comprised of stratified squamous cells<sup>33</sup>.

**Animal model:** The animal model that reliably mimics peptides and protein absorption in humans, both in the presence and absence of penetration enhancers, has yet to be established. Preliminary evidence suggests that, at least in the case of the nasal absorption of insulin, the extent of enhancement may be a function of the animal model chosen<sup>34-35</sup>.

Since absorption enhancers promote peptide and protein absorption by perturbing membrane integrity, it is inevitable that a varying extent of damage would occur to these mucosal tissues which are in intimate contact with an enhancer. To date only few studies have been undertaken to determine the nature of the potentially exaggerated effects or toxicities caused by the various types of penetration enhancers<sup>36-38</sup>. Overall non-surfactant type enhancers when used on an acute basis appear to cause less severe and more readily reversible morphological changes than their surfactants counterparts. Aside from morphological changes, there is virtually no information on the biochemical changes that may occur locally or the

systemic toxicity which may result from the absorption of the penetration enhancers themselves and from the absorption of toxins at the mucosal surfaces that otherwise would be excluded from the systemic circulation. Membrane damaging effects have been demonstrated by protein release<sup>39</sup>, ciliostasis<sup>40</sup>, and histological examinations<sup>41</sup>. Such toxicity may eventually preclude use of these agents in clinical trials.

Another emerging class of absorption enhancers is phospholipids. They occurs in cell membranes and can be classified as acidic or zwitter ions.

Lysophospholipids are surface active zwitter ion compounds generated naturally in biological membranes by action of phospholipases. They are active in low concentrations and are converted within the membrane to normal cell components<sup>42</sup>. L- $\alpha$ -Lysophosphatidylcholine (LPC) has been shown to increase the permeability of rat ileum to macromolecules<sup>43</sup> and has more recently been investigated as an enhancer for the intranasal absorption of insulin in rats<sup>44</sup> and of human growth hormone in rats, rabbits and sheep<sup>45-46</sup> although its histopathological effects on the nasal tissue have not been reported. The mechanisms of absorption promotion for LPC is poorly understood but it possesses "membrane activity" and has shown absorption enhancing properties<sup>45</sup>. Dimyristoylphosphatidylglycerol (DMPG) is an acidic phospholipid which has been shown to interact with certain polypeptides like glucagon, insulin, calcitonin, parathyroid, secretin, and growth hormone to form an amphipathic helix<sup>47</sup>. Circular Dichroism spectroscopy studies show that sCT forms an amphipathic helix and adopts a conformation of higher helical content in the presence of phospholipid DMPG<sup>48</sup>. This hydrophilic lipoprotein complex (DMPG-sCT) has been associated with an increase in the hypocalcemic activity<sup>49-50</sup>. However there have been no literature citation of

describing the use of DMPG as an nasal absorption enhancer for calcitonin.

Calcitonin is a 32 amino acid amidated peptide used for over two decades to treat certain degenerative bone diseases and the hypercalcemia associated with malignancy. Before 1989, calcitonin was administered exclusively by the parenteral route. The bioavailability of an intravenous dose is absolute. However the chronic nature of the ailment and the high cost and discomfort associated with parenteral administration decreases patient compliance. The feasibility of the administration of calcitonin via the nasal route has been documented and the results have been encouraging<sup>51</sup>.

Nasal spray formulations of salmon calcitonin currently are on the market in Europe and Japan and have been extremely well received. Similar approvals are pending in the United States. These products have provided a much needed alternative for patients, however extremely low bioavailability of less than 3% of the free drug when administered nasally necessitates the use of absorption enhancers<sup>52</sup>.

The objective of this investigation was to study (i) the effect of DMPG on the intranasal absorption of sCT in the presence of various NaCl salt concentrations in a range of 0.045-0.3 M (ii) the effect of DMPG on the calcium lowering effect of sCT (iii) whether the addition of a viscosity enhancer to these formulations would further increase the intranasal bioavailability of sCT.

## **EXPERIMENTAL**

### **Materials**

Salmon calcitonin (sp. activity 5384 IU/mg) was obtained from Armour Pharmaceuticals, Kankanee, IL. Dimyristoylphosphatidylglycerol (DMPG) was purchased from Avanti Lipids, Alabaster, AL. RIA Kits were procured from Diagnostics Systems laboratories, Webster, TX. Acetic acid and Sodium acetate, analytical grade were purchased from Fisher Scientific, Fair lawn, NJ. Anesthetics acepromazine maleate, ketamine hydrochloride were purchased from Aveco Co. Inc., Fort Dodge, IA. DMA Calcium Plus Reagent®, DMA, Houston, TX, Methylcellulose, Methocel A15C, Dow Chemical Company, Midland MI. Chlorbutanol, Sigma Chemical Company, St. Louis MO, Catheters and Serum separators, Baxter Health Care Corporation, Deerfield, IL, Heparin Sodium injection, USP, Elkins-Sinn Inc., Cherry Hill, NJ were purchased and used as recieved. Distilled water was used for all nasal formulations.

A metered nasal spray pump was used to spray 50 µl at one time. Nasal micron spray pump was employed to deliver formulations of higher viscosity (76 cps), methylcellulose 1% w/w. The volume delivered was 35 µl at one time. Both the pumps were provided by Pfeiffer GMBH, Radolfzell, Germany.

### **Methods**

#### **Preparation of formulations containing DMPG and sCT at low viscosity**

##### **(1 cps)**

DMPG was added as a solid to a 0.03 M acetate buffer (pH 4) and the mixture was stirred. Calcitonin was added and it was allowed to interact at room temperature for 5-6 hrs. The ratio of DMPG to salmon calcitonin was 50:1, this

particular ratio provided a dose of sCT of 2000 IU and 1% w/w of DMPG. The affinity of sCT for DMPG has been shown to be independent of temperature within experimental error, between 25 and 38° C and the reaction is very spontaneous.

#### **Preparation of formulation containing DMPG and sCT at high viscosity (76 cps)**

It has also been demonstrated<sup>53</sup> that the addition of a viscosity enhancing agent increases the contact time of the dosage form with the mucosal membrane and may help to enhance absorption by reducing the mucociliary clearance of the administered drug.

To prepare formulations containing DMPG and sCT at a higher viscosity of 76 cps, methylcellulose was used as a viscosity enhancing agent at a concentration of 1% w/w. The formulations were prepared using a hot-cold technique. The methylcellulose was dispersed in hot water and then ice cold water was added and stirred for 4-5 hrs for methylcellulose to swell and hydrate and reach the specified viscosity. The pH was adjusted to 4. DMPG and calcitonin were added. The mixture was allowed to interact at room temperature for 5-6 hrs.

#### **Administration of nasal formulations to the rabbits**

It has been shown<sup>54</sup> that by adjusting the volume and the concentration of intranasal drug, a significant difference in bioavailability can be obtained. The bioavailability of desmopressin from 2X50 µl dose was 20% greater than after administration of a 1X100 µl dose. Instilling the drug as a smaller volume in both nostrils as compared to larger volume in one nostril demonstrated

slower clearance from the nose.

However the metered nasal sprays are limited in their ability to spray only low viscosity solutions (< 5 cps) as fine spray. The nasal micron spray pump, a prototype device, designed to deliver solutions of higher viscosity as a fine spray was investigated for its use in nasal drug delivery<sup>55</sup>.

A metered nasal spray pump based on the mechanism of ball and spring with a flanged actuator using a spray insert of 35° spray angle was used to spray 50 µl of low viscosity solutions (1 cps) at one time, and the total volume of the dose instilled was 200 µl by spraying 2 times into each nostril. The nasal micron spray pump was employed to deliver formulations of higher viscosity (76 cps). The volume delivered was 35 µl at one time, and a total of 210 µl dose was instilled by spraying 3 times into each nostril.

### **In vivo study**

#### **Animal Model**

New Zealand white rabbits were obtained from Millbrook farms (Amherst, MA) with a mean weight of about 3 kg. The study protocol was reviewed and approved by the Institutional Animal Care and Use Committee at the University of Rhode Island, Kingston, RI

#### **Experimental procedure**

The rabbits were fasted for 36 hours prior to each experiment with free access to water. The rabbits were anesthetized with an i.m. dose of 0.25 mg/kg of ketamine hydrochloride and 2.5 mg/kg of acepromazine maleate. Rabbits were kept lying on their backs on thermal rugs during each experiment. A

catheter was placed in the rabbit's median ear artery and a 0.8-1 ml of blood samples was collected at -5, 10, 20, 30, 40, 50, 60, 75, 90, 120, 180, 240 and 300 minutes after administration of the formulations. The samples were allowed to clot for at least 30 minutes at room temperature and then centrifuged at 3000 rpm for 10 minutes and stored at -20° C until assayed.

For control formulations, rabbits were anesthetized and blood was withdrawn at the same intervals as mentioned above and serum analyzed. This was called control group as no calcitonin was administered and was used to measure any interference from the trauma on administration of anesthesia on the inherent calcitonin level. The values obtained were subtracted from the levels obtained after administration of calcitonin formulations.

### **Study Design**

The various formulations of calcitonin were administered as spray to the rabbits as per a randomized cross-over design. A wash out period of at least one week was allowed between treatments.

### **Analytical methods**

#### **Serum calcitonin**

All serum samples were assayed with a commercial RIA kit developed by Diagnostics Systems Laboratories (Webster, TX). Guinea pig anti-sCT antibodies were used as primary antibodies and have demonstrated less than 2% w/w cross reactivity. The sigmoidal standard curve of B/Bo versus [sCT], typical of competitive binding assays was linearized using a log transformation of sCT. The best fit line was determined using a non weighted least square regression analysis. The radioimmunoassay quantitation range was 100-5000 pg/ml. The

coefficient of determination ( $r^2$ ) in this range for the B/Bo versus log [sCT] plots was greater than 0.993 for all assays performed in duplicate.

### **Serum calcium levels**

Serum calcium levels were determined using a DMA Calcium Plus Procedure® utilizing arsenazo III to bind the calcium at an acid pH and form a bluish purple complex. Color intensity was measured spectrophotometrically at 650 nm. The range of the linearity was determined to be between 5-25 mg/dl. The coefficient of determination ( $r^2$ ) in this range for the absorbance versus concentration (mg/dL) was 0.999 for all the assays performed in duplicate during the course of study.

### **Data Analysis**

The nasal bioavailability of calcitonin was calculated relative to the serum calcitonin levels after injecting calcitonin intravenously.  $C_{max}$  values are the peak serum calcitonin concentrations observed at time  $T_{max}$  after administration of calcitonin.

$$\% \text{Absolute Bioavailability} = \frac{AUC_{in} - AUC_{control}}{AUC_{iv} - AUC_{control}} \times \frac{Dose_{iv}}{Dose_{in}} \times 100$$

where  $AUC_{in}$ ,  $AUC_{iv}$  and  $AUC_{control}$  refer to the area under the curve calculated by the linear trapezoidal rule.  $AUC_{in}$  refers to area under the curve calculated after the intranasal administration of calcitonin formulations.  $AUC_{iv}$  refers to area under the curve calculated after the intravenous administration of calcitonin formulations.  $AUC_{control}$  refers to area under the curve calculated without administration of calcitonin.

The hypocalcemic effect was measured in terms of percent maximal decrease (%max<sub>d</sub>); which is defined as the highest percentage of reduction in calcium levels as compared to the basal values. The total decrease in serum calcium level (D%) was calculated by a modified method, from Hirai et al, 1981.

$$D\% = \frac{AUC_{control}-AUC_{in}}{AUC_{control}} \times 100$$

where  $AUC_{in}$ ,  $AUC_{control}$  and  $AUC_{iv}$  refer to the area under the curve calculated by the linear trapezoidal rule after the intranasal administration of calcitonin formulations. The time at which this reduction takes place is  $T_{max}$ . Statistical analysis was performed using ANOVA by Statistical Analytical Software (SAS). Formulation effects were compared with the controls using Dunnett's test. Multiple comparison among the treatment effects were determined using Scheffe's multiple comparison test. Differences among the treatment were assumed to be significant for values of  $p < 0.05$ .

### **Selection of best formulations**

The primary aim of this investigation was to study the effect of DMPG on the intranasal bioavailability of sCT. The influence of DMPG was investigated at the three levels of salt concentration of 0.045 M, 0.15 M and 0.3 M. In order to optimize the effect of these salt concentrations and the concentration of DMPG on area under the curve, response surface analysis was applied. The SAS/RSREG was used to fit the parameters of the complete quadratic response surface obtained at the three levels of salt concentration incorporated in the formulations and analyze the fitted surface to determine the level of the salt concentration needed for the optimum formulations. The predicted optimal value can be found from the estimated surface, which usually shows a minima, maxima or saddle point, Since the independent variable should be

at three or more levels, salt concentration was designated as independent variable with area under the curve as a response surface and the concentration of DMPG was used as a covariate investigated at two levels.<sup>56</sup>

## RESULTS AND DISCUSSION

Currently, two classes of nasally delivered therapeutic compounds are on the market. The first and largest class encompasses low molecular weight drugs designed to treat local inflammation of the nasal mucosa and sinuses. This class includes topical steroids, cromolyn sodium, and most over the counter nasal products. These compounds are generally hydrophobic in their unionized form, and readily partition into the mucosal membrane. The second class includes those drugs intended for systemic delivery after intranasal administration. Despite the attractiveness of the nasal cavity for systemic administration, the second class includes only a few compounds: vasopressin, oxytocin, LH-RH analogs, calcitonin, and vitamin B-12. Except for vitamin B-12, the compounds are all peptides and show very low bioavailability. Although the bioavailability of these peptides is low, therapeutic activity may often be sufficient because of the high biological potency of these compounds.

