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Concentrations, Fluxes and Residence Time of PBDEs Across the Tropical Atlantic Ocean

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Concentrations, Fluxes and Residence Time of PBDEs Across the Tropical Atlantic Ocean

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25 **INTRODUCTION**

26 Oceans have emerged as an important buffer and final sink for a wide range of persistent 27 organic pollutants $(POPs)$ ¹⁻⁴ There are four major pathways affecting POPs in the Oceans. 28 The most important pathway for persistent hydrophobic and lipophilic compounds is the 29 biological pump, which moves POPs to depth by partitioning into phytoplankton and settling 30 to depth 4.5 . POPs can also be moved to depth via the subduction of surface water 3 . Another 31 'physical' pathway is the movement to depth via eddy diffusion, which will be most important 32 for polar compounds which do not sorb strongly to organic carbon in the water column $⁶$.</sup> 33 Lastly, compounds can be prone to degradation, either biodegradation $⁷$ or by direct and</sup> 34 indirect photolysis in the surface water 8.9 . Potential sources of POPs to the Atlantic combine 35 atmospheric deposition and riverine transport from terrestrial sources. In the case of 36 polybrominated diphenylethers (PBDEs), there is the possibility that debromination reactions 37 results in the production of lower brominated congeners in situ 10 . Due to their low vapour 38 pressure, and high lipophilicity, their transport on particles in atmosphere and seawater is 39 likely the dominating process.

40 This is particularly true for the Atlantic Ocean, which is affected by emissions from America, Africa and Europe.^{11,12} It has been extensively studied on several transects for legacy 42 pollutants, such as polychlorinated biphenyls $(PCBs)^{11}$ and polycyclic aromatic hydrocarbons 43 (PAHs)¹³, polychlorinated dibenzo-p-dioxins and furans¹ and hexachlorocyclohexanes $14-16$. 44 Most of these transects occurred on European research vessels on their biannual migration 45 from the Arctic to the Antarctic and vice versa. These transects typically follow the western 46 side of the North Atlantic. Sampling on these transects was invariably affected by continental 47 emissions off Europe and Africa, making extrapolations across the entire Atlantic Ocean 48 difficult and fraught with uncertainties.

49 More recently, results of newer contaminants, such as brominated flame retardants and 50 perfluorinated alkyl acids and sulfonates have been published.¹⁷⁻²⁰ Yet we still know very 51 little about their sources, transport and fate on the ocean-scale. Our recent results suggested 52 that the remote oligotrophic Ocean has become a secondary source of $PCBs^{21,22}$. Contrarily, 53 PAHs continue to deposit into the Atlantic Ocean²³, possibly linked to their reactivity in the 54 surface water.¹³ Recent results showed the widespread presence of brominated flame 55 retardants, particularly polybrominated diphenylethers (PBDEs), at trace levels in air and 56 water across the Atlantic Ocean.^{17,24,25} Air-water exchange gradients implied on-going net 57 deposition of PBDEs into the surface waters. 17

58 A research cruise of the *R/V Endeavor* in July-August 2009 from Namibia via Barbados back 59 to her home port in Narragansett (RI), USA offered us a unique opportunity to determine 60 PBDE concentrations in air, water and air-water exchange gradients across the tropical 61 Atlantic Ocean (Figure 1). The cruise track covered both large east-west and north-south 62 gradients, and was mostly far away from shore. In addition to the traditional sampling of 63 PBDEs using active high-volume air and water sampling, we also deployed passive 64 polyethylene (PE) samplers as complementary approaches to measure truly dissolved 65 concentrations. Previous results suggested good agreement between active and passive 66 sampling (e.g., within a factor of 2 for PCBs).²¹

67 In summary, we collected and analyzed samples across the tropical Atlantic to deduce 68 whether (i) there were significant east-west and north-south gradients of selected PBDE 69 congeners; (ii) the tropical Atlantic Ocean was a net sink or secondary source of PBDEs; and 70 (iii) to compare active and passive sampling approaches for PBDEs; and (iv) estimate settling 71 fluxes of PBDEs in air and water to determine the turnover of PBDEs in these compartments.

73 **METHODS**

74 **Sampling**

75 *High volume water sampling*

76 Active air and water sampling were taken as detailed previously.²² In brief, a total of 57 water 77 samples were collected in the ship's laboratory from the ship's seawater pipe using a 78 sampling train, equipped with a pre-combusted glass fiber filter (GFF) and 3 polyurethane 79 foam (PUF) plugs in series. Water samples were collected continuously. Initially (samples 1- 80 33) collected 600-1100 L of water, after Barbados, 200-500 L were filtered (samples 34- 57).

81

82 *High volume air samples*

83 Air sampling was conducted when the wind came from across the bow to collect 84 uncontaminated marine boundary layer air. Previous work has shown that sampling of POPs 85 can easily be contaminated from the ship's emission (either indoor air or stack emission). 26 86 During this transect, winds often shifted away from the bow, such that only 47 air samples 87 were collected using a high-volume air sampler, equipped with a pre-combusted GFFs and 2 88 pre-extracted PUF plugs. The active air sampler was set-up at the front of the monkey's level 89 (above the bridge), facing the wind. Up to sample 23, volumes of $400 - 570$ m³ (~ 12 hours 90 each) were collected, after Barbados, volumes were $\sim 230 - 350$ m³ (~ 8 hours each).

91

92 *PE sheet samplers*

93 Passive samplers were deployed to (a) the indoor air for 1 -2 weeks each (n=7); (b) the marine 94 air for consecutive 48 hours (n=12); (c) to flowing seawater in the ship's laboratory for 95 consecutive 48 hours (n=13); and (d) samplers were towed behind the ship for 48 hours 96 outside country's exclusive economic zones (limiting these to n=9).

97 The air PE samplers were exposed in inverted stainless steel bowls ('UFO disk') on the ship's 98 main (chimney) mast 17.5 m above sea level (i.e., several meters above the high-volume 99 sampler). PE samplers were exposed continuously, even when hi-volume air sampling was 100 paused due to adverse winds. The water PE was fixed in a steel pipe connected to the flowing 101 seawater in side the ship's special purpose laboratory, at a nominal flow rate of 10 L min⁻¹. 102 PE samplers were towed on a line via the A-frame, ca. 100 m behind the ship, for 40-70 hours 103 each.

