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Use of passive samplers to detect Organochlorine Pesticides in air and water at Wetland Mountain region sites (S-SE Brazil)

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1 **Use of passive samplers to detect Organochlorine Pesticides in air and**
2 **water at Wetland Mountain region sites (S-SE Brazil)**

3

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23 **Abstract**

24 Low-density polyethylene (LDPE) passive samplers were deployed in upland surface
25 waters and the overlying atmosphere during May and June 2012, to determine the
26 transport and trends of freely dissolved and gaseous organochlorine pesticides (OCPs)
27 along altitudinal gradients in mountain regions in south and southeast Brazil. Gaseous
28 OCP concentrations were dominated by hexachlorobenzene (3.0 to 29 $\text{pg}\cdot\text{m}^{-3}$) and
29 endosulfans ($\Sigma = \alpha\text{-endosulfan} + \beta\text{-endosulfan} + \text{endosulfan sulphate}$, 170 to 260 $\text{pg}\cdot\text{m}^{-3}$)
30 ³), whereas freely dissolved endosulfans were significantly higher than all other OCPs
31 ($p < 0.001$). The presence of some target pesticides at the highest elevation sites
32 indicated their efficient high-altitude transport from regional sources. Air-water
33 exchange gradients indicated net deposition of most volatile and recently banned OCPs
34 (e.g., HCB, endosulfan) over Brazilian mountains. Moreover, the exposure of these sites
35 to large-scale continental airflows with varying source contributions may partly explain
36 the atmospheric deposition of selected OCPs over upland freshwaters at tropical and
37 subtropical mountains sites in Brazil. These findings, coupled with LDPE passive air
38 and water sampling measurements, point out the potential inputs from distant sources of
39 semi-volatile chemicals to the two high-altitude sites.

40

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42

43 **Keywords:** Low density polyethylene (LDPE) passive samplers, organochlorine
44 pesticides, long-range atmospheric transport, Brazil.

45

46 **1. Introduction**

47 Mountain cold-trapping of persistent organic pollutants (POPs) has been
48 receiving considerable attention over the last decades (Calamari et al., 1991; Blais et al.,
49 1998; Wania and Westgate, 2008). POPs are recalcitrant chemicals with relatively high
50 potential for bioaccumulation and environmental mobility, and display toxic effects on
51 organisms and humans even at relative low concentration levels (UNEP, 2002). Their
52 re-emissions, long-range atmospheric transport (LRAT), and deposition are controlled
53 by meteorological and geographical parameters such as orographic winds, high
54 precipitation rates and low temperatures, typically found in mountainous regions (Daly
55 and Wania 2005; Kirchner et al., 2009; Sheng et al., 2013).

56 It is well-established that higher precipitation rates coupled with transport of
57 pollutants from sources in lowland crop soils –either current use or revolatilisation of
58 previously applied compounds– could have an important role in the atmospheric
59 deposition of semi-volatile organic pollutants in tropical and subtropical mountain
60 regions (Daly et al., 2007; Estellano et al. 2008). Although several POPs -including
61 organochlorine pesticides (OCPs) and polychlorinated biphenyls (PCBs)- are already
62 regulated and even banned in many South American countries, recent studies have
63 highlighted the atmospheric transport of OCPs to tropical and subtropical remote
64 mountain regions, such as the Bolivian and Chilean Andes, and also Brazilian uplands
65 (Estellano et al., 2008; Meire et al., 2012; Pozo et al., 2004).

66 Numerous studies have shown the feasibility of using passive samplings to
67 detect POPs in several environmental compartments, to monitor the exposure and
68 evaluate the risk of wild-life and humans at regional and global scales (Harner et al.,
69 2006; Levy et al., 2009; Lohmann et al., 2010; Pozo et al., 2009). Low density

70 polyethylene passive samplers (LDPE) have been widely used in recent years to detect a
71 range of trace hydrophobic organic contaminants (HOCs) in aquatic and air media,
72 including legacy and emerging pollutants (Khairy and Lohmann 2012, 2013; Sacks and
73 Lohmann, 2011). HOCs strongly partition into the LDPE matrix at a high-accumulation
74 rate (Sacks and Lohmann, 2011). This feature offers advantages compared to the
75 traditional environmental measurements, such as lower detection limits and easier and
76 less costly laboratory methods, making LPDE a very useful tool for first-order risk
77 assessment.

78 The aims of this study were to (i) assess the presence of legacy and recently
79 banned OCPs in air and water in two National Parks in Brazil; (ii) investigate the
80 altitudinal variation of target OCPs over shallow freshwaters in tropical and subtropical
81 mountain regions in Brazil, and (iii) determine whether the OCPs had reached
82 equilibrium between lakes and atmosphere. To achieve these aims, we deployed passive
83 air and water samplers consisting of low-density polyethylene strips during fall and
84 winter seasons (May-June 2012), following altitudinal transects in south and southeast
85 Brazil. The current work presents original data of OCPs in the waters over Brazilian
86 uplands.

87

88 **2. Material and Methods**

89 *2.1. Site description*

90 We conducted this study in the mountain ranges of “Serra do Mar” (National
91 Park of Itaiaia - NPIT, lat.: 22°22’38”S, lon.: 44°41’35”W, Rio de Janeiro State) and
92 “Serra Geral” (National Park of São Joaquim - NPSJ, lat.: 28°00’49”S, lon.:
93 49°35’17”W, Santa Catarina State), located in southeast and south Brazil, respectively.

94 The two national parks comprise rainforests and high-altitude ecosystems and are
95 considered important conservation areas for biodiversity and endemic species (Myers et
96 al., 2000).

