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## North Atlantic Deep Water formation inhibits high Arctic contamination by continental perfluorooctane sulfonate discharges

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1 **North Atlantic Deep Water Formation Inhibits High Arctic Contamination by**  
2 **Continental Perfluorooctane Sulfonate (PFOS) Discharges**

3

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14 Key points:

- 15 • Perfluorooctane sulfonate (PFOS) shows promise as a new ocean tracer but is also a  
16 potent toxicant
- 17 • PFOS concentrations in the upper North Atlantic Ocean peaked between 2002 and 2005  
18 but are still increasing in the deep ocean.
- 19 • Sinking, convection and southward transport of subpolar surface waters greatly reduces  
20 PFOS inputs to the high Arctic

## 21 **Abstract**

22 Perfluorooctane sulfonate (PFOS) is an aliphatic fluorinated compound with eight carbon atoms  
23 that is extremely persistent in the environment and can adversely affect human and ecological  
24 health. The stability, low reactivity, and high water solubility of PFOS combined with the North  
25 American phase-out in production around the year 2000, make it a potentially useful new tracer  
26 for ocean circulation. Here we characterize processes affecting the lifetime and accumulation of  
27 PFOS in the North Atlantic Ocean and transport to sensitive Arctic regions by developing a 3-D  
28 simulation within the MITgcm. The model captures variability in measurements across  
29 biogeographical provinces ( $R^2 = 0.90$ ,  $p=0.01$ ). In 2015, the North Atlantic PFOS reservoir was  
30 equivalent to 60% of cumulative inputs from the North American and European continents (1400  
31 Mg). Cumulative inputs to the Arctic accounted for 30% of continental discharges, while the  
32 remaining 10% was transported to the tropical Atlantic and other regions. PFOS concentrations  
33 declined rapidly after 2002 in the surface mixed-layer (half-life: 1-2 years) but are still increasing  
34 below 1000 m depth. During peak production years (1980-2000), plumes of PFOS enriched  
35 seawater were transported to the Subarctic in energetic surface ocean currents. However, Atlantic  
36 Meridional Overturning Circulation (AMOC) and deep ocean transport returned a substantial  
37 fraction of this northward transport (20%, 530 Mg) to southern latitudes and reduced cumulative  
38 inputs to the Arctic (730 Mg) by 70%. Weakened AMOC due to climate change is thus likely to  
39 increase the magnitude of persistent bioaccumulative pollutants entering the Arctic Ocean.

40

41 Index terms: Biogeochemical cycles, processes, and modeling

## 42 1. Introduction

43 Perfluorooctane sulfonate (PFOS) is an aliphatic organofluorine molecule containing  
44 eight carbon atoms, a highly fluorinated “tail” and a hydrophilic sulfonate “head” group. It has  
45 been transported in wastewater and rivers to the global oceans as a terminal sink after use in  
46 furniture and carpet coatings, food packaging, non-stick cookware, outdoor clothing/gear, and  
47 industrial applications since the 1950s [Wang *et al.*, 2017]. Biomonitoring studies have revealed  
48 high PFOS concentrations in many marine food webs and exposures pose risks to both human  
49 and ecological health [Tomy *et al.*, 2004; Dassuncao *et al.*, 2017]. PFOS does not appreciably  
50 degrade under environmental conditions due to the strength of carbon-fluorine bonds [Wang *et al.*, 2015]. It is not volatile and has much higher water solubility and lower affinity for  
51 partitioning to particles than other hydrophobic organic pollutants such as polychlorinated  
52 biphenyls (PCBs) [Muir and Lohmann, 2013; Zareitalabad *et al.*, 2013; Rayne *et al.*, 2009].  
53 Thus, Yamashita *et al.* [2008] proposed it as a novel tracer for ocean circulation. However, little  
54 is known about the distribution and accumulation of PFOS in the global oceans. Here we  
55 investigate processes driving the lifetime and accumulation of PFOS in the North Atlantic Ocean  
56 and its transport to vulnerable Arctic regions by developing a new simulation within a 3-D ocean  
57 circulation model (MITgcm).  
58

59 Globally, chemical manufacturing by the 3M Company in Decatur, Alabama, USA  
60 accounted for most (~80%) of the 66,000 tonnes of PFOS produced between 1958-2000  
61 [Armitage *et al.*, 2009a]. 3M voluntarily discontinued manufacturing of the PFOS parent  
62 chemical between 2000-2002, amid concerns about human and ecological health impacts [Land  
63 *et al.*, 2015]. Some PFOS has been produced in Asia after 2000 but most chemical  
64 manufacturing has shifted to other fluorinated compounds with different carbon chain lengths

65 and head groups in the larger family of poly- and perfluoroalkyl substances (PFASs) [*Wang et*  
66 *al.*, 2017]. The phase out in North American PFOS production adds to its potential utility as a  
67 tracer for ocean water mass ages and circulation.

68 On the continental scale, wastewater is thought to account for the majority of PFOS  
69 releases (~85%) to the environment [*Buser and Morf*, 2009; *Earnshaw et al.*, 2014].  
70 Atmospheric emissions of unstable PFOS precursors can also occur during chemical  
71 manufacturing but are estimated to account for only 1% of total inputs to the ocean [*Armitage et*  
72 *al.*, 2009a]. How changes in PFOS production have affected continental discharges and  
73 subsequent accumulation in the ocean is poorly understood. Prior work suggests vertical  
74 diffusion and sorption to settling particles are important for oceanic transport of PFASs  
75 [*Lohmann et al.*, 2013; *Sanchez-Vidal et al.*, 2015]. Seawater PFOS measurements are limited  
76 and modeling studies have not previously resolved basin-scale inputs and circulation [*Armitage*  
77 *et al.*, 2009a; *Schenker et al.*, 2008]. The only prior ocean model for PFOS contained 18°  
78 latitudinal bands for the globe and did not distinguish the Atlantic and Pacific basins [*Armitage*  
79 *et al.*, 2009a]. Previous modeling studies for other PFASs similarly do not consider spatial  
80 variability in oceanic concentrations and transport or lack detailed release estimates for  
81 continental discharges [*Armitage et al.*, 2009b; *Wania et al.*, 2007; *Stemmler and Lammel*, 2010].

82 The main objectives of this work are to: 1) quantify the magnitude and spatial distribution  
83 of continental PFOS discharges to the North Atlantic Ocean, and 2) better characterize the roles  
84 of lateral and vertical transport processes for the distribution and lifetime of PFOS in the marine  
85 water column. We develop an inventory of cumulative continental PFOS releases from  
86 wastewater and rivers in North America and Europe between 1958-2010 and use it to force a  
87 new 3-D simulation for the North Atlantic Ocean within the MITgcm [*Marshall et al.*, 1997]. We

88 evaluate the model using available seawater PFOS measurements and apply it to investigate how  
89 changing releases have affected spatial and temporal variability in concentrations and inputs to  
90 the Arctic Ocean. This information is needed to further assess the potential viability of PFOS as  
91 a tracer for oceanic transport processes as well as its significance as a toxicant in marine food  
92 webs since biological exposures are driven by seawater concentrations.

## 93 **2. Model description**

94 The MITgcm is an ocean general circulation model (<http://mitgcm.org>) that contains a  
95 biogeochemical and ecological model for cycling of carbon, nitrogen, silica, phosphorus and iron  
96 through inorganic, living plankton (phyto and zoo), and dead organic matter [*Dutkiewicz et al.*,  
97 2012]. It has a horizontal resolution of  $1^\circ \times 1^\circ$  and includes 23 vertical levels. Ocean state  
98 estimates are based on the Estimating Circulation and Climate of the Ocean (ECCO-GODAE)  
99 data product [*Marshall et al.*, 1997; *Wunsch and Heimbach*, 2007]. We added tracers for the  
100 dissolved and particle-bound forms of PFOS that are transported laterally and vertically by  
101 oceanic circulation, mixing and particle settling.