104

105 **Sample analysis**

106 *High volume samples (PUFs, GFFs)*

107 Analytical standards for PBDEs (calibration sets, natives and 13 C labeled) were purchased 108 from Wellington Laboratories (Canada). PUFs and GFFs were extracted using automated 109 warm Soxhlet extraction (40 minutes warm Soxhlet followed by 20 minutes of solvent 110 rinsing) with dichloromethane (DCM) in a B-811 extraction unit (Büchi, Switzerland). The 111 extracts of water PUF a GFF samples were first dried using $Na₂SO₄$, concentrated and then 112 split into 2 portions, $\frac{1}{4}$ of extract was used for PAHs analysis, $\frac{23}{12}$ the remaining $\frac{3}{4}$ of extract 113 were used for PBDEs, indicator PCBs and OCPs analysis.²¹

114

115 *Analysis*

116 Several ¹³C BDEs (28, 47, 99, 100, 153, 154, 183 and 209, all from Wellington Laboratories, 117 Canada) were added prior to extraction. The extract was cleaned-up on a $H₂SO₄$ modified 118 (30% w/w) silica column. Analytes were eluted with 40 mL DCM/*n*-hexane mixture (1:1). 119 The eluate was concentrated using stream of nitrogen in a TurboVap II concentrator unit and 120 transferred into an insert in a vial. The syringe standards $(^{13}C$ BDEs 77 and 138 and native 121 PCB 121) were added to all samples, the final volume was 50 uL.

122 HRGC/HRMS instrumental analysis for PBDEs was performed on 7890A GC (Agilent, USA) 123 equipped with a 15 m x 0.25 mm x 0.10 µm DB5 column (Agilent J&W, USA) coupled to 124 AutoSpec Premier MS (Waters, Micromass, UK). The MS was operated in EI+ mode at the 125 resolution of >10 000. For BDE 209, the MS resolution was set to >5 000. Injection was 126 splitless 1 μ L at 280°C, with He as carrier gas at 1 mL min⁻¹. The GC temperature program 127 was 80°C (1 min hold), then 20°C min⁻¹ to 250°C, followed by 1.5°C min⁻¹ to 260°C (2 min 128 hold) and 25° C min⁻¹ to 320°C (4.5 min hold).

129

130 *PE sheet samplers*

131 Blanks and exposed sheets of PE were rinsed with Millipore water, dried with a disposable 132 tissue and soaked for 24 hours in 200 mL of *n*-hexane followed by 24 hours in 200 mL of 133 DCM. The two solvents were then pooled; concentrated using stream of nitrogen in a 134 TurboVap II concentrator and the extract was split into 2 portions and processed using the 135 same procedure as the high volume samples.

136

137 **Quality assurance, Quality control**

138 The results for PBDEs in high volume samples are recovery corrected (recoveries ranged 139 from 34 to 110%, Table S7). Method performance was tested prior to sample preparation. 140 Four field blanks were extracted for water and air GFFs each; Four PUF blanks were 141 extracted for air and water samples each (one PUF blank was excluded for air samples) and 142 combined for blank correction and method detection limit (MDL) determination. The MDL 143 was calculated as 3 standard deviations of blank concentrations (for more details, see SI and 144 Tables S2-S6).

145

146 **Physicochemical properties**

147 Best-fit values of PE-water (K_{PEw}) equilibrium partition constants were taken from Lohmann $(2012)^{27}$, and corrected for average sampling temperature and salinity (Table S1). PE-air 149 (*K*PEa) values were calculated based on a linear regression of subcooled-liquid vapour pressure 150 (*p*L) established for PAHs, OCPs and PCBs (data not shown). Octanol-water partition constant 151 (K_{ow}) and p_L values were calculated on the correlation with molecular weight presented by 152 Dugani and Wania (2003).²⁸ The Setschnow constant was taken as 0.35, as reported 153 elsewhere.²⁹ Average values were taken for enthalpies of PE-water (25 kJ mol⁻¹) and PE-air 154 exchange (80 kJ mol⁻¹), close to calculated values of internal octanol-water and octanol-air 155 exchange. $30,31$

156

157 **Concentrations and fluxes from PE samplers**

158 Truly dissolved concentrations of PBDEs in water, C_{diss} (pg L^{-1} H₂O) and gas-phase, C_{gas} (pg 159 m^3) were derived from PE-normalized concentrations, C_{PE} (pg L⁻¹ PE):

160
$$
C_{\text{diss/gas}} = \frac{C_{PE}}{K_{PEw \text{ (or PEa)}} \times \left(1 - e^{\frac{R_s \times t}{K_{PEw \text{ (or PEa)}} \times V_{PE}}}\right)}
$$
(1)

161 Where R_s (m³ day⁻¹) is the overall sampling rate, t is the deployment time (days), V_{PE} the 162 volume of the PE sampler (m^3) and K_{PEw} (or K_{PEa}) is corrected for the average temperature 163 and salinity of the deployment (for details, see SI). Sampling rates were estimated based on 164 PRC loss and typical uptake rates.²¹ For towed samplers, we assumed equilibration was 165 achieved. In view of the high molecular weights (MW) of BDEs, R_s was modulated relative to 166 BDE 28 as a function of MW^{-0.4}, representing an average of previous relationships between R_s 167 and molecular size. 32

168 • Air-water exchange fluxes, F_{aw} (pg m⁻² day⁻¹), were calculated according to the two-film 169 theory. The air-water exchange velocity, k_{ol} (m d⁻¹), was estimated as detailed previously. ²³

170
$$
F_{a/w} = C_{diss} - \frac{C_{gas}}{K_{aw}} \times k_{ol}
$$
 (2)

171

172 **Meteorological and sea surface auxiliary measurements**

173 From the ship's routine measurements, we averaged values of latitude, longitude, surface 174 water temperature (T_{water}) , salinity, and fluorescence of the flow-through seawater; air 175 temperature (T_{air}) , relative humidity (RH), relative and absolute wind speed and direction 176 recorded every minute for each sampling period (see SI). Back – trajectories were calculated 177 for 5 days with 6 h steps at 300 m above sea level using HYSPLIT.³³ Concentrations of 178 dissolved organic carbon (DOC) were derived from relationships by Siegel et al. (2002), 179 mostly around 1 mg DOC L^{-1} . 34

180

181 **Cruise track**

182 On the water side, the cruise covered different major currents of the tropical Atlantic Ocean 183 (Figure 1). Currents were identified based on a combination of changes in temperature (and 184 salinity for the Amazon plume) and typical current fields 35 (see Table S9). In the southern 185 hemisphere, these were the Benguela, South Equatorial and North Brazil Current. Discharges 186 from the Amazon, the North Equatorial Current and Gulf Stream affected samples in the 187 northern hemisphere. We obtained back-trajectories to confirm air mass origin.²¹ Initially, we 188 encountered southeasterly trade winds, moving air masses in a westerly direction towards the 189 equator. We then passed the intertropical convergence zone (ITCZ), which was situated at 190 approximately 10 ° N during our cruise. The cruise then continued in the northeasterly trade 191 winds moving air masses towards the west along the equator in the NH. The last few samples 192 were affected by the westerlies, moving air masses eastwards across the Atlantic Ocean.

194 **RESULTS AND DISCUSSION**

195

196 **PBDE concentrations in the water**

197 *Particle-bound PBDE water concentrations*

198 Typical particle-bound concentrations of PBDEs in the surface water were low, at $\lt 1$ pg L^{-1} , 199 with the exception of BDE 209 (mean 7 pg L^{-1} , median 1.8 pg L^{-1}) (Table S18). The most 200 abundant BDE congeners were 47, 99, 100 and 85, which were detected in approximately 2/3 201 of all samples. In terms of concentrations, BDEs 47 and 99 displayed highest concentrations 202 in most samples. When above the MDL, BDE 209 dominated overall aqueous concentrations 203 of BDEs on particles.

