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Concentrations, Fluxes and Residence Time of PBDEs Across the Tropical Atlantic Ocean

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1	Concentrations, fluxes and residence time of PBDEs across the tropical Atlantic
2	Ocean
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9	Abstract
10	Little is known about the fate of polybrominated diphenylethers (PBDEs) across the Oceans.
11	Air and water were sampled using both active and passive polyethylene samplers on an East-
12	West transect across the tropical Atlantic Ocean in 2009, and analyzed for PBDEs. Typical
13	particle-bound concentrations of PBDEs in the surface water were low, at < 1 pg L^{-1} . Truly
14	dissolved concentrations from passive samplers were ~ 0.5 pg L^{-1} for BDE 47 and around 0.1
15	pg L ⁻¹ for BDEs 28, 99 and 100 (results from active samples were compromised). In the
16	atmosphere, particle-bound BDE 209 dominated overall concentrations (median 1.2 pg m ⁻³),
17	followed by BDE 99 (0.13 pg m ^{-3}). Gas-phase concentrations based on passive samplers were
18	1–8 pg m ⁻³ for BDE 47, and \leq 4 pg m ⁻³ for BDE 99. Net air-water exchange gradients
19	strongly favoured gas-phase deposition of PBDEs into the water. Net gas-phase deposition
20	fluxes ranged from 10s of pg m ⁻² day ⁻¹ for BDEs 28 and 85 to around 1 ng m ⁻² day ⁻¹ for BDE
21	47, 99 and 209. Settling fluxes of particle-bound PBDEs in atmosphere and surface water
22	were around 50 pg m ⁻² day ⁻¹ for BDE 47, and < 10 pg m ⁻² day ⁻¹ for the other congeners.
23	

25 INTRODUCTION

Oceans have emerged as an important buffer and final sink for a wide range of persistent 26 organic pollutants (POPs).¹⁻⁴ There are four major pathways affecting POPs in the Oceans. 27 The most important pathway for persistent hydrophobic and lipophilic compounds is the 28 29 biological pump, which moves POPs to depth by partitioning into phytoplankton and settling to depth ^{4,5}. POPs can also be moved to depth via the subduction of surface water ³. Another 30 31 'physical' pathway is the movement to depth via eddy diffusion, which will be most important for polar compounds which do not sorb strongly to organic carbon in the water column 6 . 32 Lastly, compounds can be prone to degradation, either biodegradation 7 or by direct and 33 indirect photolysis in the surface water ^{8,9}. Potential sources of POPs to the Atlantic combine 34 atmospheric deposition and riverine transport from terrestrial sources. In the case of 35 polybrominated diphenylethers (PBDEs), there is the possibility that debromination reactions 36 results in the production of lower brominated congeners in situ¹⁰. Due to their low vapour 37 pressure, and high lipophilicity, their transport on particles in atmosphere and seawater is 38 39 likely the dominating process.

40 This is particularly true for the Atlantic Ocean, which is affected by emissions from America, Africa and Europe.^{11,12} It has been extensively studied on several transects for legacy 41 pollutants, such as polychlorinated biphenyls (PCBs)¹¹ and polycyclic aromatic hydrocarbons 42 $(PAHs)^{13}$, polychlorinated dibenzo-*p*-dioxins and furans¹ and hexachlorocyclohexanes¹⁴⁻¹⁶. 43 44 Most of these transects occurred on European research vessels on their biannual migration 45 from the Arctic to the Antarctic and vice versa. These transects typically follow the western side of the North Atlantic. Sampling on these transects was invariably affected by continental 46 emissions off Europe and Africa, making extrapolations across the entire Atlantic Ocean 47 48 difficult and fraught with uncertainties.

49 More recently, results of newer contaminants, such as brominated flame retardants and perfluorinated alkyl acids and sulfonates have been published.¹⁷⁻²⁰ Yet we still know very 50 little about their sources, transport and fate on the ocean-scale. Our recent results suggested 51 that the remote oligotrophic Ocean has become a secondary source of PCBs^{21,22}. Contrarily, 52 PAHs continue to deposit into the Atlantic Ocean²³, possibly linked to their reactivity in the 53 surface water.¹³ Recent results showed the widespread presence of brominated flame 54 retardants, particularly polybrominated diphenylethers (PBDEs), at trace levels in air and 55 water across the Atlantic Ocean.^{17,24,25} Air-water exchange gradients implied on-going net 56 deposition of PBDEs into the surface waters.¹⁷ 57

A research cruise of the *R/V Endeavor* in July-August 2009 from Namibia via Barbados back 58 59 to her home port in Narragansett (RI), USA offered us a unique opportunity to determine PBDE concentrations in air, water and air-water exchange gradients across the tropical 60 61 Atlantic Ocean (Figure 1). The cruise track covered both large east-west and north-south gradients, and was mostly far away from shore. In addition to the traditional sampling of 62 63 PBDEs using active high-volume air and water sampling, we also deployed passive 64 polyethylene (PE) samplers as complementary approaches to measure truly dissolved 65 concentrations. Previous results suggested good agreement between active and passive sampling (e.g., within a factor of 2 for PCBs).²¹ 66

In summary, we collected and analyzed samples across the tropical Atlantic to deduce whether (i) there were significant east-west and north-south gradients of selected PBDE congeners; (ii) the tropical Atlantic Ocean was a net sink or secondary source of PBDEs; and (iii) to compare active and passive sampling approaches for PBDEs; and (iv) estimate settling fluxes of PBDEs in air and water to determine the turnover of PBDEs in these compartments.