102 PFOS has a  $pK_a$  of negative three and is extremely stable in solution [*Campbell et al.*,  
103 2009; *Cheng et al.*, 2009; *Vecitis et al.*, 2008]. Partitioning between the aqueous phase and  
104 suspended particles is simulated using an empirically derived organic carbon normalized  
105 partition coefficient ( $\log K_{OC} = 2.6$ ) [*Higgins and Luthy*, 2006]. Particle concentrations,  
106 composition and export fluxes are based on the ecology simulation embedded in the MITgcm  
107 [*Dutkiewicz et al.*, 2012] and used to simulate vertical PFOS transport associated with settling  
108 particles.

109 The model is initialized by assuming zero oceanic PFOS concentrations before 1958, the  
110 onset of chemical production. For the years 1958 to 2010, the model is forced with  $1^\circ \times 1^\circ$



111 gridded inputs to the North Atlantic Ocean between 20°N and 60°N from rivers and wastewater  
112 treatment plants in North America and Europe using the inventory developed here. The  
113 simulation is continued with zero inputs between 2010-2038 to capture the phase-out in PFOS  
114 production. We divide annual inputs equally over each 3-hour time step in the model simulation  
115 because available monitoring data do not show significant seasonal differences (Supporting  
116 Information (SI), Table S1 [*Boulanger et al.*, 2005; *Schultz et al.*, 2006; *Loganathan, et al.*,  
117 2007; *Plumlee et al.*, 2008; *Erickson*, 2009; *Furl et al.*, 2011; *Furdui et al.*, 2008; *D'eon et al.*,  
118 2009; *Guerra et al.*, 2014]).

### 119 3. Continental PFOS release inventory (1958-2010)

120 We developed a time dependent inventory of PFOS inputs from wastewater treatment  
121 plants and rivers to the North Atlantic between 1958 and 2010. We neglect PFOS originating  
122 from degradation of atmospheric precursors because atmospheric inputs to the ocean are thought  
123 to be small [*Armitage et al.*, 2009a].

124 PFOS inputs from European rivers to the North Atlantic (Figure S1) are based on  
125 monitoring data for 2006-2007 and the relationship derived by *Pistocchi and Loos* [2009]  
126 between measured PFOS concentrations in river water ( $E_R$ , kg a<sup>-1</sup>) and population within a  
127 watershed basin ( $P_B$ ):

$$128 \quad E_R = 9.6 \times 10^{-9} P_B^{1.0115} \quad (\text{Equation 3})$$

129 To estimate temporal changes in loading, we scaled per-capita releases calculated from gridded  
130 European population data [*CIESIN*, 2005] by relative changes in annual production of the parent  
131 chemical to PFOS (perfluorooctane sulfonyl fluoride: POSF) for all years between 1958 and  
132 2010 [*Armitage et al.*, 2009a; *Paul et al.*, 2009; *Zhang et al.*, 2012].

133 Comprehensive riverine PFOS monitoring data are not available for North America for  
134 any year. Thus, we compiled PFOS wastewater monitoring data (Table S1, Figure S2) from 19  
135 Canadian facilities between 2009-2010 [Guerra *et al.*, 2014] and 36 U.S. facilities in 2007  
136 [Erickson *et al.*, 2009]. Following Pistocchi and Loos [2009], we related PFOS releases ( $E_w$ ,  $\mu\text{g}$   
137  $\text{d}^{-1}$ ) to population served ( $P_w$ ) using power regression analysis:

$$138 \quad E_w = \beta P_w^\alpha \quad (\text{Equation 4})$$

139 where,  $\beta$  represents the baseline per-capita PFOS discharge rate, which is equal to 0.21 for  
140 Canada and 0.47 for the U.S. (Figure S3). The parameter  $\alpha$  is a scaling factor representing a non-  
141 linear increase in PFOS releases with population served, equal to  $1.24 \pm 0.09$  for Canada and  
142  $1.21 \pm 0.10$  for the U.S (Figure S3).

143 Only 9.9% of the municipalities in Canada [Environment Canada, 2009] and 1.8% of  
144 wastewater treatment plants in the U.S. [U.S. EPA, 2008] discharge directly into the North  
145 Atlantic (Figure S4, Table S2). To calculate wastewater PFOS inputs to river basins that enter  
146 the North Atlantic Ocean (Figure S5, Table S3), we summed PFOS discharges derived from the  
147 total populations served by wastewater treatment facilities within each catchment using ArcGIS  
148 [Environment Canada, 2009; Global Runoff Data Centre, 2007; U.S. EPA, 2008]. We also  
149 included direct releases (360 Mg during 1958-2002) from the 3M Decatur (Alabama) facility to  
150 the Mississippi watershed that accounted for a large fraction of global production in 2002  
151 [Armitage *et al.*, 2009a; Filipovic *et al.*, 2013; Scott *et al.*, 2009; 3M, 2000]. We neglected other  
152 environmental release pathways such as leaching from landfills because they comprise <10% of  
153 total releases on the continental scale, even though they can be important for inland rivers and  
154 streams [*e.g.*, Zhang *et al.*, 2016]. We derived basin wide releases ( $\mu\text{g person}^{-1} \text{ a}^{-1}$ ) based on  
155 wastewater treatment plants within each catchment area [CIESN, 2005]. Temporal changes in

156 global historical inventories [Armitage *et al.*, 2014a; Wang *et al.*, 2014a] were used to scale  
157 population-based releases over the full history of production.

158         There are no data available for riverine or wastewater PFOS discharges in North Africa,  
159 Mexico and Central America. We assumed population-based PFOS emission factors were  
160 approximately one order of magnitude lower than those in Northern Europe based on wastewater  
161 monitoring data (Figure S6) from other developing areas [Filipovic *et al.* 2013; Xie *et al.* 2013].  
162 We scaled temporal changes in population based inputs using global historical release  
163 inventories in the same manner as other regions [Armitage *et al.*, 2014a; Wang *et al.*, 2014a].

164         We assumed all riverine PFOS will enter the ocean within one year except for systems  
165 where flow is mediated by the presence of a large inland water body (i.e., the St. Lawrence  
166 River/Lake Ontario, the Nelson River/Lake Winnipeg, and the Danish Straits/Baltic Sea). We  
167 estimated transport times for these systems by calibrating the lag time in a one-box model to  
168 measured PFOS concentrations leaving these systems, given inputs estimated from this work  
169 (Table S4). Resulting lag times ranged between 1.5-2.5 years. We did not use the residence time  
170 for each of these water bodies because we do not expect PFOS to be well mixed.

### 171 *3.1 Uncertainty in release estimates*

172         We estimate a 5-fold uncertainty in continental PFOS discharges presented here based on  
173 the 95% confidence intervals of regression relationships for the U.S. and Canada (Figure S3).  
174 We propagate this uncertainty for evaluation of model results and use available measurements to  
175 constrain inputs. Uncertainty estimated here is slightly lower than the approximately one order of  
176 magnitude uncertainty for other PFAS inventories. In prior work, uncertainty estimates reflected  
177 limited information on the magnitude of PFASs emissions from products containing these  
178 compounds, assumptions about yields of atmospheric precursors, and ranges in industrial release

179 estimates [Prevedorous *et al.*, 2006, Paul *et al.*, 2009, Armitage *et al.*, 2009a; 2009b; Wang *et*  
180 *al.*, 2014a; 2014b].