204 A strong hemispherical gradient was apparent for particle-bound aqueous PBDEs. The ratio of 205 the mean concentrations was approximately 5-times higher in the northern than in the 206 southern hemisphere. For example, mean concentrations of BDEs 47 and 99 were 0.04 pg L^{-1} 207 in the southern hemisphere, but 0.3 pg L^{-1} in the northern hemisphere. Similarly, the 208 frequency of detection for all congeners was only 21% in the southern hemisphere, but 49% 209 in the northern hemisphere. Clearly, the southern tropical Atlantic Ocean is much cleaner (on 210 average 5-times) with respect to particle-bound PBDEs in the surface water than the northern 211 hemisphere tropical Atlantic. We note that the increase in particle-bound PBDEs began at 5 – 212 10 °N, not the equator itself (Table S18). This was probably the result of the ITCZ having 213 shifted northwards during the northern summer, as reflected in the back-trajectories. This also 214 implies that particle-bound PBDEs reflected fairly recent deposition events.

215 A closer look at the spatial distribution of particle-bound PBDEs showed that highest 216 concentrations were detected in the Amazon plume, the Gulf Stream Eddy and in the sample 217 closest to the U.S. coast (Figure 2). BDE 209 was only detected in a few samples from the

218 Amazon River plume, where concentrations reached 36 pg L^{-1} . In the sample taken in the Gulf 219 Stream, particle-bound BDE congener concentrations were not elevated (Figure 2).

220 Xie et al. (2011) published dissolved and particle-bound concentrations of PBDEs along a 221 North-South transect of the Eastern Atlantic Ocean in 2008.¹⁷ Particle bound BDE 222 concentrations were ≤ 0.1 pg L⁻¹ (BDE 209 was not analyzed). Highest concentrations were 223 observed close to Europe and off South Africa. Particle-bound concentrations of both cruises 224 were similar in magnitude, implying that contamination-free sampling of particles was 225 achieved on both cruises.

226

227 *Truly dissolved PBDE concentrations from passive sampling*

228 PBDEs were detected in the towed passive samplers deployed in the water, likely due to their 229 extremely high K_{PFW} values (causing strong enrichment in the polyethylene films) (see SI). 230 Total mass of PBDEs accumulated in the towed PE samplers were at least 10 times higher 231 than those in the flow-through water exposed PE samplers, highlighting the boundary-layer 232 limitations we encountered in the laboratory. This was also evident in the detection frequency 233 of PBDEs in both types of deployments. Whereas almost all PBDE congeners were regularly 234 detected in the towed passive sampler (overall detection frequency was 90%, except BDE 66, 235 which was never $>$ MDL), only $<$ 5 % of PBDEs were $>$ MDL in the flow-through passive 236 sampler exposure. Towing of passive samplers could be a potentially transformative way of 237 sampling truly dissolved concentrations in the open Oceans.³⁶ Concentrations of PBDEs were 238 greater by a factor of approximately 2 in the northern and western tropical Atlantic compared 239 to the southern and eastern section based on the towed passive sampler results.

240 The dominant congeners were BDEs 47 and 99 (Figure 3). Typical concentrations in the 241 towed PE were ~ 0.5 pg L⁻¹ for BDE 47 and around 0.1 pg L⁻¹ for BDEs 28, 99 and 100 242 (Table S12). Overall, the PE-derived concentrations were remarkably similar to those reported 243 by Xie et al. (2011) for the Eastern Atlantic Ocean on a *R/V Polarstern* cruise in 2008 (Figure 244 3).¹⁷ On that cruise, samples were taken with a classical active sampling train using PAD 245 resins and GFFs. Mean values reported by Xie et al. (2011) were 0.1 pg L^{-1} for BDEs 47 and 246 99, and a factor of 10 lower for other PBDEs (BDE 209 was not analyzed).¹⁷

247 BDE concentrations started to diverge for the highest molecular weight BDE congeners 248 (Figure 3), with PBDE concentrations from towed PE samplers suggesting much lower truly 249 dissolved concentrations than those from Xie et al. $(2011)'$ s active sampling.¹⁷ This could be 250 due to the assumption of all PBDEs having reached equilibrium between the water and the 251 samplers during the 2-day tows. The good agreement between both types of sampling and 252 cruises (~ factor of 2) for congeners with up to five bromines (BDE 100) suggests that 253 compounds with a log $K_{ow} \le 7$ had equilibrated. For congeners with six or more bromines, a 254 correction for non-equilibrium could be necessary, or a longer deployment time (reducing the 255 boundary layer through faster passive sampler movement seems unlikely to be achieved in the 256 field, unless through special design features). Alternatively, active sampling results could be 257 biased high due to the co-sampling of PBDEs bound to DOC.

258

259 *Dissolved PBDE concentrations from active sampling*

260 Dissolved concentrations from active sampling were, by and large, 8 times lower in the South 261 Atlantic than in the North Atlantic (Table S16). The dominant PBDE congeners across all 262 samples were 99, 47 and 85. In the South Atlantic, concentrations ranged from below MDL in 263 most samples towards 90 pg L^{-1} for PBDE 99. In the North Atlantic, concentrations increased 264 markedly in the Amazon River plume, where several PBDE congeners reached hundreds of 265 . pg L^{-1} . Two other regions of elevated concentrations were in the Gulf Stream Eddy, and the 266 sample closest to the US coast, where BDEs 47 and 99 reached several thousands pg L^{-1} . 267 These concentrations are far greater than reported by Xie et al $(2011)^{17}$, and the 268 concentrations deduced from our towed passive samplers (both below 1 pg L^{-1}). In fact, the 269 concentrations from active sampling on this cruise exceeded those reported for coastal 270 environments or Lake Michigan: dissolved ΣPBDEs of up to 60 pg L^{-1} were reported in the 271 coastal waters of Hong Kong³⁷, 100 – 200 pg L^{-1} for Izmir Bay³⁸, while Streets et al. (2006) 272 reported average concentration of ΣPBDEs of 18 pg L^{-1} for Lake Michigan³⁹.

273 The strong discrepancy between the dissolved PBDE concentrations reported by Xie et al. $(2011)^{17}$ and our BDE concentrations from passive sampling with our results from active 275 sampling is reason for concern. Apparently dissolved PBDE concentrations from active 276 sampling on our study were elevated by $10 - 1,000$ fold. We examined whether the collection 277 of colloids or microplastics in our PUF-based sampling approach could explain the difference. 278 We assumed 1 mg DOC L^{-1} during the cruise. DOC-water partition constants, K_{DOC} , were 279 taken either from Burkhard $(2000)^{40}$ for natural colloids, or values extrapolated from sediment 280 partitioning experiments by Wang et al. $(2011)^{41}$. In either case only up to 50% of the 281 congeners dominating total dissolved concentrations (BDEs 47, 100 and 99) were bound to 282 colloids. The subtropical gyres in the Atlantic Ocean are known to accumulate small pieces of 283 plastic debris, or microplastics.⁴² If half of the [DOC] consisted of microplastics with sorption 284 properties similar to PE, around 50% of BDE 47 would be colloidal-bound, and 80% of BDEs 285 99 and 100. These calculations imply that co-sampling of colloids or microplastics cannot 286 explain the observed active sampling results. Most likely, active sampling results became 287 contaminated before, during or after sampling.