73 **METHODS**

74 Sampling

75 *High volume water sampling*

Active air and water sampling were taken as detailed previously.²² In brief, a total of 57 water samples were collected in the ship's laboratory from the ship's seawater pipe using a sampling train, equipped with a pre-combusted glass fiber filter (GFF) and 3 polyurethane foam (PUF) plugs in series. Water samples were collected continuously. Initially (samples 1-33) collected 600-1100 L of water, after Barbados, 200-500 L were filtered (samples 34- 57).

81

82 *High volume air samples*

83 Air sampling was conducted when the wind came from across the bow to collect uncontaminated marine boundary layer air. Previous work has shown that sampling of POPs 84 can easily be contaminated from the ship's emission (either indoor air or stack emission).²⁶ 85 86 During this transect, winds often shifted away from the bow, such that only 47 air samples 87 were collected using a high-volume air sampler, equipped with a pre-combusted GFFs and 2 88 pre-extracted PUF plugs. The active air sampler was set-up at the front of the monkey's level (above the bridge), facing the wind. Up to sample 23, volumes of $400 - 570 \text{ m}^3$ (~ 12 hours 89 each) were collected, after Barbados, volumes were ~ $230 - 350 \text{ m}^3$ (~ 8 hours each). 90

91

92 *PE sheet samplers*

Passive samplers were deployed to (a) the indoor air for 1 -2 weeks each (n=7); (b) the marine air for consecutive 48 hours (n=12); (c) to flowing seawater in the ship's laboratory for consecutive 48 hours (n=13); and (d) samplers were towed behind the ship for 48 hours outside country's exclusive economic zones (limiting these to n=9). 97 The air PE samplers were exposed in inverted stainless steel bowls ('UFO disk') on the ship's 98 main (chimney) mast 17.5 m above sea level (i.e., several meters above the high-volume 99 sampler). PE samplers were exposed continuously, even when hi-volume air sampling was 100 paused due to adverse winds. The water PE was fixed in a steel pipe connected to the flowing 101 seawater in side the ship's special purpose laboratory, at a nominal flow rate of 10 L min⁻¹. 102 PE samplers were towed on a line via the A-frame, ca. 100 m behind the ship, for 40-70 hours 103 each.

104

105 Sample analysis

106 High volume samples (PUFs, GFFs)

Analytical standards for PBDEs (calibration sets, natives and ¹³C labeled) were purchased from Wellington Laboratories (Canada). PUFs and GFFs were extracted using automated warm Soxhlet extraction (40 minutes warm Soxhlet followed by 20 minutes of solvent rinsing) with dichloromethane (DCM) in a B-811 extraction unit (Büchi, Switzerland). The extracts of water PUF a GFF samples were first dried using Na₂SO₄, concentrated and then split into 2 portions, ¹/₄ of extract was used for PAHs analysis,²³ the remaining ³/₄ of extract were used for PBDEs, indicator PCBs and OCPs analysis.²¹

114

115 Analysis

Several ¹³C BDEs (28, 47, 99, 100, 153, 154, 183 and 209, all from Wellington Laboratories, Canada) were added prior to extraction. The extract was cleaned-up on a H_2SO_4 modified (30% w/w) silica column. Analytes were eluted with 40 mL DCM/*n*-hexane mixture (1:1). The eluate was concentrated using stream of nitrogen in a TurboVap II concentrator unit and transferred into an insert in a vial. The syringe standards (¹³C BDEs 77 and 138 and native PCB 121) were added to all samples, the final volume was 50 uL. 122 HRGC/HRMS instrumental analysis for PBDEs was performed on 7890A GC (Agilent, USA) 123 equipped with a 15 m x 0.25 mm x 0.10 μ m DB5 column (Agilent J&W, USA) coupled to 124 AutoSpec Premier MS (Waters, Micromass, UK). The MS was operated in EI+ mode at the 125 resolution of >10 000. For BDE 209, the MS resolution was set to >5 000. Injection was 126 splitless 1 μ L at 280°C, with He as carrier gas at 1 mL min⁻¹. The GC temperature program 127 was 80°C (1 min hold), then 20°C min⁻¹ to 250°C, followed by 1.5°C min⁻¹ to 260°C (2 min 128 hold) and 25°C min⁻¹ to 320°C (4.5 min hold).

129

130 *PE sheet samplers*

Blanks and exposed sheets of PE were rinsed with Millipore water, dried with a disposable tissue and soaked for 24 hours in 200 mL of *n*-hexane followed by 24 hours in 200 mL of DCM. The two solvents were then pooled; concentrated using stream of nitrogen in a TurboVap II concentrator and the extract was split into 2 portions and processed using the same procedure as the high volume samples.

136

137 **Quality assurance, Quality control**

The results for PBDEs in high volume samples are recovery corrected (recoveries ranged from 34 to 110%, Table S7). Method performance was tested prior to sample preparation. Four field blanks were extracted for water and air GFFs each; Four PUF blanks were extracted for air and water samples each (one PUF blank was excluded for air samples) and combined for blank correction and method detection limit (MDL) determination. The MDL was calculated as 3 standard deviations of blank concentrations (for more details, see SI and Tables S2-S6).