#### 181 **4. Model evaluation**

182 The median discharge scenario based on wastewater and riverine monitoring data  
183 suggests that  $2.7 \times 10^3$  Mg PFOS has cumulatively entered the North Atlantic since 1958 (low  
184 bound:  $5.3 \times 10^2$  Mg, high bound  $1.3 \times 10^4$  Mg, Figure 1). North American coastal releases  
185 accounted for 59% of total releases and the remaining 41% was from Europe (Figure 1). Inputs  
186 to the North Atlantic peaked in the year 2000 and declined to negligible amounts by 2010  
187 (Figure 1). For the median release scenario, an additional  $1.5 \times 10^3$  Mg of PFOS flowed into the  
188 North Atlantic indirectly from the Gulf of Mexico and Mediterranean Sea.

189 Direct release values from two previous PFOS inventories based on chemical production  
190 and product use are comparable to the low and median bounds for this work. Armitage *et al.*  
191 [2009a] suggested cumulative direct global releases to the ocean of  $2.8 \times 10^2$  to  $2.3 \times 10^3$  Mg  
192 between the years 1957-2002. Most releases were concentrated in North America and Europe  
193 between 36-54°N [Armitage *et al.*, 2009]. Paul *et al.* [2009] similarly estimated total global  
194 historical PFOS emissions to air and water to be  $4.5 \times 10^2$  to  $2.7 \times 10^3$  Mg between 1970-2002.  
195 Indirect (atmospheric) PFOS inputs to the global oceans are thought to be smaller (5-230 Mg)  
196 based on work by Armitage *et al.* [2009a]. We find the high estimate of PFOS releases derived  
197 here is implausible based on these global production constraints [Armitage *et al.*, 2009; Paul *et*  
198 *al.*, 2009].

199 Figure 2a compares modeled surface seawater (10 m) concentrations for the year 2010  
200 based on the median scenario for PFOS releases (Figure 1) with observed surface seawater  
201 concentrations between 2009 and 2011. Relatively high concentrations are apparent next to

202 continental source regions in eastern North America and the North Sea and relatively lower  
203 levels are observed in the eastern subtropical region (Sargasso Sea). We divided the North  
204 Atlantic Ocean into the biogeographical provinces characterized by *Longhurst* [2007] to assess  
205 performance of the model in different regions. We do not consider coastal monitoring data  
206 because the model is not intended to capture the fine-scale circulation and dynamics of inland  
207 seas and estuaries.

208 Modeled seawater concentrations capture most of the variability in the observed data ( $R^2$   
209 = 0.90,  $p=0.01$ ) across biogeographical provinces (Figure 2b). The grey dashed line in Figure 2b  
210 indicates perfect agreement between measured and modeled results and the regression  
211 relationship indicates the model has a high mean bias of 75% relative to observations for all  
212 regions. Measured PFOS concentrations are lower in the western portion of the North Atlantic  
213 Subtropical Gyral Province (NASW) in the Sargasso Sea, and the southern North Atlantic  
214 Tropical Gyral Province (NATR), than those in the eastern North Atlantic Subtropical Gyral  
215 Province (NASE). The model successfully captures these differences. Only two surface water  
216 measurements ranging from  $<10$  pg L to  $59$  pg L<sup>-1</sup> are available for the Gulf Stream Province  
217 (GFST) [*Ahrens et al.*, 2010a; *Benskin et al.* 2012]. The model shows large ranges in PFOS  
218 concentrations in the Gulf Stream due to a plume of PFOS enriched seawater transported  
219 offshore (Figure 2a). Modeled seawater PFOS concentrations are elevated ( $>50$  pg L<sup>-1</sup>) in the  
220 North Atlantic Drift Province (NADR) and the subarctic regions below Greenland. Surface  
221 observations near the European shelf appear to support this spatial pattern (Figure 2).

222 *Yamashita et al.* [2008] collected five vertical profiles from the Labrador Sea and mid-  
223 Atlantic Ocean in 2004, which we compare to modeled seawater PFOS concentrations from the  
224 same regions and year. The median PFOS release scenario (solid red line in Figure 3) reasonably

225 captures observed surface enrichment at stations AO2, AO4 and AO5 and depletion at depth  
226 (Figure 3). However, the median emissions scenario overestimates PFOS concentrations  
227 throughout the vertical profiles (Figure 3), which is consistent with model bias identified for  
228 surface water concentrations (Figure 2b). The model does not capture some of the fine scale  
229 vertical variability at stations AO3-5 due to its coarse resolution ( $1^\circ \times 1^\circ$ ). The low release  
230 scenario from Figure 1 (dashed red line in Figure 3) better matches the observed magnitude of  
231 seawater PFOS concentrations in the Labrador Sea but underestimates those in the mid-Atlantic.

232 This analysis suggests a mean model bias correction of 75% based on Figure 2b is  
233 appropriate for all results. Therefore, cumulative PFOS inputs to the North Atlantic Ocean of  
234  $2.4 \times 10^3$  Mg are most consistent with available observational constraints from seawater  
235 measurements. Releases of this magnitude produce mean modeled surface water (10 m)  
236 concentrations of  $39 \pm 14$  pg L<sup>-1</sup> in 2010 compared to an observed value across all offshore  
237 sampling locations of  $43 \pm 21$  pg L<sup>-1</sup> between 2009 and 2011. Hereon, this bias corrected release  
238 scenario is used for all modeling and mass budget calculations.

## 239 **5. Model uncertainty and sensitivity analysis**

240 For all ocean inputs, we conducted sensitivity analyses allowing for additional lag times  
241 of 2, 5 and 10 years between PFOS distribution in products and releases from watersheds to the  
242 ocean (Figure S7). Some studies have suggested there may be a substantial lag between  
243 wastewater PFOS releases to aquatic systems and transport to marine areas due to slow transport  
244 in groundwater [Filipovic *et al.*, 2013]. We find introducing an additional lag exacerbates  
245 overestimates for recently measured surface seawater concentrations (Figure 2b, Figure 3). Thus,  
246 we retain the base assumption that most PFOS releases in wastewater will be transported to the  
247 ocean within one year.

248 We consider uncertainty in the empirically measured partition coefficient for PFOS using  
249 sensitivity analysis on the upper and lower bounds reported in the literature ( $\log K_{oc} = 2.4$  to  $3.8$ )  
250 [Ahrens *et al.*, 2011; Ferrey *et al.*, 2012]. Results of our simulation suggest the mass of vertically  
251 transported PFOS in the ocean associated with settling particles is negligible compared to that  
252 from lateral and vertical ocean circulation (Figure S8). Modeled concentrations of PFOS bound  
253 to particles (Figure S8) are consistent with limited measurements that were all  $<10 \text{ pg L}^{-1}$   
254 [Ahrens *et al.*, 2009a]. We examined the sensitivity of this finding to the range in partition  
255 coefficients reported in the literature. The maximum  $\log K_{oc}$  value reported in prior work is  $3.8$   
256 [Ahrens *et al.*, 2011], compared to the value of  $2.6$  [Higgins and Luthy, 2007] specified in our  
257 base simulation. This does not change our initial finding as the fraction of particle bound PFOS  
258 in seawater is negligible across expected ranges of  $K_{oc}$  and potential POC concentrations in  
259 seawater (Figure S9). Previous studies that considered PFOS fluxes associated with settling of  
260 suspended particles in isolation suggest this may be a substantial oceanic transport process  
261 [Sanchez-Vidal *et al.*, 2015]. Our model results indicate this is not the case for the North  
262 Atlantic.