288

289 **PBDE concentrations in the atmosphere**

290 *Particle-bound PBDEs in the atmosphere*

291 A wide range of PBDEs were routinely detected on particles in the marine boundary layer, 292 including BDEs 66, 100, 99, 154, 153 and 183, and often BDE 209 (Figure 4). When 293 detected, BDE 209 dominated overall concentrations (median 1.2 pg m⁻³), followed by BDE 294 99 (0.13 pg m⁻³). For all other BDEs, mean and median concentrations were < 0.1 pg m⁻³. 295 There were no significant differences between BDE concentrations on particles between both 296 hemispheres (Table S10).

297 Concentrations were slightly greater than those reported by Xie et al. (2011) across the 298 Eastern Atlantic Ocean (BDE 209 was not measured in that study).¹⁷ During their cruise, only 299 BDE 47 had a median concentration greater 0.1 pg $m⁻³$, with other congeners mostly below 300 detection limits.¹⁷

301

302 *True gas-phase PBDE concentrations from passive sampling*

303 Most PBDEs were detected routinely in the air passive samplers (Table S14). Greatest 304 concentrations were obtained for BDE 47 at $1 - 8$ pg m⁻³, followed by BDE 99 (< 4 pg m⁻³), 305 with the other BDEs < 1 pg m⁻³ (Figure 4). The intermittent detection of BDE 209, ranging $f(306 \text{ from } < \text{MDL} - 260 \text{ per } \text{m}^3)$, could be due to its episodic transport, or reflect contamination on-307 board. There was no significant difference between the two hemispheres in terms of gas-phase 308 PBDE concentrations. These concentrations are comparable to, or slightly higher than those 309 reported by Xie et al. (2011) for gas-phase PBDEs across the Eastern Atlantic (Figure 4).¹⁷ Li 310 et al. (2011) published combined gas+particle phase concentrations across the south-eastern 311 Atlantic Ocean.⁴³ Mean concentrations were a few pg $m⁻³$, similar to the concentrations 312 reported here.

313

314 *Gas-phase PBDE concentrations from active sampling*

315 In the southern hemisphere air samples, gas-phase PBDEs were routinely below detection 316 limits (on the order of 1 pg m⁻³ for most congeners, see Tables S16). In several southern hemisphere samples, all PBDE congeners were detected, with concentrations reaching 1 ng m⁻ 317

 $318³$ for BDEs 47 and 99 in sample NAM-AP-020. During the collection of that particular 319 sample, the high volume sampler motor was shut off for 12 hours due to adverse wind 320 conditions, but the PUFs and GFF left in place. The exceedingly elevated concentrations from 321 this sample imply ship's emission as a strong contamination source of PBDEs.

322 Concentrations of gas-phase BDEs increased in the northern hemisphere (i.e., once the ship 323 crossed the ITCZ), with most congeners being detected above MDL regularly. Typical 324 concentrations were around 100 pg m⁻³ for BDEs 47 and 99 and at or below 10 pg m⁻³ for the 325 other congeners. A strong north-to-south gradient was apparent, with ratios of mean or 326 median concentrations implying that concentrations of PBDEs were greater in the NH by 327 factors of 2-8. The comparison with the passive sampling results, and the BDE concentrations 328 reported previously by Xie et al. $(2011)^{17}$ imply that gas-phase samples were routinely 329 contaminated.²⁶

330

331 *Air concentrations inside the ship*

332 The PE samplers deployed inside the ship displayed elevated concentrations of PBDEs, 333 particularly BDE 47 at hundreds pg $m⁻³$ (Table S15). Other prominent congeners were BDE 99, 28 and 100 at tens pg m⁻³ on average. The sum BDEs averaged 600 pg m⁻³ inside the ship, 335 comparable to other published concentrations of PBDEs in indoor air. For example, a mean 336 concentration of 940 pg m⁻³ for ΣPBDEs was reported for indoor air in Toronto, with other 337 studies detecting $100 - 200$ pg m⁻³ for indoor air.⁴⁴ Much higher concentrations were reported 338 for air in offices in Sweden, with concentrations ranging up to 4,700 pg m^{-3,45}

339

340 **The ship as a major source of PBDE contamination of samples and the environment**

341 There are several lines of evidence that the *R/V Endeavor* itself was a major source of PBDEs,

342 and partially contaminated active samples taken on-board. First, the indoor air displayed

343 elevated concentrations of a wide range of PBDEs, in-line with concentrations observed in 344 office and residential buildings on-land.^{36,37} Second, field PUF blanks taken during active air 345 and water sampling resulted in elevated BDE concentrations (Tables S5-S6). Third, during a 346 couple of active sampling events, the high volume sampler (hi-vol) was shut off (but the PUFs 347 left in place) when the wind was coming from the side or stern of the ship. These samples 348 displayed strongly elevated concentrations of PBDEs. Fourth, active sampling resulted in 349 much greater PBDE concentrations than passive sampling. The gas-phase PE sampler was 350 mounted several meters above the hi-vol, implying that diffusive contamination from the ship 351 did not reach the PE sampler. Fifth, there is a noticeable increase in PBDE concentrations in 352 the PE field blanks, with lowest concentrations for the atmospheric PEs (exposed to marine 353 air) at 170 pg per sample. The PEs mounted in the ship's laboratory were accompanied by 354 field blanks containing on average 540 pg per sample, while the towed PEs accumulated on 355 average 5,300 pg per sample. Towed PEs were mounted on the back of the ship, where the 356 ship's circulated air and engine room ventilation could often be felt.

357

358 **Pollutant gradients across the Atlantic Ocean**

359 Our results suggest that the north-western tropical Atlantic carried 5-times higher 360 concentrations of PBDEs on particles, and twice as high truly dissolved concentrations than 361 the south-eastern Atlantic Ocean. Yet our atmospheric results do not show any significant 362 differences between both hemispheres and regions. Our results seem to contradict PBDE 363 concentrations reported by Xie et al. (2011), who reported higher atmospheric concentrations 364 during their northern Atlantic Ocean transect relative to the samples taken further south.¹⁷ For 365 their aqueous samples, there was no significant difference between dissolved or particle- 366 bound PBDE concentrations between the North and South Atlantic.¹⁷ We suggest that the 367 gradients observed on the different cruises have as much to do with the particular regions 368 being crossed rather than solely reflecting on northern versus southern hemisphere. The fact 369 that we observed greater aqueous concentrations of PBDEs in NH is mostly due to our cruise 370 transecting the remote southern tropical Atlantic (with low PBDE concentrations) but ended 371 up sampling the Amazon River, warm core, Gulf Stream, and US coastal seaboard in the 372 northern hemisphere.