145

146 **Physicochemical properties**

Best-fit values of PE-water (K_{PEw}) equilibrium partition constants were taken from Lohmann 147 (2012)²⁷, and corrected for average sampling temperature and salinity (Table S1). PE-air 148 (K_{PEa}) values were calculated based on a linear regression of subcooled-liquid vapour pressure 149 $(p_{\rm I})$ established for PAHs, OCPs and PCBs (data not shown). Octanol-water partition constant 150 (K_{ow}) and p_{L} values were calculated on the correlation with molecular weight presented by 151 Dugani and Wania (2003).²⁸ The Setschnow constant was taken as 0.35, as reported 152 elsewhere.²⁹ Average values were taken for enthalpies of PE-water (25 kJ mol⁻¹) and PE-air 153 exchange (80 kJ mol⁻¹), close to calculated values of internal octanol-water and octanol-air 154 exchange.^{30,31} 155

156

157 Concentrations and fluxes from PE samplers

158 Truly dissolved concentrations of PBDEs in water, C_{diss} (pg L⁻¹ H₂O) and gas-phase, C_{gas} (pg 159 m⁻³) were derived from PE-normalized concentrations, C_{PE} (pg L⁻¹ PE):

160
$$C_{diss/gas} = \frac{C_{PE}}{K_{PEw (or PEa)} \times \left(1 - e^{-\frac{R_s \times t}{K_{PEw (or PEa)} \times V_{PE}}}\right)}$$
(1)

161 Where R_s (m³ day⁻¹) is the overall sampling rate, t is the deployment time (days), V_{PE} the 162 volume of the PE sampler (m³) and K_{PEw} (or K_{PEa}) is corrected for the average temperature 163 and salinity of the deployment (for details, see SI). Sampling rates were estimated based on 164 PRC loss and typical uptake rates.²¹ For towed samplers, we assumed equilibration was 165 achieved. In view of the high molecular weights (MW) of BDEs, R_s was modulated relative to 166 BDE 28 as a function of MW^{-0.4}, representing an average of previous relationships between R_s 167 and molecular size.³²

168 Air-water exchange fluxes, F_{aw} (pg m⁻² day⁻¹), were calculated according to the two-film 169 theory. The air-water exchange velocity, k_{ol} (m d⁻¹), was estimated as detailed previously.²³

170
$$F_{a/w} = C_{diss} - \frac{C_{gas}}{K_{aw}} \times k_{ol}$$
(2)

171

172 Meteorological and sea surface auxiliary measurements

173 From the ship's routine measurements, we averaged values of latitude, longitude, surface 174 water temperature (T_{water}), salinity, and fluorescence of the flow-through seawater; air 175 temperature (T_{air}), relative humidity (RH), relative and absolute wind speed and direction 176 recorded every minute for each sampling period (see SI). Back – trajectories were calculated 177 for 5 days with 6 h steps at 300 m above sea level using HYSPLIT.³³ Concentrations of 178 dissolved organic carbon (DOC) were derived from relationships by Siegel et al. (2002), 179 mostly around 1 mg DOC L⁻¹.³⁴

180

181 Cruise track

182 On the water side, the cruise covered different major currents of the tropical Atlantic Ocean 183 (Figure 1). Currents were identified based on a combination of changes in temperature (and salinity for the Amazon plume) and typical current fields ³⁵ (see Table S9). In the southern 184 185 hemisphere, these were the Benguela, South Equatorial and North Brazil Current. Discharges 186 from the Amazon, the North Equatorial Current and Gulf Stream affected samples in the northern hemisphere. We obtained back-trajectories to confirm air mass origin.²¹ Initially, we 187 188 encountered southeasterly trade winds, moving air masses in a westerly direction towards the 189 equator. We then passed the intertropical convergence zone (ITCZ), which was situated at 190 approximately 10 ° N during our cruise. The cruise then continued in the northeasterly trade 191 winds moving air masses towards the west along the equator in the NH. The last few samples 192 were affected by the westerlies, moving air masses eastwards across the Atlantic Ocean.

194 **RESULTS AND DISCUSSION**

195

PBDE concentrations in the water

197 Particle-bound PBDE water concentrations

Typical particle-bound concentrations of PBDEs in the surface water were low, at < 1 pg L^{-1} , with the exception of BDE 209 (mean 7 pg L^{-1} , median 1.8 pg L^{-1}) (Table S18). The most abundant BDE congeners were 47, 99, 100 and 85, which were detected in approximately 2/3 of all samples. In terms of concentrations, BDEs 47 and 99 displayed highest concentrations in most samples. When above the MDL, BDE 209 dominated overall aqueous concentrations of BDEs on particles.

204 A strong hemispherical gradient was apparent for particle-bound aqueous PBDEs. The ratio of 205 the mean concentrations was approximately 5-times higher in the northern than in the southern hemisphere. For example, mean concentrations of BDEs 47 and 99 were 0.04 pg L^{-1} 206 in the southern hemisphere, but 0.3 pg L^{-1} in the northern hemisphere. Similarly, the 207 208 frequency of detection for all congeners was only 21% in the southern hemisphere, but 49% 209 in the northern hemisphere. Clearly, the southern tropical Atlantic Ocean is much cleaner (on 210 average 5-times) with respect to particle-bound PBDEs in the surface water than the northern 211 hemisphere tropical Atlantic. We note that the increase in particle-bound PBDEs began at 5 – 212 10 °N, not the equator itself (Table S18). This was probably the result of the ITCZ having 213 shifted northwards during the northern summer, as reflected in the back-trajectories. This also 214 implies that particle-bound PBDEs reflected fairly recent deposition events.

A closer look at the spatial distribution of particle-bound PBDEs showed that highest concentrations were detected in the Amazon plume, the Gulf Stream Eddy and in the sample closest to the U.S. coast (Figure 2). BDE 209 was only detected in a few samples from the Amazon River plume, where concentrations reached 36 pg L^{-1} . In the sample taken in the Gulf Stream, particle-bound BDE congener concentrations were not elevated (Figure 2).