This study was designed to evaluate the phospholipid, DMPG, as an absorption enhancer and to determine its ability to enhance the bioavailability of salmon calcitonin. Phospholipids may be directly involved in the binding of many drugs and hormones to specific cell-surface receptors sites<sup>47-48</sup>. There is evidence that binding to lipid is required for the activity of peptides including enkephalin<sup>57</sup> and thyrotropin<sup>58</sup>. Calcitonin is a membrane active peptide and its interaction with the lipid may be related to

its biological activity. Circular Dichroism spectroscopy studies show that salmon calcitonin (sCT) forms an amphipathic helix and adopts a conformation of higher helical content in the presence of the phospholipid, Dimyristoylphosphatidylglycerol (DMPG). This hydrophilic lipoprotein complex has been associated with an increase in the hypocalcemic activity<sup>48-50</sup>.

### **Pharmacokinetic studies**

Since salts are often used in nasal formulations to adjust ionic strength, tonicity etc, it was decided to investigate the effect of salt concentration on the absorption of sCT with or/without DMPG. Table I summarizes the T<sub>max</sub>, C<sub>max</sub>, AUC and Bioavailability for the formulations investigated. In order to determine the absolute bioavailability of intranasal sCT formulations, the AUC for the control formulation and i.v solution was also determined and was calculated to be 401±75.3 and 1102.1±19.8 ng.min/ml.

### **Low viscosity formulations**

#### **Effect of 0.045 M NaCl**

The influence of DMPG on the absorption of sCT in a formulation containing 0.045 M NaCl at a viscosity of 1 cps is shown in figure 1. The C<sub>max</sub> obtained for the formulation containing only sCT was 26±7.6 ng/ml at 38 minutes whereas the C<sub>max</sub> elicited by the formulation containing only DMPG and sCT increased to 82±30 ng/ml at 90 minutes. The AUC elicited by sCT formulation at 0.045 M NaCl was 3171±258 ng.min/ml and an absolute bioavailability of 0.71 whereas the AUC elicited by formulation incorporating DMPG and sCT at 0.045 M NaCl was 6017±706 ng.min/ml with an absolute bioavailability of 1.4. The presence of DMPG and sCT in the formulation demonstrated a two fold

**Figure 1 THE INFLUENCE OF A SALT CONCENTRATION OF 0.045 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN**

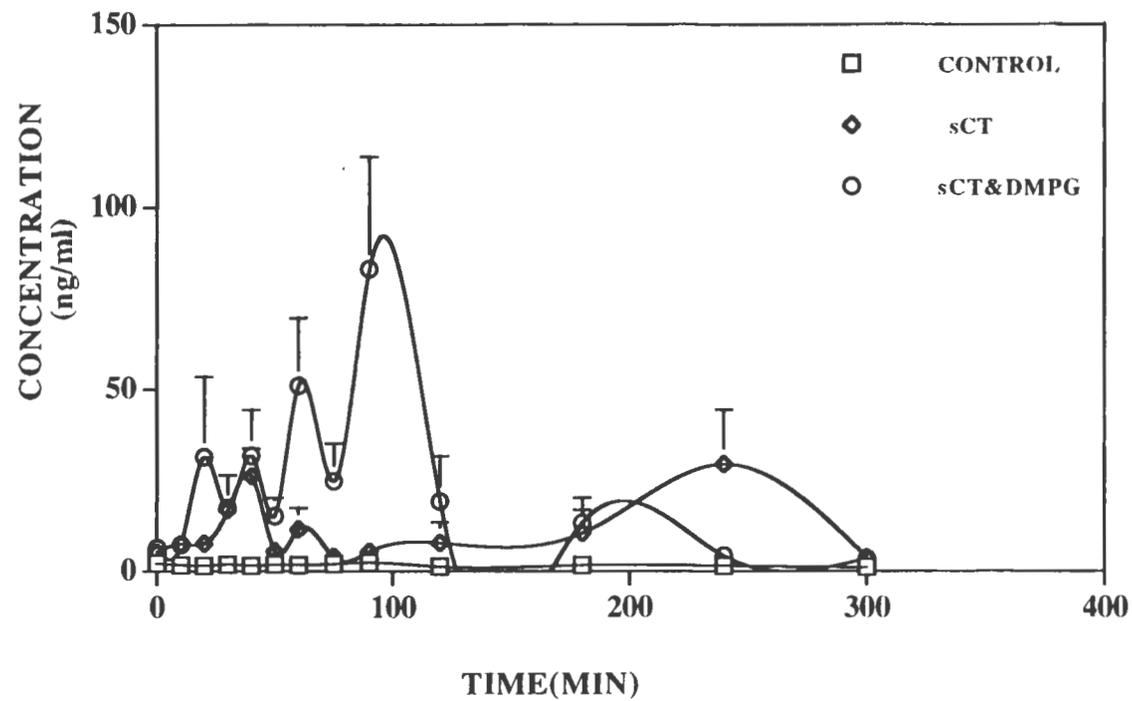


TABLE I PHARMACOKINETIC PARAMETERS OF LOW VISCOSITY SALMON CALCITONIN FORMULATIONS AFTER INTRANASAL ADMINISTRATION IN NEW ZEALAND RABBITS (N = 5)

VARIABLE	T <sub>max</sub> (min)	C <sub>max</sub> (ng/ml)	AUC(ng.min/ml)	%Bioavailability
CONTROL			401 ± 75.3	
I.V (5 I.U.)	14	25 ± 2.9	1102.1 ± 19.8 <sup>a</sup>	100
INTRANASAL sCT DOSE 2000 I.U.				
0.045 M	38	26 ± 7.6	3171 ± 258 <sup>b</sup>	0.71
0.15 M	40	10 ± 4.2	714.2 ± 15.7 <sup>c</sup>	0.16
0.3 M	90	42 ± 19.6	3508 ± 1280	0.80
INTRANASAL DMPG-sCT DOSE 2000 I.U.				
0.045 M	90	82 ± 30	6017 ± 706 <sup>b</sup>	1.4
0.15 M	10	22 ± 30	1521 ± 275 <sup>c</sup>	0.34
0.3 M	63	28 ± 7.5	3692 ± 1369	0.84

a significantly different from control at p<0.05

b significantly different in AUC for DMPG formulations at 0.045 M NaCl at p<0.05

c significantly different in AUC for DMPG formulations at 0.15 M NaCl at p<0.05

increase in the absorption of sCT as compared to formulation containing only sCT at a NaCl concentration of 0.045 M and a viscosity of 1 cps.

#### **Effect of 0.15 M NaCl**

The influence of DMPG on the absorption of sCT from formulations prepared with 0.15 M NaCl at a viscosity of 1 cps is shown in figure 2. The C<sub>max</sub> was found to be 10± 4.2 ng/ml at 40 minutes for the formulations containing only sCT whereas the C<sub>max</sub> increased to 22 ng/ml at 10 minutes for the formulations incorporating DMPG and sCT. The AUC elicited was 714.2 ng.min/ml and an absolute bioavailability of 0.16 for the sCT formulation as compared to an AUC of 1521 ng.min/ml and an absolute bioavailability of 0.34 for the formulations containing DMPG and sCT. These results demonstrated that even at a NaCl salt concentration of 0.15 M and a viscosity of 1 cps, the bioavailability elicited by the presence of DMPG in sCT formulations increased two fold compared to sCT formulations without DMPG. However the bioavailability elicited by sCT with a 0.045 M NaCl salt concentration and a viscosity of 1 cps was higher as compared to formulations containing sCT at a 0.15 M NaCl salt concentration and a viscosity of 1 cps (1.4% versus 0.34%).

#### **Effect of 0.3 M NaCl**

The influence of increasing the NaCl salt concentration to 0.3 M with a viscosity of 1 cps in the absence or presence of DMPG on the bioavailability of sCT is shown in figure 3. The C<sub>max</sub> was 42±19.6 ng/ml at 90 minutes for the sCT formulations. The C<sub>max</sub> was reduced to 28±7.5 ng/ml at 63 minutes in the presence of DMPG in the sCT formulations. The AUC for the sCT administration was 3508±1280 ng.min/ml with an absolute bioavailability of

Figure 2 THE INFLUENCE OF A SALT CONCENTRATION OF 0.15 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN

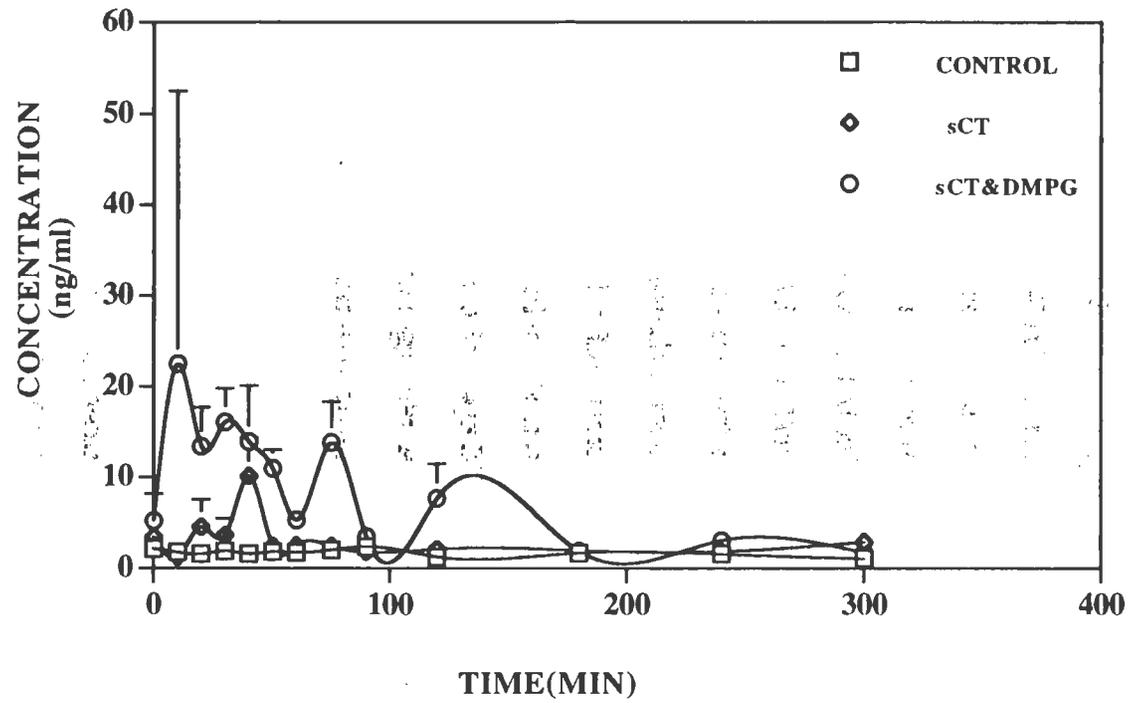
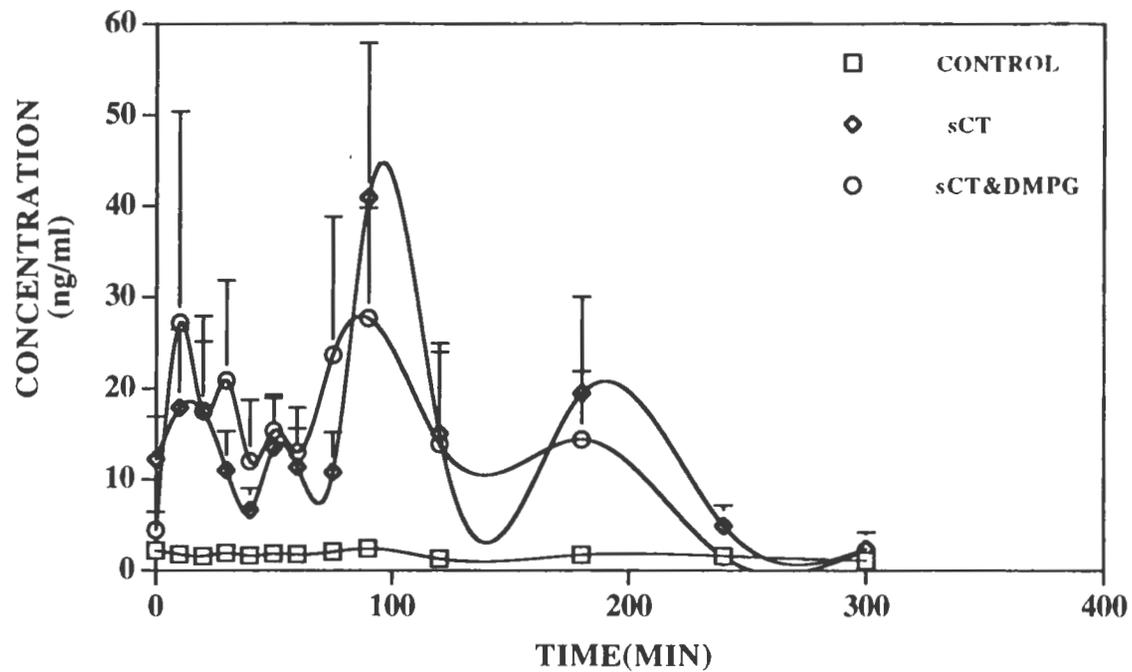


Figure 3 THE INFLUENCE OF A SALT CONCENTRATION OF 0.3 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN



0.80 as compared to an AUC of  $3692 \pm 1369$  ng.min/ml and an absolute bioavailability of 0.84 in the presence of DMPG in sCT formulations. Therefore in the presence of higher salt concentration of 0.3 M NaCl, the absorption of sCT was not affected by the presence of DMPG, in fact it was approximately the same magnitude as seen without addition of DMPG. This indicated that DMPG did not have any absorption enhancing effect on sCT absorption at 0.3 M. Therefore maximum bioavailability elicited by DMPG and the sCT formulations was obtained at the lowest concentration of NaCl.