373 Xie et al. (2011) sampled the atmosphere closer to Europe and Africa for most of their 374 samples taken in the northern hemisphere, but farther from shore during their southern 375 transect, which might explain the difference in atmospheric PBDE concentrations they 376 reported. 17

377

378 **Air-water exchange**

379 Air-water exchange gradients were calculated based on simultaneous passive sampler 380 deployments in air and water. Gradients were based on PBDE concentrations in PE (ng g^{-1}) 381 PE) at equilibrium.⁴⁶ In short, passive samplers, such as the PE we used, reflect the chemical 382 activity of the BDEs in their respective matrix (air and water in our case). The ratio of those 383 BDE activities (corrected for non-equilibrium) is the activity gradient across the air-water 384 interface.

385 In the water, results from towed PE samplers were used; in the air, the equilibrium-corrected 386 PBDE concentrations from the passive sampler deployed on the ship's mast were used. In 387 approximately 50% of possible cases, PBDEs were > MDL in both phases simultaneously 388 (Table S19). Net air-water exchange gradients strongly favoured deposition of PBDEs from 389 the gas-phase into the water phase. Gradients increased with increasing MW. For BDEs 28 390 and 47, net air-to-water exchange gradients ranged from 3-13. For the higher MW BDEs 85, 391 99 and 100, gradients increased to be in the range 10-50, and even greater for the higher MW

392 congeners. Our results are similar to the gradients by Xie et al. (2011), who reported net 393 deposition for all PBDEs along the eastern Atlantic Ocean.¹⁷

394 – Air-water exchange velocities ranged from $\lt 1$ to 36 cm day⁻¹, and decreased with increasing 395 MW (Table S22). Net gas-phase deposition fluxes ranged from 10s of pg $m⁻²$ day⁻¹ for BDEs 28 and 85 to around 1 ng m^{-2} day⁻¹ for BDE 47, 99 and 209 (Table S23). Similar values were 397 reported by Xie et al. (2011) . ¹⁷

398 We calculated the net annual gas-phase deposition of PBDEs, assuming these air-water 399 exchange fluxes were representative across the Atlantic Ocean. For all PBDEs reported here, 400 the gas-phase deposition flux is approximately 40 tons annually, of which 20 t are the median 401 flux for BDE 209.

402

403 **Deposition of PBDEs across the Atlantic Ocean**

404 We calculated the deposition flux of PBDEs on particles, $F_{part,dep}$ (pg m⁻² day⁻¹), across the 405 Atlantic Ocean as:

$$
406 \tFpart,dep = Cpart \times vdep
$$
\n(3),

407 where C_{part} is the particle-bound PBDE concentration (pg m⁻³) from this cruise and v_{dep} is the 408 deposition velocity taken as 0.1 cm s^{-1} as reported by DelVento and Dachs (2007).⁴⁷ This 409 translates into a half-live of particle-bound PBDEs in a 1000 m marine boundary layer of 410 around 6 days. Calculated deposition fluxes were dominated by BDE 209 (median flux of 100 411 pg m⁻² day⁻¹), BDEs 47 and 99 (median fluxes of 46 and 11 pg m⁻² day⁻¹). These fluxes 412 compare well to those calculated by Xie et al. (2011) for PBDEs across the Eastern Atlantic 413 Ocean.¹⁷

414 If the results reported here are representative across the Atlantic Ocean, the annual deposition 415 of particle-bound BDEs 47, 99, 100 and 85 will be around a ton scaled to the entire Atlantic 416 Ocean (surface area \sim 75 x 10⁶ km²). The fact that particle-bound PBDE concentrations from 417 our cruise were similar to those by Xie et al. (2011) support that assumption. The median 418 annual deposition flux for BDE 209 is ~ 7,500 kg for the Atlantic Ocean. The fluxes derived 419 above signal a significant annual deposition (gas-phase deposition of ~40 t and particulate-420 bound deposition of ~10 t) of PBDEs into the Atlantic Ocean .

421 Yet the current deposition flux of PBDEs pales in comparison to production volumes of these 422 flame retardants. The estimated cumulative use of the commercial pentaBDE mixture alone 423 was estimated to 100 000 t since 1970.⁴⁸ This indicates that more is to come, as PBDEs 424 slowly make their way from products into the environment. Relative to PCBs, the gas-phase 425 deposition fluxes of PBDEs are similar in magnitude. 4 Yet the particle-bound deposition flux 426 of PBDEs to the Atlantic Ocean already exceeds that of PCBs, which was estimated to be 427 around 2.2 tons per year.⁴⁹ A greater fraction of PBDEs is bound to particles, explaining the 428 derived greater importance of their particle-bound deposition relative to PCBs.

429

430 **Settling fluxes of PBDEs out of the mixed layer**

431 We estimated settling fluxes of PBDEs out of the mixed layer depth across the tropical 432 Atlantic Ocean. The settling flux of particulate organic carbon (POC) at 100 m depth, *FOC,100*, 433 was estimated based on its relationship with Chlorophyll *a* (Chl a)⁵⁰:

434 Log
$$
(F_{OC,100)}
$$
 = 2.09 + 0.81×Log(Chl α) (4)

435

436 Resulting settling fluxes of OC at 100 m depth were typically $30 - 80$ mg m⁻² day⁻¹, typical 437 for the tropical oceans.⁵¹ To derive a settling rate of particles out of the mixed layer depth, we 438 estimated POC concentrations based on published 10-year averages for July as reported by 439 Stramska (2009).⁵² We took POC concentrations as 80 mg m⁻³ up to the equator, 40 from the 440 equator to 5 °N, 30 from 5 to 10 °N, 20 from 10 to 30 °N, 30 from 30 to 35 °N and 80 north 441 of 45 °N. Dividing $F_{OC,100}$ by POC concentrations yields the settling velocity⁵³, which ranged 442 from $0.5 - 2.0$ m day⁻¹. Lastly, the product of settling velocity and particle-bound PBDE 443 concentrations yielded the PBDE removal flux from the surface ocean layer. This PBDE 444 export flux rarely exceeded 1 ng m^{-2} day⁻¹ for most congeners, except for BDE 209, which 445 reached fluxes of tens ng m⁻² day⁻¹ (Figure S3). Settling fluxes were greater in the northern 446 hemisphere than south of the equator, driven by the greater frequency of detection and 447 concentrations of PBDE in the particle-phase (see above).