## 263 **6. Spatial distribution of PFOS in seawater**

264 The major surface currents in the Atlantic Ocean result in predominantly eastern and  
265 northern transport of coastal PFOS pollution sources toward the Subarctic (Figure 4). Surface  
266 water PFOS concentrations in 1980 and 2000 varied by more than an order of magnitude in  
267 PFOS enriched seawater plumes from the St. Lawrence River, Florida Current, and North Sea  
268 (Figure 4a). The Gulf Stream transports PFOS entrained in the Florida Current into the central  
269 and eastern Atlantic Ocean, where it merges with the North Atlantic Current (NAC) and diverges  
270 both north and east [Pickard and Emery, 1990]. This is visible in Figure 4a/b by offshore dilution

271 and diffusion north and east of elevated PFOS concentrations in the Gulf Stream. These results  
272 help to explain the relatively lower PFOS concentrations measured in seawater from the western  
273 part of the North Atlantic Subtropical Gyral Province (NASW, Figure 2).

274 Transport patterns for PFOS enriched surface waters (Figure 2) imply the North Atlantic  
275 is a large source to the Arctic. North Atlantic seawater enters the Arctic at multiple depths. This  
276 circulation is critical for transfer of heat into polar regions and meridional overturning of the  
277 ocean [Askenov *et al.*, 2010]. PFOS discharges from the North Sea (Figure 4) are predominantly  
278 transported north into the Norwegian and Greenland Seas [Smith *et al.*, 2011]. Surface waters  
279 from this area enriched in PFOS become dense during winter cooling, sink and form the lower  
280 limb of the North Atlantic Deep Water (NADW) that flows south as part of global thermohaline  
281 circulation [Pickard and Emery, 1990]. The Norwegian and Greenland Sea regions supply an  
282 estimated 80% of the NADW [Harvey and Theodorou, 1986; Pickard and Emery, 1990].

283 Similarly, surface waters entrained in the Labrador Current that captures PFOS  
284 discharges from the Hudson Bay sink in “convective chimneys” from the center of the Labrador  
285 Gyre [Clarke and Gascard, 1983] and are transported south in NADW primarily in the western  
286 Atlantic in the Deep Western Boundary Current (DWBC) [Rhein *et al.*, 2010]. Apparent ages of  
287 NADW determined from tracers around 55°N are generally less than a decade, even at depths  
288 greater than 3500 m [Doney and Bullister, 1992; Rhein *et al.*, 2010]. This explains the  
289 penetration of PFOS to the ocean floor in this region (mean: 12 pg L<sup>-1</sup> below 1000 m) based on  
290 the vertical profiles measured by Yamashita *et al.* [2008] in the Labrador Sea in 2004 (Figure 3).  
291 For vertical profiles collected by Yamashita *et al.* [2008] between 25°N and 30°N (Figure 3),  
292 seawater below 1000 m also contained detectable but lower levels of PFOS (generally <5 pg L<sup>-1</sup>,  
293 Figure 3). These water depths primarily contain NADW transported south and may have some



294 influence from Mediterranean outflow [*Pickard and Emery, 1990*]. *Rhein et al.* [2015] reported  
295 the western Atlantic portion of NADW around these latitudes (in the Sargasso Sea) has an  
296 apparent age of around 30 years, which suggests an influence from surface inputs of PFOS in the  
297 early-mid 1970s.

## 298 **7. Temporal changes in PFOS concentrations and mass budgets for the North Atlantic**

299 Figure 5 shows the temporal evolution of PFOS concentrations in seawater from the  
300 surface mixed layer (0-10 m), subsurface (360-510 m), and near the permanent thermocline  
301 (985-1335 m). Concentrations between 0-10 m depth peak in 2001 (basin-wide median: 66 pg L<sup>-1</sup>)  
302 <sup>1</sup>), one year after maximum releases from rivers and wastewater. Spatial heterogeneity in surface  
303 water concentrations is reflected by the wide variability in 95<sup>th</sup> percentile confidence intervals  
304 (CI) shown in Figure 5 (2001: 9-633 pg L<sup>-1</sup>) that become more homogeneous over time (2015:  
305 22-53 pg L<sup>-1</sup>).

306 PFOS concentrations in subsurface seawater (365-510 m) are lower and less variable than  
307 surface waters due to dilution with lower concentration subsurface seawater (Figure 5). The lag  
308 time between peak releases (ca. 2000) and peak subsurface concentrations (ca. 2005, 41 pg L<sup>-1</sup>,  
309 95<sup>th</sup> CI: 27-66) at these depths is 2-3 years. This likely reflects strong winter convection in the  
310 subpolar North Atlantic, which results in mixing with subsurface waters up to ~600 m depth  
311 [*McCartney and Talley, 1982*]. By 2015, median modeled concentrations declined to 37 pg L<sup>-1</sup>  
312 and variability in ocean concentrations is much smaller than earlier years (95<sup>th</sup> CI: 30-47, Figure  
313 5).

314 A longer lag after peak releases is observed for peak PFOS concentrations in deeper  
315 waters around the permanent thermocline (985-1335 m) (Figure 5). Modeled concentrations  
316 plateau in 2040 around a median of 26 pg L<sup>-1</sup>. The effects of water mass circulation in different

317 regions are evident in Figure 5, where the 95<sup>th</sup> percentile concentrations, representing more  
318 recently ventilated water masses, peaked in 2009 at 47 pg L<sup>-1</sup> (a 9-year lag) and have declined  
319 since this time. By contrast, the 5<sup>th</sup> percentile of PFOS concentrations in the North Atlantic are  
320 continuing to slowly increase and are expected to reach 9 pg L<sup>-1</sup> in 2040 (lag time more than  
321 three decades). Similar patterns have been reported for other pollutants in the North Atlantic  
322 such as lead (Pb), due to time required for subduction and ventilation of surface seawater into the  
323 thermocline [Wu and Boyle, 1997].

324 Cumulative PFOS inputs to the North Atlantic between 1958 and 2015 estimated in this  
325 study are 2.4×10<sup>3</sup> Mg. The North Atlantic PFOS reservoir in 2015 was equivalent to  
326 approximately 60% of inputs in 2015 (1.4×10<sup>3</sup> Mg, Figure 6), with the remaining fractions lost  
327 to the Arctic Ocean (30%), tropical Atlantic and other regions (10%). Over time, the mass  
328 distribution of PFOS has shifted toward deeper waters from >80% in the upper ocean (top 1000  
329 m) in 2000 to approximately 50% in 2015, consistent with Figure 5. This has important  
330 implications for reducing biological exposures as most marine life is found in the upper 1000 m  
331 of the ocean [Ryther, 1969].

332 Lifetimes of PFOS against losses (reservoir/gross losses) from this work range from 1-3  
333 years (half-life:  $t_{1/2} = 1-2$  years) in the surface mixed layer of the North Atlantic to one to two  
334 decades in the deep ocean (Table 1). Loss processes are mainly internal circulation and outflow  
335 to the Arctic (>60°N), Mediterranean Sea, and the tropical Atlantic Ocean (<20°N) (Figure 6). By  
336 2015, 85% of total losses from the Atlantic Ocean had been transported into the Arctic, with  
337 smaller outflow to other regions. Basin-wide lifetimes are shorter in 2000 than 2015, reflecting  
338 the entrainment of PFOS source regions in energetic currents in the ocean such as the Gulf  
339 Stream (Figure 4). Gonzalez-Gaya et al. [2014] estimated a 5.8 year half-life in the North

340 Atlantic based on observations for another PFAS with the same number of carbon atoms. For  
341 seawater between 100-1000 m, modeled half-lives for PFOS in this study were approximately  
342 three years.