### **Pharmacodynamic effects**

It has been suggested that the presence of DMPG in an sCT formulation enhances the helical activity of sCT and may be related to the increase in the hypocalcemic activity<sup>48-50</sup>. Thus the parameters of hypocalcemic activity were determined for formulations containing only sCT and those containing DMPG along with sCT. The results are summarized in Table II as T<sub>max</sub> (min), %max<sub>d</sub> (percent maximal decrease), AUC (mg.min/dL) and D% (total hypocalcemic activity).

### **Low viscosity formulations**

#### **Effect of 0.045 M NaCl**

The influence of DMPG on the calcium level after intranasal administration of sCT in the presence of 0.045 M NaCl at a viscosity of 1 cps is shown in figure 4. The maximal decrease, %max<sub>d</sub> was 28.12% at 180 minutes for sCT formulations as compared to %max<sub>d</sub> of 34.58% at 120 minutes for formulations incorporating DMPG and sCT. The AUC was  $3308 \pm 42.45$  mg.min/dL and a total lowering of calcium, D% of 18% for sCT formulations as compared to a AUC of  $3209.55 \pm 21.66$  mg.min/dL with D% of 19% in the

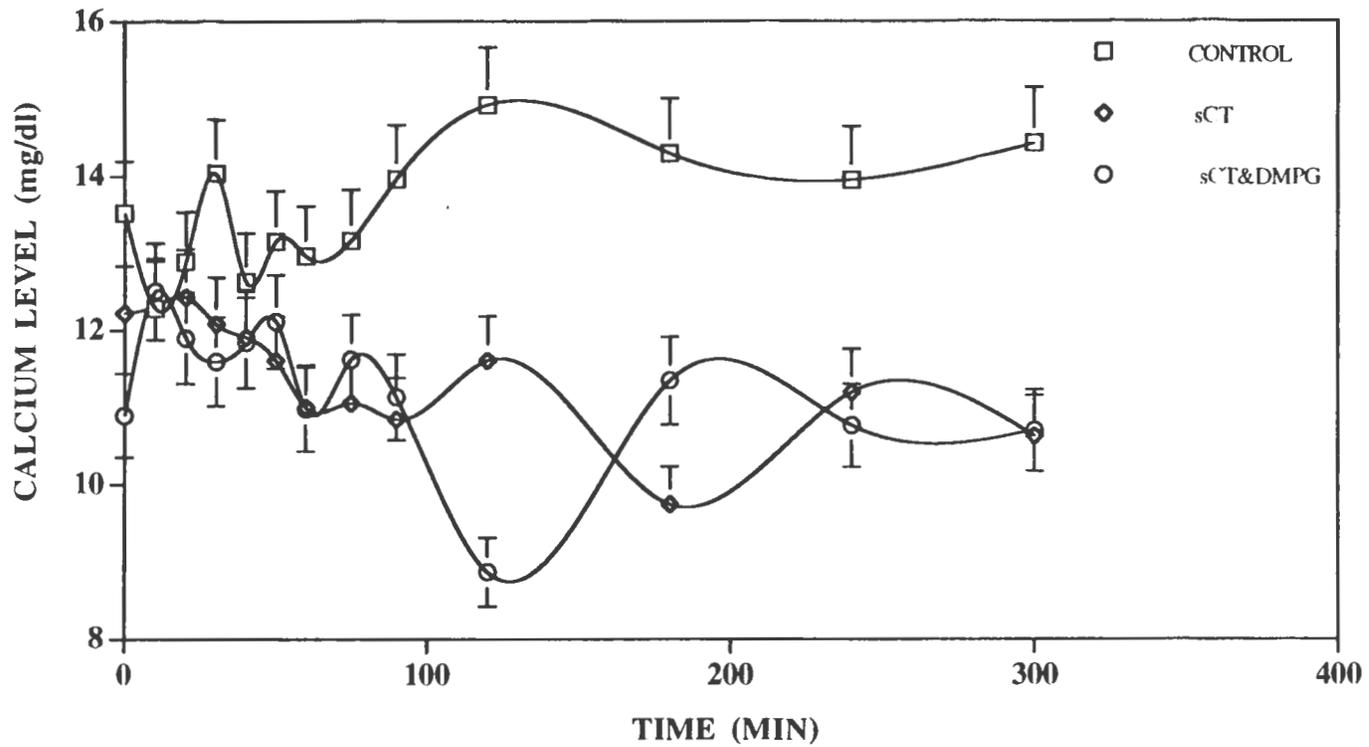
TABLE II PHARMACODYNAMIC PARAMETERS OF LOW VISCOSITY SALMON CALCITONIN FORMULATIONS AFTER INTRANASAL ADMINISTRATION INTO NEW ZEALAND RABBITS (N =5); DOSE = 2000 I. U.

<u>VARIABLE</u>	<u>T<sub>max</sub>(min)</u>	<u>Max<sub>d</sub>(%)</u>	<u>AUC (mg.min/dL)</u>	<u>% D</u>
CONTROL			4045.40 ± 255.17	
I.V (5 I.U.)	30	28	2732 ± 271.39 <sup>a</sup>	32
<u>sCT</u>				
0.045 M	180	28.12	3308.0 ± 42.45	18
0.15 M	240	12.96	3724.12 ± 120.99 <sup>b</sup>	7.9
0.3 M	180	28.83	3267.74 ± 31.31	19
<u>DMPG-sCT</u>				
0.045 M	120	34.58	3209.55 ± 21.66	19
0.15 M	50	38.89	2884.70 ± 53.24 <sup>b</sup>	20
0.3 M	90	38.49	3208 ± 102.49	20

a = significantly different from control at  $p > 0.05$

b = significantly different in calcium levels at  $p > 0.05$

FIGURE 4 THE INFLUENCE OF A SALT CONCENTRATION OF 0.045 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN



sCT formulations incorporating DMPG. Thus for formulations containing 0.045 M NaCl salt and a viscosity of 1 cps, the maximal decrease in calcium elicited by the presence of DMPG and sCT was higher (34.15% versus 28.12%), but  $D\%$ , which is the overall measure of the hypocalcemic activity of sCT was similar (18% versus 19%) to sCT formulations. Thus there was no significant difference in the total hypocalcemic activity elicited by formulations with or without the enhancer at  $p < 0.5$ .

#### **Effect of 0.15 M NaCl**

The influence of DMPG on the calcium level after intranasal administration of sCT in the presence of 0.15 M NaCl at a viscosity of 1 cps is shown in figure 5. The  $\%max_d$  was 12.96% at 240 minutes and 38.89% at 50 minutes in the absence and presence of DMPG in the sCT formulations respectively. The AUC was  $3724.12 \pm 120.99$  mg.min/dL with a total lowering of calcium,  $D\%$  of 7.9% for sCT formulations as compared to the AUC of  $2884.70 \pm 53.24$  mg.min/dL with a  $D\%$  of 20% in the presence of DMPG in sCT formulations. The comparison of AUC and a  $D\%$  for these two formulations demonstrated that there was a significant difference on the total lowering of calcium level in the presence of DMPG in sCT formulations at a salt concentration of 0.15 M and a viscosity of 1 cps as compared to sCT formulations (20% vs 7.9%) at  $p < 0.05$ .

#### **Effect of 0.3M NaCl**

The influence of DMPG on the calcium level after intranasal administration of formulations containing sCT in the presence of 0.3 M NaCl at a viscosity of 1 cps is shown in figure 6. The maximal decrease was 28.83% at 180 minutes for formulations containing only sCT as compared to 38.49% at 90 minutes for

FIGURE 5 THE INFLUENCE OF A SALT CONCENTRATION OF 0.15 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN

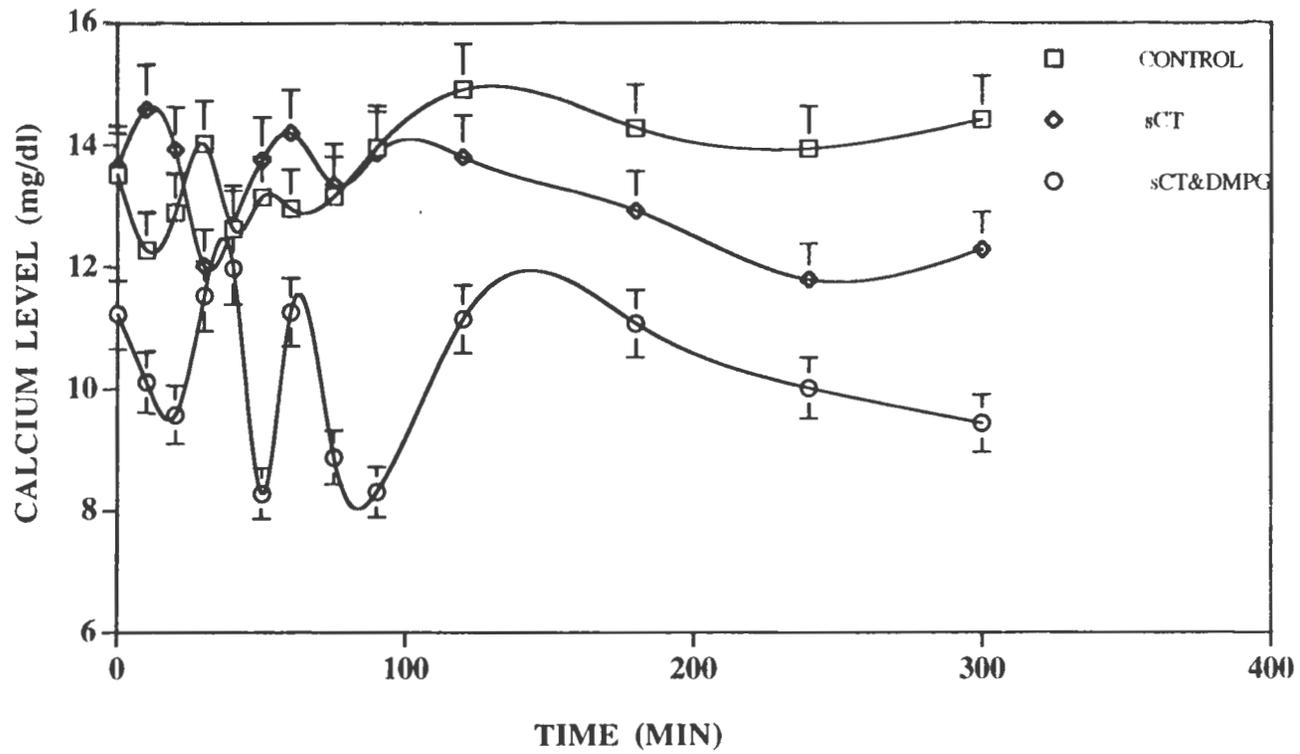
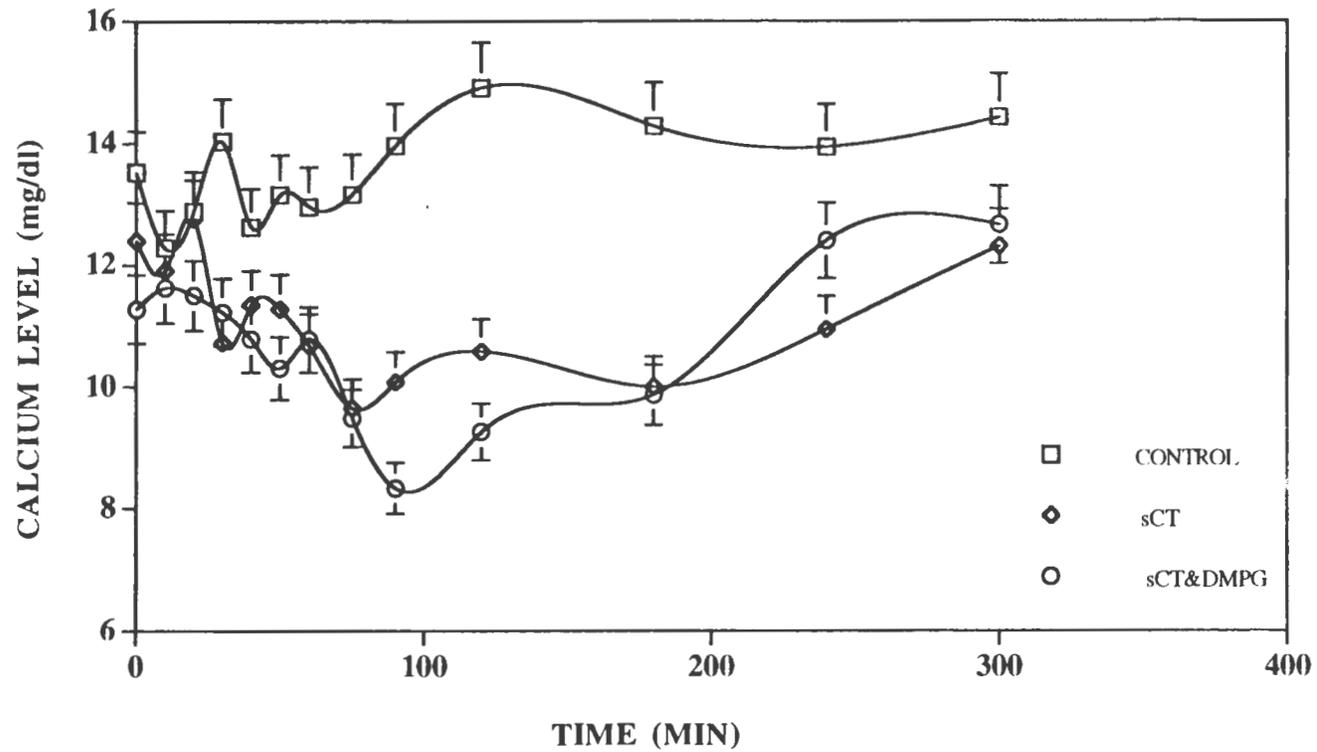