448

449 **Comparison of PBDE fluxes**

450 We compared the fluxes of gas-phase and particle-bound PBDEs out of the atmosphere to 451 those out of the Atlantic Ocean's mixed layer depth. Net gas-phase deposition fluxes were 452 much greater than the particle-bound settling fluxes, implying additional removal from the 453 surface mixed layer, probably via photolytic and microbial degradation.⁵ The median values 454 of both particle-bound settling fluxes agreed surprisingly well in view of the assumptions 455 behind the settling velocities in both atmosphere and surface water (Figure 6). Settling fluxes 456 were around 50 pg m^{-2} day⁻¹ for BDE 47 in both atmosphere and surface water, and < 10 pg 457 m⁻² day⁻¹ for other congeners. For PBDEs 47 through 99, settling fluxes from the surface 458 water exceeded atmospheric deposition fluxes, while for the higher brominated congeners, 459 median settling fluxes out of the mixed layer depth (MLD) were zero due to PBDEs being < 460 MDL in the water.

461

462 **Residence time of particle-bound PBDE in surface water**

463 The MLD was taken from published 10-year averages for July as reported by Stramska 464 (2009).⁵² It ranged from 80 m up to the equator, 40 m from the equator to 20 °N, and 20 m 465 north of 20 °N. The product of MLD and total (sum of truly dissolved + particle-bound) 466 PBDEs vielded the mass loading of PBDEs on particles (ng m^{-2}) in the surface water. The 467 mass loading divided by the PBDE removal flux (ng m^{-2} day⁻¹) gave the residence time (days) 468 of particle-bound PBDEs in the surface water of the Atlantic Ocean, assuming particle settling 469 was the only removal pathway. The tight coupling of particle-bound PBDE fluxes from 470 atmosphere and out of the mixed layer could indicate that different fates of particle-bound 471 PBDEs (settling) and those derived from gaseous diffusion, probably due to chemical and 472 biological degradation of dissolved molecules in the water column. In the northern 473 hemisphere, median PBDEs' residence were several weeks, but were around one year in the 474 southern hemisphere (Figure S5). These values suggest that settling on particles is an 475 important removal pathway in the northern hemisphere, whereas biological and photolytic 476 degradation are probably determining the fate of PBDEs in the southern hemisphere.

477

478 **Implications**

479 Our results imply that PBDEs continue to be deposited across the Atlantic Ocean. Our 480 research vessel proved to be a significant source of contamination for PBDEs itself during 481 active sampling of air and water. Yet the use of passive samplers enabled us to retrieve valid 482 samples for both gas-phase and truly dissolved PBDEs. Other researchers have shown that 483 other contamination-free sampling can be achieved, though on a different research vessels. 17,43 484 Our results suggest that atmospheric deposition of particle-bound PBDEs is probably 485 slightly lower but similar in magnitude to their removal flux on settling particles from the 486 surface mixed layer. Net gas-phase deposition of PBDEs exceeded particle-bound deposition, 487 suggesting that there are other important removal pathways for dissolved BDEs, such as 488 microbial and photolyic degradation. The total deposition of PBDEs to the Atlantic Ocean, 489 around tens of tons per year, pales in comparison to their historical production volumes. This 490 might indicate the less efficient long-range transport of these higher molecular weight 491 compounds, compared to, for example, polychlorinated biphenyls. It could also indicate that 492 PBDEs are only slowly emitted from their current use, guaranteeing continuous fluxes into 493 the oceans in the future.⁵⁴

494

495 **Supporting Information Available**

496 The Supporting information includes additional details, such as physicochemical constants,

497 instrumental analysis and QC, sampling set-up and sample details, air and water 498 concentrations. This material is available free of charge via the Internet at http://pubs.acs.org.