Xie et al. (2011) published dissolved and particle-bound concentrations of PBDEs along a North-South transect of the Eastern Atlantic Ocean in 2008.¹⁷ Particle bound BDE concentrations were ≤ 0.1 pg L⁻¹ (BDE 209 was not analyzed). Highest concentrations were observed close to Europe and off South Africa. Particle-bound concentrations of both cruises were similar in magnitude, implying that contamination-free sampling of particles was achieved on both cruises.

226

227 Truly dissolved PBDE concentrations from passive sampling

228 PBDEs were detected in the towed passive samplers deployed in the water, likely due to their extremely high K_{PEw} values (causing strong enrichment in the polyethylene films) (see SI). 229 230 Total mass of PBDEs accumulated in the towed PE samplers were at least 10 times higher 231 than those in the flow-through water exposed PE samplers, highlighting the boundary-layer 232 limitations we encountered in the laboratory. This was also evident in the detection frequency 233 of PBDEs in both types of deployments. Whereas almost all PBDE congeners were regularly 234 detected in the towed passive sampler (overall detection frequency was 90%, except BDE 66, which was never > MDL), only < 5 % of PBDEs were > MDL in the flow-through passive 235 236 sampler exposure. Towing of passive samplers could be a potentially transformative way of sampling truly dissolved concentrations in the open Oceans.³⁶ Concentrations of PBDEs were 237 greater by a factor of approximately 2 in the northern and western tropical Atlantic compared 238 239 to the southern and eastern section based on the towed passive sampler results.

The dominant congeners were BDEs 47 and 99 (Figure 3). Typical concentrations in the towed PE were ~ 0.5 pg L^{-1} for BDE 47 and around 0.1 pg L^{-1} for BDEs 28, 99 and 100 (Table S12). Overall, the PE-derived concentrations were remarkably similar to those reported by Xie et al. (2011) for the Eastern Atlantic Ocean on a *R/V Polarstern* cruise in 2008 (Figure 3).¹⁷ On that cruise, samples were taken with a classical active sampling train using PAD resins and GFFs. Mean values reported by Xie et al. (2011) were 0.1 pg L^{-1} for BDEs 47 and 99, and a factor of 10 lower for other PBDEs (BDE 209 was not analyzed).¹⁷

247 BDE concentrations started to diverge for the highest molecular weight BDE congeners (Figure 3), with PBDE concentrations from towed PE samplers suggesting much lower truly 248 dissolved concentrations than those from Xie et al. (2011)'s active sampling.¹⁷ This could be 249 250 due to the assumption of all PBDEs having reached equilibrium between the water and the 251 samplers during the 2-day tows. The good agreement between both types of sampling and 252 cruises (~ factor of 2) for congeners with up to five bromines (BDE 100) suggests that 253 compounds with a log $K_{ow} \le 7$ had equilibrated. For congeners with six or more bromines, a 254 correction for non-equilibrium could be necessary, or a longer deployment time (reducing the 255 boundary layer through faster passive sampler movement seems unlikely to be achieved in the 256 field, unless through special design features). Alternatively, active sampling results could be 257 biased high due to the co-sampling of PBDEs bound to DOC.

258

259 Dissolved PBDE concentrations from active sampling

260 Dissolved concentrations from active sampling were, by and large, 8 times lower in the South 261 Atlantic than in the North Atlantic (Table S16). The dominant PBDE congeners across all 262 samples were 99, 47 and 85. In the South Atlantic, concentrations ranged from below MDL in most samples towards 90 pg L^{-1} for PBDE 99. In the North Atlantic, concentrations increased 263 264 markedly in the Amazon River plume, where several PBDE congeners reached hundreds of pg L⁻¹. Two other regions of elevated concentrations were in the Gulf Stream Eddy, and the 265 sample closest to the US coast, where BDEs 47 and 99 reached several thousands pg L^{-1} . 266 These concentrations are far greater than reported by Xie et al $(2011)^{17}$, and the 267

268 concentrations deduced from our towed passive samplers (both below 1 pg L⁻¹). In fact, the 269 concentrations from active sampling on this cruise exceeded those reported for coastal 270 environments or Lake Michigan: dissolved Σ PBDEs of up to 60 pg L⁻¹ were reported in the 271 coastal waters of Hong Kong³⁷, 100 – 200 pg L⁻¹ for Izmir Bay³⁸, while Streets et al. (2006) 272 reported average concentration of Σ PBDEs of 18 pg L⁻¹ for Lake Michigan³⁹.

273 The strong discrepancy between the dissolved PBDE concentrations reported by Xie et al. $(2011)^{17}$ and our BDE concentrations from passive sampling with our results from active 274 275 sampling is reason for concern. Apparently dissolved PBDE concentrations from active 276 sampling on our study were elevated by 10 - 1,000 fold. We examined whether the collection 277 of colloids or microplastics in our PUF-based sampling approach could explain the difference. We assumed 1 mg DOC L^{-1} during the cruise. DOC-water partition constants, K_{DOC} , were 278 taken either from Burkhard (2000)⁴⁰ for natural colloids, or values extrapolated from sediment 279 partitioning experiments by Wang et al. (2011)⁴¹. In either case only up to 50% of the 280 281 congeners dominating total dissolved concentrations (BDEs 47, 100 and 99) were bound to colloids. The subtropical gyres in the Atlantic Ocean are known to accumulate small pieces of 282 plastic debris, or microplastics.⁴² If half of the [DOC] consisted of microplastics with sorption 283 284 properties similar to PE, around 50% of BDE 47 would be colloidal-bound, and 80% of BDEs 285 99 and 100. These calculations imply that co-sampling of colloids or microplastics cannot 286 explain the observed active sampling results. Most likely, active sampling results became 287 contaminated before, during or after sampling.