FIGURE 6 THE INFLUENCE OF A SALT CONCENTRATION OF 0.3 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN



formulations incorporating DMPG and sCT. The AUC was  $3267.74 \pm 31.31$  mg.min/dL and a total lowering of calcium, D% of 19% for sCT formulations as compared to AUC of  $3208 \pm 102.49$  mg.min/dL with a D% of 20% for formulations containing DMPG and sCT. This indicated that the total lowering in calcium was comparable, in the absence and presence of DMPG at a salt concentration of 0.3 M (19% versus 20%). Therefore the calcium lowering effects of sCT were demonstrated to be not affected by presence of DMPG when the salt concentration was 0.3 M.

The hypocalcemic activity was comparable for formulations containing only sCT and for formulations incorporating DMPG with sCT at a NaCl salt concentration of 0.045 M and 0.3 M at the low viscosity of 1 cps. However at a NaCl salt concentration of 0.15 M with a viscosity of 1 cps, the formulations containing DMPG with sCT showed a significant difference in the calcium lowering effect of salmon calcitonin. However in all the three salt concentrations the total lowering of calcium ranged from 7.9% to 20%. The  $T_{max}$  calculated as a pharmacodynamic parameter for the maximum lowering of calcium ranged from 50 to 240 minutes indicating the pharmacodynamic activity of salmon calcitonin in lowering of calcium is sustained and does not correlate with the presence of calcitonin in the serum.

### **Influence of viscosity**

It was hypothesized that higher viscosity formulations may act in synergism with the absorption enhancing effect of DMPG at a NaCl salt concentration of 0.045-0.3 M. The pharmacokinetic parameters obtained after administering the absorption enhancer, DMPG in sCT formulations at the higher viscosity of 76 cps were compared to the low viscosity formulations of 1 cps, with and

without the DMPG. The study was designed to evaluate the use of high viscosity (~ 75 cps) formulations obtained by using 1% w/w methylcellulose delivered as a fine spray with a prototype device, the nasal micron spray pump, to facilitate the uniform distribution of spray in the nasal cavity<sup>55</sup>.

It has been demonstrated<sup>53</sup> that the addition of methylcellulose produced a more sustained and slower absorption as indicated by the difference in time to reach maximum plasma concentrations. However the AUC data indicated a similar bioavailability for the two formulations. It is thought that the addition of a viscosity enhancing agent such as methylcellulose increases the contact time of the dosage form to the mucosal membrane and may help in enhancing the absorption by reducing the mucociliary clearance of the administered formulation and also changing the pattern of deposition and clearance<sup>59</sup>. An increase in viscosity increases the particle size in a spray formulation. It also leads to a more localized deposition on the anterior part of the nose and thus facilitates slower clearance<sup>60</sup>.

On the other hand, with regard to mucociliary clearance, results are difficult to interpret. According to some authors, an increase in viscosity results in a decrease in the clearance rate<sup>61</sup>. Viscosity also influences the rate of diffusion of the active ingredient resulting in delayed release without any improvements in bioavailability<sup>62</sup>.

The influence of viscosity on formulations containing DMPG as compared to formulations without DMPG at a NaCl concentration of 0.045-0.3 M are shown in figure 7-9. The pharmacokinetic parameters of these formulations are summarized in Table III.

TABLE III PHARMACOKINETIC PARAMETERS OF HIGH VISCOSITY SALMON CALCITONIN FORMULATIONS AFTER INTRANASAL ADMINISTRATION IN NEW ZEALAND RABBITS (N = 5)

<u>VARIABLE</u>	Tmax (min)	Cmax(ng/ml)	AUC(ng.min/ml)	%Bioavailability
CONTROL			401 ± 75.3	
I.V (5 I.U.)	14	25 ± 2.9	1102.1 ± 19.8 <sup>a</sup>	100
INTRANASAL sCT DOSE 2000 I.U.				
0.045 M	120	23 ± 2.4	3579 ± 103	0.71
0.15 M	58	12 ± 9.7	604.64 ± 97.74 <sup>b</sup>	0.14
0.3 M	90	32 ± 19.6	3589 ± 25 <sup>c</sup>	0.72
INTRANASAL DMPG-sCT DOSE 2000 I.U.				
0.045 M	90	54 ± 24	4203 ± 384.1	0.96
0.15 M	10	13 ± 7	1244 ± 275 <sup>b</sup>	0.28
0.3 M	63	8 ± 7.5	2134 ± 1369 <sup>c</sup>	0.56

a significantly different from control at  $p < 0.05$

b significantly different in AUC for DMPG formulations at 0.15 M NaCl at  $p < 0.05$

c significantly different in AUC for DMPG formulations at 0.3M NaCl at  $p < 0.05$

### High Viscosity formulation

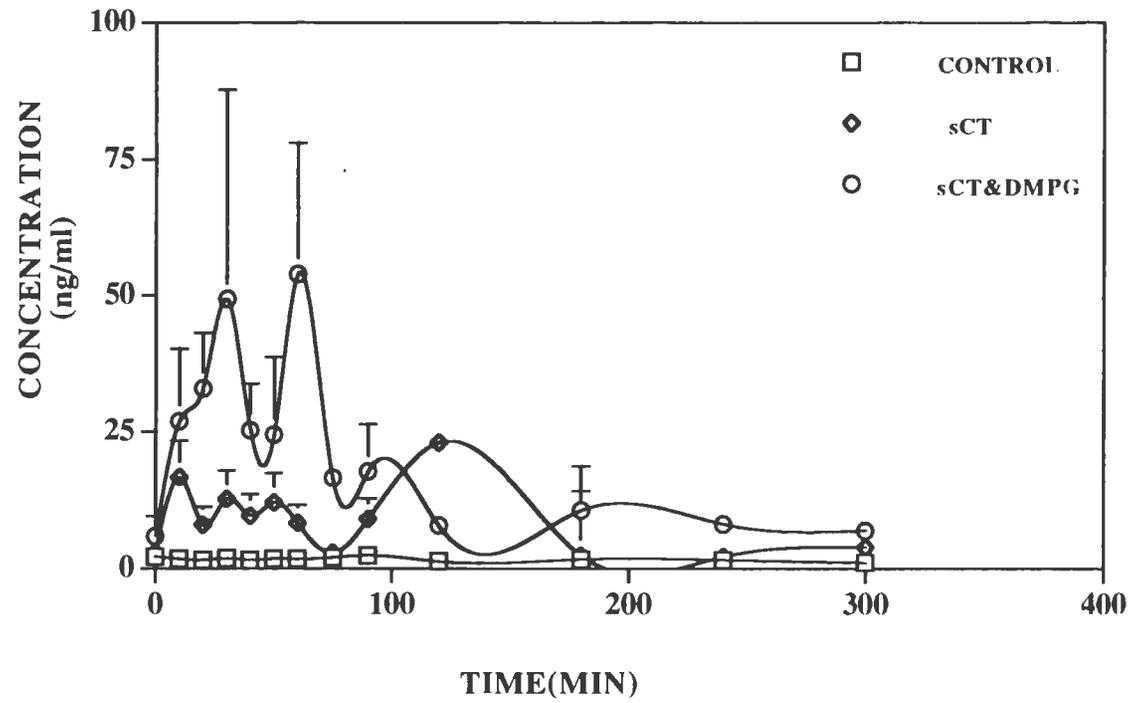
#### **Effect of 0.045 M NaCl**

The influence of DMPG on the absorption of sCT in formulations containing 0.045 M NaCl at a viscosity of 76 cps is shown in figure 7. The C<sub>max</sub> was 23 ng/ml at 120 minutes for the sCT formulations as compared to the C<sub>max</sub> of 54±24 ng/ml at 55 minutes elicited by the presence of DMPG in sCT formulations. The AUC was 3579±103.0 ng.min/ml in the absence of DMPG and 4203±706 ng.min/ml in the presence of DMPG with an absolute bioavailability of 0.81 and 0.96 respectively. The presence of DMPG at a high viscosity of 76 cps did not make a significant difference in the bioavailability of sCT at a NaCl concentration of 0.045 M (0.96 vs 0.81). However DMPG at a low viscosity of 1 cps elicited a two fold increase in the bioavailability (1.4 vs 0.71). This can be explained by the fact that increased viscosity reduces the diffusion of the drug from the formulation to the site of absorption. Thus the t<sub>max</sub> is increased to 60 and 120 minutes for the two high viscosity formulations as compared to the 38 minutes for the low viscosity formulations without much improvement in bioavailability.

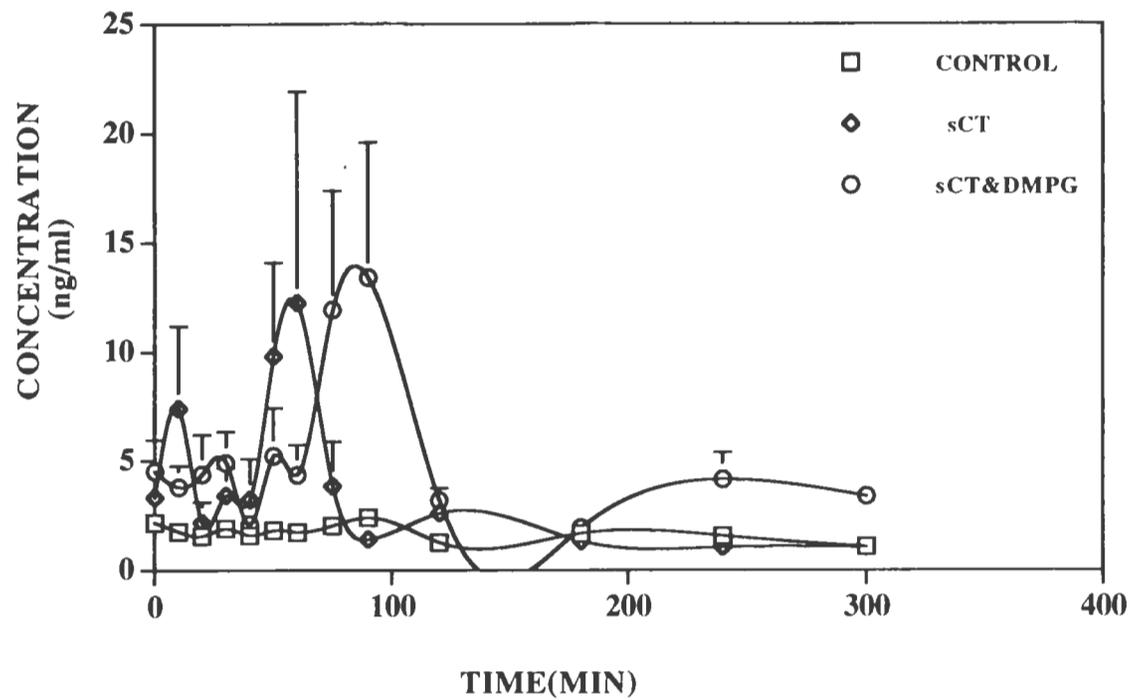
#### **Effect of 0.15 M NaCl**

The effect of the formulation containing DMPG and sCT at a NaCl concentration of 0.15 M at 76 cps is shown in figure 8. The C<sub>max</sub> obtained for formulations containing only sCT was 12±9.7 ng/ml at 58 minutes as compared to the C<sub>max</sub> for formulations containing DMPG and sCT of 13±7 ng/ml at 90 minutes. The AUC was 604.64 ng.min/ml and an absolute bioavailability of 0.14 for sCT formulation whereas an AUC of 1244.914±275.65 ng.min/ml and a bioavailability of 0.28 was obtained for the formulations containing DMPG and sCT. At NaCl salt concentration of 0.15 M, there was a

Figure 7 THE INFLUENCE OF A SALT CONCENTRATION OF 0.045 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN AT HIGH VISCOSITY



**Figure 8 THE INFLUENCE OF A SALT CONCENTRATION OF 0.15 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN AT HIGH VISCOSITY**



two fold increase (0.28 vs 0.14) in the bioavailability of sCT in the presence of DMPG at a high viscosity of 76 cps . These results were very similar to those obtained at low viscosity formulations of 1 cps at a salt concentration of 0.15 M.