97

98 *2.2. Field Deployment*

99 Four sampling sites were established along altitudinal gradients at each National
100 Park during late fall and early winter seasons (May and June) in 2012. Passive LDPE
101 sheets were deployed in the surface waters and overlying atmosphere concomitantly
102 over four to five weeks (Figure 1). Water LDPE samplers were deployed 1 m below the
103 surface attached to small buoys, whilst air samplers were fixed at 1.5 m above ground,
104 inside inverted stainless steel bowls to form a clam-shell shape. The bowl design also
105 protects the sampling medium (LDPE sheet) from direct radiation and rainfall, and
106 diminishes the wind speed effect (Shoeib and Harner, 2002). Site and deployment
107 details are given in Table S1. Additionally, five field blanks were collected at both
108 parks to assess possible sample contamination during transport and storage. Methods for
109 preparation and extraction of LDPEs as well as the instrumental analyses of the extracts
110 and quality assurance/quality control are given in supplementary material (SM).

111

112 **3. Theory/calculation**

113 *3.1. Atmospheric and water concentrations derived from LDPEs*

114 We added performance reference compounds (PRCs) added to the samplers
115 before deployment in order to gauge whether organochlorine pesticides had reached
116 equilibrium and to adjust for disequilibrium in polyethylene samplers. Assuming that

117 uptake and elimination rates are equivalent, the freely dissolved (or gas-phase)
118 concentrations of individual OCPs ($C_{water\ or\ air}$) were then calculated by the following
119 equation:

120

$$121 \quad C_{water\ (air)} = C_{LDPE} / K_{LDPE-water\ (LDPE-air)} \times [1 - \exp^{-R_s \times t / K_{LDPE-water\ (LDPE-air)} \times m_{LDPE}}] \quad (1)$$

122

123 where C_{LDPE} is the OCP concentration in the LDPE ($L.kg^{-1}$), R_s is the sampling rate
124 [$m^3.day^{-1}$], t is the deployment period (days), m_{LDPE} is the mass of the LDPE sheet (kg)
125 and $K_{LDPE-water}$ (*or* $K_{LDPE-air}$, when applicable) is the sampler-water (air) partitioning
126 coefficients ($m^3.kg^{-1}$). Sampling rates (Table S3) were calculated using the PRCs as
127 follows:

$$128 \quad f = \exp(-R_s \times t / K_{LDPE-water\ (LDPE-air)} \times m_{LDPE}) \quad (2)$$

129 where f is the fraction of PRC retained in the passive sampler after deployment. Details
130 on the PRC method have been described elsewhere (Booij et al., 2002; Khairy et al.,
131 2014). Physico-chemical properties used to calculate the sampler-water partitioning
132 coefficients ($K_{LDPE-water}$ s) are given in supplementary material (SM).

133

134 3.2. Air mass back trajectory analysis.

135 To investigate the influence of atmospheric large-scale circulation on the
136 concentrations of air pollutants at the highest altitudinal sites, 5-day backward air
137 trajectories were calculated using the Hybrid Single Particle Lagrangian Integrated
138 Trajectory Model (HYSPLIT) (Draxler et al., 2012). In this study, we used cluster
139 analysis to identify source regions and to investigate how different atmospheric flow

140 patterns influence the concentrations of OCPs at the monitored sites. The Supporting
141 Material (SM) provides additional details on the model and input dataset.

142

143 *3.3. Air-Water Exchange Gradients*

144 Air-water exchange gradients were calculated as the ratio of OCP concentrations
145 measured in the LDPE (ng/LDPE) in each reservoir (water and air), corrected for non-
146 equilibrium, minus one:

147

$$148 \quad \text{Air – Water exchange} = (C_{LDPE (Water)} / C_{LDPE (air)}) - 1 \quad (3)$$

149

150 In effect, the LDPE samplers reflect dissolved and freely OCPs in water or air,
151 respectively, and thus they easily enable to derive the gradient across the air-water
152 interface. Positive values indicate net volatilisation of OCPs and negative values
153 indicate net deposition. We did not apply any temperature correction, as suggested by
154 Lohmann et al. (2011), so the gradients indicate conditions in the field.

155

156 **4. Results and Discussion**

157 *4.1. Sampling rates and OCP equilibrations in field-deployed LDPE samplers*

158 Sampling rates (R_s) for the atmospheric and water samplers are given in Table
159 S3. Atmospheric R_s ranged from 8.0 to 43 m³.day⁻¹, which is consistent with R_s values
160 measured for atmospheric OCPs (Khairy and Lohmann, 2013). Greater R_s values were
161 observed at the highest altitudes at both National Parks, however the relationship
162 between R_s and altitude was statistically insignificant.

163 Tables S5 and S6 depict calculated % equilibrium for gaseous OCPs. HCB, α -
164 and γ - HCHs ($\log K_{LDPE-air} \leq 7.15$) were the only detected OCPs that approached

165 equilibrium (> 97 %) in all deployed samplers. All the other detected OCPs were still
166 far from reaching equilibrium (< 50 %). Khairy and Lohmann (2014) observed a similar
167 pattern for gaseous OCPs in the USA.

168 *Rs* values for the water samplers ranged from 18 to 81 L.day⁻¹. A similar range
169 was observed for pyrene and higher molecular weight polycyclic aromatic hydrocarbons
170 (PAHs) sampled in Narragansett Bay (Lohmann et al., 2011) and in the Great Lakes
171 (McDonough et al., 2014), and for extended OCP studies (15 to 80 L.day⁻¹) in the
172 Lower Great Lakes (Lakes Ontario and Erie) (Khairy et al., 2014). The sampling rates
173 for National Park of Itatiaia were significantly higher ($p = 0.006$) than those for
174 National Park of São Joaquim, and the calculated % equilibrium values for freely
175 dissolved OCPs showed variations between the two parks. At the National Park of
176 Itatiaia, all the detected OCPs (except *p,p'*-DDT) approached equilibrium by the end of
177 the deployment period, whilst at the National Park of São Joaquim, only HCB, α -
178 endosulfan and endosulfan sulphate were at equilibrium (Table S5 and S6). This
179 difference is a consequence of different *Rs* values (higher *Rs* values indicate faster
180 equilibration).