288

PBDE concentrations in the atmosphere

290 Particle-bound PBDEs in the atmosphere

A wide range of PBDEs were routinely detected on particles in the marine boundary layer, including BDEs 66, 100, 99, 154, 153 and 183, and often BDE 209 (Figure 4). When detected, BDE 209 dominated overall concentrations (median 1.2 pg m⁻³), followed by BDE
99 (0.13 pg m⁻³). For all other BDEs, mean and median concentrations were < 0.1 pg m⁻³.
There were no significant differences between BDE concentrations on particles between both
hemispheres (Table S10).

297 Concentrations were slightly greater than those reported by Xie et al. (2011) across the 298 Eastern Atlantic Ocean (BDE 209 was not measured in that study).¹⁷ During their cruise, only 299 BDE 47 had a median concentration greater 0.1 pg m⁻³, with other congeners mostly below 300 detection limits.¹⁷

301

302 True gas-phase PBDE concentrations from passive sampling

303 Most PBDEs were detected routinely in the air passive samplers (Table S14). Greatest concentrations were obtained for BDE 47 at $1 - 8 \text{ pg m}^{-3}$, followed by BDE 99 ($\leq 4 \text{ pg m}^{-3}$), 304 with the other BDEs $< 1 \text{ pg m}^{-3}$ (Figure 4). The intermittent detection of BDE 209, ranging 305 from < MDL - 260 pg m⁻³, could be due to its episodic transport, or reflect contamination on-306 307 board. There was no significant difference between the two hemispheres in terms of gas-phase 308 PBDE concentrations. These concentrations are comparable to, or slightly higher than those reported by Xie et al. (2011) for gas-phase PBDEs across the Eastern Atlantic (Figure 4).¹⁷ Li 309 310 et al. (2011) published combined gas+particle phase concentrations across the south-eastern Atlantic Ocean.⁴³ Mean concentrations were a few pg m⁻³, similar to the concentrations 311 312 reported here.

313

314 Gas-phase PBDE concentrations from active sampling

In the southern hemisphere air samples, gas-phase PBDEs were routinely below detection limits (on the order of 1 pg m⁻³ for most congeners, see Tables S16). In several southern hemisphere samples, all PBDE congeners were detected, with concentrations reaching 1 ng m⁻³ ³ for BDEs 47 and 99 in sample NAM-AP-020. During the collection of that particular sample, the high volume sampler motor was shut off for 12 hours due to adverse wind conditions, but the PUFs and GFF left in place. The exceedingly elevated concentrations from this sample imply ship's emission as a strong contamination source of PBDEs.

322 Concentrations of gas-phase BDEs increased in the northern hemisphere (i.e., once the ship crossed the ITCZ), with most congeners being detected above MDL regularly. Typical 323 concentrations were around 100 pg m⁻³ for BDEs 47 and 99 and at or below 10 pg m⁻³ for the 324 325 other congeners. A strong north-to-south gradient was apparent, with ratios of mean or 326 median concentrations implying that concentrations of PBDEs were greater in the NH by 327 factors of 2-8. The comparison with the passive sampling results, and the BDE concentrations reported previously by Xie et al. (2011)¹⁷ imply that gas-phase samples were routinely 328 contaminated.²⁶ 329

330

331 Air concentrations inside the ship

The PE samplers deployed inside the ship displayed elevated concentrations of PBDEs, particularly BDE 47 at hundreds pg m⁻³ (Table S15). Other prominent congeners were BDE 99, 28 and 100 at tens pg m⁻³ on average. The sum BDEs averaged 600 pg m⁻³ inside the ship, comparable to other published concentrations of PBDEs in indoor air. For example, a mean concentration of 940 pg m⁻³ for Σ PBDEs was reported for indoor air in Toronto, with other studies detecting 100 – 200 pg m⁻³ for indoor air.⁴⁴ Much higher concentrations were reported for air in offices in Sweden, with concentrations ranging up to 4,700 pg m⁻³.⁴⁵

339

340 The ship as a major source of PBDE contamination of samples and the environment

341 There are several lines of evidence that the *R/V Endeavor* itself was a major source of PBDEs,

342 and partially contaminated active samples taken on-board. First, the indoor air displayed

343 elevated concentrations of a wide range of PBDEs, in-line with concentrations observed in office and residential buildings on-land.^{36,37} Second, field PUF blanks taken during active air 344 345 and water sampling resulted in elevated BDE concentrations (Tables S5-S6). Third, during a couple of active sampling events, the high volume sampler (hi-vol) was shut off (but the PUFs 346 347 left in place) when the wind was coming from the side or stern of the ship. These samples 348 displayed strongly elevated concentrations of PBDEs. Fourth, active sampling resulted in 349 much greater PBDE concentrations than passive sampling. The gas-phase PE sampler was 350 mounted several meters above the hi-vol, implying that diffusive contamination from the ship 351 did not reach the PE sampler. Fifth, there is a noticeable increase in PBDE concentrations in 352 the PE field blanks, with lowest concentrations for the atmospheric PEs (exposed to marine 353 air) at 170 pg per sample. The PEs mounted in the ship's laboratory were accompanied by 354 field blanks containing on average 540 pg per sample, while the towed PEs accumulated on 355 average 5,300 pg per sample. Towed PEs were mounted on the back of the ship, where the ship's circulated air and engine room ventilation could often be felt. 356

357

358 **Pollutant gradients across the Atlantic Ocean**

359 Our results suggest that the north-western tropical Atlantic carried 5-times higher 360 concentrations of PBDEs on particles, and twice as high truly dissolved concentrations than 361 the south-eastern Atlantic Ocean. Yet our atmospheric results do not show any significant 362 differences between both hemispheres and regions. Our results seem to contradict PBDE 363 concentrations reported by Xie et al. (2011), who reported higher atmospheric concentrations during their northern Atlantic Ocean transect relative to the samples taken further south.¹⁷ For 364 365 their aqueous samples, there was no significant difference between dissolved or particlebound PBDE concentrations between the North and South Atlantic.¹⁷ We suggest that the 366 gradients observed on the different cruises have as much to do with the particular regions 367

368 being crossed rather than solely reflecting on northern versus southern hemisphere. The fact 369 that we observed greater aqueous concentrations of PBDEs in NH is mostly due to our cruise 370 transecting the remote southern tropical Atlantic (with low PBDE concentrations) but ended 371 up sampling the Amazon River, warm core, Gulf Stream, and US coastal seaboard in the 372 northern hemisphere.