#### **Effect of 0.3 M NaCl**

The effect of the formulation containing DMPG and sCT at a NaCl concentration of 0.3 M and 76 cps is shown in figure 9. The C<sub>max</sub> was 32±19.6 ng/ml at 90 minutes for the salmon calcitonin formulations. The C<sub>max</sub> was 8±7.5 ng/ml at 63 minutes for DMPG and sCT formulations. The absolute bioavailability of 0.72 was obtained for sCT formulations in the presence of 0.3 M salt as compared to a bioavailability of 0.56 for formulations containing DMPG and sCT in the presence of 0.3 M as shown in Table III. This demonstrated that in the presence of higher salt concentration of 0.3 M NaCl the DMPG did not elicit an absorption enhancing effect. These results correlated with results obtained for low viscosity formulations at 0.3 M NaCl salt concentration.

#### **Pharmacodynamic effects**

##### **High viscosity formulations**

The parameters for the hypocalcemic activity of the high viscosity formulations with and without DMPG are summarized in Table IV as T<sub>max</sub>(min), %max<sub>d</sub> (maximal percent decrease), AUC (mg.min/dL) and D% (total hypocalcemic activity). A review of the pharmacodynamic effect of sCT of lowering of calcium is illustrated in figures 10-12.

Figure 9 THE INFLUENCE OF A SALT CONCENTRATION OF 0.3 M AND DMPG ON THE INTRANASAL ABSORPTION OF SALMON CALCITONIN

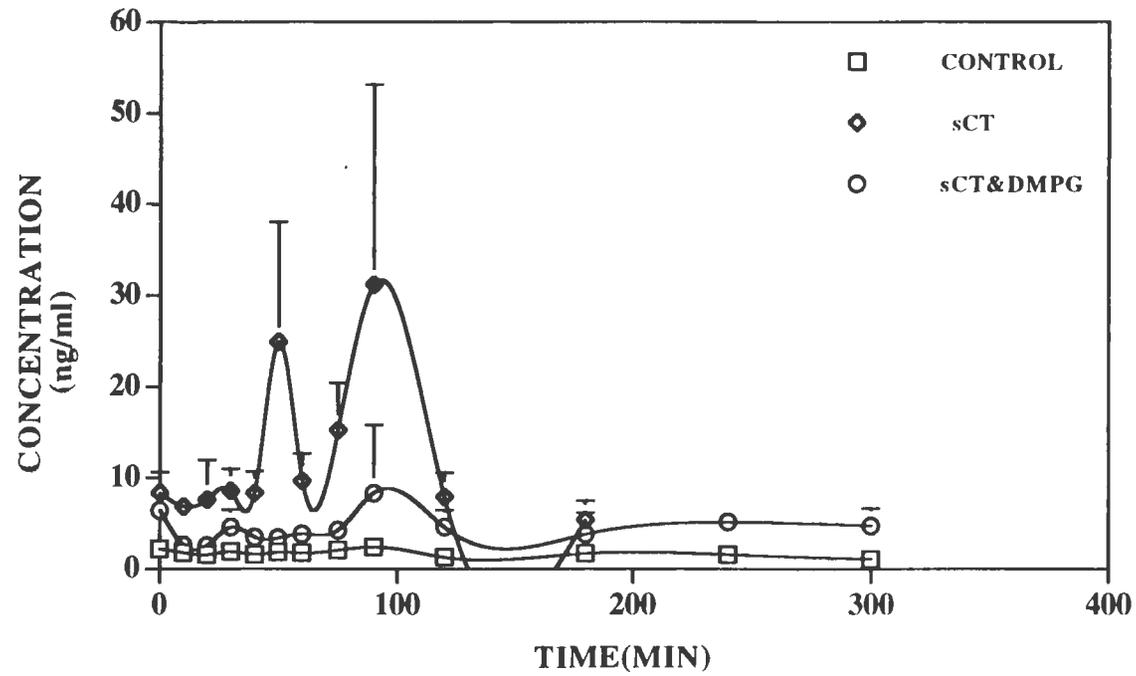


TABLE IV PHARMACODYNAMIC PARAMETERS OF HIGH VISCOSITY SALMON CALCITONIN FORMULATIONS AFTER INTRANASAL ADMINISTRATION INTO NEW ZEALAND RABBITS (N =5); DOSE = 2000 I. U.

VARIABLE	T <sub>max</sub> (min)	%Max <sub>d</sub>	AUC (mg.min/dL)	% D
CONTROL			4045.40 ± 255.17	
I.V (5 I.U.)	30	28	2732 ± 271.39 <sup>a</sup>	32
INTRANASAL <u>sCT</u> DOSE 2000 I.U.				
0.045 M	120	34.58	3154.47	22
0.15 M	240	28.81	3242.83 <sup>b</sup>	19
0.3 M	90	23.52	3696.28	9
INTRANASAL <u>DMPG &amp; sCT</u> DOSE 2000 I.U.				
0.045 M	90	28.36	3331	17
0.15 M	180	27.09	3161.67	21
0.3 M	75	29.01	3510.80	13

a significantly different from control at p<0.05

b significantly different in AUC for DMPG formulations at 0.15 M at p<0.5

#### **Effect of 0.045 M NaCl**

The influence of DMPG on the calcium level after intranasal administration of sCT in the presence of 0.045 M NaCl and at a viscosity of 76 cps is shown in figure 10. The maximal decrease was 34.58% at 300 minutes and 28.36% at 90 minutes in the absence and presence of DMPG respectively. A total lowering of calcium of 22% for formulations containing sCT whereas a D% of 17% for formulations incorporating DMPG in sCT was obtained. There was no significant difference on the calcium level in the absence or presence of DMPG at 0.045M NaCl at a viscosity of 76 cps.

#### **Effect of 0.15 M NaCl**

The influence of DMPG on the calcium level after intranasal administration of sCT in the presence of 0.15 M NaCl and at a viscosity of 76 cps is shown in figure 11. The %max<sub>d</sub> was 28.81% at 240 minutes and 27.09% at 180 minutes in the absence and presence of DMPG respectively. The AUC was 3242.83 mg.min/dL with a D% of 19% for formulations containing sCT and 3161.679 mg.min/dL with D% of 21% for formulations containing DMPG and sCT. The comparison of AUC and a D% of these two formulations demonstrated that there was no significant difference in the calcium level by high viscosity formulations at a salt concentration of 0.15 M containing enhancer or without enhancer. These results were different from those obtained at a low viscosity of 1 cps where there was a significant difference in the overall hypocalcemic activity of formulations with and without enhancer. The reason for this difference is not known.

#### **Effect of 0.3 M NaCl**

The influence of DMPG on the calcium level after intranasal administration

FIGURE 10 THE INFLUENCE OF A SALT CONCENTRATION OF 0.045 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN AT HIGH VISCOSITY

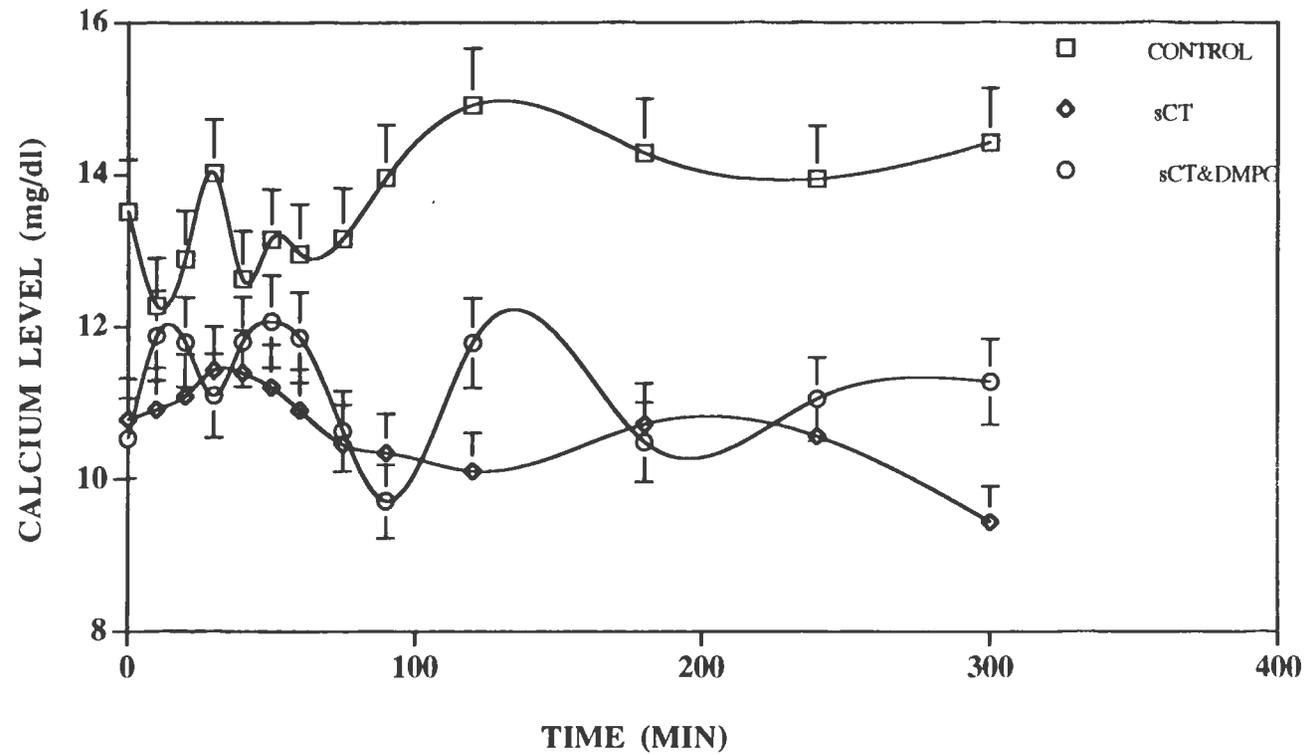


FIGURE 11 THE INFLUENCE OF A SALT CONCENTRATION OF 0.15 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN AT HIGH VISCOSITY