181

182 4.2. Air and water concentrations of target pesticides

183 At both National Parks, the most frequently detected compounds in air were
184 HCB and endosulfans (α -endosulfan, β -endosulfan and endosulfan sulphate). The
185 highest atmospheric concentrations were detected for HCB (4.0 to 29 pg.m⁻³) followed
186 by α -endosulfan (1.8 to 20 pg.m⁻³), β -endosulfan (<LOD to 7.1 pg.m⁻³), α -HCH (<LOD
187 to 3.2 pg.m⁻³), endosulfan sulphate (0.5 to 2.5 pg.m⁻³), endrin (<LOD to 1.9 pg.m⁻³) and
188 *p,p'*-DDT (<LOD to 0.6 pg.m⁻³). All the other investigated OCPs were below detection

189 limits. Figure 2, and Tables S7 and S8 of the SM summarise the air concentrations of
190 target pesticides.

191 Similar to the atmospheric findings, endosulfans accounted for highest
192 frequency detection in shallow waters amongst target OCPs (100s pg.L^{-1}) (Tables S7
193 and S8). However, amongst the less frequently detected pesticides (*p,p'*- DDT, *p,p'*-
194 DDD, *p,p'*- DDE and *trans*-nonachlor) dieldrin stands out with intermediate
195 concentration levels at the National Park of São Joaquim (< LOD to 31 pg.L^{-1}) (Figure
196 2). The greatest concentrations were reported for endosulfan sulphate (162 to 260 pg.L^{-1}),
197 followed by dieldrin (15 to 31 pg.L^{-1}), α -endosulfan (<LOD to 6.5 pg.L^{-1}), HCB (1.2
198 to 2.3 pg.L^{-1}) and DDTs (0.1 to 0.6 pg.L^{-1}). The results for individual pesticides are
199 discussed below. We chose to present geometric mean (GM) values are presented rather
200 than arithmetic means to prevent extreme values (outliers) from skewing the mean.

201

202 4.2.1. Hexachlorobenzene (HCB)

203 Although HCB has been widely used in the past, its commercial use has been
204 banned worldwide (Barber et al., 2005). However, HCB is currently released as an
205 intermediate and by-product in the manufacture of a large number of chlorinated
206 solvents, aromatics and pesticides and continues to volatilize from contaminated soils to
207 the atmosphere (Bailey, 2001). In Brazil, there are well-known hotspots and stockpiles
208 of HCB (e.g. Cubatão, São Paulo State) (UNEP, 2002) that may still represent a
209 significant local environmental sources. On top of that, Brazil imported ca. 800 t of
210 HCB in the middle of 60s (UNEP 2002).

211 HCB was detected at all sampling sites at both National Parks, and exhibited the
212 highest air concentrations among the target compounds (Figure 2). HCB concentrations
213 in air ranged from 21 to 29 pg.m^{-3} (GM 25 pg.m^{-3}) and from 4.3 to 9.9 pg.m^{-3} (GM 8.0

214 pg.m^{-3}) for NPIT and NPSJ, respectively. These air concentrations are much lower than
215 air concentrations measured in high-mountain areas worldwide, such as: Central
216 Pyrenees, High Tatras (36 to 120 pg.m^{-3}) (Van Drooge et al., 2004) and Swiss Alps (102
217 pg.m^{-3}) (Shunthirasingham et al., 2013), Canadian Rocky mountains (21 to 149 pg.m^{-3})
218 (Daily et al., 2007), Tibetan plateau, ($>50 \text{ pg.m}^{-3}$) (Kang et al., 2009; Liu et al., 2010)
219 and above the Arctic circle (48 to 71 pg.m^{-3}) (Wong et al., 2011). Only few studies have
220 reported the presence of HCB in South American mountain regions (Barra et al., 2004;
221 Shunthirasingham et al., 2011). Generally, air concentrations are in the same range as
222 those reported in the atmosphere of coastal/open Southern Atlantic (Bidleman et al.,
223 1993; Montone et al., 2005), Pacific Oceans, and Antarctic Peninsula region (Zhang et
224 al., 2010). These results suggest that lower air concentration levels of HCB are fairly
225 uniformly distributed across the Southern Hemisphere (Table S10), which indicates that
226 its historical use in this hemisphere was lower than in the Northern Hemisphere.

227 HCB exhibited fairly uniform vertical distributions at all monitoring sites for
228 both monitored National Parks. The HCB distribution pattern is consistent with
229 previous mountain studies along altitudinal transects (10 to 4400 m.a.s.l.) in the
230 Southern and Northern Hemispheres (Shunthirasingham et al., 2011; Wong et al., 2011).
231 Generally, these findings demonstrate well-mixed conditions of HCB in the atmosphere
232 and the lack of strong source-receptor relationships.

233 HCB concentrations in the water were in a narrow range, from 2.1 to 2.3 pg.L^{-1}
234 and from 1.2 to 2.3 pg.L^{-1} for NPIT and NPSJ, respectively. These values are
235 comparable to concentrations found at other remote mountain lakes worldwide (<10
236 pg.L^{-1}) (Barber et al., 2005; Catalan et al., 2004; Vilanova et al., 2001). Low HCB
237 concentrations in water have been recently reported along the South Atlantic (1.9 to 3.3
238 pg.L^{-1}), and South Pacific (0.4 to 0.8 pg.L^{-1}) open water transects (Booij et al., 2007;

239 Zhang et al., 2010) (Table S11). However, these HCB levels are much lower than
240 concentrations reported in South American contaminated freshwaters (1,000-10,000
241 ng.L⁻¹) (UNEP, 2002).