373 Xie et al. (2011) sampled the atmosphere closer to Europe and Africa for most of their 374 samples taken in the northern hemisphere, but farther from shore during their southern 375 transect, which might explain the difference in atmospheric PBDE concentrations they 376 reported.¹⁷

377

378 Air-water exchange

Air-water exchange gradients were calculated based on simultaneous passive sampler deployments in air and water. Gradients were based on PBDE concentrations in PE (ng g⁻¹ PE) at equilibrium.⁴⁶ In short, passive samplers, such as the PE we used, reflect the chemical activity of the BDEs in their respective matrix (air and water in our case). The ratio of those BDE activities (corrected for non-equilibrium) is the activity gradient across the air-water interface.

In the water, results from towed PE samplers were used; in the air, the equilibrium-corrected PBDE concentrations from the passive sampler deployed on the ship's mast were used. In approximately 50% of possible cases, PBDEs were > MDL in both phases simultaneously (Table S19). Net air-water exchange gradients strongly favoured deposition of PBDEs from the gas-phase into the water phase. Gradients increased with increasing MW. For BDEs 28 and 47, net air-to-water exchange gradients ranged from 3-13. For the higher MW BDEs 85, 99 and 100, gradients increased to be in the range 10-50, and even greater for the higher MW congeners. Our results are similar to the gradients by Xie et al. (2011), who reported net
 deposition for all PBDEs along the eastern Atlantic Ocean.¹⁷

Air-water exchange velocities ranged from < 1 to 36 cm day⁻¹, and decreased with increasing MW (Table S22). Net gas-phase deposition fluxes ranged from 10s of pg m⁻² day⁻¹ for BDEs 28 and 85 to around 1 ng m⁻² day⁻¹ for BDE 47, 99 and 209 (Table S23). Similar values were reported by Xie et al. (2011). ¹⁷

We calculated the net annual gas-phase deposition of PBDEs, assuming these air-water exchange fluxes were representative across the Atlantic Ocean. For all PBDEs reported here, the gas-phase deposition flux is approximately 40 tons annually, of which 20 t are the median flux for BDE 209.

402

403 **Deposition of PBDEs across the Atlantic Ocean**

404 We calculated the deposition flux of PBDEs on particles, $F_{part,dep}$ (pg m⁻² day⁻¹), across the 405 Atlantic Ocean as:

$$406 F_{\text{part,dep}} = C_{\text{part}} \times v_{\text{dep}} (3),$$

407 where C_{part} is the particle-bound PBDE concentration (pg m⁻³) from this cruise and v_{dep} is the 408 deposition velocity taken as 0.1 cm s⁻¹ as reported by DelVento and Dachs (2007).⁴⁷ This 409 translates into a half-live of particle-bound PBDEs in a 1000 m marine boundary layer of 410 around 6 days. Calculated deposition fluxes were dominated by BDE 209 (median flux of 100 411 pg m⁻² day⁻¹), BDEs 47 and 99 (median fluxes of 46 and 11 pg m⁻² day⁻¹). These fluxes 412 compare well to those calculated by Xie et al. (2011) for PBDEs across the Eastern Atlantic 413 Ocean.¹⁷

414 If the results reported here are representative across the Atlantic Ocean, the annual deposition 415 of particle-bound BDEs 47, 99, 100 and 85 will be around a ton scaled to the entire Atlantic 416 Ocean (surface area ~ $75 \times 10^6 \text{ km}^2$). The fact that particle-bound PBDE concentrations from 417 our cruise were similar to those by Xie et al. (2011) support that assumption. The median
418 annual deposition flux for BDE 209 is ~ 7,500 kg for the Atlantic Ocean. The fluxes derived
419 above signal a significant annual deposition (gas-phase deposition of ~40 t and particulate420 bound deposition of ~10 t) of PBDEs into the Atlantic Ocean .

421 Yet the current deposition flux of PBDEs pales in comparison to production volumes of these 422 flame retardants. The estimated cumulative use of the commercial pentaBDE mixture alone was estimated to 100 000 t since 1970.⁴⁸ This indicates that more is to come, as PBDEs 423 slowly make their way from products into the environment. Relative to PCBs, the gas-phase 424 deposition fluxes of PBDEs are similar in magnitude. ⁴ Yet the particle-bound deposition flux 425 426 of PBDEs to the Atlantic Ocean already exceeds that of PCBs, which was estimated to be around 2.2 tons per year.⁴⁹ A greater fraction of PBDEs is bound to particles, explaining the 427 derived greater importance of their particle-bound deposition relative to PCBs. 428

429

430 Settling fluxes of PBDEs out of the mixed layer

431 We estimated settling fluxes of PBDEs out of the mixed layer depth across the tropical 432 Atlantic Ocean. The settling flux of particulate organic carbon (POC) at 100 m depth, $F_{OC,100}$, 433 was estimated based on its relationship with Chlorophyll *a* (Chl *a*)⁵⁰:

434
$$\log (F_{OC,100}) = 2.09 + 0.81 \times \log(Chl \alpha)$$
 (4)

435

436 Resulting settling fluxes of OC at 100 m depth were typically 30 - 80 mg m⁻² day⁻¹, typical 437 for the tropical oceans.⁵¹ To derive a settling rate of particles out of the mixed layer depth, we 438 estimated POC concentrations based on published 10-year averages for July as reported by 439 Stramska (2009).⁵² We took POC concentrations as 80 mg m⁻³ up to the equator, 40 from the 440 equator to 5 °N, 30 from 5 to 10 °N, 20 from 10 to 30 °N, 30 from 30 to 35 °N and 80 north 441 of 45 °N. Dividing $F_{OC,100}$ by POC concentrations yields the settling velocity⁵³, which ranged from $0.5 - 2.0 \text{ m day}^{-1}$. Lastly, the product of settling velocity and particle-bound PBDE concentrations yielded the PBDE removal flux from the surface ocean layer. This PBDE export flux rarely exceeded 1 ng m⁻² day⁻¹ for most congeners, except for BDE 209, which reached fluxes of tens ng m⁻² day⁻¹ (Figure S3). Settling fluxes were greater in the northern hemisphere than south of the equator, driven by the greater frequency of detection and concentrations of PBDE in the particle-phase (see above).

448

449 **Comparison of PBDE fluxes**

450 We compared the fluxes of gas-phase and particle-bound PBDEs out of the atmosphere to 451 those out of the Atlantic Ocean's mixed layer depth. Net gas-phase deposition fluxes were 452 much greater than the particle-bound settling fluxes, implying additional removal from the surface mixed layer, probably via photolytic and microbial degradation.⁵ The median values 453 454 of both particle-bound settling fluxes agreed surprisingly well in view of the assumptions 455 behind the settling velocities in both atmosphere and surface water (Figure 6). Settling fluxes were around 50 pg m⁻² day⁻¹ for BDE 47 in both atmosphere and surface water, and < 10 pg 456 m⁻² day⁻¹ for other congeners. For PBDEs 47 through 99, settling fluxes from the surface 457 458 water exceeded atmospheric deposition fluxes, while for the higher brominated congeners, 459 median settling fluxes out of the mixed layer depth (MLD) were zero due to PBDEs being < 460 MDL in the water.

461

462 **Residence time of particle-bound PBDE in surface water**

The MLD was taken from published 10-year averages for July as reported by Stramska (2009).⁵² It ranged from 80 m up to the equator, 40 m from the equator to 20 °N, and 20 m north of 20 °N. The product of MLD and total (sum of truly dissolved + particle-bound) PBDEs yielded the mass loading of PBDEs on particles (ng m⁻²) in the surface water. The

mass loading divided by the PBDE removal flux (ng $m^{-2} day^{-1}$) gave the residence time (days) 467 468 of particle-bound PBDEs in the surface water of the Atlantic Ocean, assuming particle settling 469 was the only removal pathway. The tight coupling of particle-bound PBDE fluxes from 470 atmosphere and out of the mixed layer could indicate that different fates of particle-bound 471 PBDEs (settling) and those derived from gaseous diffusion, probably due to chemical and 472 biological degradation of dissolved molecules in the water column. In the northern 473 hemisphere, median PBDEs' residence were several weeks, but were around one year in the 474 southern hemisphere (Figure S5). These values suggest that settling on particles is an 475 important removal pathway in the northern hemisphere, whereas biological and photolytic 476 degradation are probably determining the fate of PBDEs in the southern hemisphere.

477

478 Implications

479 Our results imply that PBDEs continue to be deposited across the Atlantic Ocean. Our 480 research vessel proved to be a significant source of contamination for PBDEs itself during 481 active sampling of air and water. Yet the use of passive samplers enabled us to retrieve valid 482 samples for both gas-phase and truly dissolved PBDEs. Other researchers have shown that 483 other contamination-free sampling can be achieved, though on a different research vessels. ^{17,43} Our results suggest that atmospheric deposition of particle-bound PBDEs is probably 484 485 slightly lower but similar in magnitude to their removal flux on settling particles from the 486 surface mixed layer. Net gas-phase deposition of PBDEs exceeded particle-bound deposition, 487 suggesting that there are other important removal pathways for dissolved BDEs, such as 488 microbial and photolyic degradation. The total deposition of PBDEs to the Atlantic Ocean, 489 around tens of tons per year, pales in comparison to their historical production volumes. This 490 might indicate the less efficient long-range transport of these higher molecular weight 491 compounds, compared to, for example, polychlorinated biphenyls. It could also indicate that 492 PBDEs are only slowly emitted from their current use, guaranteeing continuous fluxes into
 493 the oceans in the future.⁵⁴

494

495 Supporting Information Available

496 The Supporting information includes additional details, such as physicochemical constants,

497 instrumental analysis and QC, sampling set-up and sample details, air and water498 concentrations. This material is available free of charge via the Internet at http://pubs.acs.org.