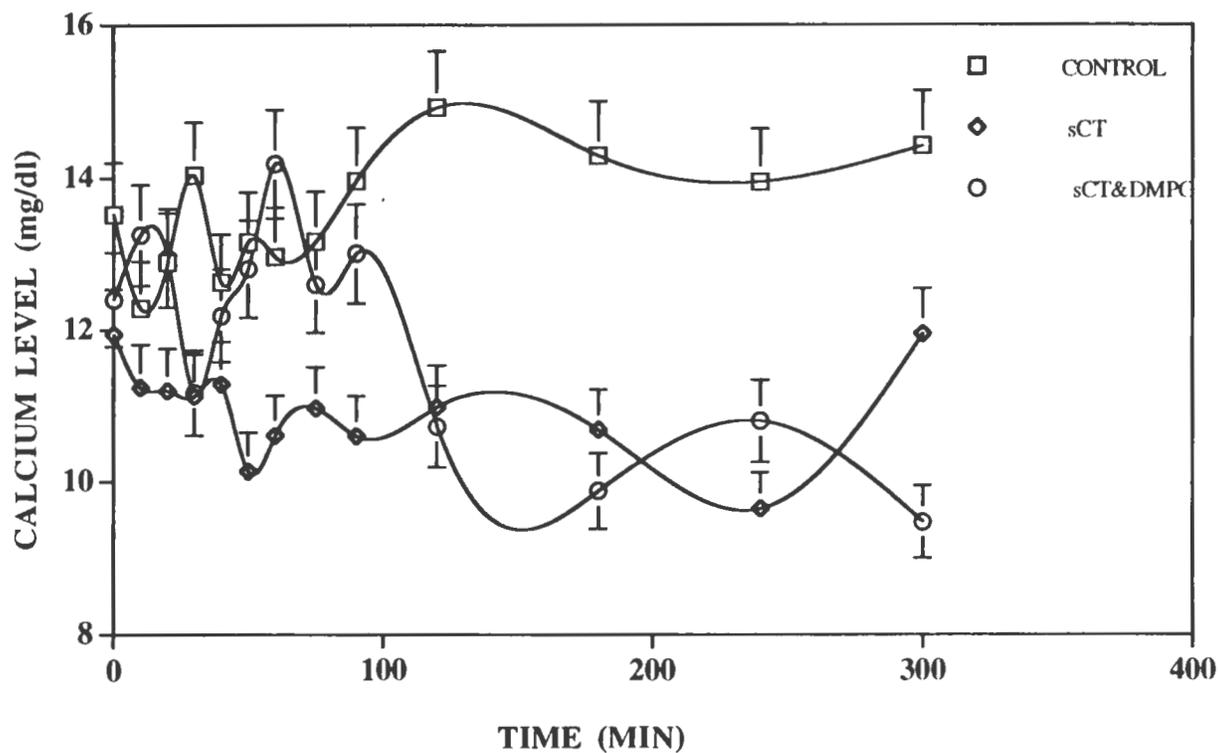
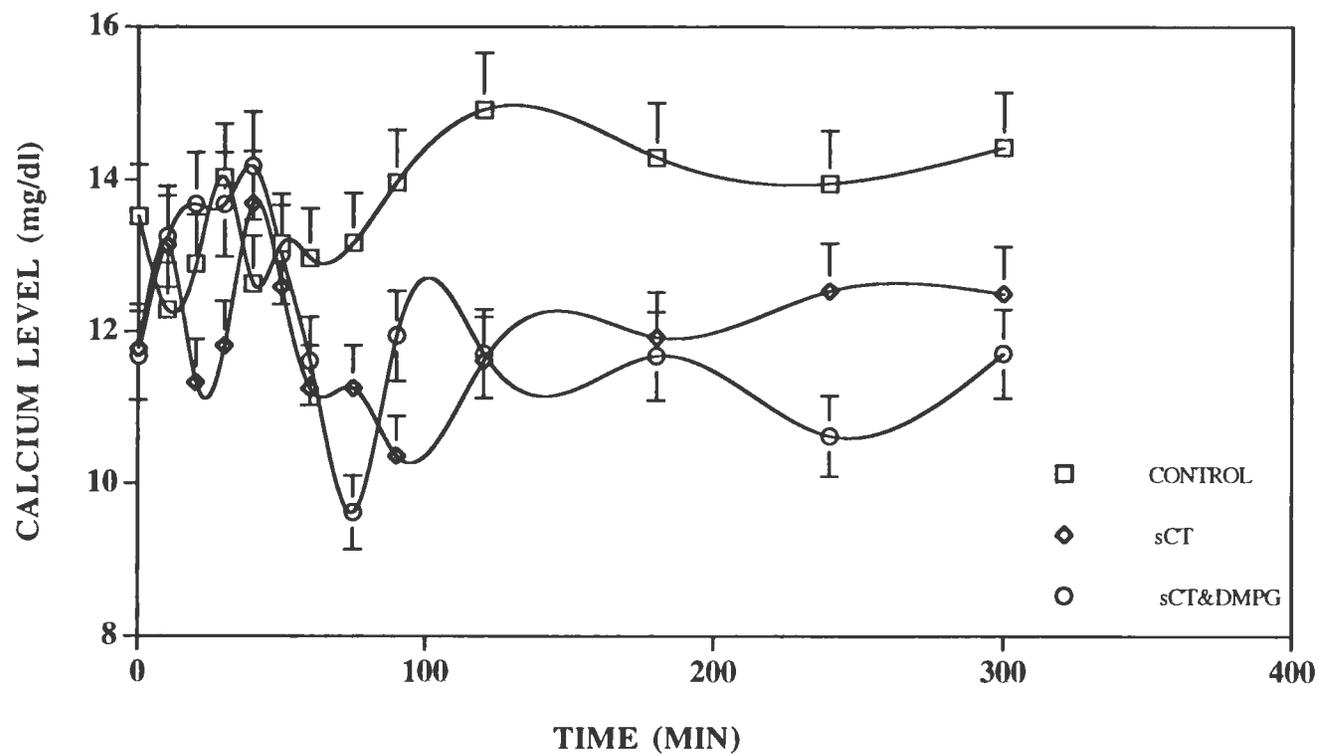


FIGURE 12 THE INFLUENCE OF A SALT CONCENTRATION OF 0.3 M AND DMPG ON THE CALCIUM LEVEL AFTER INTRANASAL ADMINISTRATION OF SALMON CALCITONIN AT HIGH VISCOSITY



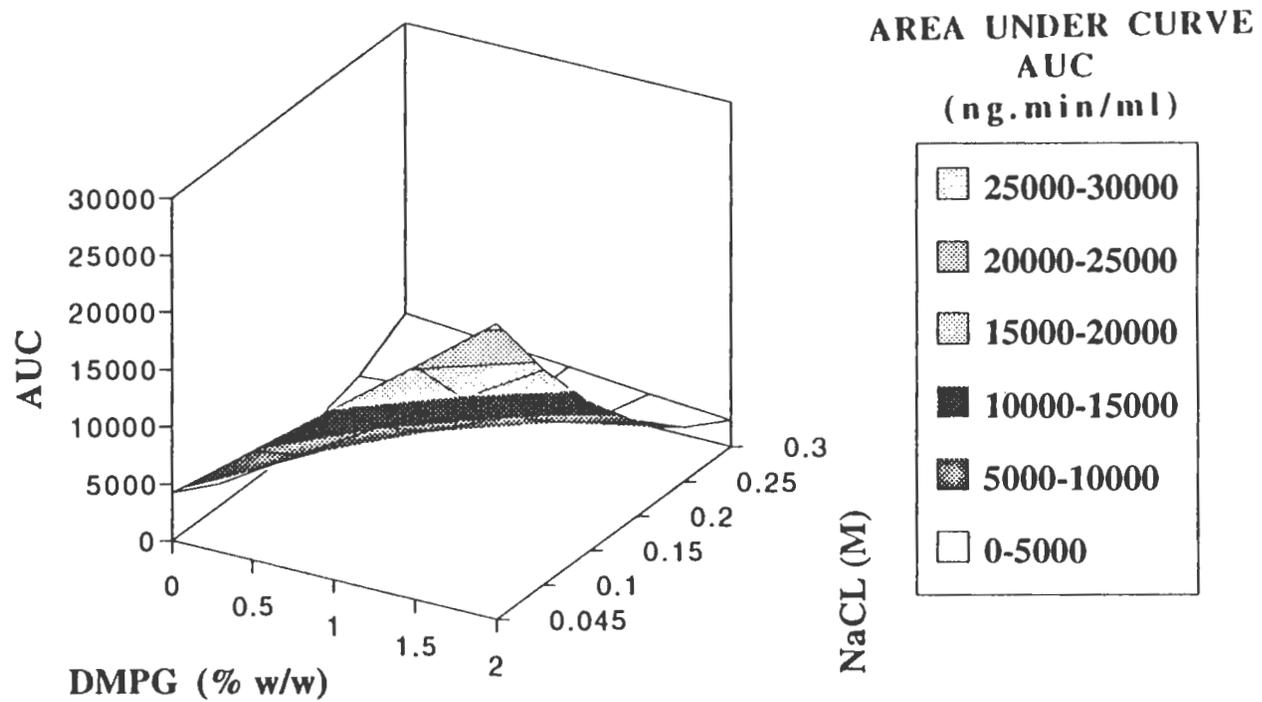
of sCT in the presence of 0.3 M NaCl and a viscosity of 76 cps is shown in figure 12. The %maxd was 23.52% at 90 minutes and 29.01% at 75 minutes in the absence and presence of DMPG respectively. The AUC was 3696.28 mg.min/dL with a D% of 9% was calculated for formulations containing only sCT as compared to the AUC of 3510.801 mg.min/dL with a D% of 13% obtained by formulations containing containing DMPG and sCT. Again there was no significant difference in the total calcium lowering effect of sCT in the absence and presence of DMPG (9% vs 13%).

The similarity in the extent of the hypocalcemic effect observed for the different formulations of sCT especially those which demonstrated significant differences in absorption of sCT may be attributed to the acute homeostatic mechanism between calcitonin and parathyroid hormone. The parathyroid hormone counteracts the effects of calcitonin once a critical level of hypocalcemia is attained. Thus the extent of the hypocalcemic effect is controlled by the hormonal balance. The acute control of blood calcium is accomplished by the feedback action of calcitonin and parathyroid hormone<sup>63</sup>. This complex regulation of blood calcium concentration makes it difficult to evaluate the efficacy of sCT delivery systems by observation of the hypocalcemic effect. Additionally the short half life of sCT makes it more difficult to correlate serum sCT and its biological effects<sup>64</sup>.

### **Selection of the best formulations**

Response surface methodology was used as a optimization technique to select the levels of the variables that could best describe the most desirable attributes of the formulation. The response surface analysis plot demonstrated that a maxima was obtained, which suggests that there is a

Figure 13 THREE DIMENSIONAL SURFACE PLOT SHOWING  
MAXIMA FOR DMPG AND SALT CONCENTRATION WITH  
RESPECT TO AREA UNDER THE CURVE



optimum salt concentration to deliver DMPG. The three dimensional plot included concentration of NaCl on X-axis, concentration of DMPG on Y -axis and AUC on Z-axis. Figure 13 indicates an AUC of 25000-30000 ng.min/ml for 0.045M formulations with DMPG at a concentration of 2% and a AUC of 0-50000 ng.min/ml for formulations with a NaCl of concentration > 0.25M and a DMPG of 0-0.5% w/w concentration. Thus the analysis suggests that the most effective formulation should contain DMPG at a salt concentration of 0.045M to provide the best absorption enhancing effect of DMPG on salmon calcitonin.

To overcome the low bioavailability of higher molecular weight polypeptides in intranasal delivery, it becomes necessary to include an absorption enhancer. However the choice of absorption enhancer is very important as it should be able to increase the uptake of the peptide, at the same time it should be safe, nontoxic in long term use and should not alter the physiology, histology of nasal cavity. This study not only investigated the role of DMPG in enhancing the absorption but also the effect of other formulation factors such as viscosity and salt concentration on the efficacy of this enhancer. Very little work has been done on phospholipids as absorption enhancers, especially acidic absorption enhancers. It is extremely important to study their effects after chronic use on ciliary beat frequency, histology and physiological functions before they can be approved as an excipient in the formulations meant for nasal delivery.

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## CONCLUSIONS

1. A study of delivery devices and components which used the Malvern laser sizer to determine the qualitative and quantitative spray profile indicated that the currently available metered nasal devices were capable of dispensing low viscosity solutions as a fine spray and maintaining a Gaussian distribution. The devices also demonstrated an excellent total dose delivery and reproducibility for all types of solutions.
2. The nasal micron spray pump, a prototype device, that was developed for pulmonary drug delivery is very versatile and can be extremely valuable in nasal drug delivery. It is capable of generating a narrower droplet size distribution and a fine spray for viscous solutions which cannot be achieved with currently available spray pumps.
3. Among the formulation variables investigated, tonicity elicited a significant difference in achieving a higher concentration of salmon calcitonin in the serum and enhancing the bioavailability of salmon calcitonin.
4. Viscosity studied at the two levels indicated that although the droplet spray distribution was significantly affected, no significant improvement in the bioavailability of salmon calcitonin was found for solutions differing in viscosity.

5. The use of dimyristoylphosphatidylglycerol, an acidic phospholipid, a novel permeation enhancer demonstrated an increase in the bioavailability of salmon calcitonin.
6. The absorption enhancing activity of dimyristoylphosphatidylglycerol when studied as a function of salt concentration indicated that a low salt concentration was required for the optimal activity.
7. The pharmacodynamic effect of salmon calcitonin of lowering of calcium indicated that although it was a useful parameter for indirectly measuring the efficacy of salmon calcitonin formulations, it did not correlate directly with the plasma concentration of the drug and was very difficult to interpret the results. This inconsistency was attributed to the fact that the hypocalcemic response is controlled by the parathyroid gland which correctively responds to calcitonin plasma concentration.
8. The low bioavailability of higher molecular weight polypeptides such as salmon calcitonin can be enhanced by a optimization of both formulation parameters and nasal administration device variables.