242

243 4.2.2. Endosulfans

244 Only recently listed on the Stockholm Convention, endosulfan was one of the
245 most widely used organochlorine insecticides globally. Since its introduction in the
246 1950s, the cumulative global use of endosulfan on crops is estimated to be 308 kt (1950
247 - 2000) and its isomers have been intensely used in South American countries (Li and
248 McDonald et al., 2005). In this study, endosulfans ($\Sigma = \alpha + \beta + \text{sulphate}$) were detected
249 at all sampling sites at the two National Parks. Gaseous endosulfan concentrations
250 ranged from 5.8 to 29 pg.m⁻³ (GM 15 pg.m⁻³) and from 2.9 to 8.7 pg.m⁻³ (GM 5.0 pg.m⁻³)
251 for NPIT and NPSJ, respectively. These concentrations are considerably lower than
252 reported in other studies in Chilean (63 to 100 pg.m⁻³), Bolivian (28 to 1800 pg.m⁻³) and
253 even Brazilian (40 to 5600 pg.m⁻³) uplands (Estellano et al., 2008; Meire et al., 2012;
254 Pozo et al., 2004) (Table S10). However, previous studies in South American
255 mountainous regions have generally associated the highest air concentration levels of
256 recently banned pesticides with intense seasonal applications on lowland crops, mainly
257 during summer periods (Estellano et al., 2008; Meire et al., 2012). These findings
258 suggest that fall-winter seasons monitored in this study reflect the expected lower air
259 concentrations levels of endosulfans at all upland sites.

260 For NPIT, the air concentrations of endosulfans at the highest monitored sites
261 (ITA3 and ITA4) were about 2- to 5-fold higher than those measured at lowest
262 altitudinal sites (figure 3). This is consistent with previous findings for other tropical
263 Brazilian mountains in the vicinity of the sites investigated here (e.g., Rio de Janeiro

264 state), especially during similar seasons. However, no clear trends of endosulfan and its
265 isomers were observed along altitudinal gradients at NPSJ sites. These results conflict
266 with previous studies conducted at the same National Park, which reported a strong
267 correlation between air concentrations of endosulfans and altitude (Meire et al., 2012).
268 Recent studies in the South American mountain regions have in general shown strong
269 upslope enrichment for endosulfans, which were attributed to atmospheric transport
270 from regional and continental agricultural sources (Meire et al., 2012;
271 Shunthirasingham et al., 2011).

272 Technical grade endosulfan contains α -endosulfan and β -endosulfan as the main
273 isomers (>95%). Depending on the technical mixture, the ratio of Endo- α/β in technical
274 formulations ranges from 2.0 to 2.3 (Weber et al., 2010). β -endosulfan is less stable in
275 the atmosphere and, thus, an elevated α/β ratio represents an aged signature scenario
276 (Yao et al., 2006). In this study, the ratios of α/β ranged from 1.2 to 3.7 (average 2.0)
277 which suggests fresh applications of endosulfans (lower ratios <2.3) (figure 3). Other
278 studies have shown a seasonal variation of α/β values at Brazilian and Bolivian
279 mountain sites with higher atmospheric ratio levels (>2.3), especially during the
280 summer period (Estellano et al., 2008; Meire et al., 2012). According to Meire et al.
281 (2012), this seasonal trend may be attributed to more rapid or enhanced weathering of
282 endosulfan during the summer. The winter period contains mainly fresh endosulfan,
283 however at lower emission levels. As shown in Figure 3, the predominance of α -
284 endosulfan over the total endosulfan profile (60%, average value) was observed at
285 almost all sites with higher air concentrations (>10 $\text{pg}\cdot\text{m}^{-3}$) followed by β -endosulfan
286 (25%) and endosulfan sulphate (13%). The lower air concentration levels for endosulfan
287 sulphate (< 3 $\text{pg}\cdot\text{m}^{-3}$) were similar to previous studies (Estellano et al., 2008; Meire et
288 al., 2012; Pozo et al., 2009).

289 Endosulfans were detected at high concentrations in water, ranging from 190 to
290 264 pg.L⁻¹ (GM 219 pg.L⁻¹) and from 166 to 236 pg.L⁻¹ (GM 185 pg.L⁻¹) at NPIT and
291 NPSJ, respectively. These findings are consistent with concentrations found in
292 mountain waters as well as snowpack samples across Western US National Parks (44 to
293 1500 pg.L⁻¹), Canadian Rocky Mountains (> 100 pg.L⁻¹), Alps (40 to 288 pg.L⁻¹),
294 Pyrenees (520 to 1770 pg.L⁻¹), and Caledonian Mountains (100 to 140 pg.L⁻¹) (Blais et
295 al., 1998, 2001; Brandford et al., 2010; Hageman et al., 2006; Vilanova et al., 2001).
296 Even lower concentrations of endosulfans (12 to 24 pg.L⁻¹) were recently reported in
297 waters of the Tatra Mountains, Slovakia (Arellano et al., 2011) (Table S11).

298 In this study, the mixture of endosulfan in freshwater samples were dominated
299 by endosulfan sulphate (162 to 260 pg.L⁻¹), which accounted for up to 80% of this
300 dissolved insecticide group in upland waters (Figure S1). α -endosulfan exhibited the
301 lowest concentrations (<10 pg.L⁻¹) while β -endosulfan was not detected in water
302 samples. These results correlate well with endosulfan sulphate as the predominant
303 metabolite in aquatic systems worldwide (Booij et al., 2007). The microbiologic
304 oxidation of the technical-grade endosulfan in soils, and its further re-emission to air is
305 regarded as the main source to the atmosphere. In natural aquatic ecosystems,
306 endosulfan sulphate degrades at a slower rate and hence is more persistent than the
307 parent isomers; with a half-life in water of several weeks (Weber et al., 2010). Much
308 greater concentrations of endosulfan sulphate (>100 ng.L⁻¹) than its parental isomers (α -
309 and β -endosulfan) have been detected in contaminated waters close to intensive crop
310 production areas in Argentina and Brazil (Bonansea et al., 2013; Laabs et al., 2002),
311 (Table S11) reinforcing the historical use of this insecticide in South America.