- 499
- 500

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507

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Figure 1: Cruise track of EN 464 with chlorophyll a concentrations derived from MODIS ⁵⁵

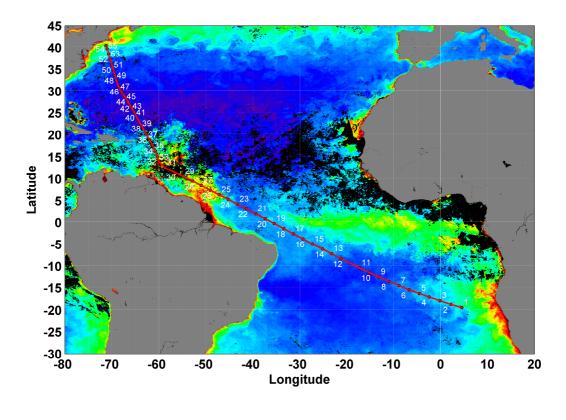
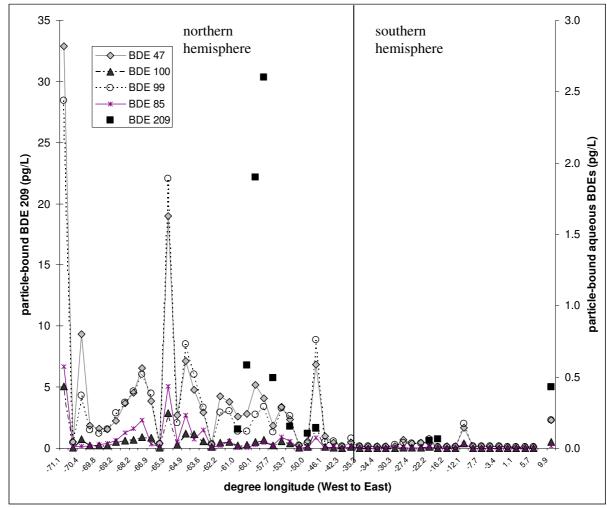


Figure 2: Concentrations of aqueous particle-bound PBDE congeners (pg L⁻¹) in 2009 672 673 674



675 676

Note: concentrations of BDE 209 are displayed on right-hand side

Figure 3: Dissolved concentrations of PBDE congeners derived from towed passive PE
samplers (pg L⁻¹) across the tropical Atlantic Ocean in 2009. Displayed are average
concentrations for the southern and northern hemisphere, and mean and median results
from Xie et al. (2011) ¹⁷ for comparison.



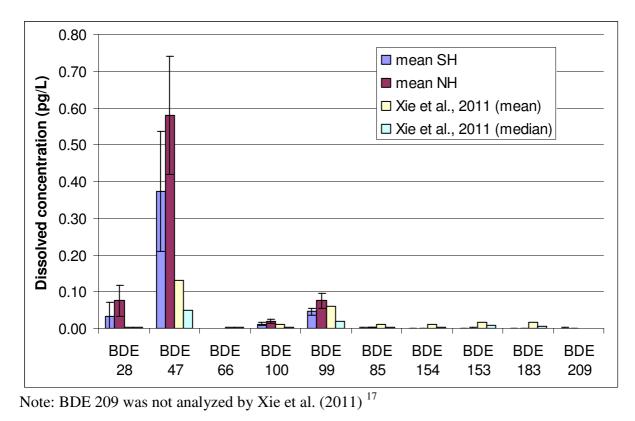
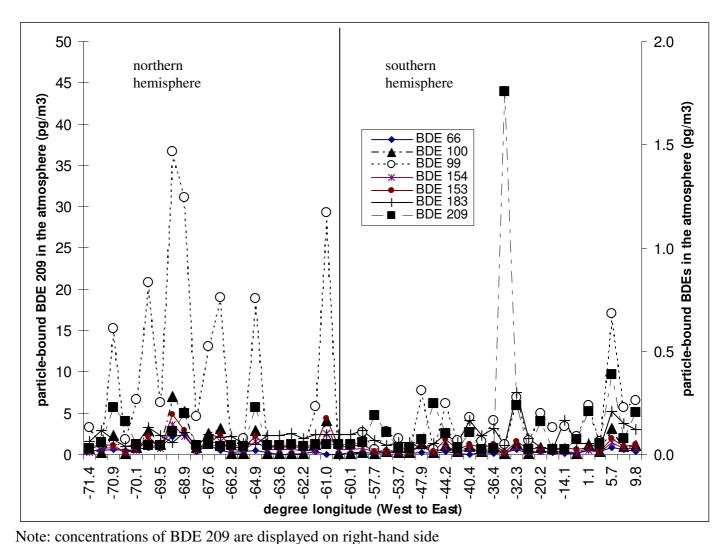


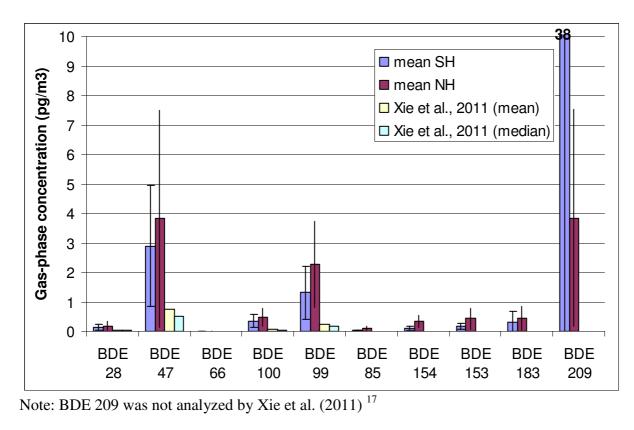
Figure 4: Particle-bound atmospheric concentrations of PBDE congeners (pg m⁻³) in 687



689

Figure 5: Gas-phase atmospheric concentrations of PBDE congeners derived from
 passive PE samplers (pg m⁻³) across the tropical Atlantic Ocean in 2009. Displayed are
 average concentrations for the southern and northern hemisphere, and mean and
 median results from Xie et al. (2011)¹⁷ for comparison.





- Figure 6: Comparison of median fluxes for PBDEs (pg m⁻² day⁻¹) from net gas-phase and particle-bound deposition out of the atmosphere and via particle-bound settling out of

the mixed layer depth