343 Variable lag-times and vertical differences in seawater concentrations (Table 1, Figure 5)  
344 help to explain mixed temporal trends in biological PFOS concentrations measured in the North  
345 Atlantic and Subarctic [Bossi *et al.*, 2005; Butt *et al.*, 2010; Dassuncao *et al.*, 2017; Houde *et al.*,  
346 2014]. Our results imply declining PFOS exposures for fish and marine mammals that  
347 predominantly forage in the ocean surface mixed layer since 2001. Temporal changes are  
348 expected to be more variable for complex food-webs with mixed prey items and biota that forage  
349 from deeper seawater such as North Atlantic pilot whales [*Globicephala melas*] that mainly  
350 consume squid between 400-700 m depth [Hoydal *et al.*, 1993; Li *et al.*, 2014]. A reported  
351 increase in PFOS concentrations in pilot whale muscle between 1987 to present is consistent  
352 with trends in mid-depth to thermocline waters shown in Figure 5 [Dassuncao *et al.*, 2017].

### 353 **9. Implications for PFOS inputs to the Arctic Ocean**

354 Figure 7 shows gross PFOS flows at differ seawater depths into and out of the Arctic  
355 (above 60°N) from the North Atlantic between 1958 and 2040. Most inputs to the Arctic Ocean  
356 are from mid-depth inflow (20-1000 m), which cumulatively accounted for  $1.1 \times 10^3$  Mg PFOS  
357 between 1958 and 2015 (surface inflow, 0-20 m = 60 Mg). Some of the North Atlantic inflow  
358 sank and mixed with deep water from the Arctic before traveling south, thereby reducing inputs  
359 to the high Arctic.

360 Circulation of PFOS transported beyond 60°N that returns south in deep water inflows  
361 (>1000 m depth, blue colors in Figure 7) cumulatively totaled  $4.1 \times 10^2$  Mg between 1958 and  
362 2015 (Figure 7). NADW formation in the Labrador and Greenland Seas (44-65°W, 50-65°N)

363 between 1958 and 2015 resulted in  $4.8 \times 10^2$  Mg PFOS transported below 1000 m. We find 25%  
364 of this flux sinks below 1000 m south of  $60^\circ\text{N}$ , resulting in a total removal of  $5.3 \times 10^2$  Mg PFOS  
365 associated with Atlantic meridional overturning circulation (AMOC). The annual downward flux  
366 from the surface (100 m) of the Labrador Sea in 2004 was  $4.1 \times 10^2$  kg, which is in the same  
367 range as previously estimated (620 kg) based on measurements [Yamashita *et al.*, 2008]. AMOC  
368 has thus dramatically reduced the amount of PFOS entering the high Arctic. Weakened  
369 convection in the Labrador Sea Region and AMOC due to warming polar temperatures and  
370 melting of the Greenland ice sheet has been widely reported [Fisher *et al.*, 2010; Rahmstorf *et*  
371 *al.*, 2015; Rhein *et al.*, 2015; Yashayaev, 2007]. These climate-driven changes thus have the  
372 potential to greatly increase the magnitude of bioaccumulative pollutants entering ecologically  
373 sensitive Arctic regions.

## 374 **10. Summary and Conclusions**

375 We developed a new 3-D simulation for the fate of cumulative PFOS inputs ( $2.4 \times 10^3$   
376 Mg) to the North Atlantic Ocean between 1958 to 2015. Results suggests concentrations in  
377 surface waters (upper 1000 m) reached maximum levels several years after peak global PFOS  
378 production (before 2005) but concentrations in the deep ocean will continue to rise until 2040.  
379 Modeled temporal changes in the North Atlantic (Figure 5) contrast previous work that  
380 suggested ocean concentrations will decline slowly in all PFOS source regions and will increase  
381 in remote regions for several decades [Armitage *et al.*, 2009a]. The abrupt phase out of most  
382 chemical manufacturing of PFOS around the year 2000, its high stability in seawater, and limited  
383 transport associated with settling of suspended particles in the ocean water column reinforce its  
384 potential as a novel tracer of ocean circulation. Recent declines in PFOS concentrations in the  
385 upper ocean are important for protecting marine food webs known to bioaccumulate this

386 deleterious compound [*Dassuncao et al.*, 2017; *Routti et al.*, 2016; *Houde et al.*, 2011; *Tomy et*  
387 *al.*, 2004].

388 In 2015, we estimate the North Atlantic (20-60°N) reservoir of PFOS to be  $1.4 \times 10^3$  Mg.  
389 Model results suggest  $7.3 \times 10^2$  Mg PFOS has entered the Arctic Ocean from the North Atlantic,  
390 and smaller amounts ( $2.9 \times 10^2$  Mg, Figure 6) have been transported to the tropical and southern  
391 Atlantic Ocean and other regions. When peak discharges were occurring, surface currents in the  
392 Atlantic Ocean transported PFOS from continental source regions to the Subarctic, explaining  
393 the large Arctic inputs. However, a large fraction of the PFOS that reached high latitudes became  
394 part of the NADW and was transported south in global thermohaline circulation. We estimate  
395 AMOC has cumulatively prevented  $5.3 \times 10^2$  Mg PFOS from entering the high Arctic since the  
396 onset of its production in 1958. Weakened AMOC in an era of rapidly changing climate is thus  
397 expected to increase the burden of persistent, bioaccumulative contaminants entering  
398 ecologically sensitive Arctic regions.

399

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410

## 411 12. References

- 412 3M (2000), Sulfonated perfluorochemicals: life-cycle waste stream estimates—Final report for  
413 3M specialty materials, 1063-1072.  
414 <http://www.chemicalindustryarchives.org/dirtysecrets/scotchgard/pdfs/1226-0681.pdf>  
415 (accessed February 2014).
- 416 Ahrens, L. (2011), Polyfluoroalkyl compounds in the aquatic environment: a review of their  
417 occurrence and fate, *J Environ Monitor*, 13(1), 20-31.
- 418 Ahrens, L., J. L. Barber, Z. Y. Xie, and R. Ebinghaus (2009a), Longitudinal and Latitudinal  
419 Distribution of Perfluoroalkyl Compounds in the Surface Water of the Atlantic Ocean,  
420 *Environ Sci Technol*, 43(9), 3122-3127.
- 421 Ahrens, L., S. Felizeter, and R. Ebinghaus (2009b), Spatial distribution of polyfluoroalkyl  
422 compounds in seawater of the German Bight, *Chemosphere*, 76(2), 179-184.
- 423 Ahrens, L., W. Gerwinski, N. Theobald, and R. Ebinghaus (2010a), Sources of polyfluoroalkyl  
424 compounds in the North Sea, Baltic Sea and Norwegian Sea: Evidence from their spatial  
425 distribution in surface water, *Mar Pollut Bull*, 60(2), 255-260.
- 426 Ahrens, L., Z. Y. Xie, and R. Ebinghaus (2010b), Distribution of perfluoroalkyl compounds in  
427 seawater from Northern Europe, Atlantic Ocean, and Southern Ocean, *Chemosphere*,  
428 78(8), 1011-1016.
- 429 Ahrens, L., L. W. Y. Yeung, S. Taniyasu, P. K. S. Lam, and N. Yamashita (2011), Partitioning of  
430 perfluorooctanoate (PFOA), perfluorooctane sulfonate (PFOS) and perfluorooctane  
431 sulfonamide (PFOSA) between water and sediment, *Chemosphere*, 85(5), 731-737.
- 432 Armitage, J. M., U. Schenker, M. Scheringer, J. W. Martin, M. Macleod, and I. T. Cousins  
433 (2009a), Modeling the global fate and transport of perfluorooctane sulfonate (PFOS) and  
434 precursor compounds in relation to temporal trends in wildlife exposure, *Environ Sci  
435 Technol*, 43(24), 9274-9280.
- 436 Armitage, J. M., M. MacLeod, and I. T. Cousins (2009b), Modeling the Global Fate and  
437 Transport of Perfluorooctanoic Acid (PFOA) and Perfluorooctanoate (PFO) Emitted from  
438 Direct Sources Using a Multispecies Mass Balance Model, *Environ Sci Technol*, 43(4),  
439 1134-1140.
- 440 Askenov, Y., S. Bacon, A. C. Coward, and A. J. G. Nurser (2010), The North Atlantic inflow to  
441 the Arctic Ocean: High resolution model study, *J Marine Syst*, 79, 1-22.
- 442 Benskin, J. P., D. C. G. Muir, B. F. Scott, C. Spencer, A. O. De Silva, H. Kylin, J. W. Martin, A.  
443 Morris, R. Lohmann, G. Tomy, B. Rosenberg, S. Taniyasu, and N. Yamashita (2012),  
444 Perfluoroalkyl acids in the Atlantic and Canadian Arctic Oceans, *Environ Sci Technol*,  
445 46, 5815-5823.
- 446 Boulanger, B., J. Vargo, J. L. Schnoor and K. C. Hornbuckle (2004) Evaluation of  
447 perfluorooctane surfactants in a wastewater treatment system and in a commercial surface  
448 protection product. *Environ Sci Technol*, 39, 5524-5530.