312 Brazil is considered the biggest user of endosulfan in South America and one of
313 the top endosulfan users worldwide (Li and McDonald 2005). According to the Brazilian

314 Ministry of Development, Industry and Foreign Trade's database
315 (<http://aliceweb.mdic.gov.br>), Brazil imported approximately 23 kt of endosulfan from
316 2001 to 2011 with an estimated annual consumption of 2.3 kt.yr⁻¹. As several other
317 South American countries, Brazil started to phase out the commercial use of endosulfan
318 in 2010. However, the agriculture use of endosulfans to control insect populations in
319 sugar-cane, cocoa, coffee, soybean and cotton plantations was totally banned only in
320 2013 (Brazilian Health Surveillance Agency - ANVISA).

321

322 4.2.3. DDTs

323 For DDTs, the parent compound *p,p'*-DDT and its metabolite *p,p'*-DDE were
324 the main detected compounds for both monitored National Parks. Atmospheric DDT
325 concentrations ranged from 0.1 to 0.6 pg.m⁻³ (GM 0.2 pg.m⁻³) and from 0.1 to 0.3 pg.m⁻
326 ³ (GM 0.2 pg.m⁻³) for NPIT and NPSJ, respectively. These concentrations were much
327 lower than concentrations observed at high-mountain sites in Europe (0.2 to 12 pg.m⁻³)
328 (Van Drooge et al., 2004), North America (1.6 to 55 pg.m⁻³) (Daly et al., 2007), Asia
329 (0.3 to 75 pg.m⁻³) (Liu et al., 2010; Wang et al., 2008) and even in South America (1.0
330 to 119 pg.m⁻³) (Estellano et al., 2008; Meire et al., 2012) (Table S10).

331 Similar to atmospheric levels, freely dissolved concentrations of DDTs indicated
332 background levels (NPIT: 0.1 to 0.4 pg.L⁻¹, GM 0.2 pg.L⁻¹; NPSJ: 0.2 to 0.3 pg.L⁻¹, GM
333 0.3 pg.L⁻¹). Only few studies have pointed out the presence of DDTs in South American
334 highland waters (Barra et al., 2005; Schreiber et al., 2013). Schreiber et al. (2013)
335 reported higher dissolved DDT (*p,p'*-DDE) concentrations in upland surface waters of
336 River Xanaes in Central Argentina (70 to 340 pg.L⁻¹). On the other hand, Barra et al.
337 (2005) did not detect the presence of dissolved DDTs along altitudinal gradients in Laja
338 River Basin waters of Central Southern Chile. The concentrations detected in our

339 measurements were also generally much lower than dissolved DDT concentrations
340 reported at other mountains and remote sites worldwide (Table S11).

341 Over the past few decades (1950-1990), Brazil has experienced a historical use
342 of DDT, primarily for sanitary purposes to control insect pests that are vectors for
343 tropical diseases, such as malaria and typhus (D'Amato et al., 2002). It is estimated that
344 Brazil used and exported to other South American countries a total of 80 t of DDT until
345 the 1980s, whilst more than 30 t were imported in the same period (Almeida et al.,
346 2007).

347

348 *4.3. Air mass back trajectory analysis*

349 The cluster analysis performed on the 5-day back trajectories unveiled that about
350 40% of the trajectories arriving at both sites come from the northern sector of Brazil
351 with purely continental influences. At the NPIT, two other clusters accounted for the
352 remainder of the trajectories. A southerly cluster of trajectory (30%) was identified
353 moving from the South Atlantic Ocean with no contact with the continent. The third
354 cluster (29%) was dominated by continental trajectories that stemmed from western
355 South America.

356 At NPSJ, 35% of the trajectories were maritime, with southerly origin. These
357 trajectories had anticyclonic curvature, which suggests that the airflow associated with
358 these pathways is part of the semi-permanent anticyclonic over the southern Atlantic.
359 The third cluster (24%) consisted of fast-travelling trajectories that originated over the
360 southern Pacific and spent most of the time over Patagonia and southern Brazil before
361 reaching the monitoring site.

362 The exposures of these sites to large-scale continental airflows with varying
363 source contributions (e.g., from extensive crop areas in central Brazil and Argentina)

364 may partly explain the atmospheric deposition of selected OCPs over upland
365 freshwaters at tropical and subtropical mountains sites in Brazil. These findings,
366 coupled with LDPE passive air and water sampling measurements, point out the
367 potential inputs from distant sources of semi-volatile chemicals to the two high-altitude
368 sites (1700 m a.s.l. NPSJ; 2400 m a.s.l. NPIT). Northerly and westerly (continental)
369 back trajectories represented 70% of the airflow reaching NPIT high-altitude site, which
370 may account for higher air concentrations of endosulfans ($>10 \text{ pg.m}^{-3}$) at this site
371 compared to NPSJ site (4.0 pg.m^{-3}), which received only 41% of continental flows
372 (figures S2 and S3). A similar pattern was also observed on surface upland waters,
373 where concentration of endosulfans at NPIT site ($>200 \text{ pg.L}^{-1}$) was slightly higher than
374 in NPSJ water samples (166 pg.L^{-1}). These results highlight the role of atmospheric
375 long-range dynamics on transporting organochlorine pesticides from distant source
376 regions and probably enhancing concentrations at high altitude mountain sites in Brazil.
377