This investigation emphasized a need to study both formulation and devices variables to overcome low bioavailability. This approach is of tremendous value to the formulation scientist by increasing the probability of optimizing drug delivery and accelerating product development.

## SECTION III

**APPENDIX A**  
**TECHNIQUES USED FOR ANALYSIS**

## QUANTITATIVE DETERMINATION OF SALMON CALCITONIN IN SERUM

### Double Antibody Radioimmunoassay

#### Principle

Double antibody radioimmunoassay is a sequential competitive radioimmunoassay. The sample is first preincubated with anti-calcitonin serum.  $^{125}\text{I}$ -labeled calcitonin then competes with the calcitonin in the serum samples for antibody sites. After incubation for a fixed time, separation of bound from free is achieved by the PEG-accelerated double antibody method. The antibody-bound fraction is precipitated by centrifugation under refrigerated conditions. The precipitated pellets are counted by a gamma counter.

#### Reagents provided (Diagnostics Systems Limited, Webster, TX)

1. Calcitonin Antiserum contains guinea pig as primary antibodies.
2.  $^{125}\text{I}$ - Calcitonin
3.  $^{125}\text{I}$ - Calcitonin calibrators.

Six vials labeled A through E with 0 to 50 ng/ml concentration of salmon calcitonin.

4. Precipitating reagent

One vial of precipitating reagent consisting of guinea pig-antigoat gamma globulin in polyethylene glycol.

## 5. Calcitonin controls

Two vials labeled calcitonin controls I and II containing calcitonin in a protein based matrix containing 2 mg/ ml and 10 ng/ml as low and high control levels of salmon calcitonin.

All the samples except precipitating reagent were lyophilized. They were stored in refrigerator and were reconstituted with distilled water for 30 minutes before the use.

### **Procedure**

Serum samples were incubated with calcitonin antiserum for 6 hrs at 2-8°C. After the first incubation, radiolabeled calcitonin was added and samples reincubated for 24 hrs at 2-8°C. Separation was achieved by the addition of precipitating reagent and was followed by a third incubation for 1 hr. The samples were centrifuged for 30 minutes at 3000 rpm at 20°C. The supernatant was drained off, the rim of tubes wiped to eliminate radioactive contamination and precipitated pellets were counted in a gamma counter for 1 minute in terms of cpm (counts per minute) and analysis of each sample was performed in duplicate.

### **Calculations**

To be able to calculate the concentration of calcitonin on serum samples, the first step was to calculate the total count, non-specific binding, maximum binding and %B/Bo.

The cpm of <sup>125</sup>Iodinated Calcitonin was counted and termed as total counts. The nonspecific binding (NSB) was determined by using deionized water instead of serum samples and was subtracted from the readingt in cpm obtained for all serum samples,

$$\text{Net Count} = \text{average cpm} - \text{average NSB cpm}$$

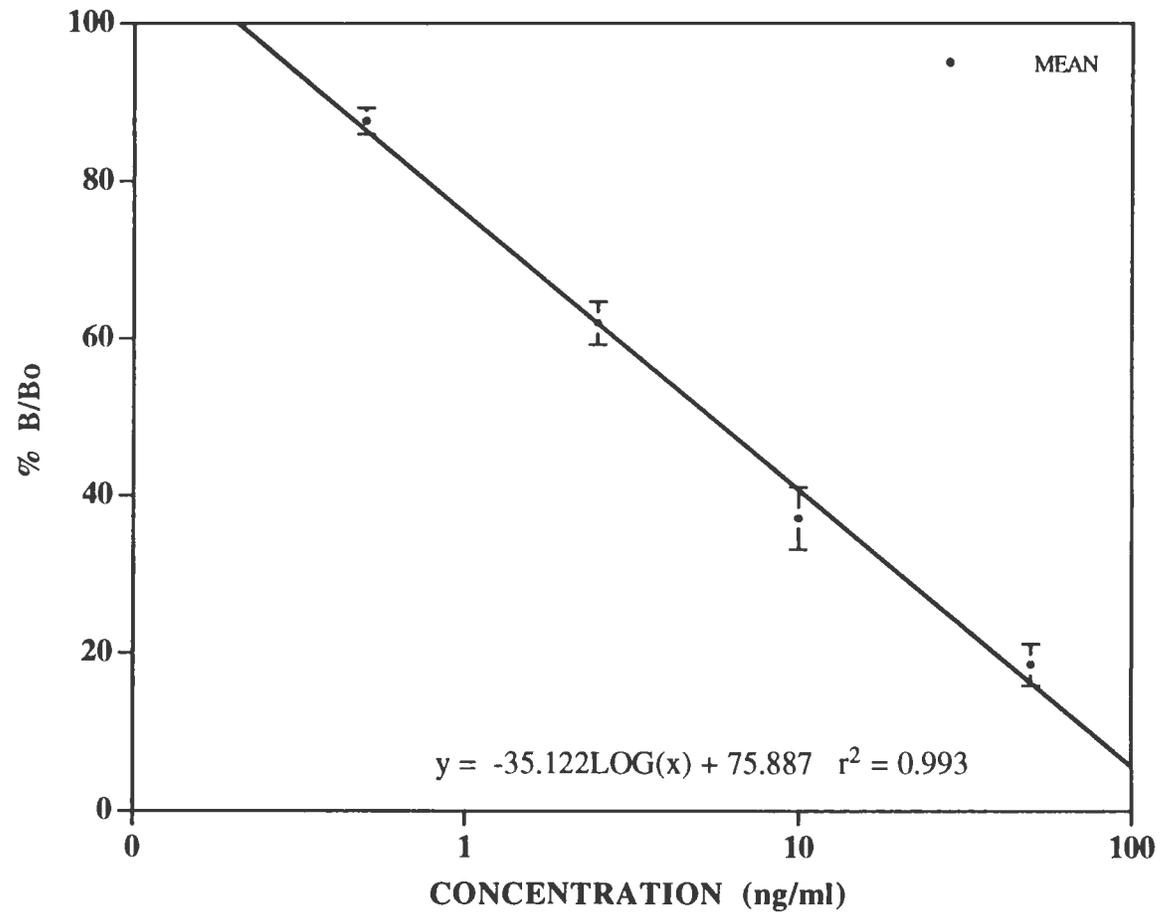
Maximum binding was provided by zero ng/ml of calcitonin standard. The counts obtained by 0ng/ml and subtracted by non-specific binding was taken to be 100% and was called maximum binding.

$$\text{Percent bound (B/Bo)} = \frac{\text{Net count}}{\text{Net maximum binding}} \times 100$$

The sigmoidal standard curve of B/Bo vs log[sCT], typical of competitive binding assays was linearized using the log transformation. Using the linear log graph paper, percent bound was plotted on the vertical axis against concentration on the horizontal axis for each of the nonzero calibrators. The best fit was determined using a non weighted least square regression analysis.

## **Results**

The radioimmunoassay quantitation range was 100 pg/ml- 5000 pg/ml. The square of the coefficient of determination on this range for the B/Bo versus log[sCT] plots was greater than 0.993 for all assays performed in duplicate.

STANDARD CURVE OF SALMON CALCITONIN  
USING RADIOIMMUNOASSAY

## QUANTITATIVE DETERMINATION OF CALCIUM IN SERUM CALCIUM PLUS® PROCEDURE

### Principle

The calcium plus® procedure developed by DMA, Houston, TX., utilizes a dye, arsenazo III which binds to calcium at an acid pH. A bluish - purple complex is formed. The intensity of the color formed is measured by an increase in absorbance of the reaction mixture at 650 nm.

### Reagents Provided

Calcium plus® reagent

Control sera of concentration of calcium 10 mg/dL

### Composition of Calcium Plus Reagent

Arsenzo III (0.136 mmol/L)

Buffer

Surfactant

Sodium azide

### Manual Procedure

1. Add 2 ml of calcium plus reagent into labeled test tubes.
2. Add 0.025 ml of each control and samples to their respective test tubes.
3. Mix well
4. Use 0.025 ml deionized water for reagent blank instead of sample or control.
5. Incubate for 2 minutes at the reaction temperature.

6. Set the wavelength of the instrument at 650 nm. Zero with the reagent blank.

7. Read and record absorbance of the samples and controls.

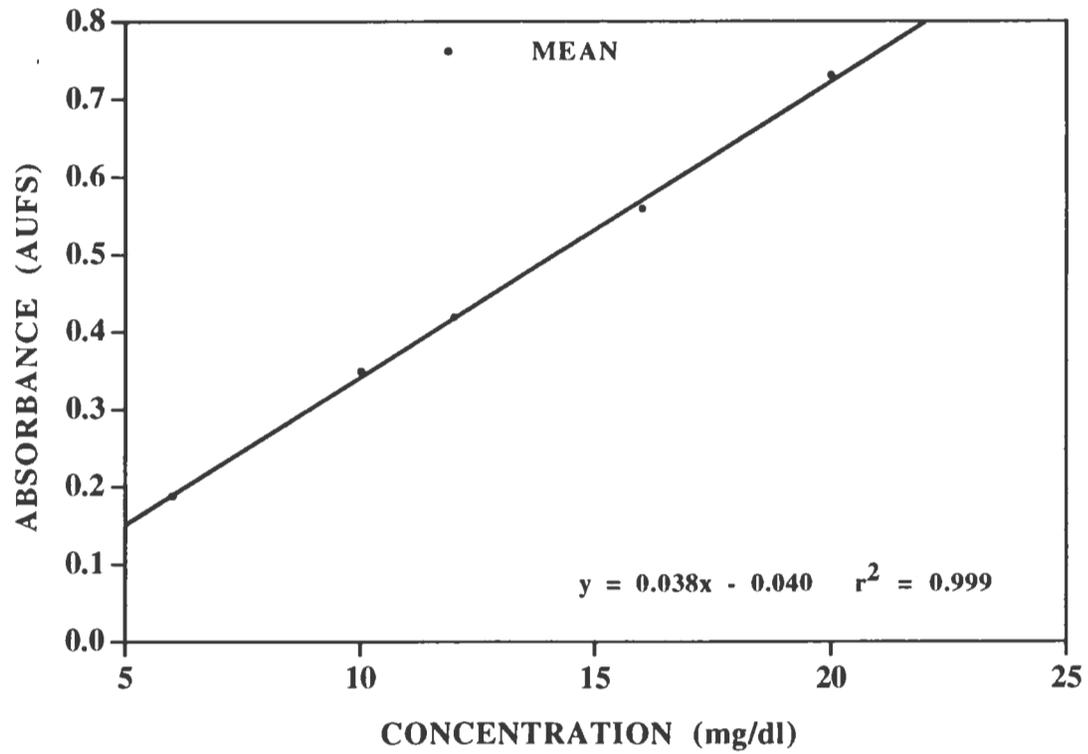
### Calculations

$$\frac{\text{Absorbance of unknown}}{\text{Absorbance of standard}} \times \text{Concentration of standard} = \text{Calcium (mg/dL)}$$

### RESULTS

Serum calcium was quantified by uv spectroscopy and the range of the linearity was determined to be between 5 mg/dL - 25 mg/dL. The square of the coefficient of determination in this range for the absorbance (AUFS) versus concentration (mg/dL) was 0.999 for all the assays performed in duplicate during the course of study.

STANDARD CURVE OF CALCIUM USING  
ARSENZO COMPLEX III MEASURED AT 650 nm.