449 Buser, A., and L. Morf (2009), Substance Flow Analysis of PFOS and PFOA, in *Perfluorinated*  
450 *surfactants perfluorooctanesulfonate (PFOS) and perfluorooctanoic acid (PFOA) in*  
451 *Switzerland*. Federal Office for the Environment (FOEN), Berne, Switzerland.

452 Butt, C. M., U. Berger, R. Bossi, and G. T. Tomy (2010), Levels and trends of poly- and  
453 perfluorinated compounds in the arctic environment, *Sci Total Environ*, 408, 2936-2965.

454 Campbell, T. Y., C. D. Vecitis, B. T. Mader, and M. R. Hoffmann (2009), Perfluorinated  
455 surfactant chain-length effects on sonochemical kinetics, *J Phys Chem A*, 113(36), 9834-  
456 9842.

457 CIESIN - Center for International Earth Science Information Network - Columbia University,  
458 United Nations Food Agriculture Programme - FAO, and Centro Internacional de  
459 Agricultura Tropical - CIAT (2005), Gridded Population of the World, Version 3  
460 (GPWv3): Population Count Grid, Future Estimates, edited, NASA Socioeconomic Data  
461 and Applications Center (SEDAC), Palisades, NY.

462 Clarke, R. A., and J. Gascard (1983), The formation of the Labrador Sea Water, part 1, Large-  
463 scale processes, *J Phys Oceanog*, 13, 1764-1778.

464 Dassuncao, C., X. C. Hu, X. Zhang, R. Bossi, M. Dam, B. Mikkelsen, and E. M. Sunderland  
465 (2017), Temporal shifts in poly- and perfluoroalkyl substances (PFASs) in North Atlantic  
466 pilot whales indicate large contribution of atmospheric precursors, *Environ Sci Technol*,  
467 51(8), 4512-4521.

468 D'eon, J. C., P. W. Crozier, V. I. Furdui, E. J. Reiner, E. L. Libelo and S. A. Mabury (2009),  
469 Perfluorinated phosphonic acids in Canadian surface waters and wastewater treatment  
470 plant effluent: Discovery of a new class of perfluorinated acids. *Environ Toxicol Chem*,  
471 28, 2101-2107.

472 Doney, S. C., and J. L. Bullister (1992), A chlorofluorocarbon section in the eastern North  
473 Atlantic, *Deep-Sea Res* 39(11/12), 1857-1883.

474 Dutkiewicz, S., M. J. Follows, and J. G. Bragg (2009), Modeling the coupling of ocean ecology  
475 and biogeochemistry, *Global Biogeochem Cy*, 23.

476 Earnshaw, M. R., A. G. Paul, R. Loos, S. Tavazzi, B. Paracchini, M. Scheringer, K.  
477 Hungerbuhler, K. C. Jones, and A. J. Sweetman (2014), Comparing measured and  
478 modelled PFOS concentrations in a UK freshwater catchment and estimating emission  
479 rates, *Environ Int*, 70C, 25-31.

480 Environment Canada (2009), Municipal Water and Wastewater Survey, [http://www.ec.gc.ca/eau-  
481 water/default.asp?lang=En&n=ED7C2D33-31](http://www.ec.gc.ca/eau-water/default.asp?lang=En&n=ED7C2D33-31) (accessed March 2014).

482 Erickson, M., M. Ferrey, P. Hoff, L. Solem, and S. Streets (2009), PFCs in Minnesota's Ambient  
483 Environment: 2008 Progress Report. Minnesota Pollution Control Agency.  
484 <http://www.pca.state.mn.us/index.php/view-document.html?gid=2855> (accessed February  
485 2014).

486 Ferrey, M. L., J. T. Wilson, C. Adair, C. M. Su, D. D. Fine, X. Y. Liu, and J. W. Washington  
487 (2012), Behavior and Fate of PFOA and PFOS in Sandy Aquifer Sediment, *Ground  
488 Water Monit R*, 32(4), 63-71.

489 Filipovic, M., U. Berger, and M. S. McLachlan (2013), Mass balance of perfluoroalkyl acids in  
490 the Baltic Sea, *Environ Sci Technol*, 47(9), 4088-4095.

491 Furdui, V. I., P. W. Crozier, E. J. Reiner, and S. A. Mabury (2008), Trace level determination of  
492 perfluorinated compounds in water by direct injection, *Chemosphere*, 73(1), S24-S30.

493 Furl, C. V., C. A. Meredith, M. J. Strynar, and S. F. Nakayama (2011), Relative importance of  
494 wastewater treatment plants and non-point sources of perfluorinated compounds to  
495 Washington State rivers, *Sci Total Environ*, 409(15), 2902-2907.

496 Fisher, J., M. Visbeck, R. Zantopp, and N. Nunes (2010), Interannual to decadal variability of  
497 outflow from the Labrador Sea, *Geophys Res Lett*, 37, L24610.

498 Global Runoff Data Centre (2007), Major River Basins of the World, The Global Runoff Data  
499 Centre, D-56002 Koblenz, Germany. <http://www.bafg.de/GRDC/> (accessed April 2014).

500 Gonzalez-Gaya, B., J. Dachs, J. Roscales, G. Caballero, and B. Jimenez (2014), Perfluorinated  
501 substances in the global tropical and subtropical surface oceans, *Environ Sci Technol*, 48,  
502 13076-13084.

503 Guerra, P., M. Kim, L. Kinsman, T. Ng, M. Alaei, and S. A. Smyth (2014), Parameters affecting  
504 the formation of perfluoroalkyl acids during wastewater treatment, *J Hazard Mater*, 272,  
505 148-154.

506 Harvey, J., and A. Theodorou (1986), The circulation of the Norwegian overflow water in the  
507 North Atlantic *Oceanol Acta*, 9, 393-402.

508 Higgins, C. P., and R. G. Luthy (2006), Sorption of perfluorinated surfactants on sediments,  
509 *Environ Sci Technol*, 40(23), 7251-7256.

510 Houde, M., A. O. De Silva, D. C. G. Muir, and R. J. Letcher (2011), Monitoring of  
511 perfluorinated compounds in aquatic biota: An updated review, *Environ Sci Technol*, 45,  
512 7962-7973.

513 Hoydal, K., and L. Lastein (1993), Analysis of Faroese catches of pilot whales (1709–1992), in  
514 relation to environmental variations. , in *Biology of northern hemisphere pilot whales*,  
515 edited by G. P. Donovan, C. H. Lockyer and A. R. Martin, p. 89–106, International  
516 Whaling Commission, Cambridge, U.K.

517 Land, M., C. A. de Wit, I. T. Cousins, D. Herzke, J. Johansson, and J. W. Martin (2015), What is  
518 the effect of phasing out long-chain per- and polyfluoroalkyl substances on the  
519 concentrations of perfluoroalkyl acids and their precursors in the environment? A  
520 systematic review protocol, *Environ Evidence*, 4, 3.