378 *4.4. Air-water Exchange Gradients*

379 In this study, HCB, α -endosulfan and endosulfan sulphate displayed air-water
380 exchange values < 0 (Eq. 3) at almost all sites, indicating net deposition along the two
381 altitudinal transects (figure 4). An exception was only observed at the lowest altitudinal
382 site (990 m. a.s.l.) of NPSJ, when HCB showed net volatilisation > 0 . These results
383 reinforce the major atmospheric input of volatile organochlorine pesticides at tropical
384 and subtropical Brazilian mountains. Similar to our results, Galbán-Malagón et al.
385 (2013) reported net deposition for HCB from the Weddell, Bransfield and
386 Bellingshausen Seas (Antarctica Peninsula Region) in two sampling cruises in 2008 and
387 2009. Net deposition or equilibrium of HCB and endosulfans (α -endosulfan) has been
388 also reported in studies in the Arctic (Lohmann et al., 2009; Wong et al., 2011).

389 In contrast, net volatilisation gradients were observed for *p,p'*-DDE at NPIT
390 monitored sites. This indicates that NPIT sites still received sufficient *p,p'*-DDE in
391 water, probably from runoff and headwater stream input, to cause net volatilisation of
392 legacy organochlorine pesticides. Air–water exchange gradients suggested net
393 volatilisation for organochlorine pesticides, including *p,p'*-DDE, along Chinese
394 coastline ocean (Lin et al., 2012). These results pointed out the surface waters net
395 volatilisation after a long period of OCP prohibition.

396

397 **5. Conclusions**

398 As atmospheric transport is faster than dilution in the water column, the volatile
399 OCPs –especially the recently banned pesticides– are experiencing net deposition over
400 pristine National Park areas. We observed this behaviour especially for endosulfan.
401 Surprisingly, even HCB was still mostly undergoing net deposition, with only DDTs
402 displaying net volatilisation. This implies that net deposition of endosulfans at these and
403 other sites will continue for the foreseeable future. Moreover, the exposure of these sites
404 to large-scale continental airflows with varying source contributions may partly explain
405 the atmospheric deposition of selected OCPs over upland freshwaters at tropical and
406 subtropical mountains sites in Brazil. These findings, coupled with LDPE passive air
407 and water sampling measurements, point out the potential inputs from distant sources of
408 semi-volatile chemicals to the two high-altitude sites.

409 Brazil is a signatory to the Stockholm and since 2001 the country has had to
410 comply with reductions in the use of POPs, including several organochlorine pesticides.
411 According to a federal law (CONAMA 430/11) the shallow waters investigated in this
412 study are classified as “special waters” and should be protected to preserve their natural
413 conditions. This study is a first effort to quantify OCPs in mountain spring waters and to

414 understand air-water exchange over pristine uplands in Brazil. Further studies are
415 necessary to address these issues, especially monitoring of recently banned and current-
416 use pesticides in waters of highland tropical sites.

417

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424

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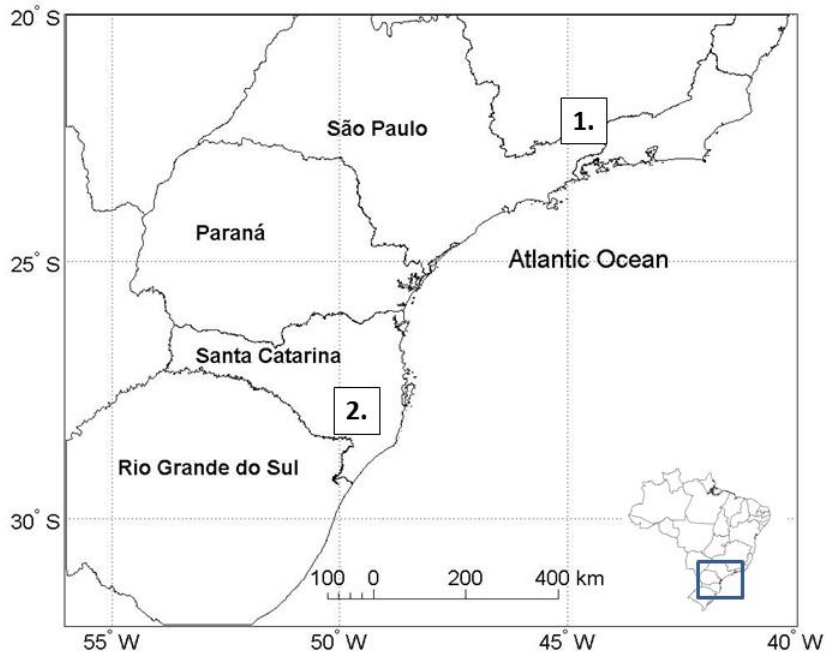
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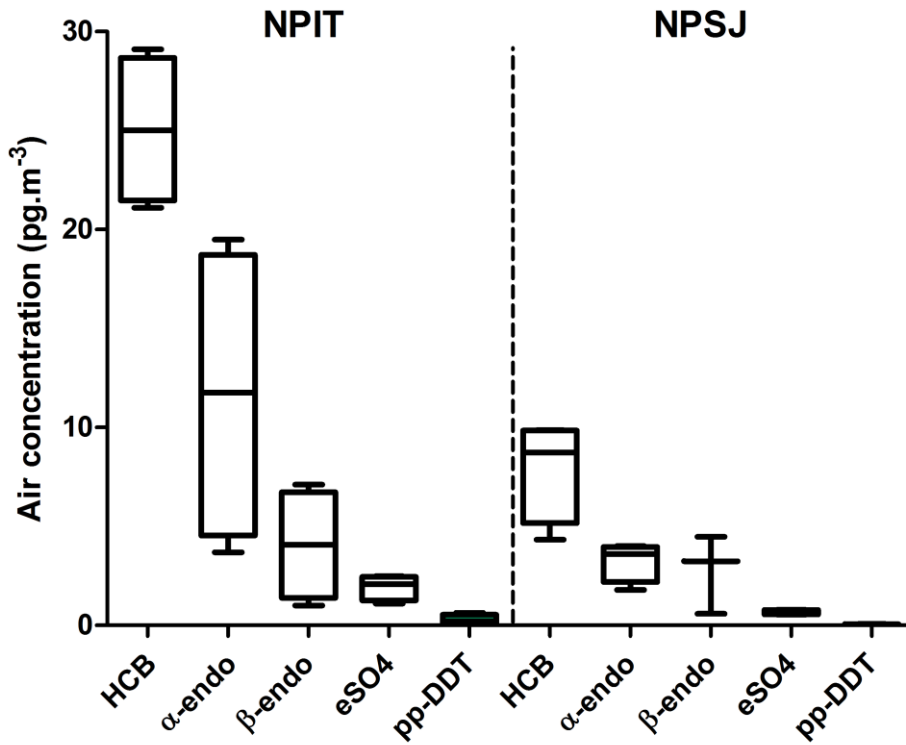
688 **Figure 1.** Map showing the south and southern regions of Brazil. The two National
689 Parks where the samplings were conducted are also indicated: 1. National Park of
690 Itatiaia – Rio de Janeiro State, 2. National Park of São Joaquim – Santa Catarina State.
691 The photographs illustrate the deployment of the passive samplers.