**APPENDIX B**

**STATISTICAL DATA SUPPORTING MANUSCRIPT I**

**STATISTICAL ANALYSIS OF MALVERN DATA  
FORMULATION VARIABLE - VISCOSITY**

**Analysis of Variance Procedure**

**Dependent Variable: M50**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	6204.0566000	3102.0283000	126.56	0.0001
Error	6	147.0588000	24.5098000		
Corrected Total	8	6351.1154000			

R-Square	C.V.	Root MSE	M50 Mean
0.976845	7.138084	4.9507373	69.356667

**Dependent Variable: M90**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	90381.751489	45190.875744	229.02	0.0001
Error	6	1183.945667	197.324278		
Corrected Total	8	91565.697156			

R-Square	C.V.	Root MSE	M90 Mean
0.987070	8.366806	14.047216	167.89222

STATISTICAL ANALYSIS OF MALVERN DATA  
FORMULATION VARIABLE - TONICITY

Analysis of Variance Procedure  
Class Level Information

Class Levels Values

TON 3 HYE HYO ISO

Number of observations in data set = 9

Dependent Variable: M50

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	10.10816089	5.05408044	2.34	0.1773
Error	6	12.95668267	2.15944711		
Corrected Total	8	23.06484356			

Source	DF	Anova SS	Mean Square	F Value	Pr > F
TON	2	10.10816089	5.05408044	2.34	0.1773

Dependent Variable: M90

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	6821.6386727	3410.8193363	12.02	0.0080
Error	6	1703.2526833	283.8754472		
Corrected Total	8	8524.8913560			

**STATISTICAL ANALYSIS OF MALVERN DATA  
FORMULATION VARIABLE - CONCENTRATION OF ENHANCER**

**Analysis of Variance Procedure**

**Dependent Variable: M50**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	1	1008.8066667	1008.8066667	177.42	0.0002
Error	4	22.7437333	5.6859333		
Corrected Total	5	1031.5504000			
		R-Square	C.V.	Root MSE	M50 Mean
		0.977952	4.097113	2.3845195	58.200000

**Dependent Variable: M90**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	1	6591.5461500	6591.5461500	275.88	0.0001
Error	4	95.5715333	23.8928833		
Corrected Total	5	6687.1176833			
		R-Square	C.V.	Root MSE	M90 Mean
		0.985708	4.891867	4.8880347	99.921667

**STATISTICAL ANALYSIS OF MALVERN DATA  
DEVICE VARIABLE - MECHANISM OF PUMP**

**Analysis of Variance Procedure**

**Dependent Variable: M50**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	1	13.89281667	13.89281667	5.30	0.0828
Error	4	10.48746667	2.62186667		
Corrected Total	5	24.38028333			
	R-Square	C.V.	Root MSE	M50 Mean	
	0.569838	3.527836	1.6192179	45.898333	

**Dependent Variable: M90**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	1	1341.6130667	1341.6130667	2.16	0.2155
Error	4	2483.1138667	620.7784667		
Corrected Total	5	3824.7269333			
	R-Square	C.V.	Root MSE	M90 Mean	
	0.350774	27.20223	24.915426	91.593333	

STATISTICAL ANALYSIS OF MALVERN DATA  
DEVICE VARIABLE - SPRAY ANGLE

Analysis of Variance Procedure

Class Level Information

Class Levels Values

ANGLE 3 60° 35° 20°

Number of observations in data set = 9

Dependent Variable: M50

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	10.13763200	5.06881600	2.45	0.1670
Error	6	12.42667200	2.07111200		
Corrected Total	8	22.56430400			

Source	DF	Anova SS	Mean Square	F Value	Pr > F
ANGLE	2	10.13763200	5.06881600	2.45	0.1670

Dependent Variable: M90

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	3143.8714729	1571.9357364	4.45	0.0653
Error	6	2118.8262373	353.1377062		
Corrected Total	8	5262.6977102			

**STATISTICAL ANALYSIS OF MALVERN DATA  
DEVICE VARIABLE - ORIFICE DIAMETER**

**Analysis of Variance Procedure**

**Dependent Variable: M50**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	10.19342222	5.09671111	2.36	0.1749
Error	6	12.93220000	2.15536667		
Corrected Total	8	23.12562222			

R-Square	C.V.	Root MSE	M50 Mean
0.440785	3.198202	1.4681167	45.904444

**Dependent Variable: M90**

Source	DF	Sum of Squares	Mean Square	F Value	Pr > F
Model	2	5899.8090667	2949.9045333	10.03	0.0122
Error	6	1764.5765333	294.0960889		
Corrected Total	8	7664.3856000			

R-Square	C.V.	Root MSE	M90 Mean
0.769769	19.42742	17.149230	88.273333

**APPENDIX C**  
**STATISTICAL DATA SUPPORTING MANUSCRIPT I & II**

General Linear Models Procedure

Scheffe's test for variable: Y

NOTE: This test controls the type I experimentwise error rate  
but generally has a higher type II error rate than REGWF  
for all pairwise comparisons

Alpha= 0.05 df= 26 MSE= 8012872

Critical Value of F= 4.22520

Minimum Significant Difference= 1939.5

Means with the same letter are not significantly different.

Scheffe Grouping	Mean	N	VIS
A	4775.8	18	1
A	2957.8	18	76

Scheffe's test for variable: Y

NOTE: This test controls the type I experimentwise error rate  
but generally has a higher type II error rate than REGWF  
for all pairwise comparisons

Alpha= 0.05 df= 26 MSE= 8012872

Critical Value of F= 3.36902

Minimum Significant Difference= 2999.7

Means with the same letter are not significantly different.

Scheffe Grouping	Mean	N	TON
A	6319	12	100
A B	3612	12	600
B	1760	12	300

General Linear Models Procedure

Scheffe's test for variable:

NOTE: This test controls the type I experimentwise error rate  
but generally has a higher type II error rate than REGWF  
for all pairwise comparisons

Alpha= 0.05 df= 26 MSE= 801287

Critical Value of F= 4.22520

Minimum Significant Difference= 1939.5

Means with the same letter are not significantly different.

Scheffe Grouping	Mean	N	DMPG
A	5331.1	18	1.0
B	2402.6	18	0

General Linear Models Procedure

T tests (LSD) for variable: Y

NOTE: This test controls the type I comparisonwise error rate  
not the experimentwise error rate.

Alpha= 0.05 df= 26 MSE= 8012872

Critical Value of T= 2.06

Least Significant Difference= 1939.5

Means with the same letter are not significantly different.

T Grouping	Mean	N	VIS
A	4775.8	18	1
A	2957.8	18	76

General Linear Models Procedure

T tests (LSD) for variable: Y

NOTE: This test controls the type I comparisonwise error rate  
not the experimentwise error rate.

Alpha= 0.05 df= 26 MSE= 8012872

Critical Value of T= 2.06

Least Significant Difference= 2375.4

Means with the same letter are not significantly different.

T Grouping	Mean	N	TON
A	6319	12	100
B	3612	12	600
B	1670	12	300

General Linear Models Procedure

T tests (LSD) for variable: Y

NOTE: This test controls the type I comparisonwise error rate  
not the experimentwise error rate.

Alpha= 0.05 df= 26 MSE= 8012872

Critical Value of T= 2.06

Least Significant Difference= 1939.5

Means with the same letter are not significantly different.

T Grouping	Mean	N	DMPG
A	5331.1	18	1.0
B	2402.6	18	0

Source	DF	Type I SS	F Value	Pr > F
VIS	1	29745956.016	3.71	0.0650
DMPG	1	77188149.634	9.63	0.0046
TON	2	130842909.196	8.16	0.0018
DMPG*TON	2	59202705.613	3.69	0.0387
VIS*TON	2	62767100.747	3.92	0.0326
VIS*DMPG	1	888473.395	0.11	0.7418

Dependent Variable: Y

Source	DF	Type II SS	F Value	Pr > F
VIS	1	29745956.016	3.71	0.0650
DMPG	1	77188149.634	9.63	0.0046
TON	2	130842909.196	8.16	0.0018
DMPG*TON	2	59202705.613	3.69	0.0387
VIS*TON	2	62767100.747	3.92	0.0326
VIS*DMPG	1	888473.395	0.11	0.7418

Dependent Variable: Y

Source	DF	Type III SS	F Value	Pr > F
VIS	1	29745956.016	3.71	0.0650
DMPG	1	77188149.634	9.63	0.0046
TON	2	130842909.196	8.16	0.0018
DMPG*TON	2	59202705.613	3.69	0.0387
VIS*TON	2	62767100.747	3.92	0.0326
VIS*DMPG	1	888473.395	0.11	0.7418

Dependent Variable: Y

Source	DF	Type IV SS	F Value	Pr > F
VIS	1	29745956.016	3.71	0.0650
DMPG	1	77188149.634	9.63	0.0046
TON	2	130842909.196	8.16	0.0018
DMPG*TON	2	59202705.613	3.69	0.0387
VIS*TON	2	62767100.747	3.92	0.0326
VIS*DMPG	1	888473.395	0.11	0.7418

```

PROC GLM;
CLASS VIS DMPG TON;
MODEL Y = VIS DMPG TON DMPG*TON TON*VIS VIS*DMPG/SS1 SS2 SS3
SS4;
MEANS VIS DMPG TON/SCHEFFE;
MEANS VIS DMPG TON/LSD;
RUN;
PROC RSREG;
MODEL Y = VIS DMPG TON / PREDICT;
RIDGE MAX;
RUN;
DATA A; DMPG = 1;
VIS = -20;
DO I = 1 TO 5;
TON = 0;
VIS = VIS+20;
DO J = 1 TO 6;
TON = TON + 100;
Y = 7610.88 -17.01*VIS + 13237*DMPG - 39.68*TON + 16.76*DMPG*VIS
- 0.034*TON*VIS - 24.07*TON*DMPG + 0.058*TON*TON;
OUTPUT;
END;
END;
PROC PRINT;

```

Coding Coefficients for the Independent Variables

Factor	Subtracted off	Divided by
VIS	38.500000	37.500000
DMPG	0.250000	0.250000
TON	350.000000	250.000000

Response Surface for Variable Y

Response Mean	3866.841889
Root MSE	3120.984519
R-Square	0.5207
Coef. of Variation	80.7115

Degrees of Type I Sum

Regression	Freedom	of Squares	R-Square	F-Ratio	Prob > F
Linear	3	137385648	0.2415	4.702	0.0088
Quadratic	1	100391366	0.1764	10.307	0.0033
Crossproduct	3	58457705	0.1027	2.000	0.1368
Total Regress	7	296234720	0.5207	4.345	0.0023

Parameter	Degrees	Parameter Estimate	Standard Error	T for H0: Parameter=0
	of Freedom			
INTERCEPT	1	7610.872353	2403.668564	3.166
VIS	1	-17.017080	29.852014	-0.570
DMPG	1	13237	4503.851317	2.939
TON	1	-39.680564	13.970208	-2.840
VIS*VIS	0	0	.	.
DMPG*VIS	1	16.757132	55.484169	0.302
DMPG*DMPG	0	0	.	.
TON*VIS	1	-0.034236	0.067505	-0.507
TON*DMPG	1	-24.075277	10.125811	-2.378
TON*TON	1	0.059433	0.018513	3.210

Parameter Estimate from Coded		
Parameter	Prob >  T	Data
INTERCEPT	0.0037	1250.697729
VIS	0.5732	-930.395314
DMPG	0.0065	1363.966125
TON	0.0083	-1353.585583
VIS*VIS	.	0
DMPG*VIS	0.7649	157.098111
DMPG*DMPG	.	0

Parameter Estimate from Coded

Parameter	Prob >  T	Data
TON*VIS	0.6160	-320.966371
TON*DMPG	0.0245	-1504.704792
TON*TON	0.0033	3714.566354

Factor	Degrees		Mean Square	F-Ratio	Prob > F
	Freedom	Sum of Squares			
VIS	3	33139861	11046620	1.134	0.3523
DMPG	3	133140423	44380141	4.556	0.0101
TON	4	188412141	47103035	4.836	0.0043

Canonical Analysis of Response Surface

(based on coded data)

Critical Value

Factor	Coded	Uncoded
VIS	47.250043	1810.376607
DMPG	17.853221	4.713305
TON	5.839590	1809.897426

Coded	Uncoded Factor Values		
Radius	VIS	DMPG	TON
0.0	38.500000	0.250000	350.000000
0.1	37.221271	0.263987	331.113469
0.2	36.525396	0.274531	308.467457
0.3	36.167287	0.282944	284.441931
0.4	35.990575	0.290160	259.959596
0.5	35.917028	0.296676	235.327136

Coded	Uncoded Factor Values		
Radius	VIS	DMPG	TON
0.6	35.906673	0.302752	210.650463
0.7	35.937585	0.308538	185.968059
0.8	35.996872	0.314121	161.293960
0.9	36.076490	0.319558	136.632759
1.0	36.171171	0.324886	111.985288

**SIMULATED DATA FOR RESPONSE SURFACE ANALYSIS**

OBS	DMPG	VIS	I	TON	J	Y
1	0.5	0	1	100	1	9637.88
2	0.5	0	1	200	2	6206.38
3	0.5	0	1	300	3	3934.88
4	0.5	0	1	400	4	2823.38
5	0.5	0	1	500	5	2871.88
6	0.5	0	1	600	6	4080.38
7	0.5	20	2	100	1	9397.28
8	0.5	20	2	200	2	5897.78
9	0.5	20	2	300	3	3558.28
10	0.5	20	2	400	4	2378.78
11	0.5	20	2	500	5	2359.28
12	0.5	20	2	600	6	3499.78
13	0.5	40	3	100	1	9156.68
14	0.5	40	3	200	2	5589.18
15	0.5	40	3	300	3	3181.68
16	0.5	40	3	400	4	1934.18
17	0.5	40	3	500	5	1846.68
18	0.5	40	3	600	6	2919.18
19	0.5	60	4	100	1	8916.08
20	0.5	60	4	200	2	5280.58

SIMULATED DATA FOR RESPONSE SURFACE ANALYSIS

OBS	DMPG	VIS	I	TON	J	Y
21	0.5	60	4	300	3	2805.08
22	0.5	60	4	400	4	1489.58
23	0.5	60	4	500	5	1334.08
24	0.5	60	4	600	6	2338.58
25	0.5	80	5	100	1	8675.48
26	0.5	80	5	200	2	4971.98
27	0.5	80	5	300	3	2428.48
28	0.5	80	5	400	4	1044.98
29	0.5	80	5	500	5	821.48
30	0.5	80	5	600	6	1757.98

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