521 Li, M., L. S. Sherman, J. D. Blum, P. Grandjean, B. Mikkelsen, P. Weihe, E. M. Sunderland, and  
522 J. P. Shine (2015), Assessing sources of human methylmercury exposure using stable  
523 mercury isotopes, *Environ Sci Technol*, 48, 8800-8806.

524 Loganathan, B. G., K. S. Sajwan, E. Sinclair, K. S. Kumar, and K. Kannan (2007),  
525 Perfluoroalkyl sulfonates and perfluorocarboxylates in two wastewater treatment  
526 facilities in Kentucky and Georgia, *Water Res.*, 41(20), 4611-4620.

527 Longhurst, A. R. (2007), *Ecological geography of the sea, 2nd Ed.*, Academic Press, Elsevier,  
528 San Diego, CA.

529 Marshall, J., C. Hill, L. Perelman, and A. Adcroft (1997), Hydrostatic, quasi-hydrostatic, and  
530 nonhydrostatic ocean modeling, *J Geophys Res-Oceans*, 102(C3), 5733-5752.

531 McCartney, M., and L. D. Talley (1982), The subpolar model water of the North Atlantic Ocean,  
532 *J Phys Oceanog*, 12, 1169-1188.

533 Muir, D., and R. Lohmann (2013), Water as a new matrix for global assessment of hydrophilic  
534 POPs, *Trac-Trend Anal Chem*, 46, 162-172.

535 Paul, A. G., K. C. Jones, and A. J. Sweetman (2009), A first global production, emission and  
536 environmental inventory for perfluorooctane sulfonate, *Environ Sci Technol*, 43(2), 386-  
537 392.



538 Pickard, G. L., and W. J. Emery (1990), *Descriptive Physical Oceanography*, Fifth ed.,  
539 Butterworth-Heinemann, Woburn, MA. 320 pp.

540 Pistocchi, A., and R. Loos (2009), A Map of European Emissions and Concentrations of PFOS  
541 and PFOA, *Environ Sci Technol*, 43(24), 9237-9244.

542 Plumlee, M. H., J. Larabee, and M. Reinhard (2008), Perfluorochemicals in water reuse,  
543 *Chemosphere*, 72(10), 1541-1547.

544 Prevedouros, K., I. T. Cousins, R. C. Buck, and S. H. Korzeniowski (2006), Sources, fate and  
545 transport of perfluorocarboxylates, *Environ Sci Technol*, 40(1), 32-44.

546 Rahmstorf, S., J.E. Box, G. Feulner, M.E. Mann, A. Robinson, S. Rutherford, and E.J.  
547 Schaffernicht (2015), Exceptional twentieth-century slowdown in Atlantic Ocean over  
548 turning circulation, *Nature Climate Change*, 5, 475-479.

549 Rayne, S.; Forest, K., Perfluoroalkyl sulfonic and carboxylic acids: A critical review of  
550 physicochemical properties, levels and patterns in waters and wastewaters, and treatment  
551 methods. *J Environ Sci Health A* 2009, 44, 1145-1199.

552 Rhein, M., D. Kieke, and R. Steinfledt (2015), Advection of North Atlantic Deep Water from the  
553 Labrador Sea to the southern hemisphere, *J Geophys Res: Oceans*, 120, 2471-2487.

554 Routti, H., G. W. Gabrielsen, D. Herzke, K. M. Kovacs, and C. Lydersen (2016), Spatial and  
555 temporal trends in perfluoroalkyl substances (PFASs) in ringed seals (*Pusa hispida*) from  
556 Svalbard, *Environ Pollut*, 214, 230-238.

557 Ryther, J.H. (1969), Photosynthesis and fish production in the sea, *Science*, 166 (3901), 72-76.

558 Sanchez-Vidal, A., M. Llorca, M. Farre, M. Canals, and D. Barcelo (2015), Delivery of  
559 unprecedented amounts of perfluoroalkyl substances towards the deep-sea, *Sci Total*  
560 *Environ*, 526, 41-48.

561 Schenker, U., M. Scheringer, M. Macleod, J. W. Martin, I. T. Cousins, and K. Hungerbuhler  
562 (2008), Contribution of volatile precursor substances to the flux of perfluorooctanoate to  
563 the Arctic, *Environ Sci Technol*, 42, 3710-3716.

564 Scott, B. F., C. Spencer, E. Lopez, and D. C. G. Muir (2009), Perfluorinated Alkyl Acid  
565 Concentrations in Canadian Rivers and Creeks, *Water Qual Res J Can*, 44(3), 263-277.

566 Schultz, M. M., D. F. Barofsky, and J. A. Field (2006), Quantitative determination of fluorinated  
567 alkyl substances by large-volume-injection liquid chromatography tandem mass  
568 spectrometry - Characterization of municipal wastewaters, *Environ. Sci. Technol.*, 40(1),  
569 289-295.

570 Smith, J. N., F. A. McLaughlin, W. M. J. Smethie, B. S. Moran, and K. Lepore (2011), Iodine-  
571 129, Cs-137, and CFC-11 tracer transit time distributions in the Arctic Ocean, *J Geophys*  
572 *Res*, 116, C04024.

573 Stemmler, I., and G. Lammel (2010), Pathways of PFOA to the Arctic: variabilities and  
574 contributions of oceanic currents and atmospheric transport and chemistry sources, *Atmos*  
575 *Chem Phys*, 10, 9965-9980.

576 Tomy, G. T., W. Budakoeski, T. Halldorson, P. A. Helm, G. A. Stern, K. Friesen, K. Pepper, S.  
577 A. Tittlemier, and A. T. Fisk (2004), Fluorinated organic compounds in an Eastern Arctic  
578 marine food web, *Environ Sci Technol*, 38, 6475-6481.

579 U.S. EPA. (2008), Databased associated with the Clean Watersheds Needs Survey (CWNS)  
580 Report to Congress.  
581 <http://ofmpub.epa.gov/portal/page/portal/CWNS%20Reports/download> (accessed March  
582 2014).

583 Vecitis, C. D., H. Park, J. Cheng, B. T. Mader, and M. R. Hoffmann (2008), Kinetics and  
584 mechanism of the sonolytic conversion of the aqueous perfluorinated surfactants,  
585 perfluorooctanoate (PFOA), and perfluorooctane sulfonate (PFOS) into inorganic  
586 products, *J Phys Chem A*, 112(18), 4261-4270.

587 Wang, Z., I. T. Cousins, M. Scheringer, R. C. Buck, and K. Hungerbuhler (2014a), Global  
588 emission inventories for C4-C14 perfluoroalkyl carboxylic acid (PFCA) homologues from  
589 1951-2030, Part 1: production and emissions from quantifiable sources, *Environ Int*, 70,  
590 62-75.

591 Wang, Z., I. T. Cousins, M. Scheringer, R. C. Buck, and K. Hungerbuhler (2014b), Global  
592 emission inventories for C4-C14 perfluoroalkyl carboxylic acid (PFCA) homologues  
593 from 1951 to 2030, part II: The remaining pieces of the puzzle, *Environ Int*, 69, 166-176.

594 Wang, Z.; I.T. Cousins, M. Scheringer (2015) Comment on "The environmental photolysis of  
595 perfluorooctanesulfonate, perfluorooctanoate, and related fluorochemicals".  
596 *Chemosphere*, 122, 301-303

597 Wang, Z., J. C. DeWitt, C. P. Higgins, and I. T. Cousins (2017), A never-ending story of per-  
598 and polyfluoroalkyl substances (PFASs)?, *Environ Sci Technol*, 51(5), 2508-2518.