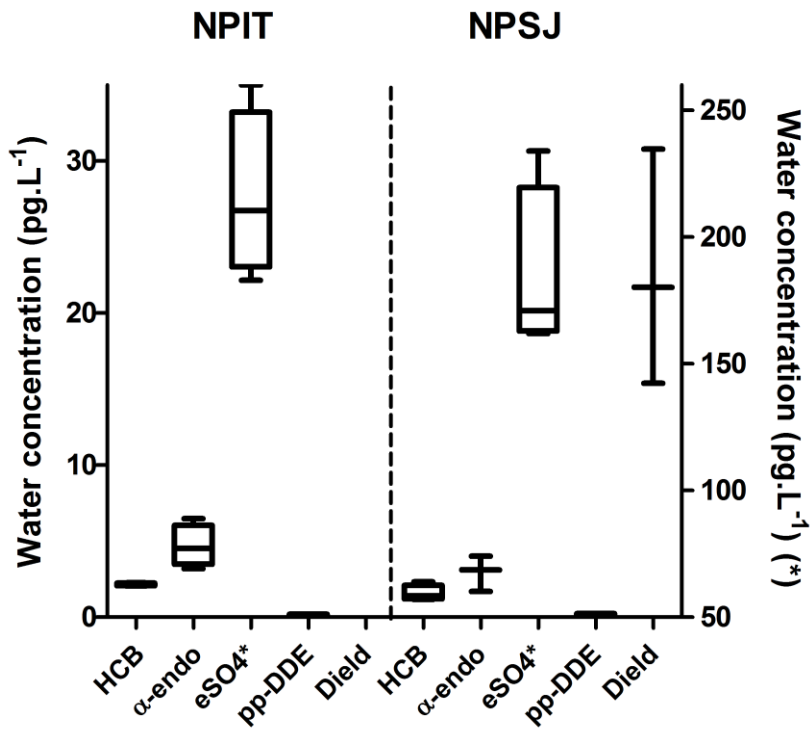


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694 **Figure 2.** “Box and whisker” plots showing summaries of the estimated atmospheric
 695 (pg.m^{-3}) and dissolved (pg.L^{-1}) concentrations of selected organochlorine pesticides at
 696 the two National Parks (NPIT – National Park of Itatiaia; NPSJ – National Park of São
 697 Joaquim.). The results represent measurements during fall-winter periods in 2012 and
 698 over altitudes ranging from 800 to 2200 m a.s.l. The following abbreviations are used:
 699 HCB = hexachlorobenzene; α -endo = alpha-endosulfan; β -endo = beta-endosulfan;
 700 eSO_4 = endosulfan sulphate (plotted in the right axis for water samples); Dield =
 701 dieldrin.





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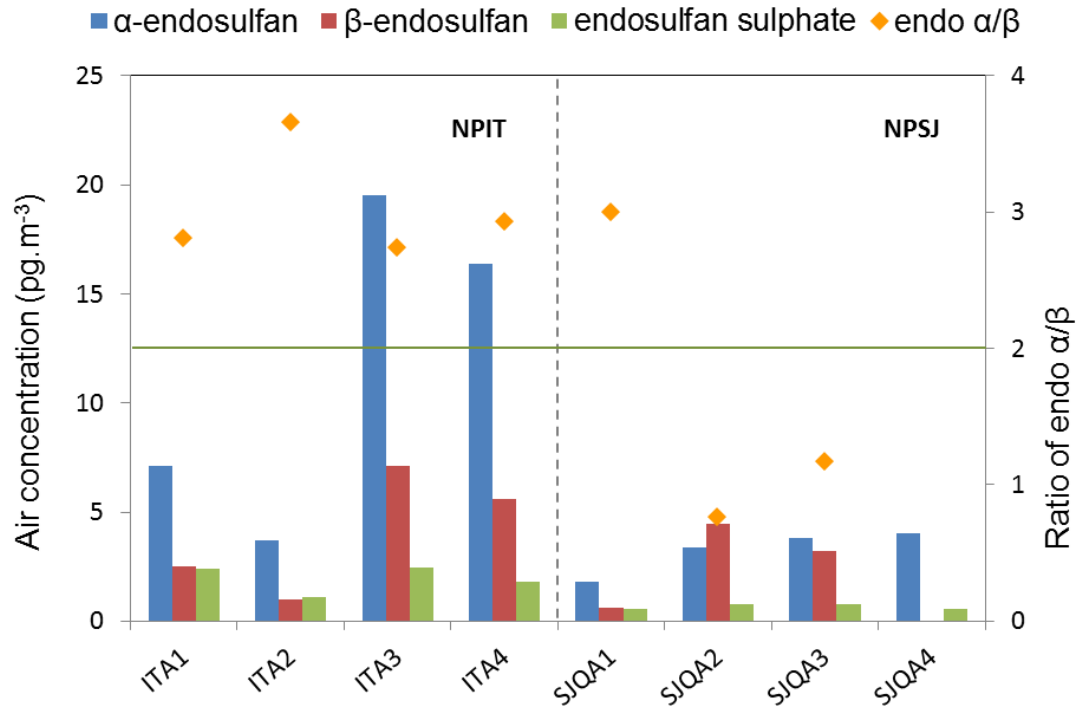
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719 **Figure 3.** Atmospheric endosulfan composition at the sampling sites. Green line
720 represents the average endo- α/β ratio value.