- 499
- 500

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507

508 **References**

- 509
- 510 (1) Nizzetto, L.; Lohmann, R.; Gioia, R.; Dachs, J.; Jones, K. C. Atlantic Ocean surface 511 waters buffer declining atmospheric concentrations of persistent organic pollutants *Environ*
- 512 *Sci Technol* **2010**, *44*, 6978–6984.
- 513 (2) Iwata, H.; Tanabe, S.; Sakai, N.; Tatsukawa, R. Distribution of Persistent
- 514 Organochlorines in the Oceanic Air and Surface Seawater and the Role of Oceans on Their 515 Global Transport and Fate *Environ. Sci. Technol.* **1993**, *27*, 1080-1098.
- 516 (3) Lohmann, R.; Jurado, E.; Pilson, M. E. Q.; Dachs, J. In *Geophysical Research Letters*, 517 2006; Vol. 33, pp L12607, 12604 PP, doi:12610.11029/12006GL025953.
- 518 (4) Dachs, J.; Lohmann, R.; Ockenden, W. A.; Mejanelle, L.; Eisenreich, S. J.; Jones, K.
- 519 C. Oceanic biogeochemical controls on global dynamics of persistent organic pollutants 520 *Environmental Science & Technology* **2002**, *36*, 4229-4237.
- 521 (5) Galban-Malagon, C.; Berrojalbiz, N.; Ojeda, M.-J.; Dachs, J. The oceanic biological 522 pump modulates the atmospheric transport of persistent organic pollutants to the Arctic *Nat* 523 *Commun* **2012**, *3*, 862.
- 524 (6) Lohmann, R.; Jurado, E.; Dijkstra, H. A.; Dachs, J. Vertical eddy diffusion as a key
- 525 mechanism for removing perfluorooctanoic acid (PFOA) from the global surface oceans 526 *Environmental Pollution* **2013**, *179*, 88-94.
- 527 (7) Galb \tilde{A} in-Malag \tilde{A} ³n, C. b. J.; Del Vento, S.; Berrojalbiz, N.; Ojeda, M.-J.; Dachs, J. 528 Polychlorinated Biphenyls, Hexachlorocyclohexanes and Hexachlorobenzene in Seawater and
- 529 Phytoplankton from the Southern Ocean (Weddell, South Scotia, and Bellingshausen Seas) 530 *Environmental Science & Technology* **2013**, *47*, 5578-5587.
-
- 531 (8) Schwarzenbach, R. P.; Gschwend, P. M.; Imboden, D. M. *Environmental Organic* 532 *Chemistry*; 2nd ed.; Wiley-Interscience: New Jersey, 2003.
- 533 (9) Latch, D. E.; Packer, J. L.; Arnold, W. A.; McNeill, K. Photochemical conversion of
- 534 triclosan to 2,8-dichlorodibenzo-p-dioxin in aqueous solution *Journal of Photochemistry and* 535 *Photobiology a-Chemistry* **2003**, *158*, 63-66.
- 536 (10) Schenker, U.; Soltermann, F.; Scheringer, M.; Hungerbühler, K. Modeling the
- 537 Environmental Fate of Polybrominated Diphenyl Ethers (PBDEs): The Importance of
- 538 Photolysis for the Formation of Lighter PBDEs *Environ Sci Technol* **2008**, *42*, 9244–9249.
- 539 (11) Gioia, R.; Eckhardt, S.; Breivik, K.; Jaward, F. M.; Prieto, A.; Nizzetto, L.; Jones, K.
- 540 C. Evidence for Major Emissions of PCBs in the West African Region *Environmental Science* 541 *& Technology* **2011**, *45*, 1349-1355.
- 542 (12) Jaward, F. M.; Barber, J. L.; Booij, K.; Jones, K. C. Spatial distribution of atmospheric
- 543 PAHs and PCNs along a north-south Atlantic transect *Environmental Pollution* **2004**, *132*, 544 173-181.
- 545 (13) Nizzetto, L.; Lohmann, R.; Gioia, R.; Jahnke, A.; Temme, C.; Dachs, J.; Herckes, P.;
- 546 Di Guardo, A.; Jones, K. C. PAHs in air and seawater along a North-South Atlantic transect:
- 547 Trends, processes and possible sources *Environmental Science & Technology* **2008**, *42*, 1580- 548 1585.
- 549 (14) Xie, Z.; Koch, B. P.; Moller, A.; Sturm, R.; Ebinghaus, R. Transport and fate of
- 550 hexachlorocyclohexanes in the oceanic air and surface seawater *Biogeosciences Discuss* **2011**, 551 *8*, 2621-2633.
- 552 (15) Lakaschus, S.; Weber, K.; Wania, F.; Bruhn, R.; Schrems, O. The air-sea equilibrium
- 553 and time trend of hexachlorocyclohexanes in the Atlantic Ocean between the Arctic and 554 Antarctica *Environmental Science & Technology* **2002**, *36*, 138-145.
- 555 (16) Xie, Z.; Koch, B. P.; Moller, A.; Sturm, R.; Ebinghaus, R. Transport and fate of
- 556 hexachlorocyclohexanes in the oceanic air and surface seawater *Biogeosciences* **2011**, *8*, 557 2621-2633.
- 558 (17) Xie, Z. Y.; Moller, A.; Ahrens, L.; Sturm, R.; Ebinghaus, R. Brominated Flame 559 Retardants in Seawater and Atmosphere of the Atlantic and the Southern Ocean
- 560 *Environmental Science & Technology* **2011**, *45*, 1820-1826.
- 561 (18) Benskin, J. P.; Muir, D. C. G.; Scott, B. F.; Spencer, C.; De Silva, A. O.; Kylin, H.;
- 562 Martin, J. W.; Morris, A.; Lohmann, R.; Tomy, G.; Rosenberg, B.; Taniyasu, S.; Yamashita,
- 563 N. Perfluoroalkyl Acids in the Atlantic and Canadian Arctic Oceans *Environmental Science &* 564 *Technology* **2012**, *46*, 5815-5823.
- 565 (19) Yamashita, N.; Kannan, K.; Taniyasu, S.; Horii, Y.; Petrick, G.; Gamo, T. A global 566 survey of perfluorinated acids in oceans *Marine Pollution Bulletin* **2005**, *51*, 658-668.
- 567 (20) Yamashita, N.; Taniyasu, S.; Petrick, G.; Wei, S.; Gamo, T.; Lam, P. K. S.; Kannan,
- 568 K. Perfluorinated acids as novel chemical tracers of global circulation of ocean waters 569 *Chemosphere* **2008**, *70*, 1247-1255.
- 570 (21) Lohmann, R.; Klánová, J.; Kukucka, P.; Yonis, S.; Bollinger, K. PCBs and OCPs on a
- 571 East-to-West transect: The importance of major currents and net volatilization for PCBs in the
- 572 Atlantic Ocean *Environ Sci Technol* **2012**, *46*, 10471–10479, Doi: 10420.11021/es203459e.
- 573 (22) Zhang, L.; Lohmann, R. Cycling of PCBs and HCB in the Surface Ocean-Lower
- 574 Atmosphere of the Open Pacific *Environmental Science & Technology* **2010**, *44*, 3832–3838.
- 575 (23) Lohmann, R.; Klanova, J.; Pribylova, P.; Liskova, H.; Yonis, S.; Bollinger, K. PAHs
- 576 on a West-to-East Transect Across the Tropical Atlantic Ocean *Environmental Science &*
- 577 *Technology* **2013**, *47*, 2570-2578.
- 578 (24) Moller, A.; Xie, Z. Y.; Sturm, R.; Ebinghaus, R. Polybrominated diphenyl ethers
- 579 (PBDEs) and alternative brominated flame retardants in air and seawater of the European 580 Arctic *Environmental Pollution*, *159*, 1577-1583.
- 581 (25) Moller, A.; Xie, Z. Y.; Sturm, R.; Ebinghaus, R. Large-Scale Distribution of
- 582 Dechlorane Plus in Air and Seawater from the Arctic to Antarctica *Environmental Science &*
- 583 *Technology*, *44*, 8977-8982.
- 584 (26) Lohmann, R.; Jaward, F. M.; Durham, L.; Barber, J. L.; Ockenden, W.; Jones, K. C.;
- 585 Bruhn, R.; Lakaschus, S.; Dachs, J.; Booij, K. Potential contamination of shipboard air
- 586 samples by diffusive emissions of PCBs and other organic pollutants: Implications and 587 solutions *Environmental Science & Technology* **2004**, *38*, 3965-3970.
- 588 (27) Lohmann, R. A critical review of low-density polyethylene's partitioning and
- 589 diffusion coefficients for trace organic contaminants and implications for its use as a passive 590 sampler *Environ Sci Technol* **2012**, *36*, 606-618, DOI: 610.1021/es202702y.
- 591 (28) Wania, F.; Dugani, C. Assessing the long-range transport potential of polybrominated
- 592 diphenyl ethers: A comparison of four multimedia models *Environmental Toxicology and* 593 *Chemistry* **2003**, *22*, 1252-1261.
- 594 (29) Jonker, M. T. O.; Muijs, B. Using solid phase micro extraction to determine salting-
- 595 out (Setschenow) constants for hydrophobic organic chemicals *Chemosphere* **2010**, *80*, 223- 596 227.
- 597 (30) Beyer, A.; Wania, F.; Gouin, T.; Mackay, D.; Matthies, M. Selecting internally
- 598 consistent physicochemical properties of organic compounds *Environmental Toxicology and* 599 *Chemistry* **2002**, *21*, 941-953.
- 600 (31) Schenker, U.; MacLeod, M.; Scheringer, M.; Hungerbuhler, K. Improving data quality
- 601 for environmental fate models: A least-squares adjustment procedure for harmonizing 602 physicochemical properties of organic compounds *Environ. Sci. Technol.* **2005**, *39*, 8434-
- 603 8441.
- 604 (32) Lohmann, R.; Booij, K.; Smedes, F.; Vrana, B. Use of passive sampling devices for 605 monitoring and compliance checking of POP concentrations in water *Environmental Science*
- 606 *and Pollution Research* **2012**, *19*, 1885-1895.
- 607 (33) NOAA: http://www.arl.noaa.gov.
- 608 (34) Siegel, D. A.; Maritorena, S.; Nelson, N. B.; Hansell, D. A.; Lorenzi-Kayser, M.
- 609 Global distribution and dynamics of colored dissolved and detrital organic materials *Journal* 610 *of Geophysical Research-Oceans* **2002**, *107*, 14.
- 611 (35) Gyory, J.; Mariano, A. J.; Ryan., E. H., 2001-2013; Vol. 2013.
- 612 (36) Allan, I. J.; Harman, C. Global Aquatic Passive Sampling: Maximizing Available
- 613 Resources Using a Novel Exposure Procedure *Environmental Science & Technology* **2011**,
- 614 *45*, 6233-6234.
- 615 (37) Wurl, O.; Lam, P. K. S.; Obbard, J. P. Occurrence and distribution of polybrominated
- 616 diphenyl ethers (PBDEs) in the dissolved and suspended phases of the sea-surface microlayer 617 and seawater in Hong Kong, China *Chemosphere* **2006**, *65*, 1660-1666.
- 618 (38) Cetin, B.; Odabasi, M. Air-water exchange and dry deposition of polybrominated
- 619 diphenyl ethers at a coastal site in Izmir Bay, Turkey *Environmental Science & Technology* 620 **2007**, *41*, 785-791.
- 621 (39) Streets, S. S.; Henderson, S. A.; Stoner, A. D.; Carlson, D. L.; Simcik, M. F.;
- 622 Swackhamer, D. L. Partitioning and Bioaccumlation of PBDEs and PCBs in Lake Michigan
- 623 *Environ Sci Technol* **2006**, *40*, 7263-7269.
- 624 (40) Burkhard, L. P. Estimating dissolved organic carbon partition coefficients for nonionic
- 625 organic chemicals *Environmental Science & Technology* **2000**, *34*, 4663-4668.
- 626 (41) Wei Wang; Delgado-Moreno, L.; Ye, Q.; Gan, J. Improved Measurements of Partition
- 627 Coefficients for Polybrominated Diphenyl Ethers *Environ Sci Technol* **2011**, *45*, 1521–1527.
- 628 (42) Kara Lavender Law; Skye Morét-Ferguson; Nikolai A. Maximenko; Giora
- 629 Proskurowski; Emily E. Peacock; Jan Hafner; Reddy, C. M. Plastic Accumulation in the 630 North Atlantic Subtropical Gyre *Science* **2010**, *329*, 1185-1188.
- 631 (43) Li, J.; Li, Q. L.; Gioia, R.; Zhang, Y. L.; Zhang, G.; Li, X. D.; Spiro, B.; Bhatia, R. S.;
- 632 Jones, K. C. PBDEs in the atmosphere over the Asian marginal seas, and the Indian and
- 633 Atlantic oceans *Atmospheric Environment* **2011**, *45*, 6622-6628.
- 634 (44) Zhang, X.; Diamond, M.; Robson, M.; Harrad, S. Sources, Emissions, and Fate of
- 635 Polybrominated Diphenyl Ethers and Polychlorinated Biphenyls Indoors in Toronto, Canada 636 *Environ Sci Technol* **2011**, *45*, 3268-3274.
- 637 (45) Bjorklund, J. A.; Thuresson, K.; Cousins, A. P.; Sellstrom, U.; Emenius, G.; de Wit, C.
- 638 Indoor Air Is a Significant Source of Tri-decabrominated Diphenyl Ethers to Outdoor Air via 639 Ventilation Systems *Environ Sci Technol* **2012**, *46*, 5876-5884.
- 640 (46) Morgan, E.; Lohmann, R. Detecting Air-Water and Surface-Deep Water Gradients of
- 641 PCBs Using Polyethylene Passive Samplers *Environ Sci Technol.* **2008**, 7248-7253.
- 642 (47) Del Vento, S.; Dachs, J. Atmospheric occurrence and deposition of polycyclic
- 643 aromatic hydrocarbons in the northeast tropical and subtropical Atlantic Ocean *Environmental*
- 644 *Science & Technology* **2007**, *41*, 5608-5613.
- 645 (48) UNEP "Report of the Persistant Organic Pollutants Review committee on the work of
- 646 its third meeting addendum, Risk management evaluation on commercial
- 647 pentabromodiphenylether," 2007.
- 648 (49) Jurado, E.; Jaward, F. M.; Lohmann, R.; Jones, K. C.; Simo, R.; Dachs, J.
- 649 Atmospheric dry deposition of persistent organic pollutants to the Atlantic and inferences for 650 the global oceans *Environmental Science & Technology* **2004**, *38*, 5505-5513.
- 651 (50) Baines, S. B.; Pace, M. L. Why does the relationshop between sinking flux and
- 652 planktonic primary production differ between lakes and oceans? *Limnology Oceanography* 653 **1994**, *39*, 213-226.
- 654 (51) Henson, S. A.; Sanders, R.; Madsen, E. Global patterns in efficiency of particulate
- 655 organic carbon export and transfer to the deep ocean *Global Biogeochemical Cycles* **2012**, *26*.
- 656 (52) Stramska, M. Particulate organic carbon in the global ocean derived from SeaWiFS
- 657 ocean color *Deep-Sea Res. Part I-Oceanogr. Res. Pap.* **2009**, *56*, 1459-1470.
- 658 (53) Lohmann, R.; Dachs, J. Deposition of Dissolved and Particulate-Bound Chemicals
- 659 from the Surface Ocean. In *Handbook of Chemical Mass Transport in the Environment*;
- 660 Thibodeaux, L. J., Mackay, D., Eds.; CRC Press: Boca Raton, FL, 2011; p 631.
- 661 (54) Harrad, S.; Diamond, M. L. New Directions: Exposure to polybrominated diphenyl
- 662 ethers (PBDEs) and polychlorinated biphenyls (PCBs): Current and future scenarios
- 663 *Atmospheric Environment* **2006**, *40*, 1187-1188.
- 664 (55) Feldman, G. C.; NASA Goddard Space Flight Center, 2013.
- 665