599 Wania, F. (2007), A global mass balance analysis of the source of perfluorocarboxylic acids in  
600 the Arctic Ocean, *Environ Sci Technol*, 41(13), 4529-4535.

601 Weisberg, S. (2014), *Applied Linear Regression*, 4<sup>th</sup> Ed., Wiley. ISBN-978-1-118-38608-8, 336

602 Wu, J., and E. A. Boyle (1997), Lead in the western North Atlantic Ocean: Completed response  
603 to leaded gasoline phaseout, *Geochim Cosmochim Acta*, 61(15), 3279-3283.

604 Wunsch, C., and P. Heimbach (2007), Practical global oceanic state estimation, *Physica D*,  
605 230(1-2), 197-208.

606 Xie, S. W., Y. L. Lu, T. Y. Wang, S. J. Liu, K. Jones, and A. Sweetman (2013), Estimation of  
607 PFOS emission from domestic sources in the eastern coastal region of China, *Environ.*  
608 *Int.*, 59, 336-343.

609 Yamashita, N., S. Taniyasu, G. Petrick, S. Wei, T. Gamo, P. K. S. Lam, and K. Kannan (2008),  
610 Perfluorinated acids as novel chemical tracers of global circulation of ocean waters,  
611 *Chemosphere*, 70(7), 1247-1255.

612 Yashayaev, I. (2007), Hydrographic changes in the Labrador Sea, 1960-2005, *Prog Oceanog*, 73,  
613 242-276.

614 Zareitalabad, P., J. Siemens, M. Hamer, and W. Amelung (2013), Perfluorooctanoic acid  
615 (PFOA) and perfluorooctanesulfonic acid (PFOS) in surface waters, sediments, soils and  
616 wastewater - A review on concentrations and distribution coefficients, *Chemosphere*,  
617 91(6), 725-732.

618 Zhang, L., J. G. Liu, J. X. Hu, C. Liu, W. G. Guo, Q. Wang, and H. Wang (2012), The inventory  
619 of sources, environmental releases and risk assessment for perfluorooctane sulfonate in  
620 China, *Environ Pollut*, 165, 193-198.

621 Zhang, X., R. Lohmann, C. Dassuncao, X. C. Hu, A. K. Weber, C. D. Vecitis, and E. M.  
622 Sunderland (2016), Source attribution of poly and perfluoroalkyl substances (PFASs) in  
623 surface waters from Rhode Island and the New York Metropolitan Area, *Environ Sci*  
624 *Technol Lett*, 3(9), 316-321.

625 Zhao, Z., Z. Y. Xie, A. Moller, R. Sturm, J. H. Tang, G. Zhang, and R. Ebinghaus (2012),  
626 Distribution and long-range transport of polyfluoroalkyl substances in the Arctic, Atlantic  
627 Ocean and Antarctic coast, *Environ Pollut*, 170, 71-77.

628

629 **Tables**630 **Table 1.** PFOS reservoirs (Mg) in surface, mid-depth and deep waters of the North Atlantic631 Ocean (20°N-60°N) in 2000 and 2015 and lifetimes against losses ( $\tau$ , a<sup>-1</sup>).

	<b>2000</b>	<b>2015</b>
<i>Reservoirs (Mg)</i>		
Surface Mixed Layer	380	95
Mid-depth (100-1000 m)	630	650
Deep (1000 m to bottom)	240	650
Full Basin (20°N-60°N)	1250	1390
<i>Lifetimes (a<sup>-1</sup>)</i>		
Surface Mixed Layer	2.6	1.6
Mid-depth (100-1000 m)	4.0	4.7
Deep (1000 m to bottom)	13.9	20.9
Full Basin (20°N-60°N)	10.8	14.1

632

633 **Figure Captions**

634 **Figure 1.** Direct continental inputs of PFOS to the North Atlantic Ocean (20°N-60°N) from  
635 European and North American (U.S. and Canada) wastewater and rivers between 1958 and 2010.  
636 Median estimates are shown as the solid line and shaded grey indicates 95% confidence intervals  
637 for North American releases. European inputs are based on *Pistocchi and Loos* [2009].

638 **Figure 2.** Modeled and observed surface seawater PFOS concentrations ( $\text{pg L}^{-1}$ ) in the North  
639 Atlantic Ocean. Panel (A) shows modeled surface (10 m) seawater concentrations for 2010  
640 compared to observations for 2009-2011. Panel (B) shows modeled and observed PFOS  
641 concentrations grouped by Longhurst Biogeographical Provinces [*Longhurst, 2007*], where error  
642 bars indicate standard error around the mean for both the model and observations. Grey dashed  
643 line indicates 1:1 fit between modeled and observed surface water concentrations. Regression  
644 slope indicates the median release scenario (Figure 1) results in a model bias relative to surface  
645 water PFOS observations of 75%. Modeled surface water concentrations capture 90% of the  
646 observed variability across biogeographical provinces [*Longhurst, 2007*]. *NADR* = North  
647 Atlantic Drift Province [*Ahrens et al., 2009a; 2010b; Ben, GFST* = Gulf Stream Province, *NASW*  
648 = North Atlantic Subtropical Province - West, *NASE* = North Atlantic Subtropical Province -  
649 East, *NATR* = North Atlantic Tropical Gyral Province. Surface PFOS measurements are from:  
650 *Ahrens et al.* [2009a; 2009b; 2010b]; *Benskin et al.* [2012]; *Gonzalez-Gaya et al.* [2014]; *Zhao et*  
651 *al.* [2012].

652 **Figure 3.** Comparison of modeled and observed vertical profiles of PFOS measured in the  
653 Labrador Sea and Mid-Atlantic Ocean by *Yamashita et al.* [2008]. Observations are shown in  
654 black and model scenarios are in red based on the median (solid line) and low (dashed line)  
655 PFOS release estimates shown in Figure 1.

656 **Figure 4.** Modeled temporal evolution of surface water (10 m) PFOS concentrations in the North  
657 Atlantic Ocean between 1980 and 2020. The model is forced with the median release scenarios  
658 from rivers and wastewater shown in Figure 1 and zero releases between 2010-2020, adjusted by  
659 the mean model bias of 75% shown in Figure 2. Major surface currents are shown as white  
660 arrows. NAC = North Atlantic Current, LC = Labrador Current, NS = North Sea.

661 **Figure 5.** Modeled PFOS concentrations in North Atlantic Seawater ( $\text{pg L}^{-1}$ ) between  $20^{\circ}\text{N}$  and  
662  $60^{\circ}\text{N}$  at different seawater depths based on the median PFOS release scenarios shown in Figure  
663 1, adjusted for mean bias (75%). Shading indicates distributions of concentrations in different  
664 ocean regions with the median across the whole basin indicated as a solid line. Darker shading  
665 indicates 25<sup>th</sup> and 75<sup>th</sup> percentile modeled concentrations and 95<sup>th</sup> percentiles are shown as the  
666 lightest blue bounds.

667 **Figure 6.** Mass flows of PFOS in the North Atlantic ( $\text{Mg a}^{-1}$ ) between  $20^{\circ}\text{N}$  and  $60^{\circ}\text{N}$  for the  
668 years 2000 (peak inputs) and 2015. Results are from the median release scenario (Figure 1) and  
669 adjusted for mean bias (75%).

670 **Figure 7.** Modeled circulation of PFOS above and below  $60^{\circ}\text{N}$  at different seawater depths  
671 between 1958-2040. Deeper waters are indicated by blue and surface waters are shown in shades  
672 of red, orange and yellow. Solid black line indicates net flow, where positive numbers indicate  
673 flow above  $60^{\circ}\text{N}$  (into the Arctic region) and negative numbers indicate flow below  $60^{\circ}\text{N}$  (into  
674 the North Atlantic).