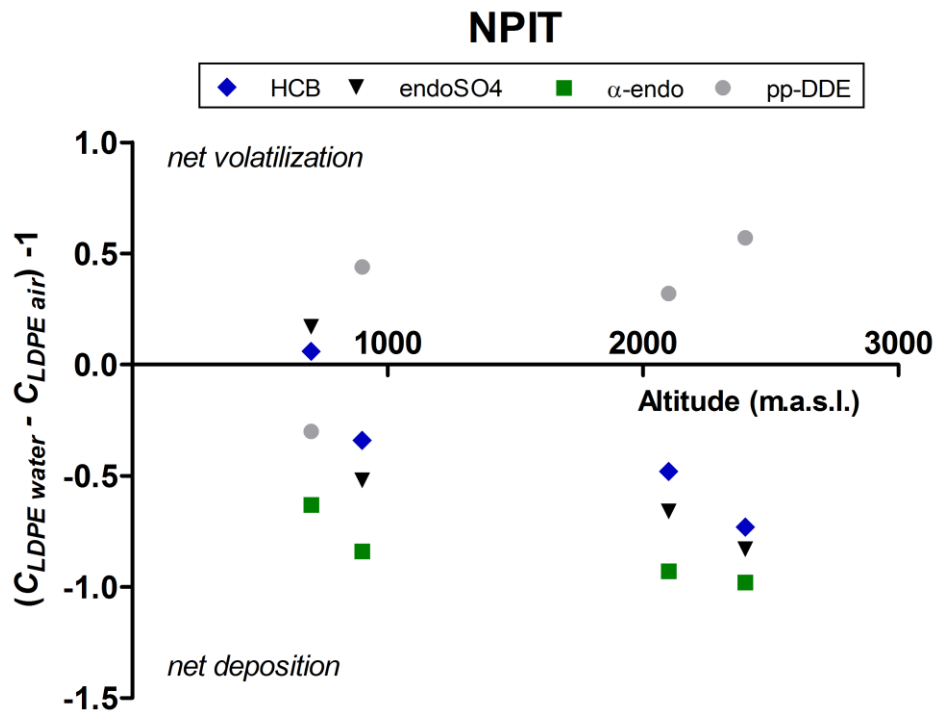
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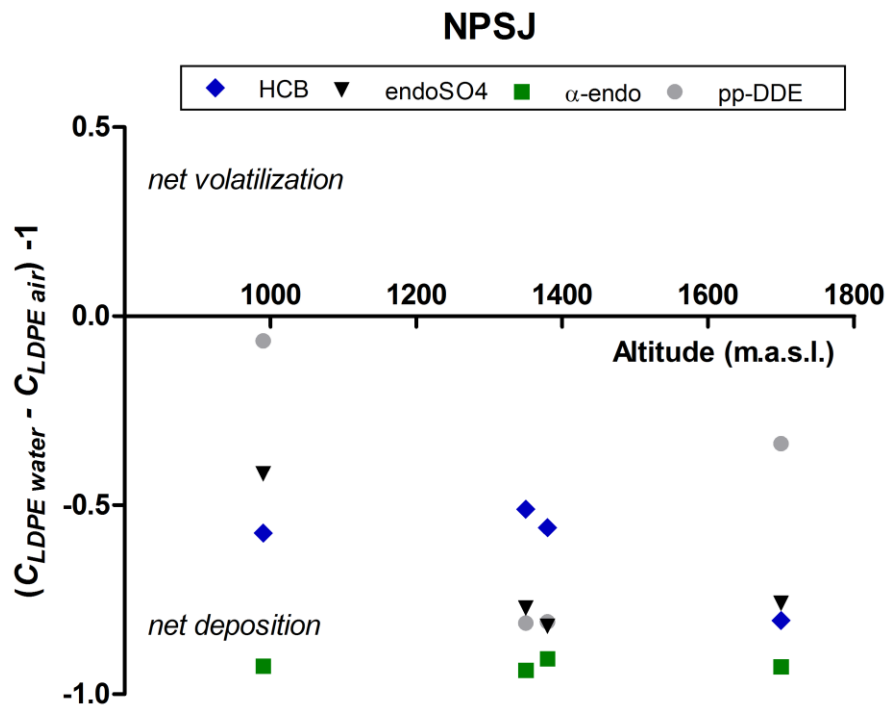
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724 **Figure 4.** Air-water exchange gradients along altitudinal transects at NPIT and NPSJ
 725 sites for HCB, α - endosulfan, endosulfan sulphate (endoSO₄), and pp-DDE. Positive
 726 values indicate net volatilisation of OCPs; negative values indicate net deposition.



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