Figure 1: Cruise track of EN 464 with chlorophyll a concentrations derived from MODIS ⁵⁵

Figure 2: Concentrations of aqueous particle-bound PBDE congeners (pg L⁻¹) in 2009 673 674

675

Note: concentrations of BDE 209 are displayed on right-hand side

677 **Figure 3: Dissolved concentrations of PBDE congeners derived from towed passive PE** 578 samplers (pg L^{-1}) across the tropical Atlantic Ocean in 2009. Displayed are average concentrations for the southern and northern hemisphere, and mean and median results concentrations for the southern and northern hemisphere, and mean and median results $\frac{680}{2000}$ from Xie et al. $(2011)^{17}$ for comparison.

682 Note: BDE 209 was not analyzed by Xie et al. $(2011)^{17}$

Figure 4: Particle-bound atmospheric concentrations of PBDE congeners (pg m-3 685 **) in** 686 **2009**

688

690

Figure 5: Gas-phase atmospheric concentrations of PBDE congeners derived from passive PE samplers (pg m⁻³) across the tropical Atlantic Ocean in 2009. Displayed are average concentrations for the southern and northern hemisphere, and mean and average concentrations for the southern and northern hemisphere, and mean and 695 median results from Xie et al. (2011)¹⁷ for comparison.

697
698

- **Figure 6: Comparison of median fluxes for PBDEs (pg m-2 day-1) from net gas-phase and**
- **particle-bound deposition out of the atmosphere and via particle-bound settling out of**

the mixed layer depth

